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Graphical Abstract





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A novel highly enantio- and diastereoselective synthesis of vitamin E side-chain

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ABSTRACT

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Cu-catalyzed cross-coupling asymmetric catalysis

A novel highly enantioselective (>99% ee) and diastereoselective (>98% de) method for the synthesis of chiral C_{15} vitamin E side-chain **1** was developed. ZACA–lipase-catalyzed acetylation protocol to provide a key α,ω -dioxyfunctional C_5 synthon **6** (≥99% ee), and two subsequent Cu-catalyzed alkyl–alkyl cross-coupling reactions of enantiomerically pure C_5 iodide **4** were employed as key steps. ¹H NMR analysis of M α NP ester was found to be a convenient method for measuring the enantiomeric purity of the C_{15} vitamin E side-chain **1**.

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Saturated isoprenoids represent a large and diverse class of naturally occurring organic compounds including vitamins,¹ pheromones,² antioxidants,³ marine natural products,⁴ and archaebacterial lipids.⁵ The saturated isoprenoids play widely varying roles in the physiological processes in living organisms such as archaea, bacteria, and human. Archaea are distinct from bacteria and eukaryotes. One characteristic feature of archaea is the chemical structure of its core membrane lipids consisting of saturated isoprenoid chains connected to glycerol through ether linkages.⁶

(3R,7R)-3,7,11-Trimethyldodecanol **1** is a very important saturated isoprenoid derivative, which can be used as a versatile intermediate for the syntheses of vitamin E,⁷ vitamin K₁,⁸ and



Figure 1. Some natural products containing chiral saturated isoprenoid units.

related antioxidants.⁹ It has also been identified as a precursor of many isoprenoid compounds in plants,¹⁰ geological sediments,¹¹ and archaebacterial lipids in archea.¹² Thus, numerous synthetic strategies have been developed for the synthesis of such important building block. Current synthetic approaches include chiral auxiliary-based methods,¹³ chiral pool synthesis,¹⁴ and enzymatic protocols.¹⁵ However, these methods have several disadvantages including long and multi-step processes, the use of stoichiometric quantities of chiral reagents, and low stereoselectivity. At present, only a very few transition metal-catalyzed methods are known, of which catalytic asymmetric hydrogenation by Noyori¹⁶ and Pfaltz¹⁷ are noteworthy. We also developed a concise and efficient synthesis of **1** via **Z**r-catalyzed **a**symmetric **c**arboalumination of **a**lkenes (ZACA).¹⁸ However, all these methods lack high (\geq 98%) stereoselectivity leading to diastereometric mixtures which are very difficult to purify.

We recently developed highly enantioselective (\geq 99% ee) and catalytic routes to a wide range of 1-alkanols including those of isotopic cryptochirality via ZACA–Cu- or Pd-catalyzed cross-coupling.¹⁹⁻²¹ The development of these methods relied on the following four critical findings: (i) ZACA–*in situ* iodinolysis of allyl alcohol to produce either (*S*)- or (*R*)-ICH₂CH(R)CH₂OH in 80–90% ee,¹⁹ (ii) ZACA–*in situ* oxidation of TBS-protected ω -alkene-1-ols to afford both (*R*)- and (*S*)- α , ω -dioxyfunctional intermediates in 80–86% ee,^{20,21} (iii) enantiomeric purification by **li**pase-catalyzed acetylation (LICA) of the above functionally rich intermediates to the \geq 99% ee level through exploitation of their high selectivity factors (*E*),²² and (iv) essentially perfect (>99%) stereochemical fidelity with retention observed widely in the subsequent Cu- or Pd-catalyzed cross-coupling reactions.^{18e,23}

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preparation of both enantiomerically and diastereomerically pure C_{15} vitamin E side-chain 1 using our ZACA–Cu-catalyzed cross-coupling strategy.

A retrosynthetic scheme for preparation of the C₁₅ vitamin E side-chain **1** is shown in Scheme 1. It was planned to couple C₅ iodide **4** and C₁₀ Grignard reagent **5** through Cu-catalyzed alkyl-alkyl cross-coupling reaction.²³ Segment **5** should be readily prepared by another Cu-catalyzed alkyl-alkyl cross-coupling reaction of isopentylmagnesium bromide with iodide **4**, which, in turn, should be derived from the key α, ω -dioxyfunctional chiral C₅ subunit **6**. We envisioned that a highly enantioselective synthesis of chiral C₅ synthon **6** could be achieved via our ZACA-LICA protocol. If the key C₅ subunit **6** could be obtained in high enantiomeric purity (\geq 99% ee), the subsequent Cu-catalyzed cross-coupling reactions would proceed with essentially full (>99%) retention of all carbon skeletal features of key C₅ synthon **6** to provide **1** with excellent enantiomeric and diastereomeric purity (ee > 99%, de > 98%).



