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Identification of Bacteria-Selective Threonyl-tRNA Synthetase Substrate Inhibitors by Structure-Based Design

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Supporting Information

ABSTRACT: A series of potent and bacteria-selective threonyl-tRNA synthetase (ThrRS) inhibitors have been identified using structure-based drug design. These compounds occupied the substrate binding site of ThrRS and showed excellent binding affinities for all of the bacterial orthologues tested. Some of the compounds displayed greatly improved bacterial selectivity. Key residues responsible for potency and bacteria/human ThrRS selectivity have been identified. Antimicrobial activity has been achieved against wild-type *Haemophilus influenzae* and efflux-deficient mutants of



wild-type Haemophilus influenzae and efflux-deficient mutants of Escherichia coli and Burkholderia thailandensis.

INTRODUCTION

Infections caused by antibiotic-resistant pathogens have led to a crisis in health care, especially for those due to Gram-negative organisms. These disease-causing microbes have developed several different classes of resistance mechanisms, one of which is target-based mutations that reduce drug affinity, as in the case of fluoroquinolones.¹ Alternatively, as seen for the aminoglyco-sides, enzymatic modification of a drug may lead to high-level resistance.² There are resistance mechanisms to all major antibacterial drugs currently in use, and many strains are resistant to multiple drug classes. Thus, new classes of agents that bind to novel targets would be a significant addition to the antimicrobial armamentarium to combat resistant pathogens.

Aminoacyl-tRNA synthetases (aaRSs) are a class of enzymes that have been validated as antimicrobial targets.³ Mupirocin, which inhibits bacterial isoleucyl-tRNA synthetase, is approved for the topical treatment of impetigo due to Staphylococcus aureus and Streptococcus pyogenes.⁴ With the exception of glutaminyl-tRNA synthetase (GlnRS), absent in many Grampositive bacteria, all of the other 19 synthetases are highly conserved and perform essential activities in clinically important microorganisms.⁵ In addition to mupirocin, whole cell screening of natural products has identified multiple aaRS inhibitors with antibacterial activity.⁶ These include borrelidin (ThrRS), granaticin (LeuRS), indolmycin (TrpRS), ochratoxin A (PheRS), and cispentacin (ProRS). In recent years, other aaRS inhibitors have emerged, including those that are mimics of the aminoacyl adenosine monophosphate intermediate (aa-AMP), such as 1 (Figure 1).⁷ Although most of the aa-AMP mimics demonstrated low nanomolar binding affinities against their corresponding aaRSs, they generally displayed weak antimicrobial activity due to poor bacterial cell permeability. ${}^{3\mathrm{f},6,8}_{\mathrm{v},8}$

In screening marine natural products for antibacterial activity, we identified borrelidin (2) as the active component of one of the extracts we prioritized for structural elucidation. Borrelidin inhibits ThrRS⁹ and has an interesting antibacterial profile that includes activity on Yersinia pestis, Haemophilus influenzae, and Francisella tularensis (unpublished results). However, the antibacterial potency and spectrum of borrelidin need to be significantly improved to generate a useful antibacterial agent that targets ThrRS. In addition, borrelidin lacks selectivity for bacterial ThrRS versus human ThrRS, raising concerns of mammalian toxicity. Indeed, borrelidin has been shown to be a potent inhibitor of angiogenesis.¹⁰ Although previous efforts sought to optimize the antibacterial activity of borrelidin, none of these efforts utilized X-ray crystallographic data to guide optimization. To address these issues, we cocrystallized borrelidin with Escherichia coli and human ThrRS (which will be reported on separately).¹¹ Comparison of the human and E. coli enzymes reveals that the borrelidin binding site is highly conserved between the two ThrRS enzymes and that the ATP/ substrate binding site displays more structural variability than the borrelidin site. The E. coli and human ThrRS selectivity ratios for a number of semisynthetic analogues of borrelidin confirm that achieving a good bacterial selectivity with borrelidin analogues may be a challenging task. Thus, we explored totally synthetic inhibitors of bacterial ThrRS that specifically target the substrate binding sites of ThrRS. In this paper we summarizes our effort to design novel small-molecule

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Figure 1. Structures of known ThrRS inhibitors: threonyl-AMP (1), borrelidin (2), threonyl-AMS (3).

antibacterial agents that selectively inhibit bacterial ThrRS enzymes.

The starting point for this program was 5'-O-[N-(threonyl)sulfamoyl] adenosine (AMS, 3), a known competitive inhibitor of ThrRS.¹² The crystal structure of 3 bound to *E. coli* ThrRS has been reported.¹¹ As an adenosine analogue, 3 lacks antibacterial activity due to poor (too polar) drug properties and lacks selectivity for the bacterial enzymes (Table 1). We aimed to address both of these issues by designing compounds with more typical heterocyclic fragments found in drugs and by tailoring the shape of the interaction sites of the inhibitors to achieve bacterial selectivity.

CHEMISTRY

Compound 4 was synthesized through a coupling of sulfonamide 5 and Boc-protected L-threonine 6 followed by deprotection of the Boc (Scheme 1). The synthesis of 10a-f and 11a was initiated with a coupling reaction between dioxaborolane 8a with 6 to form intermediate 9a (Scheme 2). Suzuki coupling of 9a with a corresponding ArBr followed by deprotection of Boc yielded 10a-f and 11a. Similarly, Suzuki coupling of **9b** with a corresponding $ArB(OH)_2$ followed by deprotection of Boc yielded 11c,d. Upon NCS treatment, 11d was converted to 11b. The synthesis of 10g,i started with 12, which was synthesized using a known procedure.¹³ The synthesis of 10j started with a reaction of 7-bromo-1,2,3,4tetrahydroisoquinoline with sulfamoyl chloride to furnish 13. A coupling reaction of 12 or 13 with 6 afforded 14a,b. A conversion of Br in 14a,b to borane pinacol ester yielded 15a,b. A Suzuki coupling reaction of 15a,b to a corresponding ArCl followed by Boc deprotection furnished 10g,i,j (Scheme 3). A coupling reaction of 6 with (3-(aminomethyl)phenyl)boronic acid by HOAT and EDCI furnished 16. After a Suzuki coupling reaction with 7-bromo-2-chloroquinazolin-4-amine followed by deprotection of the Boc, 16 was converted to 17 (Scheme 4). Compound 19 was synthesized through (3-bromophenyl)hydrazine. Conversion of the hydrazine to threonyl hydrazide 18a was effected by threonyl chloride generated in situ. Conversion of Br in 18a to borane pinacol ester yielded 18b. After a Suzuki coupling followed by deprotection of Boc, 18b was converted to 19 (Scheme 5). The synthesis of 22 started with a conversion of 6 into its primary amide, which was subsequently treated with LAH to form 20. This intermediate was treated with 3-bromobenzenesulfonyl chloride under basic conditions to form 21a. A conversion of Br 21a into borane pinacol ester yielded 21b (Scheme 6). Suzuki coupling of 21b with 7-bromo-2-chloroquinazolin-4-amine followed by deprotection of Boc yielded 22. The synthesis of 24 started with methylation of the acylsulfonamide 9b with iodomethane followed by a Suzuki coupling with 5-(4,4,5,5-tetramethyl-1,3,2dioxaborolan-2-yl)isoindolin-1-one and deprotection of the Boc (Scheme 7). The synthesis of **10h** and **26–29** started with a Heck reaction of 6-(3-bromophenyl)pyrimidin-4-amine with vinylsulfonamide to yield **25**. After coupling with a corresponding acid and deprotection of the Boc and TBS under acidic conditions, intermediate **25** furnished **10h** and **26–29** (Scheme 8).

RESULTS AND DISCUSSION

In the initial designs, we simplified 3 by replacing the polar adenosine moiety with a 4-phenoxyphenyl moiety, but retained the acylsulfonamide, as in 4. These changes resulted in loss of potency due to the loss of the H-bonding network that is present with adenosine. As a result, compound 4 had only modest inhibition at 50 μ M concentration. Enzymatic assays showed that 4 was competitive with respect to both ATP and the amino acid substrate (which will be reported on separately). Despite the low potency, we were able to obtain a cocrystal structure of 4 with E. coli ThrRS. This structure demonstrated that compound 4 binds to an "open form" of the ATP binding site that more closely resembles the apo conformation of ThrRS than the ATP-bound conformation (Figure 2). The threonylacylsulfonamide portion of the molecule binds to the Zn atom, as anticipated from the Thr-AMS structure. The 4phenoxy group is largely disordered, suggesting that it has negligible contributions to binding. These data revealed opportunities for further optimization of the binding affinity.

