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of 2,6,6-tricyano-2-cyclohexenones 3 (3-oxo-2,6,6-tricyano-cyclohexenes) and pointed at the possibility of alcoholysis of the tricyanocyclohexenone derivatives to monoesters¹.

$$R^{2}-CH_{2} \xrightarrow{CN} CN \xrightarrow{base} R^{1} \xrightarrow{CN} CN \xrightarrow{NH_{2}} H^{\oplus} \xrightarrow{R^{1}} CN \xrightarrow{CN} CN$$

$$R^{2}-CH_{2} \xrightarrow{CN} CN \xrightarrow{base} R^{2} \xrightarrow{R^{1}} R^{2} \xrightarrow{CN} CN$$

$$R^{2}-CH_{2} \xrightarrow{CN} CN \xrightarrow{base} R^{2} \xrightarrow{R^{1}} R^{2} \xrightarrow{CN} CN$$

$$R^{2}-CH_{2} \xrightarrow{CN} CN \xrightarrow{base} R^{2} \xrightarrow{R^{1}} R^{2} \xrightarrow{CN} CN$$

$$R^{2}-CH_{2} \xrightarrow{CN} R^{2} \xrightarrow{R^{1}} R^{2} \xrightarrow{CN} CN$$

$$R^{2}-CH_{2} \xrightarrow{R^{1}} R^{2} \xrightarrow{R^{1}} R^{2} \xrightarrow{R^{1}} R^{2}$$

Now we report on a selective method of synthesis of 1,3-dicyano-2-oxo-4,6,6-trialkyl-3-cyclohexene-1-carboxylates 4.

The most common method for preparation of esters from nitriles is the alcoholysis in the presence of acid². This procedure, when applied to 2,6,6-tricyano-2-cyclohexenones 3, leads to a mixture of mono-, di- and triesters which is hardly separable. Based on the fact^{1,3} that the methoxy adduct of 1-amino-2,6,6-tricyano-3,5,5-trimethyl-cyclohexene in aqueous-ethanolic hydrochloric acid solution undergoes transformation to methyl 1,3-dicyano-2-oxo-4,6,6-trimethyl-3-cyclohexene-1-carboxylate we were able to elaborate an efficient method leading to monoketoesters 4.

$$\begin{array}{c|ccccc}
R^{1} & CN & R^{3}O^{\ominus} & R^{3}O & R^{2} & CN & R^{2} & CN & R^{2} & R^{$$

We have prepared the methoxy and ethoxy adducts 5 by treating some ylidenemalononitrile dimers 2 with a base in an appropriate alcohol. The formation of methoxy adducts occurs smoothly with about 80% yield in the presence of a catalytic amount of sodium methoxide. On the other hand, the synthesis of ethoxy adducts generally requires the use of sodium hydroxide and the yield is lower in comparison to the methoxy adducts. Moreover, alkoxy adducts can be easily obtained in one step from ylidenemalononitriles as described previously! (Table 1).

The transformation of adducts 5 to monoketoesters 4 occurs via a transannular rearrangement which involves migration of the alkoxy group to the protonated cyano group at C-6. The imidic ester 6 is formed as an intermediate which undergoes further hydrolysis to the monoketoester 4. Our attempts on the rearrangement of the adducts according to the procedure described for 1-amino-3-methoxy-2,6,6-tricyano-3,5,5-trimethylcyclohexene failed. We found that the best conditions for this reaction are refluxing of the adducts 5 in dilute hydrochloric acid (18%). However, in the case of rearrangement of adducts 5i and 5j we have isolated the corresponding amide 7 as a product of hydrolysis of the intermediate imidic ester 6 by another route with the ester as a by-product (10%). We have determined that, in such cases, the addition of a 2:1 mixture of a suitable alco-

Syntheses with Unsaturated Nitriles; II. Selective Monoesterification of 2,6,6-Tricyano-2cyclohexenones by Transannular Rearrangement

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Recently, we examined the dimerization reaction of alkylidenemalononitriles 1 and developed a convenient synthesis

