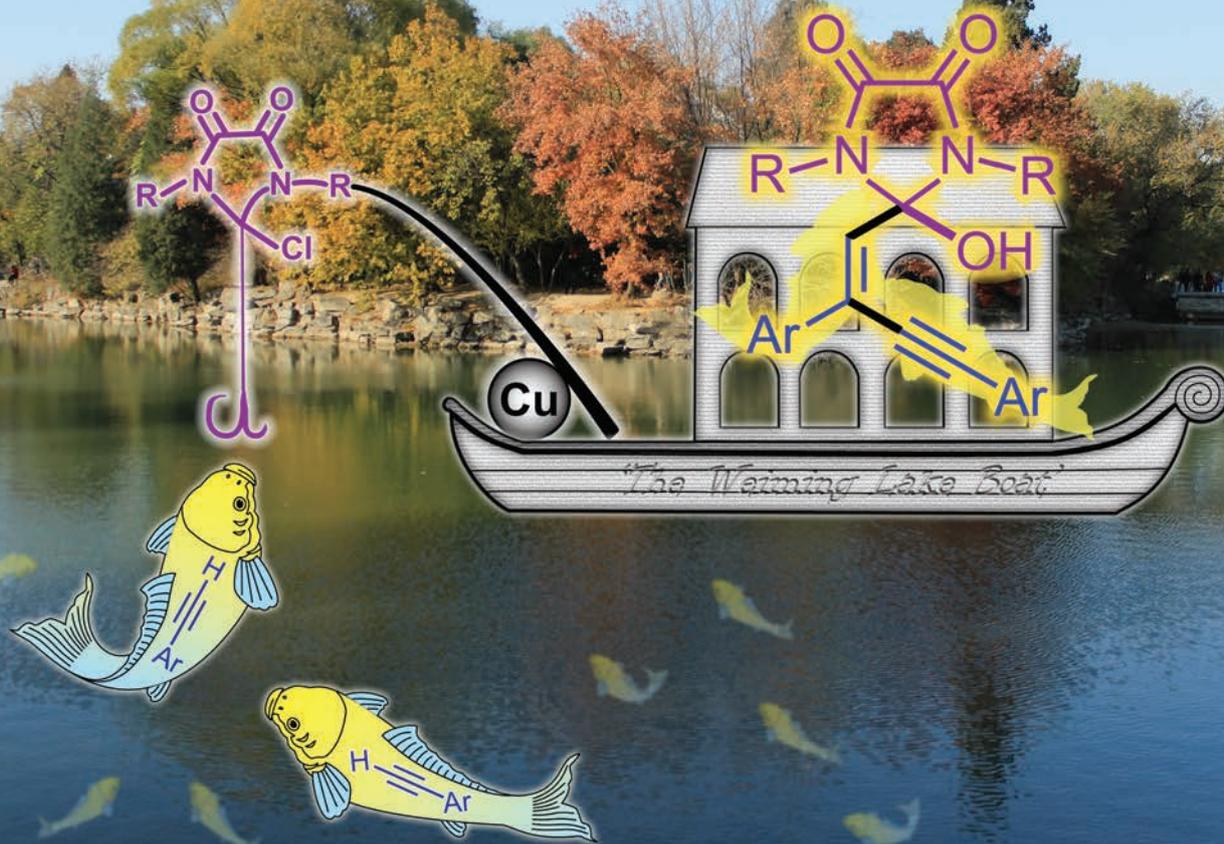


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## COMMUNICATION

Wen-Xiong Zhang *et al.*

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### Selective synthesis of (*Z*)-2-enynyl-2-hydroxy-imidazolidine-4,5-diones via Cu(I)-mediated multicomponent coupling of terminal alkynes, carbodiimides and oxalyl chloride†

Fei Zhao,<sup>a</sup> Yuexing Li,<sup>a</sup> Yang Wang,<sup>a</sup> Wen-Xiong Zhang<sup>\*a,b</sup> and Zhenfeng Xi<sup>a</sup>

