This article was downloaded by: [Pennsylvania State University] On: 19 March 2013, At: 02:00 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

Acid-Catalyzed Ether Rearrangement: Total Synthesis of Isomimosifoliol and (±)-Dihydromimosifoliol

Veera Reddy Arava $^{\rm a}$, Sashibhushan Malreddy $^{\rm a}$ & Sreenivasulu Reddy Thummala $^{\rm b}$

^a Research and Development Laboratories, Suven Life Sciences Ltd., Hyderabad, India

^b Department of Chemistry, S. K. University, Anantapur, India Accepted author version posted online: 02 Jan 2012. Version of record first published: 20 Aug 2012.

To cite this article: Veera Reddy Arava , Sashibhushan Malreddy & Sreenivasulu Reddy Thummala (2012): Acid-Catalyzed Ether Rearrangement: Total Synthesis of Isomimosifoliol and (±)-Dihydromimosifoliol, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 42:24, 3545-3552

To link to this article: <u>http://dx.doi.org/10.1080/00397911.2011.567882</u>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <u>http://www.tandfonline.com/page/terms-and-conditions</u>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



Synthetic Communications[®], 42: 3545–3552, 2012 Copyright © Suven Life Sciences Ltd. ISSN: 0039-7911 print/1532-2432 online DOI: 10.1080/00397911.2011.567882

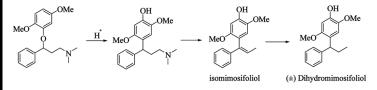
ACID-CATALYZED ETHER REARRANGEMENT: TOTAL SYNTHESIS OF ISOMIMOSIFOLIOL AND (\pm) -DIHYDROMIMOSIFOLIOL

Veera Reddy Arava,¹ Sashibhushan Malreddy,¹ and Sreenivasulu Reddy Thummala²

¹Research and Development Laboratories, Suven Life Sciences Ltd., Hyderabad, India

²Department of Chemistry, S. K. University, Anantapur, India

GRAPHICAL ABSTRACT



Abstract An acid-catalyzed rearrangement of benzyl phenyl ethers to diphenylmethane derivatives, followed by salt formation and Hofmann elimination, is a simple method for the syntheses of isomimosifoliol and dihydromimosifoliol.

Keywords Benzyl phenyl ethers; ether rearrangement; Hofmann elimination; tetraalkyl-ammonium salts

INTRODUCTION

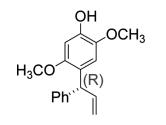
(+)-Mimosifoliol (1) was recently isolated by Wall et al.^[1] from the root of *Aeschynomene mimosifolia* Vatke (Leguminosae) along with (+)-mimosifolenone (2). Mimosifoliol (1) demonstrated weak activity in a DNA strand scission assay. A concentration of $25 \,\mu\text{g/ml}$ of 1 was approximately equal in activity to $0.1 \,\mu\text{g/ml}$ of bleomycin sulfate. Despite its activity in the assay, (+)-mimosifoliol (1) proved to be inactive against several human cancer cell lines. We herein report a total synthesis of isomimosifoliol in a new route developed by our group.

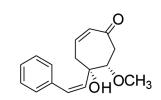
We have recently observed a novel acid-catalyzed rearrangement on ethers during the investigation of duloxetine^[2] as shown in Scheme 1.

Using this novel rearrangement as a key step, we have synthesized tolterodine in good yields^[3] as shown in Scheme 2.

Received December 18, 2010.

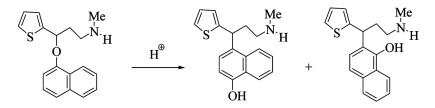
Address correspondence to Veera Reddy Arava, R&D Laboratories, Suven Life Sciences Ltd., #18, Phase-III, Jeedimetla, Hyderabad 500 055, India. E-mail: reddyvenis@rediffmail.com



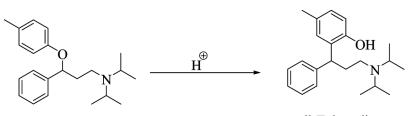


(+)-Mimosifoliol,1

(+)-Mimosifolenone, 2

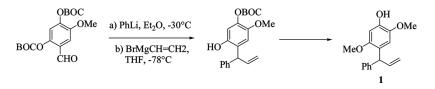


Scheme 1. Novel ether rearrangement in duloxetine.



dl-Tolterodine

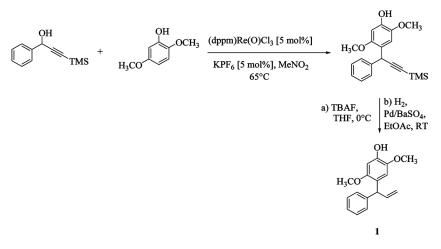
Scheme 2. Key rearrangement step in tolterodine.



