

Article

Chichibabin-type condensation of cyclic ketones with 3-R-1,2,4-triazin-5(4H)-ones

Ilya Nikolaevich Egorov, Tatyana A. Tseitler, Igor S. Kovalev, Pavel A. Slepukhin, Vladimir L. Rusinov, and Oleg Nikolaevich Chupakhin

J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/jo300697v • Publication Date (Web): 21 Jun 2012 Downloaded from http://pubs.acs.org on June 29, 2012

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



The Journal of Organic Chemistry is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

Chichibabin-type condensation of cyclic ketones with 3-R-1,2,4triazin-5(4H)-ones

Ilya N. Egorov,^{a,*} Tatyana A. Tseitler,^a Igor S. Kovalev,^a Pavel A. Slepukhin,^b Vladimir L. Rusinov^{a,b} and Oleg N. Chupakhin^{a,b}

^a Ural Federal University, Department of Organic Chemistry,
Mira St. 19, Ekaterinburg, 620 002, Russia, e-mail: <u>i.n.egorov@gmail.com</u>
^b Institute of Organic Synthesis Ural Branch of Russian Academy of Sciences,
S. Kovalevskoy St. 22, Ekaterinburg, 620 041, Russia

Abstract Graphic



Abstract

Reactions between substituted 1,2,4-triazines and ketones were investigated. General procedures for one-pot synthesis of hydrogenated derivatives of such polycyclic systems as benzo[c][1,2,4]triazino[1,6-a][2]azecine, [1,2,4]triazino[1,6-f]phenantridine and dicyclopenta[b,d]pyrido[1,2-f][1,2,4]triazine are described.

Keywords: 1,2,4-triazines, azecines, cyclic ketones, cycloaddition, fused pyridines, cyclic lactames

Introduction

Fused pyridines are widespread among natural organic compounds,¹ they are of great importance for medicine.² Pyridines, fused with two aliphatic rings draw considerable attention of researchers, as 1,2,3,4,7,8,9,10-octahydrophenantrene shows pesticide activity.³ Triazine-fused pyridines are known to exhibit antidepressant activity.⁴ The most popular synthetic route to bis-fused angular pyridines is Chichibabin's synthesis from cyclic ketones, ammonia and aldehydes. The reaction requires high pressure,^{3,5} or hydrothermal conditions in aqueous ammonium chloride.⁶ Procedures that utilize activated forms of cyclic ketones, such as the product of dimerization of cyclohexanone – 1,1'-bi(cyclohexilidene)-2-one,⁷ enamines of cyclic ketones^{2b,8} are noteworthy. Syntheses from pyrilium salts⁹ and cyclobutadienes¹⁰ are also known. The source of nitrogen in the molecule of fused pyridine in most cases is ammonia,^{5,6,9} while amides,^{7a} urea,^{7b,8d} 2-azadienes,^{8a} enamines,^{8b} carbodiimides,^{8c} ethylcyanoacetate,¹⁰ dienamides¹¹ are also used.

Results and Discussion

In this paper we describe a detailed study of synthetic methods for synthesis of 1,2,4-triazine derivatives of angular bis-fused pyridines in the reactions of cyclic ketones with 1,2,4-triazines. Compounds containing fragments of 1,2,4-triazine (6-azauridine) arouse interest because some of them show anticancer,¹² antivirus¹³ and antibacterial activity¹⁴. It is known, that 3-R-1,2,4-triazin-5(4H)-ones react with C-nucleophiles such as indoles,¹⁵ pyrroles,¹⁵ phenols,¹⁶ anilines¹⁶. We investigated the reaction between 3-R-1,2,4-triazin-5(4H)-ones and ketones. It was shown that the reaction of 3-R-1,2,4-triazin-5(4H)-ones (**1a,b**) with acetophenone (**2**) in the presence of acids leads to the formation of the expected products of nucleophilic addition to the double C=N bond of **3a,b** (Scheme 1). When acetone was used, resinification of the reaction mixture was observed, while butanone-2 and hexanone-2 did not react.



Earlier we had demonstrated that cycloaddition of 3-R-1,2,4-triazin-5(4H)-ones and cyclohexanone (4) or cyclopentanone (5) in the presence of trifluoroacetic acid leads to the formation of fused pyridines of Chichibabin's type.¹⁷ In the course of our research we investigated the process of formation of fused pyridines. The search for proper conditions of the reaction of 3-R-1,2,4-triazin-5(4H)-one (1a) with cyclohexanone (4) under the action of bases showed only methanolic solution of NaOCH₃ to be a suitable reaction medium. This reaction gives product **6a** with low yield (7%). So our studies were then switched to acidic conditions. The following conditions were modified to find an optimal combination: temperature, acid strength, addition of some oxidative and dehydrating agents (Tables 1,2).

 Table 1. Reaction of 3-Ph-1,2,4-triazin-5(4H)-one (1a) with cyclohexanone 4 under different conditions.



Acid	Solvent	Т, ⁰ С	Time	Yield of 6a , %
p-TsOH	DMF	153	0.5 h	29
p-TsOH	DMF	153	12 h	49
p-TsOH	CH ₃ CN	82	1 h	34
CF ₃ COOH	DMF	100	6 h	30
CF ₃ SO ₃ H	DMF	100	6 h	22 *
AlCl ₃	DMF	20	1 day	36
p-TsOH, DDQ	CH ₃ CN	82	3 h	28
p-TsOH, DDQ, P ₂ O ₅	CH ₃ CN	82	3 h	21
CF ₃ COOH, CAN	DMF	20	3 days	19
CF ₃ COOH	DMF	20	7 days	44 ¹⁷
H ₃ PO ₄	H_3PO_4	50	3 h	12
CH ₃ COOH	CH ₃ COOH	20	3 days	0
CH ₃ COOH	CH ₃ COOH	118	3 h	31
CF ₃ SO ₃ H	CH ₃ CN	82	6 h	31 *
CF ₃ SO ₃ H	CH ₃ OH	65	6 h	19 *
CF ₃ SO ₃ H	DMF	153	0.5 h	10 *
HCl	EtOH	20	1 day	40 **

*yield of **6a** triflate (**6aa**); **yield of the **6a** chloride (**6ab**).

Irrespective of the mechanism, Chichibabin synthesis of pyridines includes an oxidative stage, it is known that such oxidative agents as copper (II) acetate, nitrobenzene or air oxygen have no effect on the reaction.¹⁸ We demonstrated the same to be true for DDQ and CAN $((NH_4)_2Ce(NO_3)_6)$. It turned out that if protic acids in the reaction are replaced by Lewis acids, such as AlCl₃, the same product **6a** is formed. Our results indicate that the most efficient route to cycloaddition products **6,7** in zwitter-ionic form is performing the reaction of 3-R-1,2,4-triazin-5(4H)-ones (**1a-d**) with ketones **4,5** in the presence of p-TsOH either in refluxing DMF or in refluxing acetonitrile (Tables 1,2). The addition of P₂O₅ in the reaction mixture as a dehydrating agent leads to increase in the yield of **6a,c-d**.

Table 2. Reaction of 3-R-1,2,4-triazin-5(4H)-ones (**1a-d**) with ketones **4**,**5** in the presence of p-TsOH (1 equiv).



