

Cite this: DOI: 10.1039/c2cc34556k

www.rsc.org/chemcomm

COMMUNICATION

Efficient synthesis of 2'-C- α -aminomethyl-2'-deoxynucleosides†‡§Nan-Sheng Li*^a and Joseph A. Piccirilli*^{ab}

Received 25th June 2012, Accepted 12th July 2012

DOI: 10.1039/c2cc34556k

Starting from methyl 3,5-di-*O*-benzyl-2-keto- α -D-ribofuranoside, a convergent, six-step synthesis is developed to give efficiently all four 2'-C- α -aminomethyl-2'-deoxynucleosides (U, C, A, G) in 38%, 42%, 12%, 12% yield, respectively. Convergence is achieved by the glycosylation of persilylated nucleobases with methyl 2- α -phthalimidomethyl ribofuranoside.

The chemical synthesis of modified nucleosides and the incorporation of nonnatural nucleosides into RNA have become increasingly important in medical applications and as reagents for structure and function investigations of RNAs.^{1,2} A new class of nucleoside analogues, 2'-C- α -aminomethyl-2'-deoxynucleosides (Fig. 1, IV), which hold both amino and 2'-alkyl functionalities, have features common to both 2'- α -amino-2'-deoxynucleosides³ and 2'-C-branched nucleosides,⁴⁻⁸ and therefore might possess interesting antiviral and anticancer activities. Additionally, oligonucleotides containing 2'-aminomethyl nucleosides provide a new opportunity for regiospecific chemoligation reaction with a variety of amine-reactive probes to attach desired reporter groups.^{9,10} Here we describe an efficient approach to all of the four potentially important nucleoside analogues: 2'-C- α -amino-methyl-2'-deoxynucleosides (A, C, G, U).

We designed a convergent approach (glycosylation of various nucleobases with 2'-aminomethyl ribose derivatives) to prepare the four 2'-C- α -aminomethyl nucleosides (U, C, A and G)

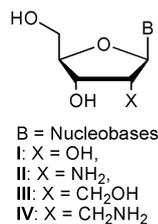


Fig. 1 Structures of ribonucleosides (I) and modified nucleosides (II–IV).

^a Department of Biochemistry & Molecular Biology, University of Chicago, 929 East 57th Street, Chicago, Illinois 60637, USA

^b Department of Chemistry, University of Chicago, 929 East 57th Street, Chicago, Illinois 60637, USA. E-mail: nli@uchicago.edu, jpiccirilli@uchicago.edu; Fax: +1 773 702 0438; Tel: +1 773 702 5236, +1 773 702 9312

† Dedicated to the memory of Professor H. Gobind Khorana.

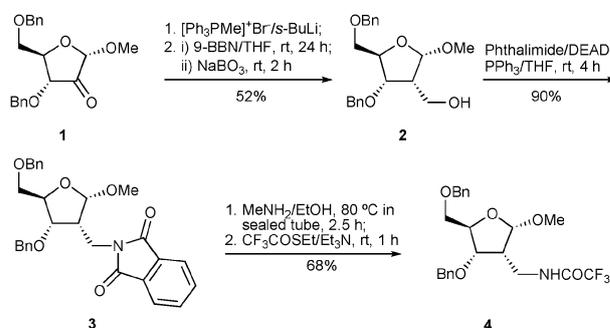
‡ This article is part of the 'Nucleic acids: new life, new materials' web-themed issue.

§ Electronic supplementary information (ESI) available: Experimental procedures and characterization of all new compounds. ¹H and ¹³C NMR spectra of **15a–d**. See DOI: 10.1039/c2cc34556k

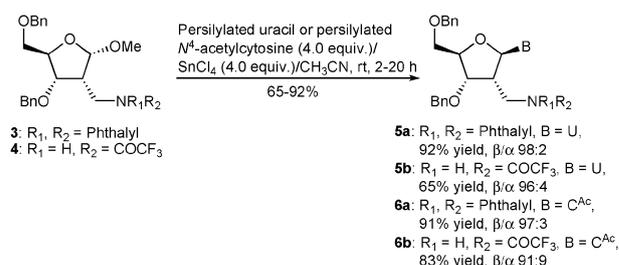
efficiently. We converted methyl 3,5-di-*O*-benzyl-2-keto- α -D-ribofuranoside (**1**)¹¹ in two steps (*via* Wittig and hydroboration reactions) to 2- α -hydroxymethyl ribofuranose derivative **2** in 52% yield (Scheme 1). Under Mitsunobu conditions,¹² **2** reacts to form 2- α -phthalimidomethyl ribofuranoside **3** in 90% yield. Deprotection of **3** with methylamine in ethanol¹³ followed by reaction with *S*-ethyl α , α , α -trifluoroacetate gives 2- α -(α , α , α -trifluoroacetyl)-aminomethyl ribofuranoside **4** in 68% yield.