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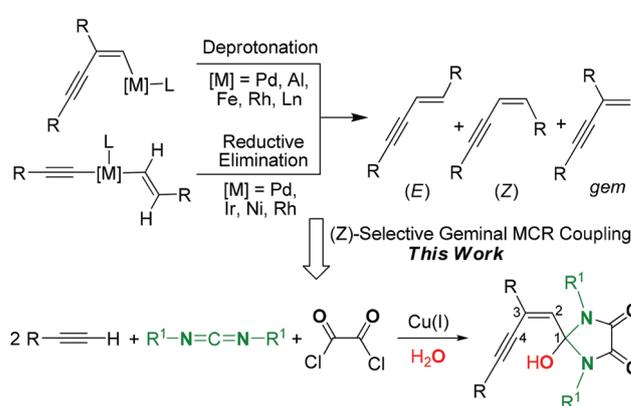
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(*Z*)-2-Enynyl-2-hydroxy-imidazolidine-4,5-diones **2** are synthesized for the first time via Cu(I)-mediated (*Z*)-selective geminal coupling among two molecules of terminal alkynes, carbodiimides, and oxalyl chloride. Further transformation of **2a** is performed to yield a highly functionalized spiro heterocyclic compound **5**.

Considerable efforts have been devoted to the dimerization of terminal alkynes because it provides a straightforward method to construct conjugated enynes, which are versatile building blocks in organic synthesis and significant components in bioactive molecules.<sup>1,2</sup> However, highly selective formation of conjugated enynes by dimerization remains limited due to the competitive formation of three possible (*E*)-, (*Z*)-, and *gem*-enynes.<sup>1,2</sup> As far as we are aware, the multicomponent coupling<sup>3</sup> via incorporating organic components into the well-established dimerization of terminal alkynes has not been reported. It is a major challenge because the deprotonation of the final enyne-containing intermediate with a terminal alkyne is a fast step or the reductive elimination of the final acetylide intermediate is more favorable in two reported mechanisms (Scheme 1). Another challenge is how to control the regio- and stereoselectivity of the corresponding enynes.

(*Z*)-2-En-4-yn-1-ols ((*Z*)-enynols for short), as a class of multifunctional organic skeletons, are of considerable interest in modern organic synthesis because of their important application in synthesis of O-containing heterocycles.<sup>4,5</sup> Although the synthesis of (*Z*)-enynols has received much attention,<sup>6–10</sup> (*Z*)-enynols bearing a heteroatomic substituent at the C1 position have not been reported because of the difficulty in introducing a



**Scheme 1** Unexpected (*Z*)-selective geminal coupling of two terminal alkynes, carbodiimides and oxalyl chloride.

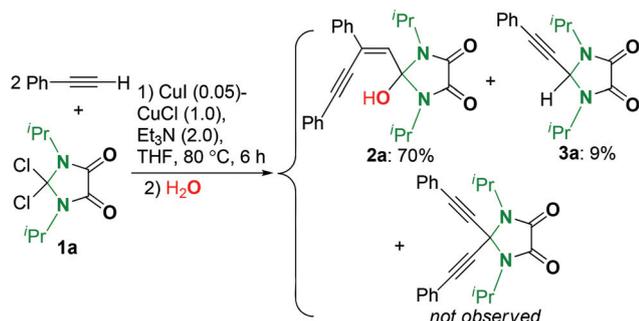
heteroatom into the starting materials. Thus, a simple and efficient method to synthesize heteroatom-incorporated (*Z*)-enynols at the C1 position remains of great importance to academia and to the pharmaceutical industry. Herein we report our new discovery of Cu(I)-mediated multicomponent coupling of two terminal alkynes, carbodiimides, and oxalyl chloride to construct the novel (*Z*)-enynols bearing a heterocyclic linker at the C1 position. In this process, the (*Z*)-selective geminal coupling of two molecules of terminal alkynes is found. Further transformation of (*Z*)-enynol was performed to yield highly functionalized spiro heterocyclic compounds.