Starting material	R	n	Conditions	Time, h	Product	Yield, %
1a	Ph	2	DMF	1	6a	44
1a	Ph	2	CH ₃ CN, P ₂ O ₅	3	6a	66
1a	Ph	2	CH ₃ CN, mol. sieves (3Å)	3	6a	21
1b	SMe	2	DMF	1	6b	11
1b	SMe	2	CH ₃ CN, P ₂ O ₅	1	6b	0
1b	SMe	2	CH ₃ CN, P ₂ O ₅	3	6b	0
1c	4-MePh	2	DMF	1	6c	24
1c	4-MePh	2	CH ₃ CN, P ₂ O ₅	3	6c	42
1d	4-ClPh	2	DMF	1	6d	15
1d	4-ClPh	2	CH ₃ CN, P ₂ O ₅	3	6d	18
1a	Ph	1	DMF	1	7a	30
1b	SMe	1	DMF	1	7b	21

Starting material	R	n	Conditions	Time, h	Product	Yield, %
1c	4-MePh	1	DMF	1	7c	24
1d	4-ClPh	1	DMF	1	7d	22

If strong acids were used in the reaction, salts of **6a-d** were formed. There was no noticed influence of acid strength on the reaction yield. The reaction of 3-R-1,2,4-triazin-5(4H)-ones (**1a-d**) with ketones **4,5** in the presence of trifluoromethanesulfonic acid leads to the salts **6aa-da** (Table 3). Reducing the amount of cyclohexanone in the reaction mixture from 2 equiv. to 1 equiv. leads to significant decrease of **6aa** yield. Increasing cyclohexanone or CF₃SO₃H concentrations does not have visible effects. Compounds **6ab-db** are soluble in acidic solutions and neutralization of reaction mixture leads to increased yield of products **6**. Triazines with aliphatic substituents in position 3 (**1e**, **R** = CH₂Ph) do not react with cyclic ketones either in the presence of CF₃SO₃H, or in ethanolic HCl solution. 1,2,4-Triazines without an oxo group in position 5 of the triazine ring (3-Ph-1,2,4-triazine, 3-SMe-1,2,4-triazine) also have no reactivity under such conditions.



Table 3. Reaction of 3-R-1,2,4-triazin-5(4H)-ones (**1a-e**) with cyclohexanone (**4**) in the presence of 1 equiv. of CF_3SO_3H (X⁻ = $CF_3SO_3^-$).

Starting	D		Time,	Dreadurat	Viald 01
material	ĸ	11	days	Product	i ieid, %
1a	Ph	2	7	6aa	8
1a	Ph	2	17	6aa	18
1a	Ph	2	35	6aa	26
1a	Ph	2	35 *	6aa	21
1a	Ph	2	35 **	6aa	3
1a	Ph	2	35 ***	6aa	25
1b	SMe	2	35	6ba	24
1c	4-MePh	2	35	6ca	26
1d	4-ClPh	2	35	6da	26

Starting material	R	n	Time, days	Product	Yield, %
1e	CH ₂ Ph	2	35	-	0

^{*}3 equiv. triflic acid; ^{**}1 equiv. cyclohexanone; ^{***}4 equiv. cyclohexanone.

It is known cyclohexanone **4** gives condensation product **9** in the presence of acids, for example in the presence of trifluoromethanesulfonic acid (Scheme 2).¹⁹ It was logical to expect the formation of products **6a-d** in the reaction of 3-R-1,2,4-triazin-5(4H)-one (**1**) with 1,1'-bi(cyclohexilidene)-2-one (**9**) and we obtained them in refluxing DMF with addition of CF₃SO₃H (15% yield), but at ambient temperature in the presence of CF₃COOH the reaction leads to lactams **10a-c,e** (Table 4).



Scheme 2

Table 4. Reaction of 3-R-1,2,4-triazin-5(4H)-ones (**1a-c,e**) with **9a** in the presence of CF_3COOH .



(c) R = 4-MePh; (e) $R = CH_2Ph$

Starting	R	Product	Yield, %	
material	K	Tioduct		
1a	Ph	10a	58	
1b	SMe	10b	71	
1c	4-MePh	10c	68	
1e	CH_2Ph	10e	53	

The Journal of Organic Chemistry

Our next step was to study the reaction of 1'-hydroxybi(cyclohexan)-2-one (8) with 3-Ph-1,2,4-triazin-5(4H)-one (1a). When the reaction was carried out at ambient temperature in the presence of CF_3COOH/CF_3SO_3H the formation of **6a** was not observed. Reaction in refluxing CH_3CN in the presence of P_2O_5 resulted in the formation of **6a** (54% yield) (Scheme 3).



Scheme 3.

When we tried to use 1,1'-bi(cyclopentilidene)-2-one (11) instead of 1,1'- bi(cyclohexilidene)-2one (9) in the reaction with 3-R-1,2,4-triazin-5(4H)-ones, only product 12c (R = 4-MePh) could be isolated and identified. In all other cases we could not obtain pure products (Scheme 4). The structure of compound 12c was assigned by analogy with compounds 10a-c, and confirmed by mass-spectrometry data, ¹H and ¹³C NMR spectra. When we attempted to carry out the reaction under other conditions, we observed no reaction in refluxing acetonitrile in the presence of CF₃COOH. The reaction was carried out in refluxing DMF in the presence of CF₃SO₃H and resulted in the formation of 7aa (19% yield) after 1 hour.



Scheme 4.

To prove the assumption that compounds 10 are intermediate compounds in the reactions leading to 6 we investigated their behavior in different conditions. It appears that 10a does not undergo transformations in refluxing DMF or in DMF at ambient temperature in the presence of CF₃COOH or p-TsOH. Refluxing of 10a in DMF in the presence of CF₃SO₃H results in the formation of triflate of triazinone 1a and the condensation product 6aa in low yields (Scheme 5). The salt of 1a is a decomposition product of 10a and could be used as starting material for reaction with cyclohexanone residue from the reaction mixture resulting in the formation of 6aa.



Scheme 5.

Liquid chromatography-mass spectrometry analysis of the reaction mixtures of 3-Ph-1,2,4-triazin-5(4H)-one (**1a**) with cyclohexanone in DMSO solutions in the presence of CF₃SO₃H after 24 hours shows intensive peaks of some molecular ions. There were observed peaks of **6ab** (ESI-MS, m/z = 332.1798 (calcld. 332.1757 for $[C_{21}H_{22}N_3O]^+$, $[M+H]^+$)), **10a** (ESI-MS, m/z = 352.2032 (calcld. 352.2020 for $[C_{21}H_{26}N_3O_2]^+$, $[M+H]^+$)), and peak of unidentified compound (ESI-MS, m/z = 258.1610 (calcld. 258.1601 for $[C_{15}H_{20}N_3O]^+$, $[M+H]^+$)). In the reaction of 3-Ph-1,2,4-triazin-5(4H)-one (**1a**) with cyclopentanone intensive peaks of **7ab** were observed (ESI-MS, m/z = 304.1488 (calcld. 304.1444 for $[C_{19}H_{18}N_3O]^+$, $[M+H]^+$)), **12a** (ESI-MS, m/z = 324.1729 (calcld. 324.1706 for $[C_{19}H_{22}N_3O_2]^+$, $[M+H]^+$)), benzonitrile (ESI-MS, m/z = 104.0520 (calcld. 104.0494 for $[C_7H_6N]^+$, $[M+H]^+$)), peak of unidentified compound (ESI-MS, m/z = 244.1470 (calcld. 244.1444 for $[C_{14}H_{18}N_3O]^+$, $[M+H]^+$)), its dimer (ESI-MS, m/z = 477.2034 (calcld. 477.2033 for $[C_{28}H_{25}N_6O_2]^+$, $[M+H]^+$)), and peak of unidentified compound (ESI-MS, m/z = 545.2625 (calcld. 545.2619 for $[C_{28}H_{33}N_8O_4]^+$, $[M+H]^+$)).