One of the strategies to enhance the binding affinity is to replace the distal phenoxy in 4 with a suitable group which can induce the "closed form" of the ATP pocket. Inhibition of the closed form of the enzyme was preferred, as it targets key highly conserved residues that are required for enzymatic activity. Because of their biochemical importance, these residues are inherently difficult to mutate, thereby making target-based resistance to the new inhibitors less likely. Furthermore, it is well appreciated that targeting the enzymatic transition states is a useful strategy for generating highly potent inhibitors.¹⁴

Computational modeling using Discovery Studio with structural information from 3 suggested that compound 10a, with an aminoquinazoline fragment, could fit the adenosine site of the closed form, as the pyrimidine ring of 10a would bind to the ThrRS enzyme in a fashion similar to that of the pyrimidine ring of 3. As predicted by the modeling, compound 10a was a strong inhibitor of *E. coli* ThrRS (Table 1). The cocrystal structure of 10a bound to *E. coli* ThrRS showed that this compound did indeed bind to the closed form of the ATP pocket with the quinazoline ring sandwiched between Arg520 and Phe379 (Figure 3). In addition, the quinazoline ring formed a number of important H-bonds with nearby residues. The amino group formed bidentate H-bonds with Glu365 and

Table 1. Binding Affinities and Bacterial ThrRS Selectivity of 3, 4, and 10a–j

П	σ	ThrRS K _i (nM)				ratio	
	K	E. coli	B. thail	Y. pestis	Human	Human/E.	
3		13.1	2.8	9.8	13.4	1	
4		36500	193000	33100	36800	1	
10a	N NH2	2.9	0.9	2.2	3.3	1.1	
10b		10.7	2.9	8.2	10.1	0.9	
10c		0.8	0.3	0.7	1.2	1.5	
10d	Me_N N NH2	17.2	12.2	9.3	72	4	
10e		20	23	25	463	23	
10f	NH2	90	89	115	2242	25	
10g	N N NH ₂	2.7	0.7	2	3	1.1	
10h	N N NH ₂	2.7	1.1	1.8	4.1	1.5	
10i		2.4	0.6	1.8	1.8	0.8	
10j	N N N N N N N N N N N N N N N N N N N	9.5	4	5.7	5.2	0.5	

Val376, similar to the amino group in the adenine. In addition, atoms N1 and N3 formed H-bonds with Ser517 and Val376, respectively. The threonyl group bound to the Zn atom, as observed for Thr-AMS. The acylsulfonamide formed a H-bond network with its nearby residues, as predicted from the threonyl-AMS binding mode. The net result of all these

interactions was that compound **10a** is a highly potent inhibitor of *E. coli* ThrRS (K_i of 2.9 nM).

Additional quinazoline analogues 10b-f, together with a few rearranged skeletons, 10g-j, were prepared to explore the structure-activity relationship (SAR) of the ATP binding pocket. Table 1 shows the ThrRS enzymatic binding activity and bacterial selectivity for these analogues. The addition of a C2 amino group, as in 10b, resulted in weaker E. coli ThrRS binding. The cocrystal structure of 10b with E. coli ThrRS showed that although this new amine group forms a hydrogen bond to the carbonyl of Leu373, this is a weak hydrogen bond due to poor geometry and length. Furthermore, compound 10b may be paying a desolvation penalty. The incorporation of a chlorine atom at C2, as shown in 10c, produced a significant improvement in potency. The distance from Cl to the carbonyl of Leu373 is 2.8 Å. This is shorter than the sum of the van der Waals radii of Cl and O (3.3 Å), indicating a halogen bond formed (Figure 4).¹⁵ Compound 10d, with a methyl group at C2, lost potency. The cocrystal structure of 10d in E. coli ThrRS demonstrated that the quinazoline ring was rotated by 180°, avoiding a steric clash between the methyl group and Leu373. To accommodate the size of the quinazoline ring of 10d, Glu365 was found in an alternative orientation (Figure 4). The binding affinity of 10d could be partially attributed to the donor-acceptor H-bonds with Val376. More dramatic loss of potency was observed when a quinazoline was changed into an isoquinoline. Isoquinolines 10e and 10f lost 30-100-fold binding affinity, at least in part due to the loss of a H-bond with Ser517. Lastly, other nonquinazolines, 10g-j, were examined, in which an aminopyrimidine or pyrrolopyrimidine replaced the aminoquinazoline ring, and a naphthyl-, a styryl-, or a tetrahydroisoquinolinyl moiety replaced the phenyl ring. These compounds retained good enzymatic potency, which indicates that different chemical scaffolds can be developed to fit tightly to the ATP binding pocket.

It is worth mentioning that **10e** and **10f**, though less potent, showed good selectivity for all bacterial ThrRS enzymes tested versus the human (cytoplasmic-1) ThrRS enzyme. These data suggest that the H-bond to Ser517 is more critical to binding to human ThrRS than to bacterial ThrRS and that a subtle change in the structure can achieve bacteria/human selectivity in the ATP binding site.

To explore the opportunity for selective bacterial ThrRS inhibitors and to fully understand the structural components necessary to achieve such a level of selectivity, we examined 11a-d. These compounds, missing the critical H-bond to Ser517, were designed to have weaker potency against human ThrRS. Table 2 shows that compounds 11a-d exploit human/ bacterial differences in a unique and dramatic way. The cocrystal structure of 11d bound to E. coli ThrRS shows that 11d induced the expected conformational changes in the E. coli enzyme that are typically seen upon substrate binding, specifically the collapse of regions around both the threonine and ATP binding sites (Figure 5a). (The same collapse has been verified with the cocrystal structure of 11d bound to the B. thailandensis enzyme.) The structure of 11d-human ThrRS enzyme, however, showed that the human enzyme undergoes the comparable collapse around the threonine site, but not in the ATP site (Figure 5b).¹⁶ These results demonstrate that the human and bacterial ATP sites are distinct when bound to this class of compounds. In particular, the conformation and lack of flexibility of Phe379, immobilized by Leu361, and therefore the inability of the polypeptide backbone (around Val376) to

Scheme 1. Synthesis of 4^a



^aReagents and conditions: (a) EDCI, DBU, HOAT, DCM, rt, 2 h, 65%; (b) 4 N HCl in dioxane, rt, 30 min, 90%.

Scheme 2. Synthesis of Analogues 10a-f and 11a-d^a



^aReagents and conditions: (a) 6, EDCI, DBU, HOAT, DCM, rt, 2 h, 55–95%; (b) (i) $PdCl_2(TPP)_2$, ArBr (for 10a–f and 11a) or $ArB(OH)_2$ (for 11c,d), Na_2CO_3 , microwave 140 °C, 5 min; (ii) 4 N HCl in dioxane, 15–79%; (c) NCS, CH_3CN , 70 °C, 1 h, 68%.

Scheme 3. Synthesis of $10g_{ij}^{a}$



"Reagents and conditions: (a) sulfamoyl chloride, DMA, 24 h, 43%; (b) 6, EDCI, DBU, HOAT, DCM, rt, 2 h, 69.7%; (c) bis(pinacolato)diboron, $PdCl_2(dppf)CH_2Cl_2$, KOAc, DMF, 80 °C, 1 h, 60–80%; (d) (i) ArCl, $Pd(TPP)_4$, Na_2CO_3 , dioxane, microwave 10 min, 130 °C; (ii) 4 N HCl in dioxane, 13–46% (two steps).

Scheme 4. Synthesis of 17^a



^{*a*}Reagents and conditions: (a) **6**, EDCI, DBU, HOAT, DCM, rt, 12 h, 99%; (b) (i) $PdCl_2(TPP)_2$, 7-bromo-2-chloroquinazolin-4-amine, Na_2CO_3 , microwave 140 °C, 6 min; (ii) 4 N HCl in dioxane, 12% (two steps).

Scheme 5. Synthesis of 19^a



^{*a*}(a) 6, ethyl chloroformate, *N*-methylmorpholine, THF, 15 min, 66%; (b) bis(pinacolato)diboron, $PdCl_2(dppf)CH_2Cl_2$, KOAc, DMF, 80 °C, 1 h, 44%; (c) (i) $PdCl_2(TPP)_2$, 7-bromo-2-chloroquinazolin-4-amine, Na_2CO_3 , microwave 140 °C, 6 min; (ii) 4 N HCl in dioxane, 31% (two steps).

Scheme 6. Synthesis of 22^{a}



^{*a*}Reagents and conditions: (a) (i) ethyl chloroformate, *N*-methylmorpholine, ammonium hydroxide, THF, rt, overnight, 96%; (ii) LAH, THF, 50 °C, 4 h, 96%; (b) 3-bromobenzenesulfonyl chloride, TEA, DCM, 2 h, 25%; (c) bis(pinacolato)diboron, PdCl₂(dppf)-CH₂Cl₂, KOAc, DMF, 80 °C, 1 h, crude; (d) (i) PdCl₂(TPP)₂, 7bromo-2-chloroquinazolin-4-amine, Na₂CO₃, microwave 140 °C, 6 min, 22%; (ii) 4 N HCl in dioxane, 39%. Scheme 7. Synthesis of 24^a



"Reagents and conditions: (a) MeI, Cs_2CO_3 , 65 °C, 6 h, DMF, 54%; (b) (i) $PdCl_2(TPP)_2$, $ArB(OH)_2$, Na_2CO_3 , microwave 140 °C, 5 min, 55%; (ii) 4 N HCl in dioxane, 47%.

Scheme 8. Synthesis of 10h and $26-29^a$



"Reagents and conditions: (a) vinylsulfonamide, Pd(OAc)₂, TPP, DMF, microwave, 120 °C, 0.5 h, 69%; (b) (i) a corresponding acid, EDCI, DBU, HOAT, DCM, rt, 2 h; (ii) 4 N HCl in dioxane, rt, 30 min, 10–68% (two steps).

hydrogen bond with the indazole portion of the molecule in this context forced **11d** to adopt a much less energetically favorable binding mode in the human enzyme (Figure 6). As a consequence, **11d** is selective for bacterial ThrRS with a $K_i > 50 \mu$ M for the human ThrRS enzyme.