Scheme 3. Pettus strategy.

The first synthesis of mimosifoliol by Pettus used o-quinone methide intermediate strategy with protection and deportation of the phenolic group^[4] as shown in Scheme 3.

The second synthesis by Toste et al. utilized rhenium-catalyzed aromatic propargylation as a key step without protection of phenols^[5] as shown in Scheme 4.



Scheme 4. Toste methodology.

RESULTS AND DISCUSSION

In our quest to develop a general methodology for derivatives and analogs of mimosifoliol (1), we adopted our recent acid-catalyzed rearrangement method^[2] as shown in Scheme 1.

Mannich adduct of acetophenone was reduced with sodium borohydride to give alcohol (3). Alcohol was reacted with 2,5-dimethoxy fluorobenzene to give an ether product (4). The ether (4) was rearranged in perchloric acid to give phenols **5a** and **5b**. The compounds **4**, **5a**, and **5b** were characterized by their spectral data and microelemental analysis. Compounds **5a** and **5b** were reacted with methyl iodide to give quaternary salts **6a** and **6b** as shown in Scheme 5.

The salt (6a) was characterized by its characteristic spectral data. The quaternary salt was subjected to Hofmann elimination conditions to get mimosifoliol (1) as shown in Scheme 6.

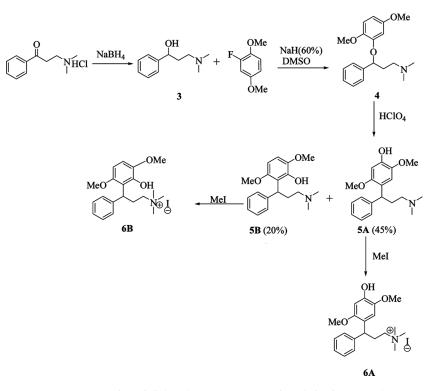
During Hofmann elimination with butyl lithium, it was observed that along with a minor dl-mimosifoliol, a major isomimosifoliol (7) was also formed. Isomimosifoliol (7) was characterized thoroughly and also converted into (\pm) -dihydromimosifoliol (8) by hydrogenation. Both compounds were characterized by spectral data and microelemental analysis.

During the rearrangement of ether (4), the *o*-isomer (5b) was also formed (20%) along with the *p*-isomer (5a) (45%). The product 5b was reacted with methyl iodide and the formed salt 6b was subjected to Hofmann elimination under 1,8-diazabicyclo [5.4.0]under-7-ese (DBU)/toluene-isopropyl alcohol (IPA) reflux conditions to give 5,8-dimethoxy-4-phenyl-chroman (or dl-cyclomimosifoliol) (9) as shown in Scheme 7, which has been characterized thoroughly.

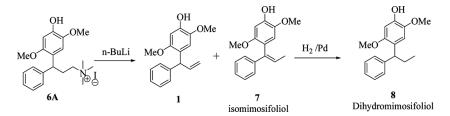
EXPERIMENTAL

¹H and ¹³C NMR spectra were recorded using a Bruker 400 spectrophotometer (400 and 100 MHz respectively) with tetramethylsilane (TMS) as internal standard.

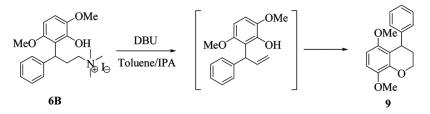
3547



Scheme 5. Preparation of diphenyl quaternary ammonium derivatives 6a and 6b.



Scheme 6. Hofmann elimination product of 6a.



Scheme 7. Hofmann elimination product of 6b.

Infrared (IR) spectra were recorded on a Perkin-Elmer spectrophotometer as KBr pellets or neat. Analytical thin-layer chromatography (TLC) was conducted on E-Merck 60F254 aluminum-backed silica-gel plates (0.2 mm). Developed plates were visualized using ultraviolet (UV) light or iodine chamber. High-performance liquid chromatographic (HPLC) spectra were recorded on a Shimadzu 2010 instrument.