Absence of the formation of lactam products **10,12** in quantities sufficient for identification in the reactions of triazines **1** with cyclic ketones probably means that there are two different ways of formation of tetracyclic compounds **6,7** and lactam compounds **10,12**. We suggested two different routes for the reactions of triazines **1** with free cyclic ketones and their dimerization products.

The high activity of 3-R-1,2,4-triazin-5(4H)-ones in the reactions with C-nucleophiles suppose initial nucleophilic addition of cyclic ketones to 1,2,4-triazine ring with formation of high reactive compounds **1A** or **1B**. The attack of the cyclic ketone on **1A** leads to the series of transformations with the final product **6**. We suppose that the product **1B** after intermolecular addition gives **1D**. Compound **1D** at ambient temperature gives 10-membered lactame **10**, the high temperature of refluxing DMF and the presence of triflic acid lead to the loss of a water molecule and formation of **6** (Scheme 6).





Conclusions

In the course of the present study new heterocyclic systems of benzo[c][1,2,4]triazino[1,6-a][2]azecine and cyclopenta[c][1,2,4] triazino[1,6-a]azonine were obtained, optimal reaction conditions were found for the condensation between 1,2,4-triazine-5(4H)-ones and cyclic ketones. It was shown that benzo[c][1,2,4]triazino[1,6-a][2]azecines **10** are not intermediates in the reaction of formation of [1,2,4]triazino[1,6-f]phenantridine derivatives **6**.

It should be noted that C-C coupling reactions between azines and cyclic ketones had been described earlier,²⁰ but intramolecular condensation was observed and investigated for the first time.

Experimental Section

3-Ph-1,2,4-triazin-5(4H)-one (1a),²¹ 3-SMe-1,2,4-triazin-5(4H)-one (1b),²² 1'hydroxybi(cyclohexan)-2-one (8),²³ 1,1'-bi(cyclohexilidene)-2-one $(9)^{24}$ and 1,1'bi(cyclopentilidene)-2-one $(11)^{25}$ were synthesized by known methods, others starting materials are commercially available. ¹H, ¹³C NMR spectra were recorded using 400MHz spectrometer; tetramethylsilane (TMS) was used as an internal standard. TOF mass analyzer was used for the HRMS.

3-(4-Tolyl)-1,2,4-triazin-5(4H)-one (1c). Methanol (10.1 ml, 0.170 mole) was added to the solution of 4-methylbenzonitrile (10.0 g, 0.085 mole) in ether (50 ml), resulting solution was chilled to $T = 0 \div 5$ °C and then dry HCl was bubbled through the solution for 2-3h. The reaction mixture was stirred for 16h at room temperature. The sediment of methyl ether of 4-methylbenzimidate hydrochloride was filtered out, washed with ether and dried. The hydrochloride was dissolved in water and 50 ml of aqueous NaOH (5N) was added to the solution. Then the product was extracted with ether, the ether solution was dried over Na₂SO₄ and ether was evaporated.

On the next stage hydrazine hydrate (2.35 g, 0.047 mole) was added to solution of 4methylbenzimidate (7.0 g, 0.047 mole) in 10 ml of methanol. The reaction mixture was stirred for 16 h, after that it was cooled to T = 5°C and glyoxylic acid monohydrate (4.32 g, 0.047 mmol) was added by portions to the solution. The temperature has to be lower 10°C. The reaction mixture was stirred 1 h at 5°C, then it was stayed in refrigerator for 16 h at 5°C. The formed yellow sediment was filtered, washed with cold methanol and dried. The yellow solid was dissolved in DMF (50ml) and refluxed for 30 min. After cooling of the solution to room temperature crystals formed. They were filtered and crystallized from ethanol. Yield 41 % (6.5 g), light brown crystals, mp 255 °C; ¹H NMR (400 MHz, DMSO-d₆) δ : 2.43 (s, 3H), 7.35 (d, *J* = 8.1 Hz, 2H), 7.63 (s, 1H), 7.98 (d, *J* = 8.2 Hz, 2H), 14.0 (br.s, 1H, NH); ¹³C NMR (100MHz, DMSO-d₆) δ : 21.0, 127.5, 127.6, 129.5, 143.0, 143.4, 158.1, 162.2; ESI-MS, m/z = 188.0819 (calcld. 188.0818 for [C₁₀H₁₀N₃O]⁺, [M+H]⁺).

3-(4-Chlorophenyl)-1,2,4-triazin-5(4H)-one (1d). The procedure is the same as for **1c**. Yield 31 % (5.4 g), yellow crystals, mp 282-283 °C; ¹H NMR (400 MHz, DMSO-d₆) δ : 7.68 (d, 2H), 7.86 (s, 1H), 8.07 (d, 2H), 14.12 (br.s, 1H); ¹³C NMR (100MHz, DMSO-d₆) δ : 129.6, 129.8, 130.1, 138.1, 144.3, 157.8, 162.5; ESI-MS, m/z = 208.0283 (calcld. 208.0272 for [C₉H₇ClN₃O]⁺, [M+H]⁺).

3-Benzyl-1,2,4-triazin-5(4H)-one (1e). The procedure is the same as for **1c**. Yield 39 % (6.2 g), yellow crystals, mp 165-166 °C. ¹H NMR (400 MHz, DMSO-d₆) δ : 3.11 (s, 2H), 6.43-6.54 (m,

5H), 6.92 (s, 1H), 13.68 (br.s, 1H); ¹³C NMR (100MHz, CD₃OD) δ : 41.0, 128.7, 130.0, 130.1, 135.9, 144.7, 165.3, 165.4; ESI-MS, m/z = 188.0812 (calcld. 188.0818 for [C₁₀H₁₀N₃O]⁺, [M+H]⁺).

6-(2-Oxo-2-phenylethyl)-3-phenyl-1,6-dihydro-1,2,4-triazin-5(4H)-one (3a). Acetophenone (270 μl, 2.310 mmol) and CF₃SO₃H (100 μl) were added to the suspension of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in 10 ml CH₃CN. The solution was refluxed for 3 h, then it was cooled, neutralized with NEt₃ and after 16 h formed sediment was filtered and crystallized from CH₃CN. Yield 35% (119 mg); yellow crystals; mp 194-195 °C; ¹H NMR (400 MHz, CDCl₃) δ: 3.30 (dd, *J* = 18.5, 10.1Hz, 1H), 3.92 (dd, *J* = 18.5, 2.4 Hz, 1H), 4.29 (dd, *J* = 10.1, 2.3 Hz, 1H), 6.39 (br.s, 1H), 7.41-7.45 (m, 3H), 7.48-7.52 (m, 2H), 7.60-7.64 (m, 1H), 7.67-7.70 (m, 2H), 8.00-8.02 (m, 2H), 8.37 (br.s, 1H); ¹³C NMR (100MHz, CDCl₃) δ: 37.1, 52.3, 124.9, 128.2, 128.8, 129.0, 130.1, 131.2, 133.9, 136.0, 139.4, 167.7, 197.6; ESI-MS, m/z = 294.1291 (calcld. 294.1237 for [C₁₇H₁₆N₃O₂]⁺, [M+H]⁺).