The same explanations account for the observed selectivity for **10e,f**. When the H-bond between N1 and Ser517 is replaced by the repulsive interactions between CH and Ser517 as in **10e,f**, the isoquinoline ring will rotate to minimize the undesired interactions. As anticipated, bacterial ThrRS enzymes can accommodate such a rotation better than the human ThrRS enzyme. These results represented an important proofof-concept that bacterial ThrRS can be targeted with good selectivity, and that the ATP site is likely the best region to achieve this.

The SAR around the threonylsulfonamide moiety has been studied. Cocrystal structures of a number of our inhibitors indicated that the acylsulfonamide, engaged in several H-bonds with nearby residues, appeared indispensable to the binding of the inhibitors (Figure 7). The relative contribution of each of the acylsulfonamide components to binding was examined with the synthesis of 17, 19, 22, and 24 (Table 3). Using 10c and 11a as reference compounds in this study, we replaced the SO₂ group with a CH₂ group as in 17 and with an NH as in 19. Both changes caused >100-fold reduction in enzymatic potency compared to that of 10c. Removal of the threonine carbonyl group as in 22 abolished all activities. The NH of the acylsulfonamide was "masked" with a methyl group as in 24. This change also abolished all activity seen in 11a.



Figure 2. Cocrystal structures of **3** (left) and **4** (right) in *E. coli*. ThrRS. The cocrystal structure of **3** in *E. coli* ThrRS reveals that the side chain containing Arg520 folds down to be closer to the adenosine moiety to form sandwiched interactions together with Phe379 (the "closed form"). In contrast, the apo form of the enzyme adapts a more open form in which Arg520 is engaged in H-bonds with Ile524 and Gln479. The 4-phenoxy moiety in **4**, due to a lack of H-bond interactions as in adenine, is largely disordered upon binding to the enzyme, which allows the ATP binding site to remain "open".



Figure 3. Part of the H-bond networks of compound 10a in *E. coli* ThrRS. Residues that interact with the quinazoline ring are shown in bold.



Figure 4. Overlay of cocrystal structures 10c (aqua) and 10d (cyan). The Glu365 (shown bold) residue moves in the presence of 10d. The halogen bond is indicated with a dashed line.

reiterated the importance of acylsulfonamide to the binding affinities of our ThrRS inhibitors. The carbonyl group appeared

Table 2. Binding Affinity and Improved Bacterial Selectivity of 11a-d

Ar	s.H	NH ₂	11a-d
	000	ОН	

			selectivity ratio			
	Ar	E. coli	B. thail.	Y. pestis	Human	Human/ E. coli
11a	HN	7.7	7.4	3.9	91	12
11b		52	77	39	1518	29
11c	Me N N H	127	143	83	5820	46
11d	N N N N N N N N N N N N N N N N N N N	182	164	132	>50,000	>270

particularly critical, possibly due to its roles in mediating the pK_a of both threonine and acylsulfonamide.

Finally, analogues that outline the importance of the Zn chelating portion of our ThrRS inhibitors were prepared. Although it is generally believed that ThrRS enzymes recognize L-threonine as a substrate, in our study we have expanded the SAR surrounding the methyl group of L-threonine. The corresponding diamine and diol had also been evaluated. Table 4 summarizes the results for these compounds. These data showed that there is a limitation in the Zn binding pocket to accommodate size, with threonine being the optimal followed by ethylthreonine 26 and isopropylthreonine 27. The human (cytoplasmic-1) and B. thailandensis ThrRS enzymes are particularly sensitive to the size of the head piece, which offers an opportunity for selectivity optimization. The recognition of threonine analogues by ThrRS appeared to be specific in this study, as compounds 28 and 29 showed no appreciable enzymatic activity.



Figure 5. Phe379 is more flexible in the bacterial ThrRS enzymes such that it can shift slightly away from the active site and accommodate 11d (a). In the human ThrRS orthologue (b), Phe379 is unable to shift, and the inhibitor is forced to adopt a very different and less energetically favorable conformation.



Figure 6. Bacterial selectivity is due to differences in underlying ThrRS residues that prevent Phe379 movement to accommodate 11d. The tilted conformation of selective inhibitors, such as 11d, presses tightly against the sides of the ATP pocket (compared with the conformation of the nonselective 10a). In the bacterial enzymes, Phe379 can accommodate this tilt due to the presence of a less rigid underlying residue, Cys361. In the human orthologue, however, the underlying residue Leu361 prevents movement of Phe379 and the inhibitor must find an alternate (less energetically favorable) conformation.



Figure 7. Acylsulfonamide H-bond network.

Through these SAR studies, we were able to evaluate antimicrobial activity for a few potent ThrRS inhibitors. This was carried out through measurement of antimicrobial activity against a panel of Gram-negative wild-type strains. In addition, antibacterial activity was measured versus *E. coli imp, E. coli tolC*, and a *B. thailandensis* pump knockout strain to measure the impact of permeability and efflux pumps on antimicrobial activity. Table 5 shows the minimal inhibitory concentration (MIC) values against clinically important pathogens and BSL2 surrogate strains of important biodefense pathogens. Compared



	ThrRS K_i (nM)					
ID	E. coli	B. thailandensis	Y. pestis	human		
10c	0.8	0.3	0.7	1.2		
17	329	140	436	935		
19	136	34	70	399		
22	>50 µM	>50 µM	$>50 \ \mu M$	>50 µM		
11a	7.7	7.4	3.9	91.0		
24	$>50 \ \mu M$	>50 µM	$>50 \ \mu M$	$>50 \ \mu M$		

to threonyl-AMS (3), the newly identified ThrRS inhibitors showed improved antibacterial activity against wild-type *H. influenzae, E. coli tolC,* and *B. thailandensis* pump mutant strains, although all compounds had poor antimicrobial activity against wild-type strains of *E. coli* and *B. thailandensis.* As these

Table 4. ThrRS Inhibitors Bearing Different Head Pieces



				ThrRS K_i (nM)			
ID	Х	Y	Z	E. coli	B. thailandensis	Y. pestis	human
10h	NH ₂	OH	CH ₃	2.7	1.1	1.8	4.1
26	NH ₂	OH	C_2H_5	4.3	14.1	2.7	32
27	NH ₂	OH	<i>i</i> -Pr	32.4	588	22.5	601
28	OH	OH	CH ₃	>50 µM	>50 µM	>50 µM	>50 µM
29	NH ₂	NH ₂	CH ₃	>50 µM	2262	$>50 \ \mu M$	12576

Table 5. MIC Values of Representative Compounds^a

	MIC (µg/mL)						
ID	E. coli	E. coli imp	E. coli tolC	H. influenzae	B. thail.	B. thail. (KO)	
3	>128	>128	>128	>128	>128	64	
10a	>128	>128	64	16	>128	8	
10b	>128	>128	64	32	>128	4	
10c	>128	128	16	8	>128	32	
10g	>128	>128	64	4	>128	4	
10h	>128	>128	64	8	>128	32	
10i	>128	>128	16	16	>128	32	

^{*a*}Values that are $\leq 32 \ \mu g/mL$ are shaded light blue.

compounds have significantly increased activity on the pumpdeficient strains, they are likely efficient substrates for efflux pumps and are more limited in this respect than in their ability to permeate cellular membranes. Further structural modifications will be needed to enhance the antimicrobial activity of the inhibitors.

In summary, we have identified a series of ThrRS synthetase inhibitors using SBDD. These compounds bind to the substrate binding site of ThrRS synthetase and show excellent binding affinities for the bacterial enzymes. Several compounds possess greatly improved bacterial selectivity. Key residues responsible for potency and bacteria/human ThrRS selectivity have been identified. Improvements in antimicrobial activity, as compared to AMS, have been achieved against *H. influenzae*. Further work is directed toward the improvement of potency and the expansion of the antibacterial spectrum. These results will be reported in due course.

EXPERIMENTAL SECTION

General Procedures. Flash chromatography was performed on silica gel (280 mesh). ¹H NMR spectra were obtained with a Bruker 300 MHz spectrometer equipped with a 5 mm QNP probe or a Bruker 600 MHz spectrometer equipped with a Bruker BioSpin TCI 1.7 mm MicroCryoProbe. Chemical shifts are reported in parts per million relative to tetramethylsilane, and the following multiplicity abbreviations were used: s, singlet; d, doublet; t, triplet; m, multiplet; br, broad. HRES-LC/MS was conducted on a Micromass Q-TOF Premier (quadrupole time-of-flight) mass spectrometer (Beverly, MA) coupled with an Agilent 1100 (Santa Clara, CA) HPLC system equipped with a UV diode array detector monitoring 254 nm. HPLC was equipped with a Phenomenex kinetex-C18 column (100 × 3.0, 2.6 μ m). Unless

indicated otherwise, reagents and solvents were obtained from commercial sources and used as received.

