

Studies in Marine Macrolide Synthesis: Stereocontrolled Synthesis of the C₁–C₁₁ and C₁₅–C₂₇ Subunits of Aplyronine A

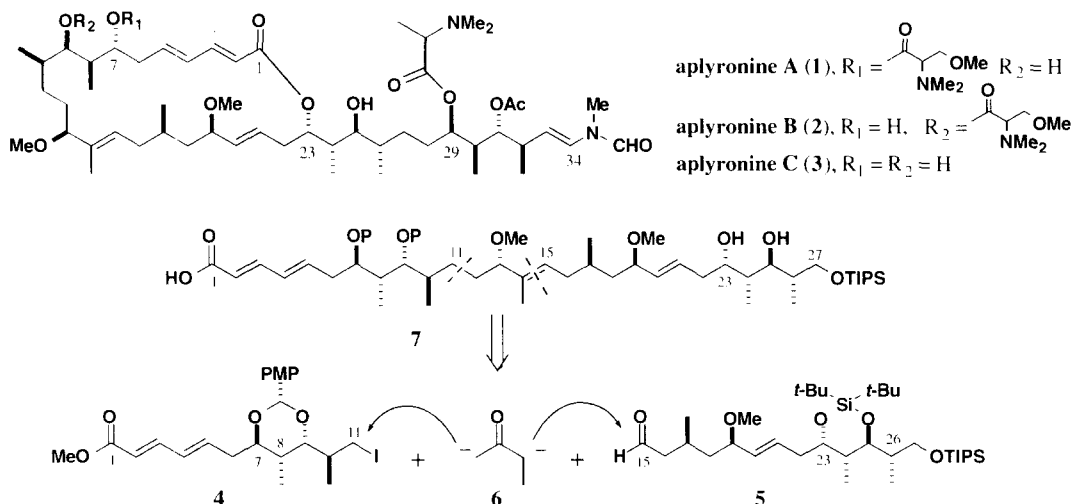
Ian Paterson,* Cameron J. Cowden and Michael D. Woodrow

University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, UK.

Received 21 May 1998; accepted 8 June 1998

Abstract: The aplyronine C₁–C₁₁ subunit **4**, containing 4 stereocentres and the (*E,E*)-diene system, was prepared in 7 steps from ethyl ketone (*R*)-**8** using a boron-mediated *anti* aldol reaction. The corresponding C₁₅–C₂₇ subunit **5**, containing 6 stereogenic centres and an (*E*)-alkene, was obtained in 10 steps from ketone (*S*)-**14** using a tin(II)-mediated *syn* aldol reaction and CBS enone reduction.
© 1998 Elsevier Science Ltd. All rights reserved.