We have focused on carbodiimide-based multicomponent reactions to construct some N-containing organic molecules.<sup>11,12</sup> Recently we have reported one-pot sequential reaction of amines, carbodiimides, and oxalyl chloride to prepare cyclic di-oxoguanidines. The 2,2-dichloroimidazoline-4,5-dione intermediate **1a** was isolated and characterized from the reaction of *N,N'*-diisopropylcarbodiimide (DIC) and oxalyl chloride (see the ESI† for its X-ray structure).<sup>12e,13</sup> The connection of four electronegative atoms in **1a** made the C2 atom highly electrophilic. So we envisioned whether two C–Cl bonds in **1a** could undergo the cross-coupling reactions with terminal

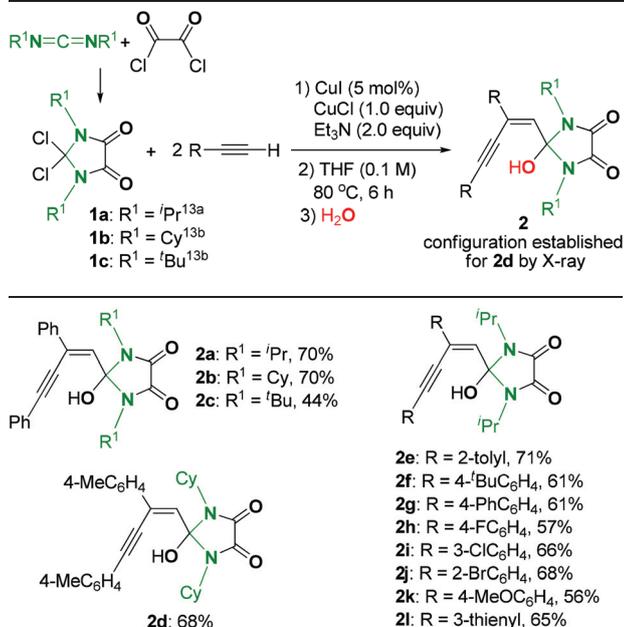
<sup>a</sup>Beijing National Laboratory for Molecular Sciences, and Key Laboratory of Bioorganic Chemistry and Molecular Engineering of Ministry of Education, College of Chemistry, Peking University, Beijing 100871, China. E-mail: wx\_zhang@pku.edu.cn; Fax: +86-10-62751708; Tel: +86-10-62759728

<sup>b</sup>State Key Laboratory of Elemento-Organic Chemistry, Nankai University, Tianjin, 300071, China

† Electronic supplementary information (ESI) available: Materials including experimental procedures, NMR spectra of all new products and X-ray data for **1a** and **2d**. CCDC 857965 and 959743. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c4ob00185k



Scheme 2 Screening of reaction conditions.

Table 1 Formation of (*Z*)-enynols<sup>a</sup>

<sup>a</sup> Byproducts 3 were formed in 5–10% yields.

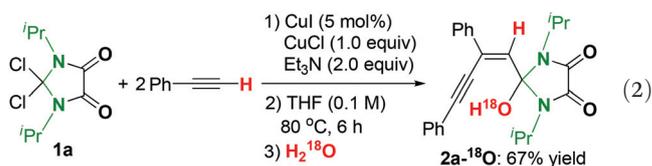
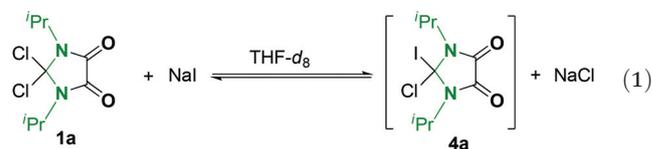
alkynes to generate 1,4-diyne. However, it was found that, in the presence of CuI and Et<sub>3</sub>N, a (*Z*)-2-enynyl-2-hydroxyimidazolidine-4,5-dione **2a**<sup>14</sup> was observed *via* the coupling of **1a** with two molecules of phenylethyne followed by a byproduct **3a**. The expected 1,4-diyne product was not observed (Scheme 2). After various reaction conditions including reaction temperature, reaction time, bases,<sup>15</sup> and the metal salts, such as CuCl, CuBr, CuI and PdCl<sub>2</sub>, were screened (see ESI† for details), an optimal condition was found and the expected **2a** was isolated in 70% yield (Scheme 2).