Preparation of 3-Dimethylamino-1-phenyl-propan-1-ol (3)

3-Dimethylamino-1-phenyl propan-1-one hydrochloride (70.0 g, 0.328 mol) was dissolved in methanol (350.0 mL) and cooled to 10–15 °C. Sodium borohydride solution [12.48 g, 0.329 mol (sodium borohydride dissolved in 130.0 mL of 10% aqueous NaOH solution)] was added over 30 min and the reaction mass was stirred for 3 h. Reaction progress was monitored by TLC, and after completion of the reaction, solvent was evaporated under vacuum below 60 °C to get the crude alcohol. The crude product was dissolved in ethyl acetate and washed with water (2 × 500.0 mL). The organic layer was dried over anhydrous sodium sulfate, and concentration of solvent under vacuum gave a colorless liquid (54.0 g, yield: 92%).

IR (neat, cm⁻¹): 3747, 3350, 2824, 1463, 515; ¹H NMR (400 MHz, CDCl₃): δ 7.39 (m, 4H), 7.24 (t, 1H, J = 7.08 Hz), 5.26 (s, 1H), 4.94 (q, 1H, J = 4.12 Hz), 2.63 (m, 1H), 2.43 (m, 1H), 2.29 (s, 6H), 1.84 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 145.09, 128.09, 126.76, 125.49, 75.49, 58.24, 45.23, 34.55; ESI-MS (+Ve mode): m/z (%) = 181.03 [M + 2]; Purity by HPLC: 99.92%.

Preparation of [3-(2,5-Dimethoxyphenoxy)3-phenylpropyl]dimethylamine (4)

Sodium hydride (15.64 g, 60%, 0.391 mol) was added to a solution of 3-dimethylamino-1-phenyl propan-1-ol (50.0 g, 0.279 mol) in dimethyl sulfoxide (250.0 mL) and stirred for 30 min at rt. 2,5-Dimethoxy-1-fluorobenzene (52.34 g, 0.335 mol) was added to the reaction mass, heated to 135–140 °C, and stirred for 4 h. Reaction progress was monitored by TLC, and after completion of the reaction, the mass was cooled to rt. Water (300.0 mL) was added, and the product was extracted into toluene $(2 \times 250.0 \text{ mL})$. The toluene layer was washed with saturated brine solution $(2 \times 100.0 \text{ mL})$ followed by water (100.0 mL). The toluene layer was dried over anhydrous sodium sulfate and concentrated under vacuum at <70 °C to get the crude product (87.5 g). The obtained crude was dissolved in ethyl acetate (200.0 mL) and 30.0 g of 85% o-phosphoric acid was added drop wise at (25-30 °C). After the addition, the reaction mixture was stirred for 30 min, cooled to $10-15 \,^{\circ}\text{C}$, and stirred for 1 h. The slurry was then filtered and the solids were washed with 50.0 mL of cold ethyl acetate (10–15 °C). The solid was stirred in a mixture of 50.0 mL dichloromethane and 100.0 mL of water at $25-30 \,^{\circ}\text{C}$, and 90.0 mL of 30%aqueous ammonium hydroxide solution was added. The reaction mixture was stirred for 10 min at 25–30 °C, and the layers were separated. The organic layer was washed with water $(50.0 \text{ mL} \times 2)$ and dried over anhydrous sodium sulfate. The solution was concentrated at 35-40 °C to get solid material (55.0 g, yield: 63%).

IR (KBr, cm⁻¹): 3433, 2946, 1690, 1878, 1609, 1509, 1460, 1231, 1162, 762; ¹H NMR (400 MHz, CDCl₃): δ 7.39 (d, 2H, J = 7.3 Hz), 7.31 (t, 2H, J = 7.3 Hz), 7.25 (t, 2H, J = 7.22 Hz), 6.78 (d, 1H, J = 8.7 Hz), 6.39 (d, 1H, J = 2.7 Hz), 6.35 (dd, 1H, $J_1 = 8.6$, $J_2 = 2.8$ Hz), 5.24 (t, 1H, J = 5.6 Hz), 3.83 (s, 3H), 3.50 (s, 3H), 2.46 (t, 2H, J = 7.2 Hz), 2.30 (m, 1H), 2.23 (s, 6H), 2.03 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 153.93, 148.66, 144.32, 141.72, 128.42, 127.47, 126.03, 113.27, 104.26, 104.15, 56.93, 55.75, 55.32, 45.39, 36.40; ESI-MS (+Ve mode): m/z (%) = 316.3 [M+1]; DSC: 57.62 °C; Purity by HPLC: 99.95%. Anal. calcd. for C₁₉H₂₅NO₃ (315.40): C, 72.35; H, 7.99; N, 4.44. Found: C, 72.0961; H, 7.8667; N, 4.7088.