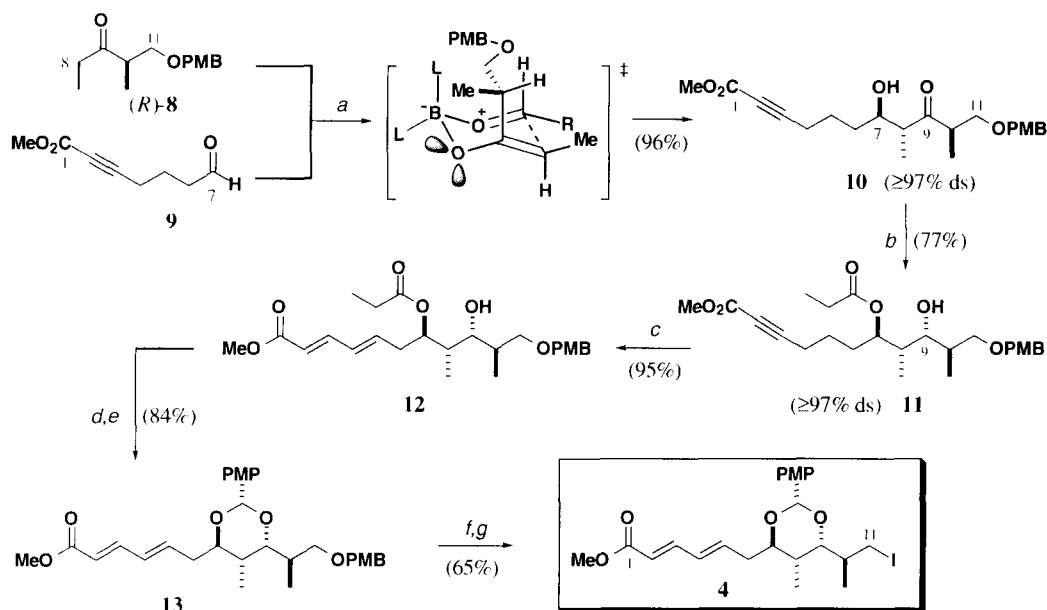
In 1993, Yamada and co-workers reported the isolation and characterisation of aplyronines A (**1**), B (**2**) and C (**3**) from the Japanese sea hare *Aplysia kurodai*.¹ Aplyronine A, which showed potent cytotoxicity against HeLa-S₃ cells (IC₅₀ 0.039 ng/mL) and pronounced activity *in vivo* against a range of tumours,¹ is a complex 24-membered macrolide with distinctive amino acid residues at C₇ and C₂₉ along with an elaborate C₂₃ side-chain, terminating in a vinyl *N*-methyl formamide group. More recently, the Yamada group confirmed the absolute stereochemistry of the aplyronines by total synthesis.²



Scheme 1

As part of our studies towards the total synthesis of this novel class of bioactive marine macrolides,³ we now report a stereocontrolled synthesis of the aplyronine C₁–C₁₁ and C₁₅–C₂₇ subunits, **4** and **5** in Scheme 1, using aldol chemistry developed in our laboratory. A key feature of the synthesis of the iodide **4** was the temporary masking of the C₁–C₅ (*E,E*)-diene ester as an alkyne ester, facilitating the use of a boron-mediated *anti* aldol coupling for the installation of the C₇ and C₈ stereocentres. In the case of the aldehyde **5**, the construction of the C₂₃–C₂₆ stereotetrad was based on the use of a tin(II)-mediated *syn* aldol reaction. In the accompanying paper,⁴ we describe the efficient coupling of these subunits through an appropriate C₁₂–C₁₄ linker **6** to generate the truncated seco acid **7**, followed by its transformation into the desired 24-membered macrolide framework of the aplyronines.

The synthesis of the C₁–C₁₁ subunit **4** is outlined in **Scheme 2**. Using our standard conditions for the generation of the (*E*)-enol borinate,⁵ enolisation of ethyl ketone (*R*)-**8**⁶ was followed by addition of aldehyde **9**,⁷ leading to formation of the *anti-anti* aldol adduct **10** in 96% yield with $\geq 97\%$ ds.⁸ Notably, the acetylenic ester was carried through this reaction without difficulty. Stereoselective reduction of the C₉ carbonyl of **10** was achieved using a modified, samarium-catalysed, Evans-Tishchenko reaction.⁹ Hence, treatment of **10** with a premixed solution of SmI₂ (15 mol%) and EtCHO gave the 1,3-*anti* reduction product **11** in 77% yield with $\geq 97\%$ ds.¹⁰ After some experimentation, it was found that the isomerisation¹¹ of the alkyne ester to the desired (*E,E*)-diene was best achieved at this stage. Use of Ph₃P in conjunction with PhOH smoothly equilibrated the alkyne in **11** to diene **12** (95%), isolated as a single geometric isomer. Next, ester cleavage (K₂CO₃ / MeOH) in **12** gave a diol which was protected as its *p*-methoxyphenyl (PMP) acetal **13**.¹² Selective deprotection¹³ of the PMB ether of **13** with DDQ was followed by conversion of the primary alcohol into the corresponding iodide **4**.

