Table I.

$$R_1$$
 NH-CO-OH COOC₂H₄

No.	R	R_1	Mp, °C	Recrystal- lization solvent ^a	Formula	Analgetic activity (mice) b, c	Probability ^d P ≤
1	H	Н	159-160	В	C ₁₆ H ₁₅ NO ₄	37.1	<0.001
2	2C1	H	154-155	Α	C ₁₆ H ₁₄ ClNO ₄	52.5	< 0.001
3	3C1	H	143-144	Α	C ₁₆ H ₁₄ CINO ₄	56.4	≤0.001
4	4C1	H	195-196	Α	$C_{16}H_{14}CINO_4$	32	< 0.005
5	2C1	3C1	157-158	С	$C_{16}H_{13}Cl_2NO_4$	64.1	< 0.001
6	2C1	4C1	164-165	С	$C_{16}H_{13}Cl_2NO_4$	74.3	< 0.001
7	2C1	5Cl	203-204	С	$C_{16}H_{13}Cl_2NO_4$	70.5	< 0.001
8	2C1	6C1	200-201	C	$C_{16}H_{13}Cl_2NO_4$	62.8	< 0.01
9	3C1	4C1	194-195	С	$C_{16}H_{13}Cl_2NO_4$	61.5	< 0.01
10	3C1	5 Cl	185-186	С	$C_{16}H_{13}Cl_2NO_4$	67.9	< 0.001
11	2CH,	H	149-150	В	$C_{17}^{10}H_{17}^{13}NO_4$	55.1	< 0.001
12	3CH ₃	H	125-126	В	$C_{17}H_{17}NO_4$	51.2	< 0.001
13	4CH ₃	H	170-171	В	$C_{17}H_{17}NO_4$	37.1	< 0.05
14	2CH ₃	3CH ₃	167-168	С	$C_{18}H_{19}NO_4$	46.1	≤0.01
15	4OCH,	н	163-164	С	$C_{12}^{12}H_{12}^{12}NO_{5}^{7}$	44.8	< 0.001
16	4OC ₂ H ₅	Н	165-166	С	$C_{18}H_{19}NO_5$	47.4	< 0.001
17	2CF ₃	H	160-161	C B	$C_{17}H_{14}F_{3}NO_{4}$	42.3	<0.001
18	2CH,	5 Cl	181-182	В	$C_{17}H_{16}CINO_4$	46.1	< 0.05
19	2CH ₃	4C1	178-179	B B	$C_{17}H_{16}CINO_4$	41	< 0.001
20	2CH ₃	3Cl	188-189	В	$C_{17}H_{16}CINO_4$	41	< 0.01
4HHA					., .J -	48.7	<0.001

^aA, MeOH; B, i-PrOH; C, AcOH. ^bIncrease of reaction time % 3 hr after treatment. ^cDoses were of 30 mg/kg for each group of 10 mice. ^dThe hot plate test counts were analyzed statistically by means of the Student t test. P was compared to controls.

0.05 mole of substituted anilines. The reaction mixture was refluxed for 2 hr and then diluted with cold $\rm H_2O$, and the crystalline reaction product was filtered off. It was washed with 5% NaHCO₃ and recrystallized.

References

- G. B. Chesher, H. O. J. Collier, F. A. Robison, Taylor, S. E. Hunt, J. I. Jones, and A. S. Lindsey, *Nature (London)*, 175, 206 (1955).
- (2) P. A. I. Janssen and A. Jageneau, J. Pharm. Pharmacol., 9, 381 (1957).
- (3) J. M. Z. Gladych and E. P. Taylor, J. Chem. Soc., 4678 (1956).

Substituted Thiazolidones as Anticonvulsants†

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In continuation of our interest^{1,2} in thiazolidones, some new 2-arylimino-3-(3,4-dimethoxyphenethyl)thiazolid-4-ones have been synthesized and tested for their anticonvulsant activity against pentylenetetrazol-induced seizures in albino mice.

Anticonvulsant activity was detd² by injecting the thiazolidone ip in a 5% aqueous suspension of gum acacia in groups of 10 mice of either sex. Pentylenetetrazol (80 mg/kg) was injected 4 hr after the administration of thiazolidones and the mice were then observed for 60 min for the occurrence of seizures. Animals devoid of even a threshold convulsion were considered protected. Anticonvulsant activity shown by substituted thiazolidones at 100 mg/kg is given