Preparation of 4-(3-Dimethylamino-1-phenyl-propyl)2,5dimethoxy-phenol (5)

To a solution of [3-(2,5-dimethoxy-phenoxy)3-phenyl-propyl]-dimethylamine **4** (60.0 g, 0.19 mol) in dichloromethane (400.0 mL) was added 100.0 mL of perchloric acid (71–73%) at 0–5 °C. The reaction mixture was stirred for 3 h at 0–5 °C, and the reaction progress was monitored by TLC. After the staring material disappeared ice water was added to the reaction mixture and basified (pH 9–10) with aqueous ammonium hydroxide solution. The reaction mixture was stirred for 30 min, and the layers were separated. The aqueous layer was extracted with dichloromethane (2 × 100.0 mL). Combined organic layers were washed with water (2 × 100.0 mL) and dried over anhydrous sodium sulfate. The solvent was separated at 40 °C to get 40.0 g of crude material. The crude product mixture was separated using column chromatography. Three compounds, *para*-isomer (**5a**, 45%), *ortho*-isomer (**5b**, 22%), and 2,5-dimethoxy phenol (5.0 g, 10%), were isolated and characterized as follows.

Para-isomer (5a). IR (KBr, cm⁻¹): 2830, 2477, 1587, 1466, 1199, 1041, 787; ¹H NMR (400 MHz, CDCl₃): δ 10.09 (s, 1H), 7.26 (m, 5H), 6.72 (s, 1H), 6.50 (s, 1H), 4.35 (t, 1H, J=7.6 Hz), 3.77 (s, 3H), 3.69 (s, 3H), 2.27 (m, 2H), 2.19 (s, 6H), 2.14 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 151.43, 144.98, 144.75, 140.67, 128.09, 127.77, 125.70, 123.67, 111.03, 99.94, 45.21, 40.90, 32.69; ESI-MS (+Ve mode): m/z (%) = 316.30 [M + 1]; DSC: 139.03 °C; Purity by HPLC: >99%. Anal. calcd. for C₁₉H₂₅NO₃ (315.40): C, 72.35; H, 7.99; N, 4.44. Found: C, 71.9882; H, 8.2287; N, 4.9499.

Ortho-isomer (5b). IR (KBr, cm⁻¹): 2936, 2860, 2775, 1597, 1480, 1359, 1244, 1151, 1094, 701; ¹H NMR (400 MHz, CDCl₃): δ 11.0 (s, 1H), 7.26 (m, 5H), 6.70 (d, 1H, J=8.7 Hz), 6.26 (d, 1H, J=7.8 Hz), 4.55 (t, 1H, J=9.0 Hz), 3.86 (s, 3H), 3.20 (s, 3H), 2.43 (m, 4H), 2.37 (s, 6H); ¹³C NMR (100 MHz, CDCl₃): δ 151.43, 144.97, 144.67, 140.60, 128.09, 127.77, 125.69, 123.74, 111.00, 99.86, 58.16, 56.47, 56.02, 45.35, 40.90, 32.75; ESI-MS (+Ve mode): m/z (%) = 316.30 [M + 1]; DSC: 118.31 °C; Purity by HPLC: >95%. Anal. calcd. for C₁₉H₂₅NO₃ (315.40): C, 72.35; H, 7.99; N, 4.44. Found: C, 72.1818; H, 8.1739; N, 4.6845.

Quaternary Salt Preparation of 4-(3-Dimethylamino-1-phenyl-propyl)-2,5-dimethoxy-phenol with Methyl lodide (6a)

To a solution of 4-(3-dimethylamino-1-phenyl-propyl)-2,5-dimethoxy-phenol **5a** (8.0 g, 0.025 mol) in dichloromethane (100.0 mL) was added methyl iodide (4.50 g, 0.0317 mol). The reaction mixture was stirred for 1 h at rt, and the reaction

progress was monitored by TLC. After the disappearance of starting material, the salt was filtered from the reaction mixture, and the cake was washed with dichloromethane (20.0 mL) to get a white compound (10.62 g, yield: 92%).