Scheme 1. Retrosynthetic analysis of C_{15} vitamin E side-chain 1.

The α, ω -dioxyfunctional C₅ synthon **6** (\geq 99% ee) was prepared as shown in Scheme 2. 3-Buten-1-ol **7** was protected with TBSCl and was subjected to ZACA reaction. A solution of 1 mol% of (+)-bis(neomenthylindenyl)zirconium dichloride [(+)-(NMI)₂ZrCl₂]²⁴ in CH₂Cl₂ was treated with trimethylaluminum (2 equiv) and H₂O (1 equiv), then TBS-protected 3-buten-1-ol **8** was added to the above solution for asymmetric methylalumination. After the completion of methylalumination, the initial alkylalane intermediate was oxidized *in-situ* with O₂ to provide **6** in 88% yield and 81% ee.



Scheme 2. The synthesis of key α, ω -dioxyfunctional C₅ synthon **6** (\geq 99% ee).

Enantiomeric purification of α, ω -dioxyfunctional intermediate **6** obtained by the ZACA reaction was carried out by lipasecatalyzed acetylation. Thus, crude intermediate **6** (81% ee) was readily purified to the level of \geq 99% ee in 52% recovery yield by Amano lipase PS-catalyzed acetylation with vinyl acetate in 1,2dichloroethane. The high enantiomeric purity of **6** was evidenced by Mosher ester analysis²⁵ as shown in Figure 2. Two diastereomeric Mosher esters of β -chiral alcohol **6** showed the expected chemical shift differences. The absence of peaks at 4.23 ppm of MTPA ester **9** derived from alcohol **6** and (*R*)-MTPA-Cl indicated the high enantiomeric purity of alcohol **6** (Figure 2).



Figure 2. ¹H NMR spectra of two diastereomeric MTPA esters of β -chiral alcohol **6** (CDCl₃, 600MHz).

With the enantiomerically pure C_5 intermediate 6 in hand, we proceeded to investigate the synthesis of C₁₅ vitamin E side-chain 1 by using 6 as the key building block. Thus, the chiral C_5 alcohol 6 was readily converted to iodide 4 with triphenylphosphine, imidazole and iodine in CH2Cl2. Next, CuCl₂-catalyzed alkyl–alkyl cross-coupling reaction²³ between iodide 4 and isopentylmagnesium bromide proceeded efficiently in the presence of 1-phenylpropyne as an additive, followed by desilylation with TBAF, providing C_{10} alcohol intermediate 10 in 93% yield, which was further transformed to the corresponding bromide 11 in 96% yield using Ph₃P/NBS. The desired C₁₅ vitamin E side-chain 1 was prepared in 70% yield via another round of CuCl₂/1-phenylpropyne-catalyzed alkyl-alkyl crosscoupling reaction²³ between C_5 iodide 4 and the Grignard reagent derived from C_{10} bromide 11, followed by desilylation with TBAF.



Scheme 3. The synthesis of C_{15} vitamin E side-chain 1.

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We then wanted to determine the enantiomeric purity of the C_{15} vitamin E side-chain **1**. Mosher ester analysis of γ -chiral alcohol **1** proved to be ineffective for determining the optical purity of **1**. We recently demonstrated that 2-methoxy-2-(1-naphthyl)propionic acid (MaNP acid)²⁶ ester analysis is a widely applicable and powerful method for chiral discrimination of various β - and more-remotely chiral primary alcohols including those of isotopomers.^{20,21} As expected, two diastereomeric MaNP esters **12a** and **12b** derived from γ -chiral alcohol **1** showed significantly different chemical shifts. In particular, the γ -methyl groups of these two diastereomers **12** showed completely separate ¹H NMR signals (Figure 3), which indicated that the enantiomeric excess of alcohol **1** was also confirmed to be greater than 99% by ¹³C NMR analysis.



Figure 3. ¹H NMR spectra of diastereomeric M α NP esters of γ -chiral alcohol **1** (CDCl₃, 600MHz).