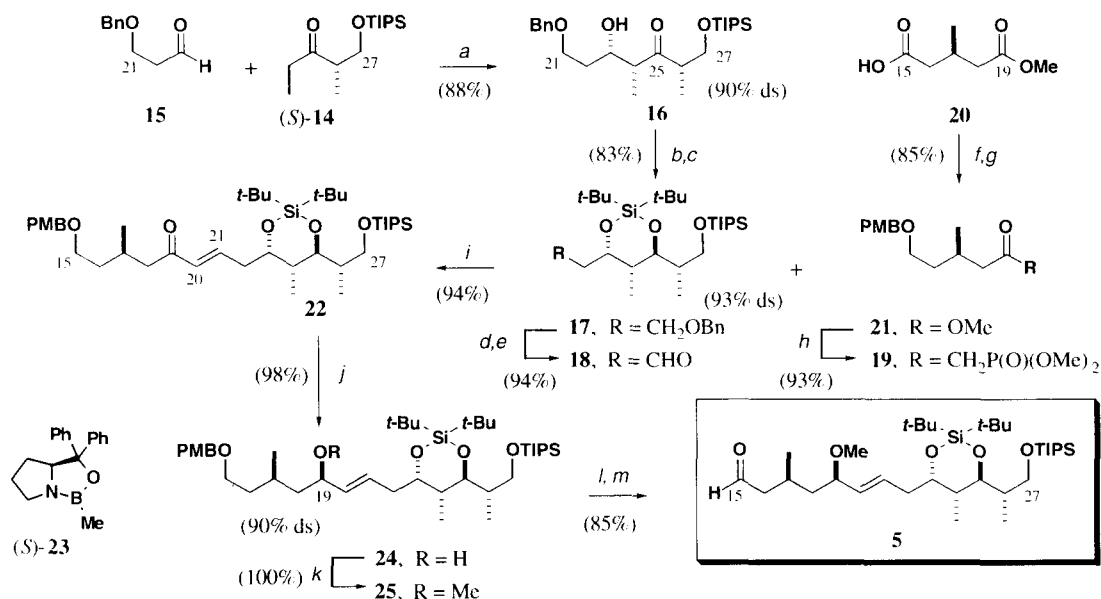


Scheme 2: (a) (*c*-Hex)₂BCl, Et₃N, Et₂O, 0 °C, 1 h; **9**, -78 → -20 °C, 12 h; H₂O₂, MeOH, pH7 buffer; (b) SmI₂ (15 mol%), EtCHO, THF, 0 °C, 15 min; **10**, 0 °C, 2 h; (c) Ph₃P, PhOH, benzene, 20 °C, 14 h; (d) K₂CO₃, MeOH, 20 °C, 2 h; (e) *p*-MeO(C₆H₄)CH(OMe)₂, PPTS, CH₂Cl₂, 20 °C, 14 h; (f) DDQ, CH₂Cl₂, pH7 buffer, 20 °C, 1 h; (g) I₂, PPh₃, imid, MeCN, Et₂O, 0 → 20 °C, 3 h.

As shown in **Scheme 3**, the C₂₃–C₂₆ stereotetrad was generated by a Sn(OTf)₂ mediated *syn*-aldol coupling¹⁴ of TIPS ether protected ketone (*S*)-**14**¹⁵ with aldehyde **15**, which gave the adduct **16** in 88% yield with 90% ds. This was followed by a Me₄NBH(OAc)₃ reduction¹⁶ of the C₂₅ ketone to generate the 1,3-*anti* diol which was subsequently protected as the di-*tert*-butyl silylene **17** (83%). Benzyl ether hydrogenolysis and Dess-Martin periodinane oxidation¹⁷ gave the aldehyde **18** (94%) in preparation for a HWE chain extension. The synthesis of the required ketophosphonate **19** started with commercially available (*R*)-methyl-3-methyl glutarate (**20**). Chemoselective reduction of the carboxylic acid (BH₃•Me₂S, THF)¹⁸ was immediately followed by hydroxyl protection (PMBOC(=NH)CCl₃, 0.3 mol% TfOH)¹⁹ to give the PMB ether **21**. Under these conditions, lactonisation of the hydroxy ester was not observed. Chain extension by condensation with lithiated dimethyl methylphosphonate²⁰ gave the C₁₅–C₂₀ segment **19** in 79% yield (3 steps).