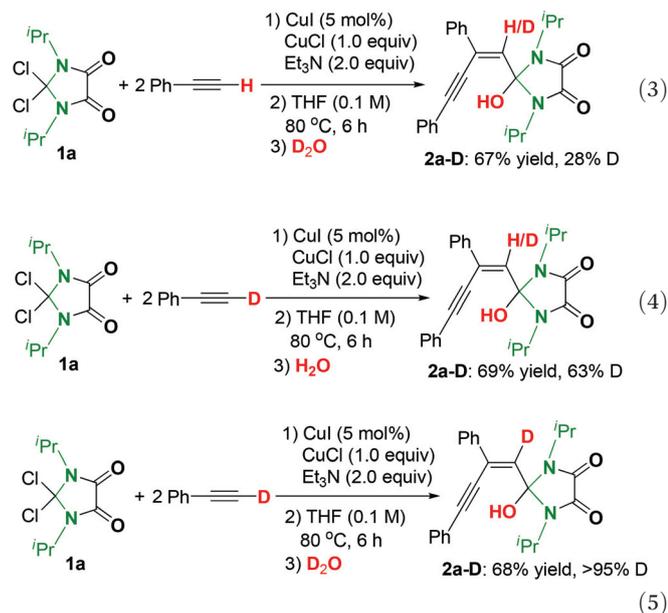
With the optimized conditions in hand, we began to explore the reaction scope. The representative results for the formation of (*Z*)-enynols **2** are summarized in Table 1. 2,2-Dichloroimidazolidine-4,5-diones **1** were generated *in situ* from carbodiimides and oxalyl chloride. Carbodiimides (RN=C=NR, R = <sup>i</sup>Pr, Cy, <sup>t</sup>Bu) were tested as suitable nitro-

gen sources for the reaction. Because of the steric hindrance of the *tert*-butyl group, <sup>t</sup>BuN=C=N<sup>t</sup>Bu gave **2c** in a significantly lower yield than other *N,N'*-dialkylcarbodiimides. As far as terminal alkynes were concerned, the reaction was not affected by the positions of the substituents at the phenyl ring of an aromatic alkyne (**2d–l**). Electron-donating groups such as alkyl (**2d–f**) and alkoxy groups (**2k**) and weak electron-withdrawing groups such as halogens (**2h–j**) would give good yields. It was noted that strong electron-withdrawing groups at the phenyl ring of an aromatic alkyne would result in no product. Heterocyclic terminal alkynes such as 3-ethynylthiophene gave the desired product **2l** in 65% isolated yield. The single crystal structure of **2d** clearly revealed the *Z*-configuration of the alkene moiety (see the ESI† for its X-ray structure).<sup>16</sup>

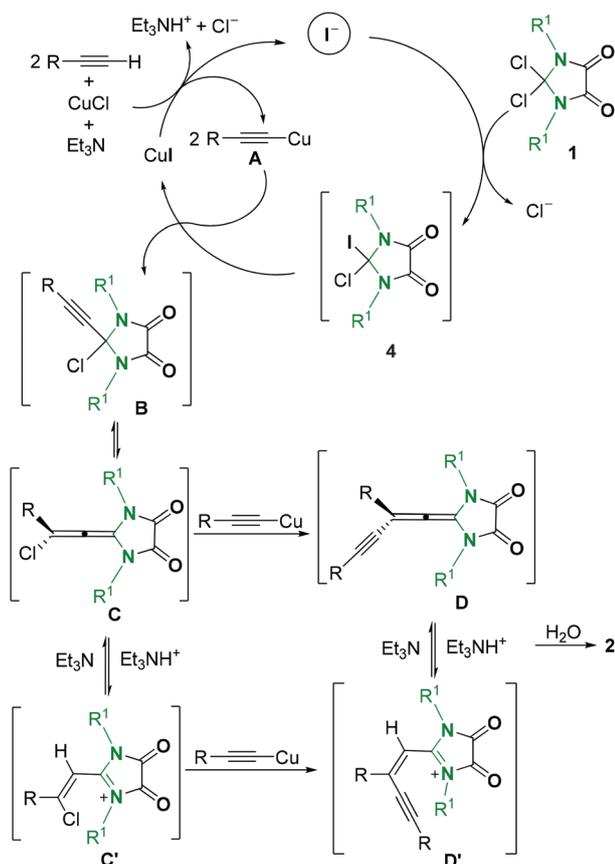
These interesting and novel results intrigued us to explore the reaction mechanism. A series of experiments were performed. First, the necessity of iodide was investigated. Iodide is usually considered to be a good nucleophile as well as a good leaving group. To obtain the evidence of the iodo-substituted intermediate, the 1 : 1 mixture of **1a** and NaI in THF-*d*<sub>8</sub> was monitored by NMR spectroscopy. Both <sup>1</sup>H and <sup>13</sup>C NMR spectra showed the formation of a new compound. The *in situ* NMR spectra also showed that the ratio of **1a** and **4a** was 1 : 0.18 and remained unchanged after a long period (*ca.* 7 days), indicating that there was an equilibrium between them (see ESI† for details). **4a** is proposed to be a monoiodo-substituted intermediate. Therefore, 2,2-dichloroimidazolidine-4,5-dione was proposed to undergo a Cl–I exchange giving an important intermediate (eqn (1)).