Table 1. 1-Amino-3-ethoxy(or -methoxy)-2,6,6-tricyanocyclohexenes 5

Produ No.	ict R¹	\mathbb{R}^2	\mathbb{R}^3	Yield ^a [%]	m.p. [°C]	Molecular formula ^b	¹H-N.M.R. δ [ppm]	M.S. <i>m/e</i> (M ⁺)
5a	H ₃ C	Н	H ₃ C	79–80	210-212°	C ₁₃ H ₁₆ N ₄ O (244.3)	1.1 (d, 6H, CH ₃); 1.42 (d, 2H, CH ₂); 1.8 (s, 3H, H ₃ C-C-); 3.7 (s, 3H, OCH ₃); 7.97 (s, 2H, NH ₂)	244
5b	H ₃ C	Н	C_2H_5	42-58	151-152°	C ₁₄ H ₁₈ N ₄ O (258.3)	1.8 (m, 6H, CH ₃ + CH ₃ CH ₂ O); 4.15 (q, $J = 10$ Hz, 2H, CH ₃ CH ₂ O); 7.92 (s, 2H, NH ₂)	258
5c	C_2H_5	Н	H ₃ C	98	103–105°	C ₁₅ H ₂₀ N ₄ O (272.3)	0.7-2.1 (m, 15 H); 3.75 (s, 3 H, OCH ₃); 5.17 (s, 2 H, NH ₂)	272
5d	C_2H_5	Н	C_2H_5	69	146–147°	C ₁₆ H ₂₂ N ₄ O (286.4)	0.8-2.1 (m, 18 H); 4.17 (q, $J=10$ Hz, 2 H, CH ₃ CH ₂ O); 5.12 (s, 2 H, NH ₂)	286
5e	n-C ₃ H ₇	Н	H ₃ C	84	116-117°	C ₁₇ H ₂₄ N ₄ O (300.4)	0.8-2.1 (m, 19 H); 3.8 (s, 3 H, OCH ₃); 5.2 (s, 2 H. NH ₂)	300
5f	n-C ₃ H ₇	Н	C_2H_5	84	142-143°	$C_{18}H_{26}N_4O$ (314.4)	$0.8-2.1$ (m, 22H); 4.2 (q, $J = 8.75$ Hz, 2H, CH ₃ C $\underline{\text{H}}_2\text{O}$); 5.15 (s, 2H, NH ₂)	314
5g	—(CH ₂)3—	H ₃ C	80	210°	C ₁₇ H ₂₀ N ₄ O (296.4)	1.3-2.5 (m, 15 H); 3.75 (s, 3 H, OCH ₃); 5.5 (s, 2 H, NH ₃)	296
5h	—(CH ₂) ₃ —		C_2H_5	72	173–174°	$C_{18}H_{22}N_4O$ (310.4)	1.2–2.5 (m, 18 H); 4.25 (q, $J = 8.75$ Hz, 2 H, CH ₁ CH ₂ O); 5.05 (s, 2 H, NH ₂)	310
5i	—(CH ₂)4	H ₃ C	79	220-222°	C ₁₉ H ₂₄ N ₄ O (324.4)	1.3-2.3 (m, 19H); 3.82 (s, 3H, OCH ₃); 4.8 (s, 2H, NH ₂)	324
5j	-(CH ₂)4	C_2H_5	70	202–206°	$C_{20}H_{26}N_4O$ (338.4)	1.3-2.4 (m, 22 H); 4.22 (q, $J = 8.75$ Hz, 2 H, CH_3CH_2O)	338

^a Yield of isolated product.

Table 2. Ethyl (or Methyl) 1,3-Dicyano-2-oxo-3-cyclohexene-1-carboxylates 4

Product Yield ^a				Yielda	m.p.	Molecular	I.R.	'H-N.M.R.	M.S.
No.	R¹	R ²	R³	[%]	[°C]	formula ^b	ν [cm ⁻¹]	δ [ppm]	m/e (M+)
4a	H ₃ C	Н	H ₃ C	90-95	140°, 142–143°	C ₁₃ H ₁₄ N ₂ O ₃ (246.3)	2260 (CN); 1750, 1700 (CO)	1.2 (d, 6H, CH ₃); 2.35 (s, 3H, H ₃ C-C-); 2.75 (d, 2H, CH ₂); 3.8 (s, 3H, OCH ₃)	246
4b	H ₃ C	Н	C_2H_5	92-93	109° 108–109°	C ₁₄ H ₁₆ N ₂ O ₃ (260.3)	2260 (CN); 1740, 1690 (CO)	1.2–1.45 (m, 9 H); 2.4 (s, 3 H, H ₃ C—C—); 2.75 (d, 2 H, CH ₂); 4.42 (q, <i>J</i> = 7.5 Hz, 2 H, CH ₃ CH ₂ O)	260
4c	C_2H_5	Н	CH ₃	95	75–77°	$C_{15}H_{18}N_2O_3$ (274.3)	2260 (CN); 1760, 1710 (CO)	0.7-1.4 (m, 11 H); 2.67 (q, $J=10$ Hz, 4H, CH ₂ C=-); 3.85 (s, 3 H, OCH ₃)	274
4d	C_2H_5	Н	C ₂ H ₅	72	69–70°	C ₁₆ H ₂₀ N ₂ O ₃ (288.3)	2260 (CN); 1750, 1700 (CO)	0.9-1.4 (m, 16 H); 2.65 (q, $J=8.75$ Hz, 2 H, CH ₂ —C—); 4.27 (q, $J=8.75$ Hz, 2 H, CH ₃ CH ₃ O)	288
4e	n - C_3H_7	Н	CH ₃	91	oil	$C_{17}H_{22}N_2O_3$ (302.4)	2260 (CN); 1760, 1700 (CO)	0.8–1.9 (m, 15 H); 2.5–3.0 (m, 4 H, CH ₂ —C—); 3.87 (s. 3 H, OCH ₃)	302
4f	n-C ₃ H ₇	Н	C_2H_5	88	97-100°	$C_{18}H_{24}N_2O_3$ (316.4)	2260 (CN); 1750, 1705 (CO)	0.8–1.85 (m, 18 H); 2.5–2.9 (m, 4 H, CH ₂ —C—); 4.30 (q, <i>J</i> = 8.75 Hz, 2 H, CH ₃ CH ₂ O)	316
4g	(CH ₂) ₃		CH ₃	83	130-131°	$C_{17}H_{16}N_2O_3$ (298.3)	2260 (CN); 1750, 1700 (CO)	1.5-2.4 + 2.7-3.3 (2 m, 15 H); 3.9 (s, 3H, OCH ₃)	298
4h	—(CH ₂) ₃ —		C_2H_5	81	125–128°	$C_{18}H_{20}N_2O_3$ (312.4)	2255 (CN); 1750, 1710 (CO)	1.1–3.2 (m, 18 H); 4.3 (q, J =8.75 Hz, 2H, CH ₃ CH ₂ O)	312
4i	(CH ₂) ₄		CH ₃	49	134°	$C_{19}H_{22}N_2O_3$ (326.4)	2260 (CN); 1760, 1710 (CO)	1.0-2.25 (m, 19 H); 3.82 (s, 3 H, OCH ₃)	326
4j	-(CH ₂)		C_2H_5	47	190–191°	$C_{20}H_{24}N_2O_3$ (340.4)	2260 (CN); 1740, 1710 (CO)	1.1–3.3 (m, 22 H); 4.27 (q, $J = 8.75$ Hz, 2H, CH ₃ CH ₂ O)	340
7	(CH ₂) ₄		$R^{3}O = NH_{2}$	40	222–225°	$C_{18}H_{21}N_3O_2$ (311.4)	3340 (NH); 2290 (CN); 1770, 1725 (CO)	1.2-2.4 (m, 16 H); 2.97 (s, 3 H, OCH ₃); 4.65 (s, 2 H, NH ₂)	311