The HWE coupling of phosphonate **19** with aldehyde **18** was best performed using Ba(OH)₂ as a mild base.²¹ This gave the (*E*)-enone **22** selectively in 94% yield. 1,2-Reduction of the enone was now required

and, not surprisingly, achiral reagents gave an *ca* 1:1 mixture of epimeric alcohols. However, a good level of reagent control was achieved using Corey's proline-derived oxazaborolidine.²² Treatment of **22** with (*S*)-**23** (10 mol%) in THF solution with BH₃•Me₂S (0.6 equiv.) gave a 98% yield of C₁₉ alcohols with 9:1 diastereoselectivity. Assignment of the configuration of the epimeric alcohols was made by Mosher ester analysis²⁴ and was in agreement with the anticipated sense of stereoinduction.²⁵ Methylation of the chromatographically separable (1*R*)-alcohol **24** gave the ether **25** and oxidative PMB removal (DDQ),²⁶ followed by oxidation, gave the aldehyde **5** (85%, 3 steps) representing the complete C₁₅–C₂₇ subunit.



Scheme 3: (a) Sn(OTf)₂, Et₃N, CH₂Cl₂; **15**, -78 °C, 2 h; (b) Me₄NBH(OAc)₃, AcOH, CH₃CN, 20 °C, 48 h; (c) (*t*-Bu)₂Si(OTf)₂, 2,6-lutidine, CH₂Cl₂, 20 °C, 12 h; (d) H₂, Pd/C, EtOAc, 20 °C, 11 h; (e) Dess-Martin periodinane, CH₂Cl₂, 20 °C, 70 min; (f) BH₃•Me₂S, THF, 0 → 20 °C, 90 min; (g) PMBOC(NH)CCl₃, TiOH (0.3 mol%), Et₂O, 20 °C, 14 h; (h) MeP(O)(OMe)₂, *n*-BuLi, THF, -78 °C, 10 min; **21**, -78 °C, 1.5 h; (i) **19**, Ba(OH)₂, THF:H₂O (40:1), 20 °C, 30 min; **18**, 20 °C, 2 h; (j) (*S*)-**23**, BH₃•Me₂S, THF, 0 °C, 40 min; (k) NaH, MeI, THF, 0 → 20 °C, 15 h; (l) DDQ, pH7 buffer, CH₂Cl₂, H₂O, 20 °C, 1 h; (m) Dess-Martin periodinane, CH₂Cl₂, 20 °C, 2 h.

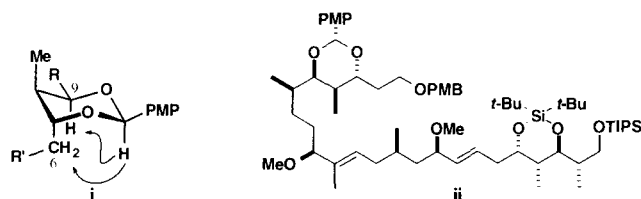
In conclusion, the C₁–C₁₁ subunit **4**, containing 4 stereocentres and the (*E,E*)-diene system, was prepared in 7 steps from ethyl ketone (*R*)-**8** in 38% yield with ≥94% ds. The corresponding C₁₅–C₂₇ subunit **5**, containing 6 stereogenic centres and an (*E*)-alkene, was obtained in 10 linear steps from ethyl ketone (*S*)-**14**, with an overall yield of 53% and 75% ds. The elaboration of these two subunits into an advanced macrolide intermediate for the aplyronines is discussed in the accompanying paper.⁴

Acknowledgement: We thank the EPSRC (GR/K54052) and Merck Sharp & Dohme for support.