3-(Methylthio)-6-(2-oxo-2-phenylethyl)-1,6-dihydro-1,2,4-triazin-5(4H)-one (3b). Acetophenone (326 μ l, 2.794 mmol) and CF₃SO₃H (100 μ l) were added to the suspension of 3-SMe-1,2,4-triazin-5(4H)-one (200 mg, 1.397 mmol) in 10 ml CH₃CN. The solution was refluxed for 3 h (emission of CH₃SH !), then it was cooled, neutralized with NEt₃ and evaporated. The residue was chromatographed by column using EtOAc as an eluent. Yield 13% (40 mg); yellow crystals; mp 132-133 °C; R_f = 0.3 (EtOAc); ¹H NMR (400 MHz, DMSO-d₆) δ : 2.33 (s, 3H), 3.10 (dd, *J* = 17.8, 6.0 Hz, 1H), 3.57 (dd, J = 17.8, 5.8 Hz, 1H), 3.86-3.89 (m, 1H), 6.56 (d, *J* = 3.0 Hz, 1H), 7.48-7.52 (m, 2H), 7.58-7.62 (m, 1H), 7.98-7.99 (m, 2H), 10.64 (s, 1H); ¹³C NMR (100MHz, DMSO-d₆) δ : 12.8, 36.4, 52.2, 128.0, 128.7, 133.3, 136.3, 139.3, 167.3, 196.6; ESI-MS, m/z = 264.0791 (calcld. 264.0801 for [C₁₂H₁₄N₃O₂S]⁺, [M+H]⁺).

Procedures for synthesis of 6a.

Method A. Cyclohexanone **4** (239 μ L, 2.310 mmol) and 1 equiv. of acid (CF₃COOH or AlCl₃, see Table 1) were added to the solution of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in the corresponding solvent (see Table 1). The solution was stirred or refluxed for the time mentioned in the Table 1. Formed sediment was filtered and crystallized from DMF. Otherwise the reaction mixture was poured to the water (50 ml) and extracted with CH₂Cl₂. The organic layer was separated, washed with water, brine and dried over Na₂SO₄. The solution in CH₂Cl₂ was evaporated and the residue was crystallized from DMF.

Method B. Cyclohexanone **4** (239 μ L, 2.310 mmol) was added to the solution of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in the phosphoric acid (5 ml). The solution was stirred for 3 h at 50 °C then it was cooled and poured in water (50 ml). Formed sediment was filtered and chromatographed by column using EtOAc as an eluent.

Method C. Cyclohexanone **4** (239 μ L, 2.310 mmol) was added to the solution of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in the acetic acid (5 ml). The reaction mixture was refluxed for 1 h and then it was evaporated. The residue was refluxed in CHCl₃ for 5 min and filtered. The chloroform solution was evaporated and the residue was crystallized from DMF.

Method D. 1,1'-Bi(cyclohexilidene)-2-one (**9**) (206 μ L, 1.155 mmol) and CF₃SO₃H (1.155 mmol, 102 μ L) were added to the solution of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in DMF (3 ml). The reaction mixture was refluxed for 1 h, neutralized with NEt₃ and then it was evaporated. The residue was crystallized from DMF.

Method E. 1'-Hydroxybi(cyclohexan)-2-one (8) (227 μ L, 1.155 mmol), p-TsOH (203 mg, 1.155 mmol) and P₂O₅ (492 mg, 1.155 mmol) were added to the mixture of 3-R-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) and CH₃CN (10 ml). Then it was divided on two parts. The solution in CH₃CN was decanted and evaporated to the half volume. NEt₃ (100 μ l) was added to the solution and after 2 h the sediment was filtered. Solid part of the reaction mixture was dissolved in water and neutralized with concentrated solution of NaHCO₃, the resulting sediment was filtered. Combined sediments were dried and crystallized from DMF.

Method F. Cyclohexanone **4** (239 μ L, 2.310 mmol) was added to the solution of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in methanolic MeONa (1M) (4 ml). The reaction mixture was stirred for 24 h, then it was poured in water (50 ml) neutralized with HCl (2M), the product extracted with CH₂Cl₂. Solution in CH₂Cl₂ was washed with water, brine and dried over Na₂SO₄. Then it was evaporated and the residue was crystallized from DMF.

Procedures for synthesis of 6a-d, 7a-d.

Method A. Ketone **4** or **5** (2 equiv.), p-TsOH (1 equiv) and P_2O_5 (3 equiv.) (or 1 g of molecular sieves 3Å) were added to the mixture of 3-R-1,2,4-triazin-5(4H)-one (150 mg) and CH₃CN (10 ml). The reaction mixture was refluxed for 3 h. Then it was divided on two parts. The solution in CH₃CN was decanted and evaporated to the half volume. NEt₃ (100 µl) was added to the solution and after 2 h the sediment was filtered. Solid part of the reaction mixture was dissolved in water and neutralized with concentrated solution of NaHCO₃, the resulting sediment was filtered. Combined sediments were dried and crystallized from DMF.

Method B. Ketone **4** or **5** (2 equiv.) and p-TsOH (1 equiv) were added to the solution of 3-R-1,2,4-triazin-5(4H)-one (200 mg) in DMF (3 ml). The solution was refluxed for 1 h. After 16 h sediment was filtered. If there was no sediment the reaction mixture was poured in water, the product extracted with CH_2Cl_2 . The organic layer was separated, washed with water, brine and dried over Na_2SO_4 . The solution in CH_2Cl_2 was evaporated and the residue was crystallized from DMF.

2-Phenyl-5,6,7,8,9,10,11,12-octahydro[1,2,4]triazino[1,6-f]phenanthridin-13-ium-4-olate

(6a). Yield 44% (169 mg), cream crystals, mp > 300 °C; ¹H NMR (400 MHz, CDCl₃) δ : 1.77-1.81 (m, 4H), 1.90-1.99 (m, 4H), 2.67 (t, *J* = 6.1 Hz, 2H), 2.71 (t, *J* = 6.2 Hz, 2H), 3.35 (t, *J* = 6.2 Hz, 2H), 3.61 (t, *J* = 6.2 Hz, 2H), 7.38-7.46 (m, 3H), 8.36-8.38 (m, 2H). The spectral data of the compound are identical to literature values.¹⁷

2-(Methylthio)-5,6,7,8,9,10,11,12-octahydro-[1,2,4]triazino[1,6-f]phenanthridin-13-ium-4-olate (6b). Yield 21% (73 mg), cream crystals, mp 249 °C; ¹H NMR (400 MHz, CDCl₃) δ : 1.76-1.97 (m, 8H), 2.54 (s, 3H), 2.74 (dd, J = 13.3, 6.8Hz, 4H), 3.19 (t, J = 6.2Hz, 2H), 3.59 (t, J = 6.2Hz, 2H); ¹³C NMR (100MHz, DMSO-d₆) δ : 13.5, 21.0, 21.2, 21.4, 21.9, 26.3, 27.1, 27.7, 28.8, 131.4, 136.1, 136.4, 142.4, 146.6, 163.8, 170.9; ESI-MS, m/z = 302.1355 (calcld. 302.1322 for [C₁₆H₂₀N₃OS]⁺, [M+H]⁺).