Glycosylation of persilylated pyrimidine nucleobases (uracil and *N*⁴-acetylcytosine) with 2'-aminomethyl ribofuranoside derivatives **3** and **4** in the presence of SnCl₄ gives the corresponding uridine and cytidine derivatives with high β -selectivity (91 : 9 to 98 : 2) (Scheme 2). Although yield and β -selectivity were higher using 2- α -phthalimidomethyl ribofuranoside **3**, the high β -selectivity of the glycosylation reaction indicates that both phthalyl and trifluoroacetyl groups can stabilize the 1-oxocarbenium ion intermediates and direct the nucleobase to attack the β face of the ribofuranoside.

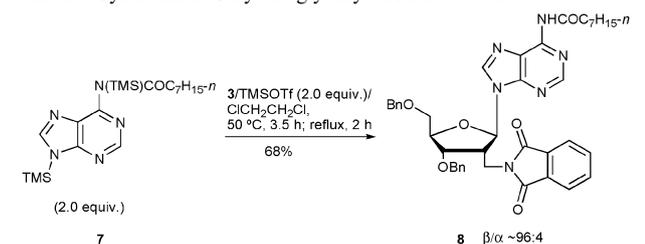
Glycosylation of purine nucleobases generally requires nonpolar solvent and the presence of TMSOTf to achieve the optimized yield and β -9 selectivity.^{14,15} Glycosylation of persilylated



Scheme 1 Synthesis of glycosylation reagents **3** and **4**.



Scheme 2 Glycosylation of persilylated pyrimidine nucleobases with **3** and **4**.

Table 1 Synthesis of **8** by the glycosylation of **7** with **3**

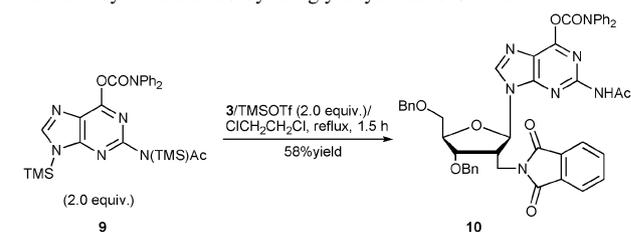
Entry	Solvent	Reaction conditions	Yield (%)	β/α
1	$\text{ClCH}_2\text{CH}_2\text{Cl}$	Reflux, 2 h	71	87 : 13
2	$\text{ClCH}_2\text{CH}_2\text{Cl}$	rt, 48 h; reflux, 1 h	68	91 : 9
3 ^a	CH_3CN	Reflux, 1.5 h	—	—
4	$\text{ClCH}_2\text{CH}_2\text{Cl}$	50 °C, 3.5 h; reflux, 2 h	68	96 : 4

^a Lower yield and lower selectivity.

*N*⁶-benzoyladenosine with **3** or **4** in the presence of TMSOTf in refluxing 1,2-dichloroethane for two hours yields the corresponding products in low yields (only ~35%) and moderate selectivity, with **3** giving better β -selectivity than with **4** (β/α 80 : 20 and 65 : 35, respectively). *N*⁶-Octanoyladenosine has been used in the glycosylation with 1-*O*-acetyl-2,3,5-tri-*O*-benzoyl- β -D-ribofuranose for its better solubility than *N*⁶-benzoyladenosine.¹⁶ It also has been used in the transglycosylation of 2'- α -trifluoroacetylaminouridine.¹⁷ We then found that use of persilylated *N*⁶-octanoyladenosine **7** significantly improved yield and β -selectivity of the glycosylation with **3**. The results are shown in Table 1. Glycosylation in nonpolar solvent (1,2-dichloroethane) generally yields the glycosylation product in high β -selectivity. In contrast, glycosylation in the polar solvent, acetonitrile, gave the product in low yield and low selectivity (Table 1, Entry 3). The optimized results with 68% yield and ~96 : 4 β -selectivity were achieved using 1,2-dichloroethane at 50 °C for 3.5 hours, then at reflux for 2 hours (Table 1, entry 4).