References and Notes

- Yamada, K.; Ojika, M.; Ishigaki, T.; Yoshida, Y.; Ekimoto, H.; Arakawa, M. *J. Am. Chem. Soc.* **1993**, *115*, 11020.
- (a) Kigoshi, H.; Ojika, M.; Ishigaki, T.; Suenaga, K.; Mutou, T.; Sakakura, A.; Ogawa, T.; Yamada, K. *J. Am. Chem. Soc.* **1994**, *116*, 7443. (b) Kigoshi, H.; Suenaga, K.; Mutou, T.; Ishigaki, T.; Atsumi, T.; Ishiwata, H.; Sakakura, A.; Ogawa, T.; Ojika, M.; Yamada, K. *J. Org. Chem.* **1996**, *61*, 5326. (c) Suenaga, K.; Ishigaki, T.; Sakakura, A.; Kigoshi, H.; Yamada, K. *Tetrahedron Lett.* **1995**, *36*, 5053.
- Norcross, R. D.; Paterson, I. *Chem. Rev.* **1995**, *95*, 2041.
- Paterson, I.; Woodrow, M. D.; Cowden, C. J. *Tetrahedron Lett.* **1998**, *38*, 6041.

5. (a) Paterson, I.; Goodman, J. M.; Isaka, M. *Tetrahedron Lett.* **1989**, 30, 7121. (b) Paterson, I.; Norcross, R. D.; Ward, R. A.; Romea, P.; Lister, M. A. *J. Am. Chem. Soc.* **1994**, 116, 11287. (c) For a review of boron-mediated aldol reactions, see: Cowden, C. J.; Paterson, I. *Org. React.* **1997**, 51, 1.
6. Ketone (*R*)-**8** was prepared in 3 steps from methyl (*R*)-3-hydroxy-2-methylpropionate with the PMB protecting group introduced using PMBC(=NH)CCl₃/TfOH (ref. 5, 19).
7. Aldehyde **9** was prepared by SO₃•pyr oxidation of the corresponding alcohol, see: (a) Tufariello, J. J.; Trybulski, E. J. *J. Org. Chem.* **1974**, 39, 3378. (b) Kita, Y.; Okunaka, R.; Honda, T.; Shindo, M.; Taniguchi, M.; Kondo, M.; Sasho, M. *J. Org. Chem.* **1991**, 56, 119.
8. All new compounds gave spectroscopic data in agreement with the assigned structures. Aldehyde **5** had: ¹H NMR δ (500 MHz, CDCl₃) 9.83 (1H, s, CHO), 5.81 (1H, dt, *J* = 15.5, 6.6 Hz, H₂₁), 5.32 (1H, dd, *J* = 15.5, 8.3 Hz, H₂₀), 4.05 (1H, m, H₂₃), 3.90 (1H, dd, *J* = 9.6, 5.8 Hz, H_{27A}), 3.80 (1H, dd, *J* = 9.1, 2.2 Hz, H₂₅), 3.59-3.53 (2H, m, H₁₉ and H_{27B}), 3.23 (3H, s, OMe), 2.44 (1H, qd, *J* = 7.2, 1.9 Hz, H₁₆), 2.37-2.29 (2H, m, H₂₄ and H_{22A}), 2.26-2.17 (3H, m, H_{22B}, H_{16B} and H₁₇), 1.89 (1H, m, H₂₆), 1.62-1.56 (1H, m, H_{18A}), 1.43 (1H, dt, *J* = 13.8, 6.3 Hz, H_{18B}), 1.10 (3H, septet, *J* = 5.0 Hz, OSi(CHMe₂)₃), 1.06 (18H, d, *J* = 5.0 Hz, OSi(CHMe₂)₃), 1.03 (3H, d, *J* = 6.8 Hz, C₂₆Me), 1.03 (3H, d, *J* = 7.0 Hz, C₁₇Me), 0.99 (18H, s, Si(CMe₃)₂), 0.85 (3H, d, *J* = 7.2 Hz, C₂₄Me); ¹³C NMR δ (62.9 MHz, CDCl₃) 202.8, 132.6, 131.6, 80.3, 76.7, 76.6, 63.6, 55.6, 50.9, 42.7, 39.0, 38.8, 34.1, 27.5, 27.3, 24.8, 21.6, 20.9, 20.4, 18.1, 15.2, 13.8, 11.9.
9. Evans, D. A.; Hoveyda, A. H. *J. Am. Chem. Soc.* **1990**, 112, 6447.
10. While the Me₄NBH(OAc)₃ reduction (ref. 16) of **10** was successful, the subsequent isomerisation of the alkyne ester proved problematic in the presence of a free hydroxyl group at C₇.
11. (a) Rychnovsky, S. D.; Kim, J. *J. Org. Chem.* **1994**, 59, 2659. (b) Trost, B. M.; Kazmaier, U. *J. Am. Chem. Soc.* **1992**, 114, 7933. See also Kazmaier, U. *Tetrahedron* **1998**, 54, 1491.
12. The PMP acetal configuration in **4** and **13** was assigned based upon the nOe's indicated for i.