2-(p-Tolyl)-5,6,7,8,9,10,11,12-octahydro-[1,2,4]triazino[1,6-f]phenanthridin-13-ium-4-olate (6c). Yield 24% (96 mg), grey crystals, mp 304-305 °C; ¹H NMR (400 MHz, CDCl₃) δ : 1.77-1.79 (m, 4H), 1.90-1.99 (m, 4H), 2.38 (s, 3H), 2.72-2.78 (m, 4H), 3.39 (t, J = 6.3Hz, 2H), 3.66 (t, J = 6.1Hz, 2H), 7.19 (d, J = 8.0 Hz, 2H), 8.22 (d, J = 8.2 Hz, 2H); ¹³C NMR (100MHz, DMSO-d₆) δ : 21.0, 21.3, 21.5, 21.9, 26.3, 27.1, 27.6, 28.8, 128.0, 128.7, 132.1, 132.9, 135.7, 136.3, 141.1, 143.3, 146.6, 161.5, 166.5; ESI-MS, m/z = 346.1903 (calcld. 346.1914 for [C₂₂H₂₄N₃O]⁺, [M+H]⁺). Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre (deposition number CCDC 859049). These data can be obtained via: www.ccdc.cam.ac.uk/data_request/cif

2-(4-Chlorophenyl)-5,6,7,8,9,10,11,12-octahydro-[1,2,4]triazino[1,6-f]phenanthridin-13-

ium-4-olate (6d). Yield 15% (63 mg), yellow crystals, mp 257-258 °C; ¹H NMR (400 MHz, CDCl₃) δ : 1.76-1.96 (m, 8H), 2.76-2.77 (m, 4H), 3.30 (t, J = 5.7Hz, 2H), 3.45 (t, J = 5.9Hz, 2H), 7.44 (d, J = 8.5Hz, 2H), 8.23 (d, J = 8.5Hz, 2H); ¹³C NMR (100MHz, CDCl₃) δ : 21.0, 21.3, 21.5, 21.9, 26.4, 27.1, 27.7, 28.9, 128.0, 129.2, 132.0, 134.1, 135.9, 136.7, 136.9, 143.6, 147.5, 160.2, 166.5; ESI-MS, m/z = 366.1354 (calcld. 366.1368 for [C₂₁H₂₁ClN₃O]⁺, [M+H]⁺).

2-Phenyl-5,6,7,8,9,10-hexahydrodicyclopenta[**3,4:5,6**]**pyrido**[**2,1-f**][**1,2,4**]**triazin-11-ium-4-olate (7a).** Yield 30% (105 mg), light green crystals, mp > 300 °C; ¹H NMR (400 MHz, CDCl₃) δ : 2.25-2.33 (m, 2H), 2.37-2.45 (m, 2H), 3.01 (t, J = 7.7Hz, 2H), 3.09-3.18 (m, 2H), 3.55 (t, J = 7.9Hz, 2H), 3.79 (t, J = 7.9Hz, 2H), 7.40-7.47 (m, 3H), 8.42-8.44 (m, 2H). The spectral data of the compound are identical to literature values.¹⁷

2-(Methylthio)-5,6,7,8,9,10-hexahydrodicyclopenta[3,4:5,6]pyrido[2,1-f][1,2,4]triazin-11-ium-4-olate (7b). Yield 21% (66 mg), grey crystals, mp 258-259 °C; ¹H NMR (400 MHz, CDCl₃) δ : 2.21-2.30 (m, 2H), 2.31-2.42 (m, 2H), 2.53 (s, 3H), 2.98 (t, *J* = 7.8 Hz, 2H), 3.10 (t, *J* = 7.6 Hz, 2H), 3.37 (t, *J* = 7.8 Hz, 2H), 3.70 (t, *J* = 7.6 Hz, 2H); ¹³C NMR (100MHz, CDCl₃) δ :

13.5, 22.1, 24.7, 30.5, 31.2, 31.5, 34.0, 129.4, 139.2, 143.2, 149.4, 152.5, 162.8, 172.3. The spectral data of the compound are identical to literature values.¹⁷

2-(p-Tolyl)-5,6,7,8,9,10-hexahydrodicyclopenta[**3,4:5,6**]**pyrido**[**2,1-f**][**1,2,4**]**triazin-11-ium-4-olate (7c).** Yield 24% (88 mg), brown crystals, mp 236-237 °C; ¹H NMR (400 MHz, CDCl₃) δ : 2.20-2.27 (m, 2H), 2.34-2.42 (m, 5H), 2.88-2.96 (m, 2H), 3.08 (t, *J* = 7.6 Hz, 2H), 3.51 (t, *J* = 7.7 Hz, 2H), 3.74 (t, *J* = 7.6 Hz, 2H), 7.23 (d, *J* = 8.0 Hz, 2H), 8.28 (d, *J* = 8.1 Hz, 2H); ¹³C NMR (100MHz, CDCl₃) δ : 21.4, 22.1, 24.5, 30.4, 31.3, 31.3, 34.1, 128.0, 128.7, 130.2, 133.0, 139.0, 141.1, 142.8, 150.4, 152.5, 162.8, 165.6; ESI-MS, m/z = 318.1620 (calcld. 318.1601 for $[C_{20}H_{20}N_3O]^+$, $[M+H]^+$).

2-(4-Chlorophenyl)-5,6,7,8,9,10-hexahydrodicyclopenta[**3,4:5,6**]**pyrido**[**2,1-f**][**1,2,4**]**triazin-11-ium-4-olate** (**7d**). Yield 22% (86 mg), yellow green crystals, mp > 300 °C; ¹H NMR (400 MHz, CDCl₃) δ : 2.21-2.28 (m, 2H), 2.34-2.41 (m, 2H), 2.94 (t, *J* = 7.7 Hz, 2H), 3.07 (t, *J* = 7.6 Hz, 2H), 3.47 (t, *J* = 7.7 Hz, 2H), 3.70 (t, *J* = 7.6 Hz, 2H), 7.34 (d, *J* = 8.6Hz, 2H), 8.26 (d, *J* = 8.6Hz, 2H). The spectral data of the compound are identical to literature values.¹⁷

Procedures for synthesis of 6aa-da.

Method A. Cyclohexanone (2 equiv.) and CF_3SO_3H (1 equiv.) were added to the solution of 3-R-1,2,4-triazin-5(4H)-one (200 mg) in DMF (3 ml). The reaction mixture was stirred at room temperature for the period of time specified in Table 3. The sediment was filtered and crystallized from DMF.

Method B. Cyclohexanone (2 equiv.) and CF_3SO_3H (1 equiv.) were added to the solution of 3-R-1,2,4-triazin-5(4H)-one (200 mg) in DMF (3 ml). The reaction mixture was refluxed for the time in Table 1 and cooled, after 12 h formed sediment was filtered and crystallized from DMF.