IR (KBr) (cm⁻¹): 3231, 1521, 1452, 1209, 730, 481; ¹H NMR (400 MHz, CDCl₃): δ 8.96 (s, 1H), 7.33 (m, 4H), 7.17 (t, 1H, J = 7.1 Hz), 6.94 (s, 1H), 6.45 (s, 1H), 4.22 (t, 1H, J = 7.6 Hz), 3.72 (s, 3H), 3.64 (s, 3H), 3.18 (m, 2H), 3.0 (s, 9H), 2.5 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 151.29, 146.36, 144.49, 141.91, 128.69, 127.90, 126.48, 121.03, 113.21, 101.15, 64.96, 57.34, 56.52, 52.61, 27.39; ESI-MS (+Ve mode): m/z (%) = 330.02 [M – I]. Melting point: 262.2–263.5 °C.

Preparation of Isomimosifoliol (7)

The quaternary salt **6a** (6.0 g, 0.013 mol) was suspended in tetrahydrofuran (THF), and the reaction mass was cooled to 0-5 °C. n-Butyl lithium in hexane (1.6 M, 36.0 mL) was added over a period of 15 min at 0-5 °C, and slowly the reaction mixture temperature was raised to 25-30 °C over a period of 1 h. It was stirred for 1 h at rt, and the reaction progress was monitored by TLC. After the starting material disappeared, ice water was added to the reaction mixture. The product was extracted with ethyl acetate (4 × 25.0 mL) and dried over anhydrous sodium sulfate. The combined organic layer was concentrated at 60 °C to get a light yellow material (4.0 g).

GC-MS of crude sample shows 31% of mimosifoliol **1** and **35%** of isomimosifoliol **7**.

Pure product was isolated from the crude mixture by column chromatography eluting with hexane and ethyl acetate (9:1) and recrystallized two times with methylenechloride and hexane mixture as solvent to get pure isomimosifoliol 7. (We could not isolate mimosifoliol in pure form; it always was mixture of 1 and 7).

IR (KBr, cm⁻¹): 3519, 2938, 2914, 2840, 2025, 1630, 1598, 1509, 1416, 1296, 1216, 1037, 655; ¹H NMR (400 MHz, CDCl₃): δ 7.27 (m, 5H), 6.66 (s, 1H), 6.58 (s, 1H), 6.29 (q, 1H, J = 6.8 Hz), 5.65 (s, 1H), 3.81 (s, 3H), 3.65 (s, 3H), 1.68 (d, 3H, J = 6.8 Hz); ¹³C NMR (100 MHz, CDCl₃): δ 151.81, 145.39, 142.14, 140.31, 138.30, 127.95, 126.38, 126.10, 124.93, 119.38, 113.95, 99.84, 56.53, 56.25, 15.63; ESI-MS (+Ve mode): m/z (%) = 271.32 [M + 1]; melting point: 101.2–104.4 °C; GC-MS: 99.28%. Anal. calcd. for C₁₇H₁₈O₃ (270.32): C, 75.53; H, 6.71. Found: C, 75.6458; H, 6.8322.

Preparation of Dihydromimosifoliol (8)

To a solution of isomimosifoliol **7** (75.0 mg, 0.277 mmol) in methanol was added 5% Pd/C (wet, 10.0 mg) in an autoclave. It was hydrogenated at 50 psi for 2 h at 35–40 °C. The reaction progress was monitored by TLC. After the starting material disappeared, the reaction mass was cooled to rt, filtered through a celite bed, and concentrated under vacuum at <60 °C to get the crude product. The product was separated by column chromatography and after bulb-to-bulb distillation gave an off-white solid (47.5 mg, yield: 63%).

REFERENCES

- Fullas, F.; Kornberg, L. J.; Wani, M. C.; Wall, M. E.; Farnsworth, N. R.; Chagwedera, T. E.; Kingharm, A. D. J. Nat. Prod. 1996, 59, 190–192.
- 2. Reddy, A. V.; Rao, S. B. U.; Reddy, B. S. Ind. J. Chem. 2007, 46B, 1695-1698.
- 3. Reddy, A. V.; Reddy, B. S.; Malreddy, S.; Golla, N. Synth. Commun. 2011, 41, 1565-1571.
- (a) Tuttle, K.; Rodringnez, A. A.; Pettus, T. R. R. Synlett. 2003, 2234–2236; (b) Selenski, C.; Pettus, T. R. R. J. Org. Chem. 2004, 69, 9196–9203; (c) Selenski, C.; Mejorado, L. H.; Pettus, T. R. R. Synlett 2004, 1101–1103.
- 5. Kennedy-Smith, J. J.; Young, L. A.; Toste, F. D. Org. Lett. 2004, 6, 1325-1327.