^a Yield of isolated product.

hol with concentrated hydrochloric acid favours the formation of monoketoesters as the main product in satisfactory yield (Table 2).

1-Amino-3-methoxy-2,6,6-tricyanocyclohexenes 5 ($R^3 = CH_3$); General Procedure:

The dimer 2 (4 mmol), methanol (10 ml), and a few drops of methanolic sodium methoxide are heated to dissolve the dimer. Then

^b The microanalyses were in good agreement with the calculated values: C, ± 0.28 ; H, ± 0.22 ; N, ± 0.19 .

^b The microanalyses were in good agreement with the calculated values: C, ± 0.23 ; H, ± 0.15 ; N, ± 0.12 .

the mixture is stirred for 2 h. The separated crystals of adduct are isolated and the rest of adduct is precipitated by adding a small amount of water to the mother liquid; yield: $\sim 80\%$. Analytical samples are crystallised from methanol or ethanol.

1-Amino-3-ethoxy-2,6,6-tricyanocyclohexenes 5 ($R^3 = C_2H_5$); General Procedure:

The dimer 2 (4 mmol), ethanol (10 ml), and sodium hydroxide (0.1 g) are treated as in the procedure described above; yield: $\sim 70\%$.

Methyl 1,3-Dicyano-2-oxo-3-cyclohexene-1-carboxylates 4 (R³ = CH₃); General Procedure:

The adduct 5 ($R^3 = CH_3$; 4 mmol) in 18% hydrochloric acid (10 ml) is heated under reflux for 1 h. The mixture is then allowed to cool and a few drops of methanol are added to solidify the oily product. The solidified product is crystallised from methanol.

Ethyl 1,3-Dicyano-2-oxo-cyclohexene-1-carboxylate 4 ($R^3 = C_2H_5$): The adduct 5 ($R^3 = C_2H_5$; 2 mmol) in 18% hydrochloric acid (10 ml) is worked up as in the procedure described above.

2',4'-Dicyano-2'-methoxycarbonyl-3'-oxo-3',4',4'a,5',6',7',8',8'a-octahydro-spiro[cyclohexane-1,1'(2'H)-naphthalene] (4i):

The adduct 5i (3 mmol) is refluxed for 1 h in a mixture of methanol (20 ml) and concentrated hydrochloric acid (10 ml). The mixture is then allowed to cool and precipitated 2'-aminocarbonyl derivative is filtered off. The ester 4i is precipitated from the mother liquid by adding water (10 ml) and crystallised from methanol.

2',4'-Dicyano-2'-ethoxycarbonyl-3'-oxo-3',4',4a',5',6',7',8',8'a-octahydro-spiro[cyclohexane-1,1'(2'H)-naphthalene] (4j):

The adduct 5j (2 mmol) is refluxed for 1 h in a mixture of methanol (20 ml) and concentrated hydrochloric acid (10 ml). The mixture is then allowed to cool and precipitated ester 4j is filtered off. The 2'-aminocarbonyl derivative is precipitated from the mother liquid by adding water (10 ml).

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