In summary, we have developed a highly enantioselective (>99% ee) and diastereoselective (>98% de) route to chiral C_{15} vitamin E side-chain 1 via ZACA-Cu-catalyzed cross-coupling from 3-buten-1-ol. The key α, ω -dioxyfunctional C₅ synthon **6** (≥99% ee) was readily prepared by ZACA–LICA protocol, which can be further functionalized at both ends. Two sequential Cucatalyzed alkyl–alkyl cross-coupling reactions of the enantiomerically pure $C_{\rm 5}$ iodide 4 were employed as the key steps for preparing the C_{15} vitamin E side-chain 1. In addition, we demonstrated that ¹H NMR analysis of M α NP ester is a convenient method for measuring the optical purity of the C₁₅ vitamin E side-chain 1. It should be noted that various (R)- and (S)- α , ω -diffunctional compounds have been successfully prepared as enantiomerically pure (≥99% ee) substances by ZACA-LICA protocol.¹⁹⁻²¹ This novel approach promises to provide a reliable method for highly enantioselective and diastereoselective syntheses of various possible stereoisomers of chiral isoprenoid natural products and analogues. Further studies along this line are currently ongoing in our laboratories.

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References

- (a) Mercier, C.; Chabardes, P. Pure Appl. Chem. 1994, 66, 1509; (b) Dowd, P.; Hershline, R.; Ham. S. W.; Naganathan, S. Nat. Prod. Rep. 1994, 11, 251.
- (a) Nakamura, Y.; Mori, K. *Eur. J. Org. Chem.* **1999**, 2175; (b) Aldrich, J. R.; Oliver, J. E.; Lusby, W. R.; Kochansky J. P.; Borges, M. J. J. *Chem. Ecol.* **1994**, 20, 1103.
- 3. Bennett, C. J.; Caldwell, S. T.; McPhail, D. B.; Morrice, P. C.; Duthie, G. G.; Hartley, R. C. *Bioorg. Med. Chem.* **2004**, *12*, 2079.
- (a) Minale, L. In Marine Natural Products: Chemical and Biological Perspectives; Scheuer, P. J., Ed.; Academic Press, 1978, Vol. 1, pp 175–240; (b) Sankaranarayanan, S.; Sharma A.; Chattopadhyay, S. Tetrahedron: Asymmetry 2002, 13, 1373.
- 5. Zhang, D.; Poulter, C. D. J. Am. Chem. Soc. 1993, 115, 1270.
- (a) Heathcock, C. H.; Finkelstein, B. L.; Aoki, T.; Poulter, C. D. Science 1985, 229, 862; (b) Zillig, W. Curr. Opin. Genet. Dev. 1991, 1, 544.
- (a) Cohen, N.; Schaer, B. J. Org. Chem. 1992, 57, 5783; (b) Hübscher, J.; Barner, R. Helv. Chim. Acta 1990, 73, 1069; (c) Netscher, T. Chimia 1996, 50, 563.
- (a) Fujisawa, T.; Sato, T.; Kawara T.; Ohashi, K. *Tetrahedron Lett.* 1981, 22, 4823; (b) Schmid, R.; Antoulas, S.; Rüttimann, A.; Schmid, M.; Vecchi, M.; Weiser, H. *Helv. Chim. Acta* 1990, 73, 1276.
- (a) Lee, H.; Lee, Y.; Kim, D.; Son, M.; Nam, T.; Jeong, B. *Tetrahedron Lett.* 2014, 55, 5895; (b) Chapelat, J.; Hengartner, U.; Chougnet, A.; Liu, K.; Huebbe, P.; Rimbach, G.; Woggon, W.-D. *ChemBioChem.* 2011, 12, 118.
- Threlfall, D. R. In Secondary Plant Products; Bell, E. A.; Charlwood, B. V., Eds.; Springer, Berlin, 1980, pp 292–298.
- 11. Rowland, S. J.; Robson, J. N. Mar. Environ. Res. 1990, 30, 191.