General Method A for the Synthesis of Acylsulfonamide. A solution of 6 (1.65 g, 6.0 mmol), HOAT (0.817 g, 6.0 mmol), and EDC (1.15 g, 6.0 mmol) in DCM (20 mL) was stirred at room temperature for 30 min, followed by the addition of sulfonamide (4.0 mmol) and DBU (1.81 mL, 12 mmol). Reaction was monitored by LC/MS and was completed in 2 h. The reaction mixture was diluted with EtOAc, washed with 1.0 N HCl and water, dried over Na_2SO_4 , concentrated, and either used as is or purified by flash chromatography (0–100% ethyl acetate/hexane). Yield: 55–95%.

General Method B for Boc Deprotection. A solution of Bocprotected threonylacylsulfonamide (0.063 mmol) was added to 4.0 N HCl in dioxane (0.5 mL) at room temperature. The reaction was stirred for 2 h, then stripped of volatiles, and purified by RPLC (10– 50% ACN in water) to get the desired compound as a TFA salt. Yield: 30-97%.

General Method C for the Suzuki Coupling Reaction. A flask was charged with 9a (108 mg, 0.20 mmol), ArCl (0.20 mmol), bis(triphenylphosphine)palladium(II) chloride (7.02 mg, 0.010 mmol), and aq sodium carbonate (0.2 mL, 1.0 M aq), vacuumed, flushed with N_2 five times, and then heated in the microwave at 140 °C for 6 min. The crude mixture was neutralized with acetic acid (1.0 N), concentrated, and purified by RPLC (0–90% ACN in water). Yield: 15–87%.

(25,3*R*)-2-Amino-3-hydroxy-*N*-((4-phenoxyphenyl)sulfonyl)butanamide (4). Using methods A and B, the title compound was synthesized in 58% yield in two steps from 6. LC: 98%. ¹H NMR (600 MHz, DMSO): δ 8.01 (br, 3H), 7.85 (t, *J* = 16.6 Hz, 2H), 7.45 (t, *J* = 7.7 Hz, 2H), 7.23 (dd, *J* = 18.1, 10.8 Hz, 1H), 7.09 (t, *J* = 9.8 Hz, 2H), 7.03 (t, *J* = 13.0 Hz, 2H), 3.95 (dd, *J* = 22.0, 16.0 Hz, 1H), 3.47 (dd, *J* = 21.2, 16.9 Hz, 1H), 1.07 (d, *J* = 6.3 Hz, 3H). HRMS (ESI): *m/z* calcd for C₁₆H₁₈N₂OSS (M + H)⁺ 351.1014, found 351.1013. tert-Butyl ((25,3*R*)-3-(tert-Butoxy)-1-oxo-1-(3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)benzenesulfonamido)butan-2-yl)carbamate (9a). Following general method A, 9a (1.05 g) was produced from 8a (0.99 g, 55%). ¹H NMR (300 MHz, DMSO): δ 12.05 (br, 1H), 8.23 (s, 1H), 8.16 (s, 1H), 8.06 (s, 2H), 7.64 (br, 2H), 7.37 (s, 1H), 6.06 (s, 1H), 3.93 (m, 1H), 3.34 (m, 1H), 1.35 (m, 21H), 1.10 (s, 3H), 0.91 (s, 9H).

tert-Butyl ((2*S*,3*R*)-1-(3-Bromobenzenesulfonamido)-3-(*tert*butoxy)-1-oxobutan-2-yl)carbamate (9b). Following general method A, 9b (1.87 g) was produced from 8b (0.94 g, 95%). ¹H NMR (300 MHz, DMSO): δ 12.21 (br, 1H), 8.04 (s, 1H), 7.94 (d, *J* = 8.0 Hz, 2H), 7.87–7.47 (t, *J* = 7.0 Hz, 1H), 6.13 (d, *J* = 9.2 Hz, 1H), 4.35–3.61 (m, 1H), 3.21 (m, 1H), 1.37 (s, 9H), 0.97 (m, 12H). HRMS (ESI): *m*/*z* calcd for C₁₉H₂₉BrN₂O₆S (M + H)⁺ 493.1008, found 493.1029.

((2*S*,3*R*)-2-Amine-*N*-((3-(4-aminoquinazolin-7-yl)phenyl)sulfonyl)-3-hydroxybutanamide (10a). Using methods C and B, 10a was produced from 9a (79%). LC: 99%. ¹H NMR (300 MHz, DMSO): δ 9.81 (br, 2H), 8.89 (s, 1H), 8.55 (d, J = 8.7 Hz, 1H), 8.27 (s, 1H), 8.13 (d, J = 8.7 Hz, 1H), 8.07–8.01 (m, 2H), 7.99 (d, J = 8.0Hz, 1H), 7.84 (br, 2H), 7.72 (t, J = 7.8 Hz, 1H), 4.01–3.87 (m, 1H), 3.25 (m, 1H), 1.10 (d, J = 6.4 Hz, 3H). HRMS (ESI): m/z calcd for C₁₈H₁₉N₅O₄S (M + H)⁺ 402.1235, found 402.1245.

(25,3Å)-2-Amino-N-((3-(2,4-diaminoquinazolin-7-yl)phenyl)sulfonyl)-3-hydroxybutanamide (10b). Using methods C and B, 10b was produced from 9a (41%). LC: 96%. ¹H NMR (600 MHz, DMSO): δ 9.10 (s, 1H), 8.87 (s, 1H), 8.49–8.28 (m, 1H), 8.17 (s, 1H), 7.86 (m, 2H), 7.82–7.67 (m, 3H), 7.63 (s, 1H), 7.48 (s, 1H), 7.35–7.06 (m, 1H), 3.94 (m, 1H), 1.08 (d, J = 6.4 Hz, 3H). HRMS (ESI): m/z calcd for C₁₈H₂₀N₆O₄S (M + H)⁺ 417.1345, found 417.1350.

(2*S*,3*R*)-2-Amino-*N*-((3-(4-amino-2-chloroquinazolin-7-yl)phenyl)sulfonyl)-3-hydroxybutanamide (10c). Using methods C and B, 10c was produced from 9a (15%). LC: 93%. ¹H NMR (300 MHz, DMSO): δ 8.45 (br, 1H), 8.38 (d, *J* = 9.0 Hz, 1H), 8.27 (s, 1H), 8.05 (d, *J* = 8.2 Hz, 1H), 7.84 (m, 3H), 7.69 (t, *J* = 7.7 Hz, 1H), 4.05– 3.77 (m, 1H), 3.5 (m, 1H), 1.10 (d, *J* = 6.4 Hz, 3H). HRMS (ESI): *m*/ *z* calcd for C₁₈H₁₈ClN₅O₄S (M + H)⁺ 436.0846, found 436.0839.

((2*S*,3*R*)-2-Amino-*N*-((3-(4-amino-2-methylquinazolin-7-yl)phenyl)sulfonyl)-3-hydroxybutanamide (10d). Using methods C and B, 10d was produced from 9a (55%). LC: 99%. ¹H NMR (CD₃OD): δ 8.34 (m, 1H), 8.3 (d, *J* = 7.2 Hz, 1H), 8.01 (d, *J* = 6 Hz, 1H), 7.89 (m, 2H), 7.81 (dd, *J*₁ = 9 Hz, *J*₂ = 3 Hz, 1H), 7.63 (t, *J* = 9 Hz, 1H), 4.13 (m, 1H), 3.56 (m, 1H), 2.50 (s, 3H), 1.25 (d, *J* = 6 Hz, 3H). HRMS (ESI): *m*/*z* calcd for C₁₉H₂₁N₅O₄S (M + H)⁺ 416.1392, found 416.1390.

(25,3*R*)-2-Amino-*N*-((3-(1-amino-3-chloroisoquinolin-6-yl)phenyl)sulfonyl)-3-hydroxybutanamide (10e). Using methods C and B, 10e was produced from 9a (15%). LC: 85%. ¹H NMR (300 MHz, DMSO): δ 8.36 (d, J = 9.0 Hz, 1H), 8.30 (s, 1H), 8.18 (m, 3H), 8.02 (m, 2H), 7.81 (d, J = 6.0 Hz, 2H), 7.10 (s, 1H), 4.05 (m, 1H), 3.76 (m, 1H), 1.11 (d, J = 6.3 Hz, 3H). HRMS (ESI): m/z calcd for C₁₉H₁₉ClN₄O₄S (M + H)⁺ 435.0894, found 435.0905.