Table I. Substituted Thiocarbamides

No.	Ar	Mp,a°C	Yield, %	Molecular formula ^b
1	C ₆ H ₅	125	85	$C_{17}H_{20}N_{2}O_{2}S$
2	o-CH ₃ C ₆ H ₄	112	65	$C_{18}H_{22}N_{2}O_{2}S$
3	m-CH ₃ C ₆ H ₄	122	78	$C_{18}H_{22}N_2O_2S$
4	p-CH ₃ C ₆ H ₄	92	85	$C_{18}H_{22}N_2O_2S$
5	$3,4-(CH_3)_2C_6H_3$	125	82	$C_{19}H_{24}N_{2}O_{2}S$
6	o-OCH ₃ C ₆ H ₄	108	62	$C_{18}H_{22}N_2O_3S$
7	p-OCH ₃ C ₆ H ₄	120	72	$C_{18}H_{22}N_2O_3S$
8	$p\text{-ClC}_6\text{H}_4$	114	80	$C_{17}H_{19}CIN_2O_2S$
9	p-BrC ₆ H ₄	135	80	$C_{17}H_{19}BrN_2O_2S$
10	α - $C_{10}H_7$	166	68	$C_{21}H_{22}N_2O_2S$

^aMelting points were taken in open capillary tubes. ^bAll compds were analyzed for C, H, and N and analyses were found within 0.4% of theory.

in Table II. Compd 2 having an o-tolyl group at position 2 afforded the maximum protection of 70%, while administration in doses above or below 100 mg/kg caused lesser anticonvulsant activity. The low toxicity of this compound was reflected by its approximate LD₅₀ (>2000 mg/kg).

Experimental Section

1-Aryl-3-(3,4-dimethoxyphenethyl)thiocarbamide. 3,4-Dimethoxyphenethylamine (0.01 mole) was mixed with a suitable aryl isothiocyanate (0.01 mole) in 15 ml of dry PhH and was refluxed on a steam bath for 2 hr. The reaction mixt was concd under reduced pressure. The solid mass which sepd on cooling was filtered, washed (Et₂O, dil HCl), dried, and recrystd from EtOH. All thiocarbamides were characterized by their sharp melting points and elemental analyses (Table I).

2-Arylimino-3-(3,4-dimethoxyphenethyl)thiazolid-4-ones. A mixt of 1-aryl-3-(3,4-dimethoxyphenethyl)thiocarbamide (0.01 mole), CICH₂COOH (0.01 mole), and anhyd NaOAc (0.015 mole) in 15 ml of glacial AcOH was refluxed for 5-6 hr. The reaction mixt was poured into H₂O and refrigerated overnight. The sepd crude product was filtered, washed several times (H₂O), and recrystd from EtOH (Table II).

[†]This investigation was supported in part with the financial assistance obtained from the Indian Council of Medical Research, New Delhi.

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No.	Ar	Mp, ^a °C	Yield, %	Molecular formulab	Protection, %	Mortality after 24 hr, %
 1	C ₆ H ₅	117	62	$C_{19}H_{20}N_{2}O_{3}S$	30	60
2	o-CH ₃ C ₆ H ₄	126	55	$C_{20}^{19}H_{22}^{20}N_{2}O_{3}^{2}S$	70	20
3	m -C H_3 C_6 H_4	118	60	$C_{20}^{20}H_{22}^{22}N_{2}O_{3}^{2}S$	30	40
4	p -CH ₃ $\overset{\circ}{C}_6\overset{\circ}{H}_4$	160	64	$C_{20}^{2}H_{22}N_{2}O_{3}S$	40	50
5	$3,4-(CH_3)_2C_6H_3$	175	62	$C_{21}H_{24}N_2O_3S$	10	70
6	o-OCH₃C ₆ H₄	90	54	$C_{20}H_{22}N_2O_4S$	50	30
7	p-OCH ₃ C ₆ H ₄	147	62	$C_{20}H_{22}N_2O_4S$	10	60
8	p-ClC ₆ H ₄	150	60	$C_{19}H_{19}CIN_2O_3S$	60	50
9	p-BrC ₆ H ₄	153	62	$C_{19}H_{19}BrN_2O_3S$	30	60
10	α-C. H.	128	58	C,,H,,N,O,S	50	40

a, b See footnotes to Table I.

Acknowledgments. The authors wish to express their thanks to Professor K. P. Bhargava and Dr. J. P. Barthwal for their advice and encouragement and to Dr. M. L. Dhar and Dr. Nitya Anand of the Central Drug Research Institute, Lucknow, for providing microanalysis facilities. Grateful acknowledgment is made to Dr. Edwin E. Hays, Gruppo Lepetit, Northridge, Calif., for a generous gift of research chemicals.

References

- R. Kumar, T. K. Gupta, and S. S. Parmar, J. Pract. Chem., 312, 201 (1970).
- (2) S. S. Parmar, C. Dwivedi, A. Chaudhari, and T. K. Gupta, J. Med. Chem., 15, 99 (1972).

Synthesis of N'-Substituted Arylsulfonylpyrazoles, Their Anthelmintic Activity, and the Cytotoxicity of Some Hydrazides†

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Our continued interest in the synthesis of biological active heterocycles has led us to study the synthesis and anthelmintic activity of N'-substituted arylsulfonyl-3,5-dimethyl-4-arylazopyrazoles. These compounds displayed anthelmintic and cytotoxicity activities of different magnitudes. All are apparently nontoxic to mice at the dosages used.