- 12. Mancuso, C. A.; Odham, G.; Westerdahl, G.; Reeve, J. N.; White, D. C.; J. Lipid Res. 1985, 26, 1120.
- (a) Chan, K.; Cohen N.; De Noble, J. P.; Specian, A. C. Jr.; Saucy, G. J. Org. Chem. 1976, 41, 3497; (b) Koreeda, M.; Brown, L. J. Org. Chem.
 1983, 48, 2122; (c) Fujiwara, J.; Fukutani, Y.; Hasegawa, M.; Maruoka, K.; Yamamoto, H. J. Am. Chem. Soc. 1984, 106, 5004; (d) Berkowitz, W. F.; Wu, Y. Tetrahedron Lett. 1997, 38, 8141.
- (a) Scott, J. W.; Bizzarro, F. T.; Parrish, D. R.; Saucy, G. *Helv. Chim. Acta* **1976**, *59*, 290; (b) Takabe, K.; Sawada; Satani, T.; Yamada, T.; Katagiri, T.; Yoda, H. *Bioorg. Med. Chem. Lett.* **1993**, *13*, 157; (c) Chen, Y. C.; Nagumo, S.; Akita, H. *Chem. Pharm. Bull.* **1996**, *44*, 2153; (d) Fleming, I.; Maiti, P.; Ramarao, C. *Org. Biomol. Chem.* **2003**, *1*, 3989.
- (a) Leuenberger, H. G. W.; Boguth, W.; Barner, R.; Schmid, M.; Zell, R. *Helv. Chim. Acta* **1979**, *62*, 455; (b) Schmid, M.; Barner, R. *Helv. Chim. Acta* **1979**, *62*, 464; (c) Zell, R. *Helv. Chim. Acta* **1979**, *62*, 474; (d) Fuganti, C.; Grasselli, P. J. Chem .Soc. Chem. Commun. **1979**, 995; (e) Nozawa, M.; Takahashi, K.; Kato, K.; Akita, H. *Chem. Pharm. Bull.* **2000**, *48*, 272.
- Takaya, H.; Ohta, T.; Sayo, N.; Kumobayashi, H.; Akutagawa, S.; Inoue, S.; Kasahara, I.; Noyori, R. J. Am. Chem. Soc. **1987**, 109, 1597.
- 17. Wang, A.; Wüstenberg, B.; Pfaltz, A. Angew. Chem. Int. Ed. 2008, 47, 2298.
- (a) Huo, S.; Shi, J.; Negishi, E. Angew. Chem. Int. Ed. 2002, 41, 2141; For the discovery of the ZACA reaction, see: (b) Kondakov, D.; Negishi, E. J. Am. Chem. Soc. 1995, 117, 10771; (c) Kondakov, D. Y.; Negishi, E. J. Am Chem. Soc.1996, 118, 1577; For the recent reviews of the ZACA reaction, see: (d) Negishi, E. Arkivoc 2011, viii, 34; (e) Negishi, E. Angew. Chem. Int. Ed. 2011, 50, 6738; (f) Xu, S.; Negishi, E. Heterocycles 2014, 88, 845.
- 19. Xu, S.; Lee, C. T.; Wang, G.; Negishi, E. Chem. Asian J. 2013, 8, 1829.
- Xu, S.; Oda, A.; Kamada, H.; Negishi, E. Proc. Natl. Acad. Sci. USA, 2014, 111, 8368.
- 21. Xu, S.; Oda, A.; Negishi, E. Chem. Eur. J. 2014, 20, 16060.
- Chen, C. S.; Fujimoto, Y.; Girdaukas, G.; Sih, C. J. J. Am. Chem. Soc. 1982, 104, 7294.
- 23. Terao, J.; Todo, H.; Begum, S. A.; Kuniyasu, H.; Kambe, N. Angew. Chem. Int. Ed. 2007, 46, 2086.
- 24. (a) Erker, G.; Aulbach, M.; Knickmeier, M.; Wingbermühle, D.; Kürger, C.; Nolte, M.; Werner, S. J. Am. Chem. Soc. 1993, 115, 4590; (b) (+)–(NMI)₂ZrCl₂ (CAS number: 641627-68-1) and (–)–(NMI)₂ZrCl₂ (CAS number: 148347-88-0) are available from Sigma-Aldrich and Wako Pure Chemicals.
- (a) Dale, J. A.; Mosher, H. S. J. Am. Chem. Soc. 1973, 95, 512; (b) Hoye, T. R.; Jeffrey, C. S.; Shao, F. Nat. Protoc. 2007, 2, 2451.
- (a) Harada, N.; Watanabe, M.; Kuwahara, S.; Sugio, A.; Kasai, Y.; Ichikawa, A. *Tetrahedron: Asymmetry* **2000**, *11*, 1249; (b) Kasai, Y.; Sugio, A.; Sekiguchi, S.; Kuwahara, S.; Matsumoto, T.; Watanabe, M.; Ichikawa, A.; Harada, N. *Eur. J. Org. Chem.* **2007**, 1811.