(25,3*R*)-2-Amino-*N*-((3-(1-aminoisoquinolin-6-yl)phenyl)sulfonyl)-3-hydroxybutanamide (10f). Using methods C and B, 10f was produced from 9a (36%). LC: 98%. ¹H NMR (300 MHz, DMSO): δ 13.54 (s, 1H), 9.33 (s, 2H), 8.79 (d, J = 8.7 Hz, 1H), 8.37 (s, 2H), 8.19 (m, 4H), 8.05 (d, J = 8.0 Hz, 1H), 7.87–7.66 (m, 2H), 7.33 (t, J = 11.5 Hz, 1H), 4.17–3.92 (m, 1H), 3.76 (br, 1H), 1.14 (d, J = 6.6 Hz, 3H). HRMS (ESI): m/z calcd for C₁₉H₂₀N₄O₄S (M + H)⁺ 401.1283, found 401.1292.

tert-Butyl ((2*S*,3*R*)-1-(7-Bromonaphthalene-2-sulfonamido)-3-(*tert*-butoxy)-1-oxobutan-2-yl)carbamate (14a). Following general method A, 14a (0.192 g) was produced from 12 (0.145 g, 69.7%). LC: 90%. ¹H NMR (300 MHz, DMSO): δ 12.19 (s, 1H), 8.64 (s, 1H), 8.55 (s, 1H), 8.17 (d, *J* = 8.8 Hz, 1H), 8.03 (d, *J* = 8.8 Hz, 1H), 7.95 (d, *J* = 8.7 Hz, 1H), 7.85 (dd, *J* = 8.8, 1.9 Hz, 1H), 6.08 (d, *J* = 9.3 Hz, 1H), 4.04–3.91 (m, 1H), 3.91–3.71 (m, 1H), 1.33 (s, 9H), 0.95 (d, *J* = 5.7 Hz, 3H), 0.89 (s, 9H). HRMS (ESI): *m*/*z* calcd for C₂₃H₃₁BrN₂O₆S (M + H)⁺ 543.1164, found 543.1174. tert-Butyl ((25,3*R*)-3-(tert-Butoxy)-1-oxo-1-(7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)naphthalene-2-sulfonamido)butan-2-yl)carbamate (15a). A solution of bis(pinacolato)diboron (0.373 g, 1.5 mmol), $PdCl_2(dppf)CH_2Cl_2(cat.)$, KOAc (0.360 g, 3.67 mmol), and 14a (0.665 g, 1.224 mmol) in DMF (7.0 mL) was flushed with argon and heated at 80 °C for 2.5 h in a sealed tube. The reaction mixture was diluted with ethyl acetate and washed with water. The organic layer was separated and concentrated to a solid (0.58 g, 80%) which was used directly in the next reaction.

(25,3*R*)-2-Amino-*N*-((7-(6-aminopyrimidin-4-yl)naphthalen-2-yl)sulfonyl)-3-hydroxybutanamide (10g). Using methods C and B, 10g was produced from 15a (45.8%). LC: 94%. ¹H NMR (300 MHz, DMSO): δ 8.89 (s, 1H), 8.75 (m, 2H), 8.64 (br, 1H), 8.27 (t, *J* = 7.9 Hz, 2H), 8.18 (m, 3H), 8.05 (dd, *J* = 8.7, 1.7 Hz, 1H), 7.21 (s, 1H), 4.10–3.94 (m, 1H), 3.72 (m, 1H), 1.10 (d, *J* = 6.4 Hz, 3H). HRMS (ESI): m/z calcd for C₁₈H₁₉N₃O₄S (M + H)⁺ 402.1236, found 402.1230.

(*E*)-2-(3-(6-Aminopyrimidin-4-yl)phenyl)ethenesulfonamide (25). Triphenylphosphine (0.03 equiv), 6-(3-bromophenyl)pyrimidin-4-amine (0.1 g, 1.0 equiv), diacetoxypalladium (0.03 equiv), ethenesulfonamide (1.2 equiv), and TEA (1.2 equiv) were mixed in DMF (1.5 mL) and degassed. The reaction was microwaved at 120 °C for 1 h. The reaction mixture was concentrated and purified by column chromatography, eluting with ethyl acetate, to give the title compound as a white solid (69%). ¹H NMR (DMSO-*d*₆): δ 8.60 (s, 1H), 8.13 (s, 1H), 7.93 (d, *J* = 9 Hz, 1H), 7.83 (d, *J* = 7.5 Hz, 1H), 7.65 (t, *J* = 7.8 Hz, 1H), 7.63 (br, 2H), 7.50 (d, *J* = 15.6 Hz, 1H), 7.25 (d, *J* = 15.0 Hz, 1H), 7.10 (s, 1H). 6.03 (br, 2H).

(25,3*R*)-2-Amino-*N*-(((*E*)-3-(6-aminopyrimidin-4-yl)styryl)sulfonyl)-3-hydroxybutanamide (10h). Following general methods A and B, the title compound was produced as a white solid from 25 (68%). HPLC: 99%. ¹H NMR (CD₃OD): δ 8.66 (s, 1H), 8.15 (s, 1H), 7.94 (t, *J* = 9 Hz, 2H), 7.8 (d, *J* = 15 Hz, 1H), 7.70 (t, *J* = 6.0 Hz, 2H), 7.50 (d, *J* = 15.6 Hz, 1H), 7.07 (s, 1H), 4.18 (m, 1H), 3.72 (d, *J* = 4.5, 1H), 1.34 (d, *J* = 6.0 Hz, 3H). HRMS (ESI): *m*/*z* calcd for C₁₆H₁₉N₅O₄S (M + H)⁺ 378.1235, found 378.1232.

(25,3*R*)-*N*-((7-(7*H*-Pyrrolo[2,3-*d*]pyrimidin-4-yl)naphthalen-2-yl)sulfonyl)-2-amino-3-hydroxybutanamide (10i). Using the same procedure as for the synthesis of 10g, 10i was produced from 15a (13%). HPLC: 99%. ¹H NMR (600 MHz, DMSO): δ 9.05 (s, 1H), 8.93 (d, J = 5.0 Hz, 2H), 8.54 (d, J = 8.4 Hz, 1H), 8.24 (t, J = 9.5Hz, 2H), 8.15 (br, 3H), 8.00 (d, J = 8.4 Hz, 1H), 7.76 (s, 1H), 7.21 (s, 1H), 5.5 (br, 1H), 4.07–3.97 (m, 1H), 3.73 (m, 1H), 1.09 (d, J = 6.3Hz, 3H). HRMS (ESI): m/z calcd for C₂₀H₁₉N₅O₄S (M + H)⁺ 426.1236, found 426.1236.

7-Bromo-3,4-dihydroisoquinoline-2(1*H***)-sulfonamide (13).** A solution of 7-bromo-1,2,3,4-tetrahydroisoquinoline (0.424 g, 2.0 mmol) in DMA (3 mL) was cooled to 0 °C. To this was added sulfamoyl chloride (0.254 g, 2.2 mmol) dissolved in DMA (2.0 mL). The reaction mixture was stirred at room temperature overnight and was purified by RPLC as a white solid (43%). HPLC: 89%. ¹H NMR (300 MHz, DMSO): δ 7.43 (s, 1H), 7.41–7.30 (d, *J* = 9.0 Hz, 1H), 7.14 (d, *J* = 9.0 Hz, 1H), 6.94 (s, 2H), 4.21 (s, 2H), 3.26 (t, *J* = 5.9 Hz, 2H), 2.94–2.79 (t, *J* = 6.0 Hz, 2H). HRMS (ESI): *m/z* calcd for C₉H₁₁BrN₂O₂S (M + H)⁺ 290.9803, found 290.9805.

tert-Butyl ((25,3R)-1-(7-Bromo-1,2,3,4-tetrahydroisoquinoline-2-sulfonamido)-3-(tert-butoxy)-1-oxobutan-2-yl)carbamate (14b). Using procedure A, 14b was synthesized from 13 (69.4%). This compound was used without further handling.

tert-Butyl ((25,3R)-3-(tert-butoxy)-1-oxo-1-(7-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1,2,3,4-tetrahydroisoquinoline-2-sulfonamido)butan-2-yl)carbamate (15b). Using the same procedure as for the synthesis of 15a, 15b was synthesized (59.7%) and was used without further handling.

(25,3*R*)-*N*-((7-(7*H*-Pyrrolo[2,3-*d*])pyrimidin-4-yl)-3,4-dihydroisoquinolin-2(1*H*)-yl)sulfonyl)-2-amino-3-hydroxybutanamide (10j). Using methods C and B, 10j was produced from 15b as a HCl salt (17%). HPLC: 77%. ¹H NMR (300 MHz, DMSO): δ 13.08 (*s*, 1H), 12.46 (*s*, 1H), 9.01 (*s*, 1H), 8.39 (br, 3H), 8.00 (d, *J* = 7.6 Hz, 2H), 7.92 (*s*, 1H), 7.49 (d, *J* = 8.1 Hz, 1H), 7.13 (*s*, 1H), 4.71 (q, *J* = 15.7 Hz, 2H), 4.17–3.96 (m, 1H), 3.84 (br, 1H), 3.75–3.51 (m, 2H), 1.20 (t, J = 6.0 Hz, 3H). HRMS (ESI): m/z calcd for $C_{19}H_{22}N_6O_4S$ (M + H)⁺ 431.1501, found 431.1498.

(25,3*R*)-2-Amino-3-hydroxy-*N*-((3-(1-oxoisoindolin-5-yl)phenyl)sulfonyl)butanamide (11a). Using methods C and B, 11a was produced from 9a as a HCl salt (36%). HPLC: 99%. ¹H NMR (300 MHz, DMSO): δ 8.67 (s, 1H), 8.25 (m, 4H), 8.09 (d, J = 7.8 Hz, 1H), 7.99 (d, J = 7.9 Hz, 1H), 7.92 (s, 1H), 7.87–7.65 (m, 3H), 4.47 (s, 2H), 4.16–3.94 (m, 1H), 3.84 (s, 1H), 1.11 (d, J = 6.4 Hz, 3H). HRMS (ESI): m/z calcd for C₁₈H₁₉N₃O₅S (M + H)⁺ 390.1123, found 390.1127.