13. Selective removal of a PMB ether in the presence of a PMP acetal has been reported previously. For example, see: Smith, A. B. III; Qiu, Y.; Jones, D. R.; Kobayashi, K. *J. Am. Chem. Soc.* **1995**, 117, 12011. In contrast, our initial route to the aplyronines was revised because of the inability to deprotect the PMB ether in **ii** without reaction at the PMP acetal.
14. Paterson, I.; Tillyer, R. D. *Tetrahedron Lett.* **1992**, 33, 4233.
15. Paterson, I.; Franklin, A. S. *Tetrahedron Lett.* **1994**, 35, 6925.
16. Evans, D. A.; Chapman, K. T.; Carreira, E. M. *J. Am. Chem. Soc.* **1988**, 110, 3560.
17. Dess, D. B.; Martin, J. C. *J. Am. Chem. Soc.* **1991**, 113, 7277.
18. Herold, P.; Mohr, P.; Tamm, C. *Helv. Chim. Acta* **1983**, 66, 744.
19. Nakajima, N.; Horita, K.; Abe, R.; Yonemitsu, O. *Tetrahedron Lett.* **1988**, 29, 4139.
20. Corey, E. J.; Kwaitowski, G. T. *J. Am. Chem. Soc.* **1966**, 88, 5654.
21. Paterson, I.; Yeung, K.-S.; Smaill, J. B. *Synlett* **1993**, 774.
22. (a) Corey, E. J.; Bakshi, R. K.; Shibata, S.; Chen, C.-P.; Singh, V. K. *J. Am. Chem. Soc.* **1987**, 109, 7925. (b) Jones, T. K.; Mohan, J. J.; Xavier, L. C.; Blacklock, T. J.; Mathre, D. J.; Sohar, P.; Jones, E. T. T.; Reamer, R. A.; Roberts, F. E.; Grabowski, E. J. J. *J. Org. Chem.* **1991**, 56, 763.
23. Use of the butyl oxazaborolidine and/or catecholborane as stoichiometric reductant did not improve the selectivity and lower yields resulted (see Corey, E. J.; Bakshi, R. K. *Tetrahedron Lett.* **1990**, 31, 611). Other asymmetric reducing agents such as those derived from LiAlH₄ also proved unsatisfactory.
24. Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. *J. Am. Chem. Soc.* **1991**, 113, 4092.
25. Precedent based on results in ref. 22(a) in conjunction with those of: (a) Bach, J.; Berenguer, R.; Garcia, J.; Vilarrasa, J. *Tetrahedron* **1995**, 36, 3425. (b) Bach, J.; Berenguer, R.; Farràs, J.; Garcia, J.; Meseguer, J.; Vilarrasa, J. *Tetrahedron: Asymmetry* **1995**, 6, 2683.
26. Horita, K.; Yoshioka, T.; Tanaka, T.; Oikawa, Y.; Yonemitsu, O. *Tetrahedron* **1986**, 42, 3021.