4-Oxo-2-phenyl-3,4,5,6,7,8,9,10,11,12-decahydro-[1,2,4]triazino[1,6-f]phenanthridin-13-

ium trifluoromethanesulfonate (6aa). Yield 25% (138 mg), green crystals, mp 252-253 °C; ¹H NMR (400 MHz, CDCl₃) δ : 1.78-2.04 (m, 8H), 2.85-2.66 (m, 4H), 3.40 (t, *J* = 6.4Hz, 2H), 3.67 (t, *J* = 6.1Hz, 2H), 7.42-7.50 (m, 3H), 8.41-8.44 (m, 2H); ¹³C NMR (100MHz, CDCl₃) δ : 21.1, 21.3, 21.6, 22.0, 26.3, 27.2, 27.6, 28.8, 128.0, 128.1, 130.8, 135.9, 136.0, 136.3, 143.4, 146.9, 161.7, 166.4; ¹⁹F NMR (100MHz, CDCl₃) δ : -78.2; ESI-MS, m/z = 332.1762 (calcld. 332.1757 for [C₂₁H₂₂N₃O]⁺, [M+H]⁺). Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre (deposition number CCDC 820437). These data can be obtained via: www.ccdc.cam.ac.uk/data_request/cif

2-(Methylthio)-4-oxo-3,4,5,6,7,8,9,10,11,12-decahydro-[1,2,4]triazino[1,6-f]phenanthridin-13-ium trifluoromethanesulfonate (6ba). Yield 24% (150 mg), light yellow crystals, mp 215-216 °C; ¹H NMR (400MHz, CDCl₃) δ: 1.74-1.80 (m, 2H), 1.84-1.95 (m, 6H), 2.52 (s, 3H), 2.72-2.77 (m, 4H), 3.16-3.19 (m, 2H), 3.54-3.57 (m, 2H); ¹³C NMR (100MHz, CDCl₃) δ: 13.6, 21.1, 21.3, 21.4, 22.0, 26.3, 27.2, 27.7, 28.9, 131.4, 136.2, 136.6, 146.2, 146.8, 163.8, 170.9; ¹⁹F NMR (100MHz, CDCl₃) δ : -78.3; ESI-MS, m/z = 302.1318 (calcld. 302.1322 for [C₁₆H₂₀N₃OS]⁺, [M+H]⁺).

2-(4-Methylphenyl)-4-oxo-3,4,5,6,7,8,9,10,11,12-decahydro-[1,2,4]triazino[1,6-

f]phenanthridin-13-ium trifluoromethanesulfonate (6ca). Yield 26% (140 mg), light yellow crystals, mp 277-278 °C; ¹H NMR (400MHz, CDCl₃) δ : 1.72-1.83 (m, 4H), 1.87-2.01 (m, 4H), 2.38 (s, 3H), 2.62-2.65 (m, 2H), 2.68-2.71 (m, 2H), 3.31-3.34 (m, 2H), 3.57-3.60 (m, 2H), 7.19 (d, *J* = 7.6 Hz, 2H), 8.22 (d, *J* = 8.0 Hz, 2H); NMR ¹³C (100MHz, CDCl₃) δ : 21.0, 21.3, 21.5, 22.0, 26.3, 27.2, 27.7, 28.9, 128.0, 128.8, 132.1, 132.9, 135.8, 136.4, 141.2, 143.5, 146.8, 161.4, 166.5; ¹⁹F NMR (100MHz, CDCl₃) δ : -78.3; ESI-MS, m/z = 346.1914 (calcld. 346.1914 for [C₂₂H₂₄N₃O]⁺, [M+H]⁺).

2-(4-Chlorophenyl)-4-oxo-3,4,5,6,7,8,9,10,11,12-decahydro-[1,2,4]triazino[1,6-

f]phenanthridin-13-ium trifluoromethanesulfonate (6da). Yield 26% (131 mg), wine red crystals, mp 252-253 °C; ¹H NMR (400MHz, CDCl₃) δ : 1.75-1.98 (m, 8H), 2.65-2.71 (m, 4H), 3.25 (t, *J* = 6.0 Hz, 2H), 3.54-3.57 (m, 2H), 7.30 (d, *J* = 8.4 Hz, 2H), 8.22 (d, *J* = 8.8 Hz, 2H); ¹³C NMR (100MHz, CDCl₃) δ : 21.0, 21.3, 21.5, 21.9, 26.4, 27.1, 27.7, 28.9, 128.0, 129.2, 132.0, 134.1, 135.9, 136.7, 136.9, 143.6, 147.5, 160.2, 166.5; ¹⁹F NMR (100MHz, CDCl₃) δ : -77.3; ESI-MS, m/z = 366.1372 (calcld. 366.1368 for [C₂₁H₂₁ClN₃O]⁺, [M+H]⁺).

4-Oxo-2-phenyl-3,4,5,6,7,8,9,10,11,12-decahydro-**Procedure** for synthesis of [1,2,4]triazino[1,6-f]phenanthridin-13-ium chloride 6ab. Cyclohexanone (239µL, 2.310mmol) was added to the solution of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155mmol) in conc. ethanol solution of HCl (5 ml). The reaction mixture was stirred at room temperature for 1 day. The sediment was filtered washed with ethanol, crystallized from DMF and dried. Yield 40% (172 mg), dark yellow crystals, mp 270-271 °C; ¹H NMR (400MHz, DMSO-d₆) δ: 1.86-2.01 (m, 8H), 2.92-2.98 (m, 4H), 3.42-3.45 (m, 2H), 3.49-3.51 (m, 2H), 7.59-7.63 (m, 2H), 7.69-7.73 (m, 1H), 8.17 (d, J = 7.2 Hz, 2H); ¹³C NMR (100MHz, DMSO-d₆) δ : 20.6, 20.8, 21.1, 21.5, 26.4, 27.8, 28.0, 28.5, 128.7, 129.1, 129.1, 129.6, 132.7, 133.8, 137.2, 140.7, 147.7, 152.7, 154.1, 157.4, 157.5; ESI-MS, m/z = 332.1769 (calcld. 332.1757 for $[C_{21}H_{22}N_3O]^+, [M+H]^+$).

Procedure for synthesis of 10a-c,e. 1,1'-Bi(cyclohexilidene)-2-one (**9**) (1 equiv) and CF₃COOH (1 equiv) were added to the solution of 3-R-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in DMF (3 ml). The reaction mixture was stirred at room temperature for 5 days, poured in Petri dish; residue from flask was washed off with ethanol (5 ml) to the same Petri dish. After 2 h formed sediment was filtered and crystallized from CH₃CN.

3-Phenyl-7,8,9,10,12,13,14,15,15a,15b-decahydro-1H-benzo[c][1,2,4]triazino[1,6-a]azecine-1,6(2H)-dione (10a). Yield 57% (232 mg), colorless crystals, mp 210-211 °C; ¹H NMR (400

MHz, CDCl₃) δ : 1.15-1.27 (m, 1H), 1.41-2.14 (m, 14H), 2.28-2.32 (m, 1H), 3.56 (dd, J = 14.3, 10.1, 1.5 Hz, 1H), 4.90 (d, J = 10.1 Hz, 1H), 5.12 (d, J = 11.4 Hz, 1H), 7.41-7.47 (m, 3H), 7.88-7.90 (m, 2H), 11.30 (s, 1H); ¹³C NMR (100MHz, CDCl₃) δ : 21.6, 24.7, 25.5, 25.8, 26.3, 27.6, 29.0, 32.4, 43.1, 57.0, 125.9, 129.0, 130.3, 130.4, 131.0, 135.1, 139.2, 167.8, 176.0; ESI-MS, m/z = 352.1996 (calcld. 352.2020 for $[C_{21}H_{26}N_3O_2]^+$, $[M+H]^+$). Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre (deposition number CCDC 859113). These data can be obtained via: www.ccdc.cam.ac.uk/data_request/cif