(25,3*R*)-2-Amino-3-hydroxy-*N*-((3-(3-methyl-1*H*-indazol-5-yl)phenyl)sulfonyl)butanamide (11c). Using methods C and B, 11c was produced from 9b as a TFA salt (18%). LC: 99%. ¹H NMR (300 MHz, DMSO): δ 8.21 (br, 1H), 8.05 (d, *J* = 5.2 Hz, 2H), 8.03 (s, 1H), 7.88 (d, *J* = 8.3 Hz, 1H), 7.72 (s, 1H), 7.70 (s, 1H), 7.67 (dd, *J* = 4.3, 2.2 Hz, 1H), 7.65 (m, 1H), 7.61 (s, 1H), 7.58 (s, 1H), 4.08−3.91 (m, 1H), 3.61 (m, 1H), 2.57 (s, 3H), 1.10 (d, *J* = 6.4 Hz, 3H). HRMS (ESI): *m*/*z* calcd for C₁₈H₂₀N₄O₄S (M + H)⁺ 389.1283, found 389.1281.

tert-Butyl ((2*S*, 3*R*)-1-(3-(1*H*-Indazol-5-yl)benzenesulfonamido)-3-(*tert*-butoxy)-1-oxobutan-2-yl)carbamate (11d). Using methods C and B, 11d was produced from 9b as a TFA salt (29%). HPLC: 98%. ¹H NMR (600 MHz, DMSO): δ 8.13 (s, 2H), 8.04 (s, 1H), 7.86 (br, 3H), 7.91–7.70 (m, 2H), 7.65 (s, 2H), 7.58 (t, *J* = 7.6 Hz, 1H), 4.01–3.84 (m, 1H), 3.48 (d, *J* = 11.0 Hz, 1H), 1.14–0.95 (d, *J* = 6.0 Hz, 3H). HRMS (ESI): *m/z* calcd for C₁₇H₁₈N₄O₄S (M + H)⁺ 375.1127, found 375.1138.

(25,3*R*)-2-Amino-*N*-((3-(3-chloro-1*H*-indazol-5-yl)phenyl)sulfonyl)-3-hydroxybutanamide (11b). A solution of 11d (0.083 mmol) and NCS (0.014 g, 0.11 mmol) in ACN (3 mL) was heated to 70 °C for 1 h. The reaction was cooled to room temperature and was concentrated to an oil. The product was purified by RPLC to give the desired product as a TFA salt (68%). HPLC: 99%. ¹H NMR (300 MHz, DMSO): δ 13.43 (s, 1H), 8.16 (s, 1H), 7.89 (d, *J* = 10.6 Hz, 2H), 7.83 (m, 4H), 7.72 (d, *J* = 9.0 Hz, 1H), 7.58 (t, *J* = 7.7 Hz, 1H), 4.01–3.82 (m, 1H), 3.5 (m, 1H), 1.10 (d, *J* = 6.4 Hz, 3H). HRMS (ESI): *m*/*z* calcd for C₁₇H₁₇ClN₄O₄S (M + H)⁺ 409.0737, found 409.0733.

(3-(((2*S*,3*R*)-3-(*tert*-Butoxy)-2-((*tert*-butoxycarbonyl)amino)butanamido)methyl)phenyl)boronic acid (16). To a roundbottom flask charged with 6 (716 mg, 2.6 mmol), (3-(aminomethyl)phenyl)boronic acid hydrochloride (375 mg, 2.00 mmol), HOAT (272 mg, 2.00 mmol), and diisopropylethylamine (1.4 mL, 8.0 mmol) in DCM (6.7 mL) was added EDCI (767 mg, 4.00 mmol) in one portion at room temperature. LC/MS after 12 h showed complete conversion. Workup: the reaction was concentrated and purified by reversed-phase liquid chromatography (5–100% acetonitrile/water). The yield of the product as a colorless foam was 99% (810 mg). ¹H NMR (300 MHz, DMSO): δ 8.21 (t, *J* = 5.6 Hz, 1H), 7.98 (s, 2H), 7.74–7.62 (m, 2H), 7.38–7.18 (m, 2H), 6.05 (d, *J* = 8.9 Hz, 1H), 4.33 (dd, *J* = 14.8, 5.9 Hz, 1H), 4.21 (dd, *J* = 14.8, 5.2 Hz, 1H), 3.92 (d, *J* = 9.4 Hz, 2H), 1.40 (s, 9H), 1.07 (s, 9H), 1.02 (d, *J* = 5.8 Hz, 3H). HRMS (ESI): *m/z* calcd for C₂₀H₃₃BN₂O₆ (M + H)⁺ 409.2510, found 409.2523.

(2*S*,3*R*)-2-Amino-*N*-(3-(4-amino-2-chloroquinazolin-7-yl)benzyl)-3-hydroxybutanamide (17). Using general methods B and C, the title compound was synthesized from 16 as a HCl salt in 12% combined yield for two steps. HPLC: 94%. ¹H NMR (300 MHz, DMSO): δ 9.06 (s, 1H), 8.71 (s, 1H), 8.37 (m, 1H), 8.16 (s, 2H), 7.88 (m, 1H), 7.84 (m, 2H), 7.49 (m, 3H), 5.61 (br, 1H), 4.48 (m, 1H), 3.96 (m, 1H), 3.51 (m, 2H), 1.19 (d, *J* = 6.0 Hz, 3H). HRMS (ESI): m/z calcd for $C_{19}H_{21}ClN_5O_2$ (M + H)⁺ 386.1384, found 386.1390.

tert-Butyl ((25,3*R*)-1-(2-(3-Bromophenyl)hydrazinyl)-3-(tertbutoxy)-1-oxobutan-2-yl)carbamate (18a). A round-bottom flask charged with 6 (275 mg, 1.00 mmol) and ethyl chloroformate (115 μ L, 1.2 mmol) was treated with 4-methylmorpholine (132 μ L, 1.2 mmol) dropwise over 5 min. After the solution was stirred for 5 min at room temperature, (3-bromophenyl)hydrazine hydrochloride (223 mg, 1.00 mmol) was added followed by 4-methylmorpholine (264 μ L, 2.4 mmol). After 15 min, LC/MS showed consumption of the starting material. Workup: the reaction was concentrated and purified by reversed-phase liquid chromatography (5–100% acetonitrile/water). The yield of the light yellow foam was 66% (292 mg). HPLC: 95%. ¹H NMR (300 MHz, DMSO): δ 9.64 (s, 1H), 8.05 (s, 1H), 7.06 (t, *J* = 6.0 Hz, 1H), 7.00 (s, 1H), 6.83 (d, *J* = 9 Hz, 1H), 6.78 (d, *J* = 6.0 Hz, 1H), 6.30 (d, *J* = 8.4 Hz, 1H), 3.99 (dd, *J* = 8.5, 4.1 Hz, 1H), 3.92–3.80 (m, 1H), 1.42 (s, 9H), 1.18 (s, 9H), 1.06 (d, *J* = 6.2 Hz, 3H). HRMS (ESI): *m*/*z* calcd for C₁₉H₃₀BrN₃O₄ (M + H)⁺ 444.1498, found 444.1506.

tert-Butyl ((2S,3R)-3-(tert-Butoxy)-1-oxo-1-(2-(3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)hydrazinyl)butan-2yl)carbamate (18b). A septum-capped vial charged with 18a (178 mg, 0.4 mmol), 4,4,4',4',5,5,5',5'-octamethyl-2,2'-bis(1,3,2-dioxaborolane) (112 mg, 0.440 mmol), and KOAc (118 mg, 1.20 mmol) in dry DMSO (1.2 mL) was vacuum flushed with nitrogen five times, then treated with bis(diphenylphosphino)ferrocene]dichloropalladium(II) complex with dichloromethane (9.8 mg, 0.012 mmol), and placed in an 80 °C heating block. LC/MS after 60 min showed ~50% conversion. After 2 h, the reaction was diluted with ether, filtered through Celite, concentrated, and purified by reversed-phase liquid chromatography (5-100% acetonitrile/water), giving 86 mg (44% yield) of the title compound. ¹H NMR (300 MHz, DMSO): δ 10.06 (s, 1H), 8.19 (s, 1H), 7.91 (s, 1H), 7.16–7.07 (m, 2H), 7.04 (t, J = 7.0 Hz, 1H), 6.93 (d, J = 7.9 Hz, 1H), 6.21 (d, J = 9.1 Hz, 1H), 4.00 (m, 1H), 3.86 (m, 1H), 1.42 (s, 9H), 1.29 (s, 12H), 1.16 (s, 9H), 1.07 (d, J = 5.4 Hz, 3H). HRMS (ESI): m/z calcd for $C_{25}H_{42}BN_3O_6$ (M + H)⁺ 492.3245, found 492.3275.

