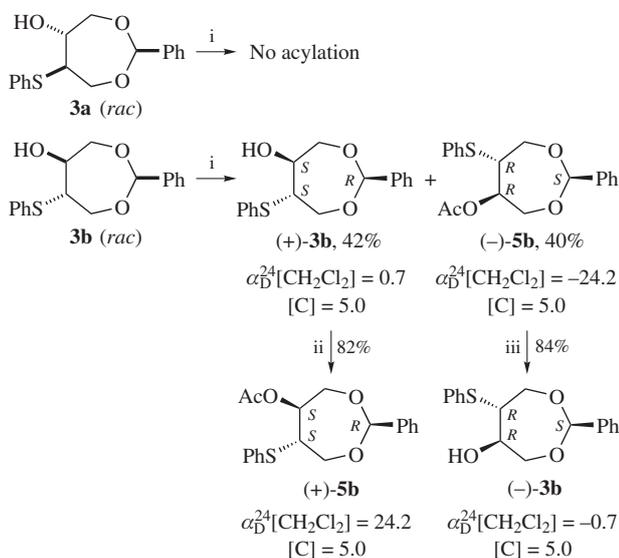


Figure 1 Crystal structure of compound **5c**.

compounds into enantiomers by lipase PS-catalyzed acylation with vinyl acetate as the source of acetyl moiety (Scheme 2).[†]

Interestingly, under the same acylation conditions diastereomeric alcohols **3a** and **3b** differed in their behaviour. Alcohol **3b** was separated into enantiomers by two-stage synthesis (Scheme 2), whereas its isomer **3a** did not undergo enzymatic acylation. The moduli of rotation angles of the obtained enantiomeric alcohols **3b** and particularly their acetates **5b** (see Scheme 2) were identical what allowed judging of high enantioselectivity of this enzymatic separation. Unfortunately, the attempted use of chiral shift-reagent to directly determine their *er* values was unsuccessful. The configuration of chiral carbon atom at the hydroxy group was ascribed on the basis of the literature data,¹³ since it is known that in the presence of lipase PS only alcohols, having the hydroxy group at the chiral carbon atom with *R*-configuration, are prone to acylation. Hence, under conditions of *trans*-opening of epoxy cycle by thiophenol the configurations of chiral carbon atoms in enantiomers obtained correspond to those presented in Scheme 2.

We have theoretically evaluated biological activity of the compounds synthesized by means of PASS Program.^{14–16} According to the data obtained, compounds **3a–c** most likely have high



Scheme 2 Reagents and conditions: i, lipase PS, vinyl acetate, THF, 40 °C, ~90 days stirring; ii, Ac₂O, DMAP, Et₃N, CH₂Cl₂; iii, K₂CO₃, MeOH, room temperature.

[‡] Crystal data for **5c**: colourless crystal, C₁₉H₂₀O₄S (*M* = 344.42) monoclinic, space group *C2/c*, at 298(2) K: *a* = 30.710(4), *b* = 5.1147(6) and *c* = 22.411(3) Å, β = 93.819(1)°, *V* = 3512.3(8) Å³, *Z* = 8, *d*_{calc} = 1.303 g cm⁻³, μ = 0.203 mm⁻¹, *R*_{int} = 0.039, θ_{max} = 27.0°, Bruker Smart Apex II CCD diffractometer, 17922 reflections collected, 2585 observed reflections with *I* > 2σ(*I*), final *R* = 0.0593, *wR*₂ = 0.1372, 3814 independent reflections with *F*² ≥ 2σ(*I*), *S* = 1.04.

CCDC 863301 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif. For details, see 'Notice to Authors', *Mendeleev Commun.*, Issue 1, 2012.

Table 1 Minimal inhibiting concentrations (MIC/mg cm⁻³) of synthesized hydroxysulfides **3a–c**.

Compound	<i>Candida albicans</i> (clinical strain)	<i>Aspergillus fumigatus</i> (typical strain)	<i>Epidermophyton floccosum</i> (clinical strain)
3a (<i>rac</i>)	~40	~40	~40
3b (<i>rac</i>)	~10	~10	~10
3c (<i>rac</i>)	>80	>80	~80
(+)- 3b	~5	~5	~5
(-)- 3b	~10	~10	~10
Fluconazole	<0.5	<0.5	~0.6

antifungal activity. This type of activity of these substances, both racemic and optically pure ones, have been investigated with the use of clinical and typical strains of some fungi (Table 1).[§]

Table 1 demonstrates that the antimycotic activity of compound **3c** is significantly lower compared to that of racemic isomers **3a** and **3b**. Meanwhile, the MICs of stereoisomers **3a** and **3b** also significantly differ. So, dioxepane **3b** with *cis*-position of hydroxy and phenyl groups in the cycle has four times higher activity than its stereoisomer **3a** with the same substituents in *trans*-position. The dextrorotatory enantiomer, whose MIC is 5 mg cm⁻³ was found to be more active as compared to levorotatory enantiomer with MIC of 10 mg cm⁻³ for all fungi investigated.

In conclusion, we have shown that compounds obtained may be considered as prospective ones concerning their antifungal activity: seven-membered isomers are more active than six-membered ones, the activity of *cis*-isomer of dioxepane series is significantly higher than that of *trans*-isomer, and the dextrorotatory enantiomer is more active as compared with levorotatory one.

Online Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.mencom.2012.05.003.

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[§] For determination of the fungal activity, see Online Supplementary Materials.