

Bioorganic & Medicinal Chemistry Letters 10 (2000) 2231-2234

Synthesis and Antiproliferative Evaluation of 7-Aminosubstituted Pyrroloiminoquinone Derivatives

Valérie Bénéteau,^a Alain Pierré,^b Bruno Pfeiffer,^c Pierre Renard^c and Thierry Besson^{a,*}

^aLaboratoire de Génie Protéique et Cellulaire, UPRES 2001, Pôle Sciences et Technologie, Université de La Rochelle, Avenue Marillac, 17042 La Rochelle cedex 1, France

> ^bInstitut de Recherche Servier, 11 Rue des Moulineaux, 92150 Surennes, France ^cA.D.I.R. et Cie, 1 Rue Carle Hébert, 92415 Courbevoie cedex, France

> > Received 19 June 2000; revised 25 July 2000; accepted 4 August 2000

Abstract—Coupling of five amines on the 7-methoxy-1,3,4,5-tetrahydropyrrolo[4,3,2-*de*]quinoline core was achieved and afforded, in particular, an opened analogue of the natural alkaloid wakayin. Evaluation of cytotoxic activity of compounds **2**, **10–13** on L1210 cells afforded IC₅₀ in the range 0.25–5.3 μ M. © 2000 Elsevier Science Ltd. All rights reserved.

Introduction

Since the discovery of discorhabdin C in 1986 until isolation and characterization of veuitamine in 1997, more than 40 pyrroloiminoquinone alkaloids have been isolated from sponges and ascidians.¹ For these marine organisms devoid of skeleton and immune system, fixed or not very mobile, secondary metabolites would have a basic role in communication, defense or attack towards other organisms.

Pyrroloiminoquinones have recently received increasing attention as a source of new and useful anticancer drugs. In particular wakayin 1, isolated from Ascidian *Clavelina* species, shows biological activities such as inhibition of topoisomerase I^2 Until now, only two approaches to structures analogous to wakayin were reported with no biological activity described.³ As part of our work on heterocyclic compounds with potential pharmacological value,⁴ we planned to synthesize pyrroloiminoquinone **2** which is structurally very close to the natural alkaloid, by the retrosynthetic pathway shown in Scheme 1, from tryptamine and 7-methoxy-3,4-dihydro-1*H*-pyrrolo[4,3,2-*de*]quinolin-8-one **3**.

Chemistry

The strategy used for building the tricyclic system 3 was inspired by the work of Joule and Alvarez.⁵ 6,7-Dimethoxy-4-methylquinoline 4 was formed by reaction between 3,4-dimethoxyaniline and methylvinylketone in refluxing acetic acid with iron(III) chloride. Nitration of 4 (concentrated nitric acid, $-50 \,^{\circ}$ C) led to the 5nitro compound 5 in a good yield (78%). According to Vismara's method, oxidation of the methyl group to aldehyde was achieved in DMSO with trifluoroacetic acid, iodine, tert-butyliodide and iron(II) chloride. The aldehyde function was then protected as a dimethylacetal to give 7 in 87% yield. The pyridine ring and the nitro group were simultaneously reduced by an excess of sodium borohydride in methanol with nickel chloride to afford diamine 8 in a very good yield (95%). Cyclization of the pyrrole ring occurred by heating compound 8 at 80 °C in aqueous HCl and THF in a 62% yield (Scheme 2).

Aminoindole 9 was treated by ceric ammonium nitrate (CAN) in aqueous acetonitrile to generate the quinone structure. Substitution of the C-7 methoxy group by the amine was achieved in situ, without isolating the very unstable intermediate 3. The product of coupling was then transformed into its trifluoroacetate salt (Scheme 3). In this way, tryptamine was condensed on the pyrroloiminoquinone core to afford compound 2. Because some studies⁶ have proved the utility of DNA-intercalating heterocyclic

^{*}Corresponding author. Fax: +33-5-46-45-82-65(or 47). E-mail: tbesson@univ-lr.fr

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Figure 1.

Scheme 1.

compounds possessing a cationic side chain, products **10** and **11** were prepared by treatment of **3** with N,N-dimethylethylenediamine and 1-(2-amino-ethyl)pyrrolidine, respectively. Finally, makaluvamines D (**12**) and I (**13**) were obtained by reaction of **3** with tyramine and ammonium chloride, respectively.⁷

Pharmacology

Compounds 2 and 10–13 were evaluated in vitro for their antiproliferative activity using the murine L1210 leukemia cell line.⁸ The results expressed as IC_{50}

(concentration reducing the cell proliferation by 50%) are reported in Table 1. Because all the IC_{50} s were less than 20 μ M, cell cycle perturbations were also investigated.