(25,3*R*)-2-Amino-*N'*-(3-(4-amino-2-chloroquinazolin-7-yl)phenyl)-3-hydroxybutanehydrazide (19). Using general methods C and B, the title compound was synthesized from 18b in 31% combined yield in two steps. HPLC: 98%. ¹H NMR (300 MHz, DMSO): δ 10.30 (s, 1H), 8.30 (m, 3H), 8.12 (s, 1H), 7.91–7.81 (m, 1H), 7.79 (s, 1H), 7.76–7.72 (m, 1H), 7.54 (t, *J* = 7.4 Hz, 1H), 7.33 (t, *J* = 8.0 Hz, 1H), 7.25–7.13 (m, 2H), 6.90 (d, *J* = 8.7 Hz, 1H), 5.70 (s, 1H), 4.00 (br, 1H), 3.66 (br, 1H), 1.25 (d, *J* = 6.2 Hz, 3H). HRMS (ESI): *m*/*z* calcd for C₁₈H₁₉ClN₆O₂ (M + H)⁺ 387.1336, found 387.133.

tert-Butyl ((2R,3R)-1-Amino-3-(tert-butoxy)butan-2-yl)carbamate (20). A solution of 6 (826 mg, 3.0 mmol) and ethyl chloroformate (1.2 equiv) in THF (15 mL) was added to TEA (1.2 equiv) slowly. After the solution was stirred at room temperature for 5 min, aq ammonium hydroxide (5 equiv) was added. The reaction was stirred overnight. LC/MS revealed the product as M + Na (297). Workup: the reaction was concentrated and purified by RPLC (0-100% ACN in water, over 15 min). The yield of the colorless foam was 787 mg (96%). To a round-bottom flask charged with the abovesynthesized foam (490 mg, 1.79 mmol) dissolved in dry THF (4.5 mL) at 0 °C was added LAH (5.36 mL, 5.36 mmol, 1.0 M solution in THF). The reaction was allowed to warm to room temperature and stirred for a total of 4 h. Reaction progress was monitored by TLC (10% MeOH/DCM with ammonia, visualized with ninhydrin stain). Workup: the reaction was cooled in an ice bath and then slowly quenched with NaOH (aq, 1.0 N) until a thick white precipitate/ emulsion was formed. The emulsion was gently extracted with ether (5 \times 10 mL) and concentrated to a colorless oil (445 mg, 96%) that was used without further purification. ¹H NMR (300 MHz, DMSO): δ 7.15 (s, 2H), 5.96 (d, J = 8.9 Hz, 1H), 3.95–3.74 (m, 2H), 3.33 (m, 2H), 1.40 (s, 9H), 1.13 (s, 9H), 1.02 (d, J = 6.2 Hz, 3H). HRMS (ESI): m/z calcd for $C_{12}H_{28}N_2O_3$ (M + H)⁺ 261.2178, found 261.2179.

tert-Butyl ((2*R*,3*R*)-1-(3-Bromobenzenesulfonamido)-3-(tertbutoxy)butan-2-yl)carbamate (21a). To a round-bottom flask charged with crude 20 (445 mg, 1.71 mmol) and 3-bromobenzene-1sulfonyl chloride (480 mg, 1.88 mmol) in DCM (5.7 mL) was added TEA (310 μ L, 2.22 mmol) dropwise over 5 min. LC/MS, after 1 h, showed complete conversion to the product. Workup: the reaction was concentrated, filtered, and purified by reversed-phase liquid chromatography (5–100% acetonitrile/water). The desired product was obtained as a colorless oil (194 mg, 25%) and was used without further handling.

tert-Butyl ((25,3R)-3-(tert-Butoxy)-1-oxo-1-(3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)benzenesulfonamido)butan**2-yl)carbamate (21b).** A Teflon-capped vial charged with **21a** (112 mg, 0.234 mmol), 4,4,4',5,5,5',5'-octamethyl-2,2'-bis(1,3,2-dioxaborolane) (65 mg, 0.257 mmol), and KOAc (69 mg, 0.70 mmol) in dry DMF (0.584 mL) was vacuum flushed with nitrogen five times, treated with bis(diphenylphosphino)ferrocene]dichloropalladium(II) complex with dichloromethane (5.7 mg, 7.0 mmol), and placed in an 80 °C heating block. After 1 h, LC/MS showed complete consumption of the starting material. Workup: the reaction mixture was concentrated, diluted with ethyl acetate, filtered through Celite, concentrated to dryness, and used in the next step without further purification.

(25,3*R*)-2-Amino-*N*-((3-(4-amino-2-chloroquinazolin-7-yl)phenyl)sulfonyl)-3-hydroxybutanamide (22). Using general methods B and C, the title compound was synthesized from 21b in 9% combined yield for the final three steps starting from 21a. HPLC: 97%. ¹H NMR (300 MHz, DMSO): δ 8.41 (br, 1H), 8.39 (d, *J* = 8.6 Hz, 1H), 8.25 (s, 1H), 8.18 (d, *J* = 7.9 Hz, 1H), 7.98 (d, *J* = 1.5 Hz, 1H), 7.95 (d, *J* = 1.7 Hz, 1H), 7.90 (d, *J* = 8.7 Hz, 2H), 7.80 (m, 2H), 5.41 (d, *J* = 6.0 Hz, 1H), 3.81 (m, 1H), 2.97 (m, 3H), 1.13 (d, *J* = 6.0 Hz, 3H). HRMS (ESI): *m*/*z* calcd for C₁₈H₂₀ClN₅O₃S (M + H)⁺ 422.1053, found 422.1051.

tert-Butyl ((2*S*,3*R*)-1-(3-Bromo-*N*-methylbenzenesulfonamido)-3-(*tert*-butoxy)-1-oxobutan-2-yl)carbamate (23). To a solution of 9b (0.17 g, 0.35 mmol) and Cs₂CO₃ (0.135 g, 0.413 mmol) in DMF (3 mL) was added iodomethane (0.15 g, 3.0 equiv). The reaction mixture was heated to 65 °C for 12 h under argon. The reaction mixture was added to water and sonicated. The solids were filtered and collected as the desired product (0.095 g, 54.3%). ¹H NMR (300 MHz, DMSO): δ 8.09 (s, 1H), 7.96 (d, *J* = 7.8 Hz, 2H), 7.60 (t, *J* = 8.0 Hz, 1H), 6.42 (d, *J* = 8.3 Hz, 1H), 4.67 (m, 1H), 3.92 (m, 1H), 3.36 (s, 3H), 1.35 (s, 9H), 1.08 (s, 9H), 0.98 (d, *J* = 6.0 Hz, 3H).

(2S,3R)-2-Amino-3-hydroxy-N-methyl-N-((3-(1-oxoisoindolin-5-yl)phenyl)sulfonyl)butanamide (24). Using the same procedure as for the synthesis of 11a, intermediate 23 produced tert-butyl ((2S,3R)-3-(tert-butoxy)-1-(N-methyl-3-(1-oxoisoindolin-5vl)benzenesulfonamido)-1-oxobutan-2-vl)carbamate in 55.4% yield. ¹H NMR (300 MHz, DMSO): δ 8.60 (s, 1H), 8.19 (s, 1H), 8.11 (d, J = 7.8 Hz, 1H), 8.01 (d, J = 7.9 Hz, 1H), 7.94 (s, 1H), 7.85 (d, J = 8.3 Hz, 1H), 7.79 (m, 2H), 6.29 (d, J = 8.7 Hz, 1H), 4.82 (m, 1H), 4.46 (s, 2H), 3.98 (m, 1H), 3.33 (s, 3H), 1.30 (s, 9H), 1.06 (s, 9H), 1.02 (d, J = 6.0 Hz, 3H). This intermediate was treated with 4.0 N HCl in dioxane at room temperature to afford product 24 in 46.8% yield. ¹H NMR (600 MHz, DMSO): δ 8.67 (s, 1H), 8.23 (s, 1H), 8.13 (d, J = 7.4 Hz, 1H), 8.10 (d, J = 7.1 Hz, 1H), 7.95 (s, 1H), 7.82 (m, 3H), 5.76 (br, 1H), 4.61 (s, 1H), 4.45 (s, 2H), 4.13 (m, 1H), 3.33 (s, 3H), 1.14 (d, J = 6.5 Hz, 3H). HRMS (ESI): m/z calcd for $C_{19}H_{21}N_3O_5S$ (M + H)⁺ 404.1280, found 404.1286.

(25,3*R*)-2-Amino-*N*-(((*E*)-3-(6-aminopyrimidin-4-yl)styryl)sulfonyl)-3-hydroxypentanamide (26). Using the same procedure as for the synthesis of 10h, the title compound was produced in 15% yield in two steps using 25 and (2*S*,3*R*)-2-((*tert*-butoxycarbonyl)amino)-3-((*tert*-butyldimethylsilyl)oxy)pentanoic acid (Annova Chem). ¹H NMR (300 MHz, DMSO): δ 8.40 (s, 1H), 8.16 (s, 1H), 7.98 (d, *J* = 6 Hz, 1H), 7.56 (d, *J* = 6.0 Hz, 1H), 7.48 (m, 1H), 7.36 (d, *J* = 15 Hz, 1H), 7.18 (d, *J* = 15.6 Hz, 1H), 7.09 (s, 1H), 4.17 (m, 1H), 3.75 (d, *J* = 4.5 Hz, 1H), 1.14 (m, 2H), 0.72 (t, *J* = 7.4 Hz, 3H). HRMS (ESI): *m*/*z* calcd for C₁₇H₂₁N₅O₄S (M + H)⁺ 392.1392, found 392.1402.