Experimental Section

Melting points, taken with a Kofler hot-stage apparatus, are uncorr. Where analyses are indicated only by symbols of the elements, analytical result obtd for those elements were within $\pm 0.4\%$ of the calcd values.

2,3,4-Pentanetrione-3-arylhydrazons, ¹ cinnamic acids, and hydrazides, ² 3-nitro-4-methoxybenzenesulfonylhydrazide, ³ 3-chloro-4-methoxybenzenesylfonylhydrazide, ³ and 2,5-dichlorobenzenesulfonylhydrazide⁴ were prepd by standard procedures.

2-Methoxy-3,5-dimethyl- and 2-Chloro-5-carboxybenzenesulfonyl Hydrazide. A soln of 2-methoxy-3,5-dimethyl- and 2-chloro-5-carboxybenzenesulfonyl chloride in EtOH was treated with $NH_2NH_2-H_2O$ (98%) at 0°. It was left at room temp for several hr, when

Table I. N¹-Arylsulfonyl-3,5-dimethyl-4-arylazopyrazoles

No.	R ₁	R ₂	Yield %	, Mp, °C	Colora	Formula ^b
	F	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\			_	
	<	_N	=N-0	CC-M	e	
			Me	C N	,R	1
				SO ₂ -(3> 1	
1	3-NO ₂ -4-OMe	2-C1	65	168-169	Ly	$C_{18}H_{16}CIN_5O_5S$
2	3-NO ₂ -4-OMe	4-OMe	95	194-195	y	$C_{19}H_{19}N_{5}O_{6}S$
3 4	3-NO ₂ -4-OMe 3-Cl-4-OMe	4-NO ₂ 2-Cl	80 65	167-168 150-151	O Y	C ₁₈ H ₁₆ N ₆ O ₇ S C ₁₈ H ₁₆ Cl ₂ N ₄ O ₃ S
5	3-Cl-4-OMe	2-O1 2-NO ₂	76	224-225	R	$C_{18}H_{16}CIN_{5}O_{5}S$
6	3-Cl-4-OMe	4-OMe	70	161-162	Y	$C_{19}H_{19}CIN_4O_4S$
7	2-OMe-5-Cl	$2-NO_2$	80	200-201	R	$C_{18}H_{16}CIN_5O_5S$
8	2,5-Cl	2-NO ₂	90	190-191	DBn	$C_{17}H_{13}Cl_{2}N_{5}O_{4}S$
9	2-Cl-5-COOH	2-NO ₂	96	220-221	BR	C ₁₈ H ₁₄ ClN ₅ O ₆ S
10	2-OMe-3,5-Me	4-OMe	96	154-155	Py	$C_{21}H_{24}N_4O_4S$

 a B, brick; Bn, brown; D, dark; L, light; O, orange; P, pale; R, red; Y, yellow. b All compds were analyzed for C, H, N, S.

Table II. Biological Activities of N^1 -Arylsulfonylpyrazoles

	%	act	ivit	y at i	highest teste	ed dosage ^b		
	In Vivo				In V	7itro		
a	Mice		Manure		was	_		
No.a	Tg	Ν	C	O	R/Lv/Ad	Hc/Ts	\mathbf{F}	Dose, ppm
1		0	0	60	75/0/0	100/100	0	100
2	0	0	0	0				400 mg/kg
3	0	0	0	0	50/0/0	100/100	0	100 mg/kg
4		0		0	0/0/0	60/90	0	100
5	0	0	0	0	0/0/0	0/50	0	100, 400 mg/kg
6	0	0	0	50				400 mg/kg
7	0				0/0/0	50/50	0	100
8	0				0/0/0	0/50	0	100
9	0	0	0	50				400 mg/kg
10	0	0	0	0				100

^aSame as Table I. ^bTg, Toxoplasma gondii-RH strain in mice prevention (of mortality); N, nematodes (trichostrongyles in mice); C, cestoses (tapeworms in mice); O, oxyurids (in mice); R, % repellency (of face fly oviposition); Lv, % contact activity on face fly larvae (prevention of pupation); Ad, % kill of adult face flies and/or pupae which fail to hatch; Hc, % inhibition of Haemonchus contorus larvae development; Ts, % inhibition of Trichostorgylus spp. larvae development; F, % inhibition of fungus growth.

crystals of the hydrazide were obtd. Recrystn from EtOH gave a colorless product, mp $116-117^{\circ}$. 2-MeO-3,5-Me₂ deriv. Anal. (C₉H₁₄N₂O₃S) C, H, N. 2-Cl-5-CO₂H deriv, mp 87° . Anal. (C₇H₇ClN₂O₄S) C, H, N.

N'-Substituted Arylsulfonyl-3,5-dimethyl-4-arylazopyrazoles. A hot soln of arylsulfonylhydrazide (0.01 mole) in EtOH (30 ml)

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[†]This work was supported by a grant from The Chemical Society, London. Taken from the dissertation submitted by N. K. in partial fulfillment of the requirements for a Master of Science degree, University of Roorkee, Roorkee, India, 1971.