Next, the sources of the alkenyl hydrogen and hydroxyl group in product **2** were explored. A series of isotopic labeling experiments were carried out. The final reaction mixture of **1a** with phenylethyne was quenched with H<sub>2</sub><sup>18</sup>O to produce the <sup>18</sup>O-labeling product **2a-<sup>18</sup>O**. This result clearly showed that the hydroxyl group in **2** should come from water (eqn (2)). Deuterium labeling experiments were performed with phenylacetylene-*d*<sub>1</sub> and/or D<sub>2</sub>O. A single deuterium source gave the deuterated product **2a-D** with low proportion of deuterium (eqn (3) and (4)). Only a combination of the two deuterium sources could lead to a fully deuterated product (eqn (5)). The results showed that the alkenyl proton should be from both terminal alkynes and water.





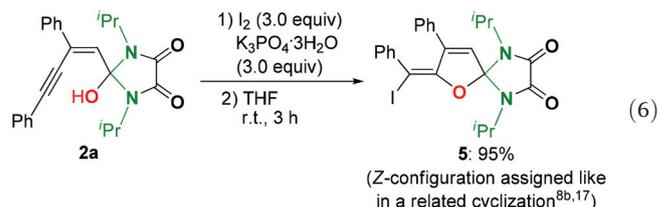
Based on the experimental results above, a plausible mechanism for the formation of **2** is proposed in Scheme 3. In the presence of Et<sub>3</sub>N, the copper acetylide (**A**) is generated from terminal alkynes and CuI/CuCl, releasing the chloride and iodide anions simultaneously. The nucleophilic substitution of **1** by iodide generates the intermediate **4**. A Sonogashira type



Scheme 3 A proposed mechanism.

cross-coupling reaction of **4** with **A** would give rise to the intermediate **B** and regenerate CuI. The regenerated CuI would participate in the next catalytic cycle. **B** then undergoes an isomerization to form chloroallene **C**, or further protonation by Et<sub>3</sub>NH<sup>+</sup> to form **C'**. A Stephens–Castro coupling of **C** or **C'** with **A** would form **D** or **D'**. **D** is quenched with water to give the final product **2**.

Further transformation of (*Z*)-enynol **2a** was tested under various conditions. A new spiro heterocyclic compound **5** was synthesized by electrophilic cyclization of **2a** with I<sub>2</sub> in THF solution with K<sub>3</sub>PO<sub>4</sub> as a base, which showed the potential of this synthetic strategy (eqn (6)).<sup>17</sup>



In conclusion, Cu(I)-mediated (*Z*)-selective geminal MCR coupling among two molecules of terminal alkynes, carbodiimides, and oxalyl chloride is achieved for the first time to afford (*Z*)-enynols bearing a heterocyclic linker at the C1 position. (*Z*)-Enynol shows the potential application in the synthesis of highly functionalized spiro heterocyclic compounds. It is noted that the multicomponent coupling *via* incorporating organic components into the well-established dimerization of terminal alkynes is realized for the first time. Further investigations on their application are ongoing.

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