(25,3*R*)-2-Amino-*N*-(((*E*)-3-(6-aminopyrimidin-4-yl)styryl)sulfonyl)-3-hydroxy-4-methylpentanamide (27). Using the same procedure as for the synthesis of 10h, the title compound was produced in 10% yield in two steps using 25 and (2*S*,3*R*)-2-((*tert*butoxycarbonyl)amino)-3-((*tert*-butyldimethylsilyl)oxy)-4-methylpentanoic acid.¹⁷ ¹H NMR (300 MHz, DMSO): δ 8.53 (s, 1H), 8.23 (s, 1H), 7.99 (d, *J* = 7.7 Hz, 1H), 7.80 (d, *J* = 7.1 Hz, 1H), 7.58 (t, *J* = 7.8 Hz, 1H), 7.45 (d, *J* = 14.0 Hz, 1H), 7.37 (m, 2H), 7.00 (s, 1H), 5.78 (d, *J* = 5.3 Hz, 1H), 3.94 (br, 1H), 3.58 (m, 1H), 1.72 (m, 1H), 0.93 (d, *J* = 5.8 Hz, 3H), 0.87 (d, *J* = 6.6 Hz, 3H). HRMS (ESI): *m/z* calcd for C₁₈H₂₃N₅O₄S (M + H)⁺ 406.1549, found 406.1546. (25,3*R*)-*N*-(((*E*)-3-(6-Aminopyrimidin-4-yl)styryl)sulfonyl)-2,3dihydroxybutanamide (28). Using the same procedure as for the synthesis of 10h, the title compound was produced in 19% yield in two steps using 25 and (2*S*,3*R*)-2,3-*O*-isopropylidene-2,3-dihydroxybutyric acid.¹⁸ ¹H NMR (300 MHz, DMSO): δ 8.63 (s, 1H), 8.29 (s, 1H), 8.00 (d, *J* = 7.7 Hz, 1H), 7.94 (d, *J* = 7.9 Hz, 1H), 7.70 (d, *J* = 15 Hz, 1H), 7.65 (d, *J* = 6.0 Hz, 1H), 7.53 (d, *J* = 15 Hz, 1H), 7.00 (s, 1H), 3.84 (m, 2H), 1.06 (d, *J* = 6.0 Hz, 3H). HRMS (ESI): *m*/*z* calcd for C₁₆H₁₈N₄O₅S (M + H)⁺ 379.1076, found 379.1088.

(2*S*,3*R*)-2,3-Diamino-*N*-(((*E*)-3-(6-aminopyrimidin-4-yl)styryl)sulfonyl)butanamide (29). Using the same procedure as for the synthesis of 10h, the title compound was produced in 15% yield in two steps using 25 and (2*S*,3*R*)-2,3-bis((*tert*-butoxycarbonyl)amino)butanoic acid.¹⁹ ¹H NMR (600 MHz, DMSO): δ 8.70 (s, 1H), 8.43 (s, 1H), 8.34 (s, 1H), 8.14 (s, 1H), 7.88 (d, *J* = 7.4 Hz, 1H), 7.81 (d, *J* = 7.1 Hz, 1H), 7.62 (t, *J* = 7.8 Hz, 1H), 7.46 (d, *J* = 18 Hz, 1H), 7.35 (d, *J* = 18 Hz, 1H), 7.35 (s, 1H), 7.26 (s, 1H), 7.18 (s, 1H), 7.09 (s, 1H), 7.05 (s, 1H), 3.81 (m, 1H), 3.59 (m, 1H), 1.18 (d, *J* = 6.5 Hz, 1H). HRMS (ESI): *m*/*z* calcd for C₁₆H₂₀N₆O₃S (M + H)⁺ 377.1396, found 377.1414.

Enzyme Assays. The tRNA synthetases catalyze a two-step reaction. The first step is the ATP-dependent amino acid activation, resulting in the formation of an amino acid adenylate. In the second step, this activated amino acid is transferred to the relevant tRNA. The enzymatic activity of ThrRS was determined using a coupled spectrophotometric assay in which the stoichiometric release of pyrophosphate during the ATP-dependent amino acid adenylation step was followed. Serine was used as a surrogate amino acid substrate due to a significantly higher rate of turnover due to the increased rate of seryl adenylate hydrolysis.²⁰

In ThrRS, the amino acid adenylation reaction proceeds via a random sequential mechanism in which either ATP or amino acid binds first to the enzyme, forming a binary complex, followed by binding of the second substrate to form a ternary enzyme–ATP– amino acid complex.²¹ The reader is referred to the numerous texts which describe techniques used to obtain the key kinetic parameters required for the determination of inhibition binding constants.²²

All reagents were obtained from Sigma-Aldrich unless indicated. Assays, typically 100 μ L, consisted of 60 mM Tris, pH 7.6, 10 mM MgCl₂, 20 mM KCl, 200 µM MESG (Berry and Associates), 4 mM ATP, 200 mM L-serine, 0.5 mM DTT, 0.1 U of inorganic pyrophosphatase, and 0.001 U of purine nucleoside phosphorylase. DMSO concentrations were maintained at 2% (v/v) in all reactions. Due to weak potency, the conditions used for compound 4 were modified and contained 25 mM L-serine, 0.5 mM ATP, and 20 nM enzyme. An enzyme concentration of 5 nM was used for all orthologues. Data were collected using a Tecan Safire II microplate reader. Inhibition data were normalized to controls lacking inhibitor (maximum activity) and no enzyme (blank). Typically, IC₅₀ values were obtained using a 16-point, 2-fold dilution series, from 50 μ M to 1.5 nM, analyzed using a 4-parameter sigmoidal log/response plot with variable slope and maximum/minimum constraint. K_i values were calculated for a bisubstrate enzyme as described by Cheng and Prusoff.

Bacterial Strains and Culture Conditions. *E. coli imp* (BAS849; permeability mutant in the MCR106 background with Imp deletion Thr332–Tyr354),²³ *E. coli tolC* (efflux mutant in the BW25113 background with a TolC deletion from Leu13–Ala480) and wild type (ATCC 25922), and *B. thailandensis* strain E264²⁴ and the isogenic $\Delta amrAB$ -oprA $\Delta bpeAB$ -oprB double knockout efflux mutant strain (KO) (H. P. Schweizer, Colorado State University) were cultured on Mueller–Hinton II cation adjusted agar (MHA) or in MH broth (MHB; BD catalog no. 212322) at 37 °C. *H. influenzae* wild type (ATCC 31517) was cultured on chocolate agar plates or in *Haemophilus* test medium broth (Remel catalog no. R112380) at 37 °C in the presence of 5% CO₂.

Minimum Inhibitory Concentration (MIC) Assays. Broth microdilution MIC assays were performed according to CLSI guidelines with the exception that compound stocks were made up at $50\times$ and the end point detection was determined colorimetrically via detection with alamarBlue (Invitrogen, catalog no. DAL1100).²⁵

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Stock solutions of all test compounds were prepared in 100% dimethyl sulfoxide (DMSO) at 6.4 mg/mL and then serially diluted in two steps. Two microliters from each dilution series was added to 98 μ L of inoculated media, resulting in final test concentrations of 0.063–128 μ g/mL. Serial dilutions of the starting inocula were plated to ensure that the assay contained roughly the desired 5 × 10⁵ CFU/mL. MIC plates were incubated overnight at 37 °C in the appropriate conditions. The following day, 10 μ L of alamarBlue was added to each well. The plates were shaken and incubated for another 1 h, and then the MICs were determined as the lowest concentration of compound where no color change was visible.

Protein Purification and Cocrystallization Conditions. The proteins were purified by metal chelate affinity chromatography, exploiting a vector-derived C-terminal $6 \times$ His tag. Compounds were cocrystallized by hanging drop vapor diffusion using a mother liquor containing 12% (w/v) PEG 4000, 21% (v/v) MPD, and 0.1 M sodium citrate, pH 5.9. The resulting crystals were of space group $P2_12_12_1$ and typically diffracted to 1.8 Å or better on a rotating anode X-ray source. Crystals of the human enzyme (cytoplasmic isoform 1) were generated with an analogous construct and purification scheme.

ASSOCIATED CONTENT

Supporting Information

Full description of the crystallographic methods, protein purification methods, enzymatic assay, and MIC conditions. This material is available free of charge via the Internet at http://pubs.acs.org.

Accession Codes

Coordinates for the cocrystal structures of compounds 10a,c,d and 11d bound to *E. coli* ThrRS and compound 11d bound to the human ThrRS have been deposited in the Protein Data Bank (http://www.pdb.org) under accession codes 4HWO, 4HWS, 4HWP, 4HWR, and 4HWT, respectively.

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Notes

The authors declare no competing financial interest.

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ABBREVIATIONS USED

B. thailandensis (KO), *B. thailandensis* pump knockout; ThrRS, threonyl-tRNA synthetase; aa-AMP, aminoacyl adenosine monophosphate; AMS, 5'-O-[N-(threonyl)sulfamoyl]-adenosine; TEA, triethylamine; SBDD, structure-based drug design; ThrRS, threonyl-tRNA synthetase

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(16) Compound 11d was cocrystallized with human ThrRS in space group $P4_12_12$ from a mother liquor containing 14% (w/v) PEG 3350, 0.2 M sodium nitrate, 0.5 M sodium chloride, and 0.1 M bis-tris propane, pH 6.1. The crystals of this complex diffracted to 2.3 Å with synchrotron radiation. Structures were solved by molecular replacement.

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