3-(Methylthio)-7,8,9,10,12,13,14,15,15a,15b-decahydro-1H-benzo[c][1,2,4]triazino[1,6-

a]azecine-1,6(2H)-dione (10b). Yield 70% (315 mg), colorless crystals, mp 244-245 °C; ¹H NMR (400 MHz, DMSO-d₆) δ : 1.11-1.21 (m, 1H), 1.33-1.41 (m, 1H), 1.46-2.12 (m, 13H), 2.28 (d, *J* = 15.0 Hz, 1H), 2.46 (s, 3H), 3.33 (dd, *J* = 14.0, 9.6 Hz, 1H), 4.90 (d, *J* = 10.4 Hz, 1H), 5.01 (d, *J* = 11.4 Hz, 1H), 11.30 (br.s, 1H); ¹³C NMR (100MHz, DMSO-d₆) δ : 13.6, 21.6, 24.5, 25.5, 25.7, 26.0, 27.4, 28.9, 32.2, 42.8, 57.5, 129.9, 135.2, 141.3, 166.1, 175.0; ESI-MS, m/z = 322.1580 (calcld. 322.1584 for [C₁₆H₂₄N₃O₂S]⁺, [M+H]⁺). Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre (deposition number CCDC 820598). These data can be obtained via: <u>www.ccdc.cam.ac.uk/data_request/cif</u>

3-p-Tolyl-7,8,9,10,12,13,14,15,15a,15b-decahydro-1H-benzo[c][1,2,4]triazino[1,6-a]azecine-1,6(2H)-dione (10c). Yield 67% (262 mg), colorless crystals, mp 205-206 °C; $[\alpha]_D = 0$ (c = 0.6, MeOH); ¹H NMR (400 MHz, DMSO-d₆) δ : 1.11-1.28 (m, 1H), 1.39-2.15 (m, 14H), 2.30 (d, J = 14.5 Hz, 1H), 2.40 (s, 3H), 3.55 (dd, J = 14.1, 10.0 Hz, 1H), 4.89 (d, J = 10.3 Hz, 1H), 5.11 (d, J = 11.4 Hz, 1H), 7.22 (d, J = 8.1 Hz, 2H), 7.77 (d, J = 8.1 Hz, 2H), 11.24 (s, 1H); ¹³C NMR (100MHz, DMSO-d₆) δ : 20.6, 21.1, 23.9, 24.9, 25.1, 25.4, 26.9, 28.3, 31.5, 41.8, 56.3, 126.2, 127.5, 129.1, 129.2, 134.4, 140.4, 140.5, 165.9, 174.7; ESI-MS, m/z = 366.2143 (calcld. 366.2176 for $[C_{22}H_{28}N_3O_2]^+$, $[M+H]^+$).

3-Benzyl-7,8,9,10,12,13,14,15,15a,15b-decahydro-1H-benzo[c][1,2,4]triazino[1,6-a]azecine-1,6(2H)-dione (10e). Yield 52% (203 mg), light brown crystals, mp 211-212 °C; ¹H NMR (400 MHz, DMSO-d₆) δ: 1.03-1.15 (m, 1H), 1.27-2.09 (m, 14H), 2.23-2.27 (m, 1H), 3.61 (s, 2H), 4.85 (d, *J* = 10.1 Hz, 1H), 4.96 (d, *J* = 11.4 Hz, 1H), 7.24-7.29 (m, 1H), 7.34-7.35 (m, 4H), 11.1 (s, 1H); ¹³C NMR (100MHz, DMSO-d₆) δ: 21.0, 23.9, 24.8, 25.0, 25.2, 26.9, 28.2, 31.3, 37.5, 41.7, 55.8, 126.8, 128.3, 128.7, 129.1, 134.3, 135.8, 142.9, 165.6, 174.4; ESI-MS, m/z = 366.2142 (calcld. 366.2176 for [C₂₂H₂₈N₃O₂]⁺, [M+H]⁺).

3-(p-Tolyl)-7,8,9,11,12,13,13a,13b-octahydro-1H-cyclopenta[c][1,2,4]triazino[1,6-a]azonine-1,6(2H)-dione (12c). 1,1'-bi(cyclopentilidene)-2-one (11) (162 μ L, 1.068 mmol) and CF₃COOH (100 μ L) were added to the solution of 3-(4-Tol)-1,2,4-triazin-5(4H)-one (1c) (200 mg, 1.068 mmol) in DMF (3 ml). The reaction mixture was stirred at room temperature for 5 days and evaporated. The residue was chromatographed by column using EtOAc as an eluent. Yield 9% (32 mg), yellow crystals, mp 229-230 °C, $R_f = 0.6$ (EtOAc); ¹H NMR (400 MHz, DMSO-d₆) δ : 1.64-1.75 (m, 4H), 1.77-1.85 (m, 1H), 2.08-2.15 (m, 1H), 2.23-2.44 (m, 7H), 2.58-2.64 (m, 1H), 2.71 (br.s, 2H), 3.71 (dd, J = 5.5, 2.1 Hz, 1H), 6.99 (d, J = 2.1 Hz, 1H), 7.14 (d, J = 8.1 Hz, 2H), 7.61 (d, J = 8.2 Hz, 2H), 10.56 (s, 1H); ¹³C NMR (100MHz, DMSO-d₆) δ : 20.5, 22.1, 24.4, 26.1, 26.7, 31.7, 33.3, 47.4, 51.7, 56.5, 99.2, 124.9, 127.3, 128.5, 129.2, 138.3, 138.9, 157.3, 165.7, 203.7; ESI-MS, m/z = 338.1849 (calcld. 338.1863 for [C₂₀H₂₄N₃O₂]⁺, [M+H]⁺).

Acknowledgements

This work was carried out with the assistance of laboratory of complex investigation and expert valuation of organic materials (Ural Federal University). This work was supported by Ministry of Science and Education of Russian Federation (State Contract # 14.740.11.1020)

Supporting Information Available: ¹H, ¹³C NMR spectra of products and other data. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- (a) Plunkett, O.; Sainsbury, M. in *Rodd's Chem. Carbon Compd.*, 2nd ed.; 1998; Vol. 4 (Part F/Part G(partial)), p 365. (b) Bailey, T.D.; Goe, G.L.; Scriven, E.F.V. *Chem. Heterocycl. Compd.* 1984, 14 (Pyridine Its Deriv., Pt. 5), 1. (c) Thummel, R.P. *Chem. Heterocycl. Compd.* 1984, 14 (Pyridine Its Deriv., Pt. 5), 253. (d) Kumar, R.; Chandra, R. *Adv. Heterocycl. Chem.* 2001, 78, 269. (e) Umeda, H.; Takeuchi, M.; Suyama, K. J. Biol. *Chem.*, 2001, 12579-12587.
- (a) Snider, B.B.; Neubert, B.J. Org. Lett., 2005, 7, 2715-2718. (b) Cappelli, A.; Anzini, M.; Vomero, S.; Canullo, L.; Mennuni, L.; Makovec, F.; Doucet, E.; Hamon, M.; Menziani, M.C.; De Benedetti, P.G.; Bruni, G.; Romeo, M.R.; Giorgi, G.; Donati, A. J. Med. Chem., 1999, 42, 1556-1575. (c) Leon, R., Marco-Contelles, J.; Garcia, A.G.; Villaroya, M. Bioorg. Med. Chem., 2005, 13, 1167-1175. (d) da Costa, J.S.; Pisoni, D.S.; da Silva, C.B.; Petzhold, C.L.; Russowsky, D.; Ceschi, M.A. J. Braz. Chem. Soc., 2009, 20, 1448-1454. (e) Pisoni, D.S.; da Costa, J.S., Gamba, D.; Petzhold, C.L.; de Amorim Borges, A.C.; Ceschi, M.A.; Lunardi, P.; Saraiva Gonçalves, C.A. Eur. J. Med. Chem., 2010, 45, 526-535. (f) Capelli, A.; Anzini, M.; Vomero, S.; De Benedetti, P.G.; Menziani, M.C.; Giorgi, G.; Manzoni, C. J. Med. Chem., 1997, 40, 2910-2921.
- 3. Schrider, M.S. U.S. Pat. 4006236A, 1977.