Results and Discussion

Compound 2, a strict analogue of wakayin, is the least active of the tested molecules on the L1210 cell line. Compared to the natural alkaloid, the enhanced flexibility of the substituent at C-7 may not be favorable for interaction with biological targets. The most cytotoxic compounds (10, 11 and 13) inhibit the growth of L1210 cells at concentrations less than $0.5 \,\mu$ M. These water soluble products are substituted by a primary amino or an aminoalkyl group. Makaluvamine D (12), bearing a hydrophobic side chain, showed a moderate antiproliferative activity in accordance with those reported for different cell lines (IC₅₀(HCT116) = 23.4 μ M, IC₅₀(xrs-6) = 19.1 μ M).⁹

Despite interesting antiproliferative properties, none of the pyrroloiminoquinone derivatives 2 and 10–13 induce significant modification of the L1210 cells cycle. This may



Scheme 2. Reaction conditions and yields: (a) AcOH, FeCl₃, reflux, 2 h, 37%; (b) concd HNO₃, -50° C, 1.5 h, 78%; (c) I₂, *t*BuI, FeCl₂, DMSO, TFA, 100 °C, 4 h, 83%; (d) HCl in Et₂O, CH₃OH, reflux, 48 h, 87%; (e) NaBH₄, NiCl₂, CH₃OH, rt, 1 h, 95%: (f) HCl 1 N, THF, 80 °C, 1 h, 62%.



Scheme 3.

Compound	Formula	$IC_{50}\left(\mu M\right)$	Effect on cellular cycle
2	C ₂₂ H ₁₉ N ₄ O ₃ F ₃	5.3	Nonspecific
10	C18H20N4O5F6	0.3	Nonspecific
11	C20H22N4O5F6	0.38	Nonspecific
12	C20H18N3O4F3	2.1	Nonspecific
13	$C_{12}H_{10}N_{3}O_{3}F_{3}\\$	0.25	Nonspecific

prove that the pyrroloiminoquinone skeleton on its own is not able to exert a specific cytotoxic action.

Conclusion

In conclusion, we achieved the synthesis of an opened analogue of wakayin whose structure is the closest to the natural compound yet published. Two products with basic side chains were also formed as well as the marine alkaloids makaluvamines D and I. These products, especially water soluble compounds **10**, **11** and **13**, proved to be quite good cytotoxic agents despite their lack of interaction with the L1210 cellular cycle.

We are now focusing our efforts on the introduction of a C–C bond at C-6 in order to mimic veiutamine, the most active pyrroliminoquinone alkaloid recently isolated from the Fijian sponge Zyzzya fuliginosa.

Acknowledgements

We thank the Comité de Charente-Maritime de la Ligue contre le Cancer for financial support.

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7. Selected data for compound **2**: IR (KBr) v 3121, 1680, 1555, 1403, 1338, 1206 cm⁻¹. ¹H NMR (DMSO-*d*₆) 2.87 (t, 2H, *J* 7.5 Hz), 3.03 (t, 2H, *J* 7.5 Hz), 3.59 (m, 2H), 3.79 (dt, 2H, *J* 7.5 Hz and *J* 3 Hz), 5.49 (s, 1H), 6.99 (t, 1H, *J* 7.3 Hz), 7.07 (t, 1H, *J* 7.3 Hz), 7.22 (d, 1H, *J* 2.4 Hz), 7.34 (d, 1H, *J* 7.3 Hz),

7.34 (s, 1H), 7.56 (d, 1H, J 7.3 Hz), 9.11 (t, 1H, J 6.2 Hz), 10.43 (s, 1H), 10.92 (s, 1H), 13.10 (s, 1H). ¹³C NMR (DMSO-d₆) 18.15; 23.30; 42.43; 44.15; 83.96; 110.62; 110.63; 111.49; 118.18; 118.42; 118.70; 121.10; 122.64; 123.12; 123.81; 126.97; 136.24; 153.12; 157.07; 167.48. MS (FAB) m/z 331 (M+H, 100%), 201 (M-C₉H₈N, 21), 188 (10). Anal. calcd for C₂₂H₁₉N₄O₃F₃: C, 59.46; H, 4.31; N, 12.61; found: C, 59.51; H, 4.35; N, 12.76. All compounds were fully characterized by

spectroscopic and elemental analysis. Structural data for compounds 13 and 14 are consistent with those previously reported.

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