- Messmer, A.; Bátori, S.; Hajós, G.; Benkó, P.; Pallos, L.; Petöcz, L.; Katalin, G.; Kosóczky, I. U.S. Pat. 4697013A, 1987.
- (a) Chichibabin, A.E. Bull. Soc. Chim. Fr., 1939, 6(3), 522-533. (b) Edgar, O.B.; Johnson, D.H. J. Chem. Soc., 1958, 3925-3944. (c) Chafetz, H.; Anderson, R.C. U.S. Pat. 3349092A, 1967. (d) Upadysheva, A.V.; Usova, E.P.; Titova, I.A.; Znamenskaya, A.P. J. Appl. Chem. USSR (Engl. Transl.), 1971, 44, 1127-1132. (e) Krishna Mohan, K.V.V.; Narender, N.; Kulkarni, S.J. Micropor. Mesopor. Mater., 2007, 106, 229-235.
- 6. Kotsuki, H.; Mehta, B.K.; Yanagisawa, K. Synlett, 2001, 8, 1323-1325.
- (a) Chafetz, H.; Patmore, E.L. U.S. Pat. 3408351A, 1968. (b) Bischoff, V.C.; Herma, H. J. Prakt. Chem., 1976, 318, 891-894.
- (a) Palacios, F.; Alonso, C.; Rubiales, G.; Ezpeleta, J.M. *Eur. J. Org. Chem.* 2001, 2115-2122.
 (b) Ruangsiyanand, C.; Rimek, H.-J.; Zymalkowski, F. *Chem. Ber.*, 1970, *103*, 2403-2410.
 (c) Noguchi, M.; Onimura, K.; Isomura, Y.; Kajigaeshi, S. *J. Het. Chem.*, 1991, 28, 885-890.
 (d) Upadisheva, A.V.; Grigorieva, N.D.; Znamenskaya, A.P. S.U. Pat. 551328A1, 1977.
- Dorofeenko, G.N.; Safaryan, G.P.; Polyakova, T.I. Chem. Het. Comp., 1972, 8, 1318-1320.
- 10. Driessen, P.B.J.; Grace, D.S.B.; Hogeveen, H.; Jorritsma, H. *Tetrahedron Lett.*, **1976**, *17*, 2263-2266.
- 11. Couture, A.; Bochu, C.; Grandclaudon, P. Tetrahedron Lett., 1989, 30, 6865-6866.
- 12. (a) Wotring, L.L.; Townsend, L.B. *Cancer Res.*, **1989**, *49*, 289-294. (b) Raić-Malić, S.;
 Grdiša, M.; Pavelic, K.; Mintas, M. *Eur. J. Med. Chem.*, **1999**, *34*, 405-413. (c) Mischra,
 R.C.; Dwivedi, N.; Tripathi, R.P.; Bansal, I.; Saxena, J.K. *Nucleosides, Nucleotides & Nucleic Acids*, **2005**, *24*, 15-35.
- 13. (a) Morrey, J.D.; Smee, D.F.; Sidwell, R.W.; Tseng, C. Antiviral Res., 2002, 55, 107-116.
 (b) Sharma, A.P.; Ollapally, A.P.; Jones, W.; Lemon, T. Nucleosides, Nucleotides & Nucleic Acids, 1992, 11, 1009-1038. (c) Kabbaj, Y.; Lazrek, H.B.; Barascut, J.L.; Imbach, J.L. Nucleosides, Nucleotides & Nucleic Acids, 2005, 24, 161-172. (d) Maslen, H.L.; Hughes, D.; Hursthouse, M.; De Clercq, E.; Balzarini, J.; Simons, C. J. Med. Chem., 2004, 47, 5482-5491. (e) Rusinov, V.L.; Egorov, I.N.; Chupakhin, O.N.; Belanov, E.F.; Bormotov, N.I.; Serova, O.A. Pharm. Chem. J., 2012, 45, 655-659 [Engl. transl. from Khim.-Farm. Zh., 2011, 45(11), 7-11].
- 14. (a) Khalil, N.S.A.M.; Mansour, A.K.; Eid, M.M. Nucleosides, Nucleotides & Nucleic Acids, 2004, 23, 1889-1910. (b) Modzelewska-Banachiewicz, B.; Kaminska, T. Eur. J. Med. Chem., 2001, 36, 93-99.

- 15. Rusinov, V.L.; Zyryanov, G.V.; Pilitcheva, T.L.; Chupakhin, O.N.; Neunhoeffer, H. J. *Heterocyclic Chem.*, **1997**, *34*, 1013-1019.
- (a) Chupakhin, O.N.; Rusinov, G.L.; Beresnev, D.G.; Neunhoeffer, H. J. Heterocyclic Chem., 1997, 34, 573-578. (b) Rusinov, G.L.; Beresnev, D.G.; Itsikson, N.A.; Chupakhin, O.N. Heterocycles, 2001, 55, 2349-2360. (c) Chupakhin, O.N.; Rusinov, G.L.; Itsikson, N.A.; Beresnev, D.G.; Fedorova, O.V.; Ovchinnikova, I.G. Russ. Chem. Bull., 2004, 53, 2308-2313. [Engl. transl. from Izv. Akad. Nauk, Ser. Khim., 2004, 2210-2215]. (d) Beresnev, D.G.; Itsikson, N.A.; Chupakhin, O.N.; Charushin, V.N.; Kodess, M.I.; Butakov, A.I.; Rusinov, G.L.; Morzherin, Yu.Yu.; Konovalov, A.I.; Antipin, I.S. J. Org. Chem., 2006, 71, 8272-8275.
- 17. Egorov, I.N.; Kovalev, I.S.; Rusinov, V.L.; Chupakhin, O.N. Z. Naturforsch., 2010, 65b, 1359-1362.
- 18. Weiss, M. J. Am. Chem. Soc., 1952, 74, 200-202.
- 19. Klumpp, D.A.; Garza, M.; Jones, A.; Mendoza, S. J. Org. Chem., 1999, 64, 6702-6705.
- 20. (a) Yadav, J.S.; Reddy, B.V.S.; Gupta, M.K.; Prathap, I.; Dash, U. *Synthesis*, 2007, 1077-1081. (b) Nikam, S.S.; Sahasrabudhe, A.D.; Shastri, R.K.; Ramanathan, S. *Synthesis*, 1983, 145-147. (c) Hamana, M.; Iwasaki, G.; Saeki, S. *Heterocycles*, 1982, 17, 177-181.
- 21. Uchutilova, V.; Fiedler, P.; Prystas, M.; Gut, J. Collect. Czech. Chem. Commun., 1971, 36, 1955.
- 22. Heinisch, L. J. Prakt. Chem., 1974, 316, 667-678.
- Iogansen, A.V.; Kurkchi, G.A.; Baeva, V.P.; Rasskazova, Z.N.; Salamatina, G.A., *Russ. J. Org. Chem.*, **1971**, 7, 2509-2511 [Engl. transl. from *Zh. Org. Khim.*, **1971**, 7, 2509].
- 24. Kelly, K.K.; Matthews, J.S. J. Chem. Eng. Data, 1969, 14, 276-277.
- 25. Martin, A. U.S. Pat. 5776884, 1998.