The glycosylation of persilylated *N*²-acetylguanosine (2 equiv.) with glycosylating agent **3** in the presence of TMSOTf (2 equiv.) gave complex results even in nonpolar solvents such as 1,2-dichloroethane, toluene and benzene. Reactions gave all of four possible products (β -9, β -7, α -9 and α -7) in low yields and in low selectivity. However, using persilylated *N*²-acetyl-*O*⁶-(diphenylcarbamoyl)-guanine (**9**)^{18,19} glycosylation in the presence of TMSOTf gave the β -9 isomer product **10** exclusively. The glycosylation of persilylated *N*²-acetyl-*O*⁶-(diphenylcarbamoyl)-guanine (**9**) with **3** under various conditions are shown in Table 2. The optimized yield (58%) is obtained if the glycosylation of *N*²-acetyl-*O*⁶-(diphenylcarbamoyl)-guanine (**9**) with **3** is carried out in refluxing 1,2-dichloroethane (80 °C oil bath) for 1.5 hours (Table 2, entry 5).

To remove the 3',5'-di-*O*-benzyl groups from the glycosylation products, we tested two debenzoylation methods (Method A: Pd(OH)₂ or Pd catalyzed hydrogenation,²⁰ and Method B: boron trichloride (Lewis acid) induced benzyl ether cleavage^{21–23}) (Table 3). Method A removed the benzyl groups from uridine derivatives **5a** and **5b** to give compounds **11a** and **11b** in 62% and 81% yield, respectively (Table 3, entries 1, 3). The uridine derivative **5a** can also be debenzoylated by Method B to give compound **11a** in 90% yield (entry 2). However, for the cytosine

Table 2 Synthesis of **10** by the glycosylation of **9** with **3**

Entry	Solvent	Reaction conditions	Yield (%)
1	$\text{ClCH}_2\text{CH}_2\text{Cl}$	rt, 62 h; reflux, 1 h	45
2	Toluene	80 °C, 3 h	15
3 ^a	$\text{ClCH}_2\text{CH}_2\text{Cl}$	reflux, 4 h	—
4 ^b	CH_2Cl_2	reflux, 4 h	—
5	$\text{ClCH}_2\text{CH}_2\text{Cl}$	reflux (80 °C), 1.5 h	58

^a Product is decomposed significantly and the yield is not determined.

^b Reaction is incomplete and the yield is not determined.

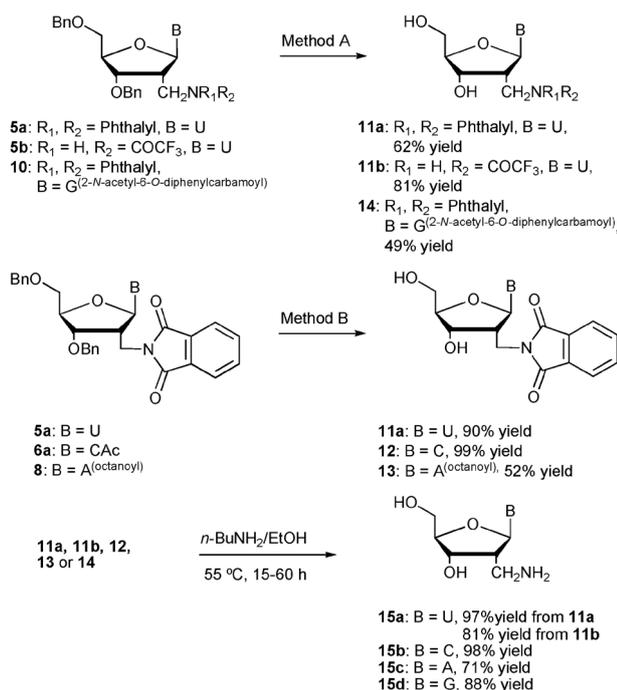
Table 3 Removal of the 3',5'-di-*O*-benzyl groups of various glycosylation products

Entry	S ^a	Method	Reaction conditions	P ^b	Yield (%)
1	5a	A	20% Pd(OH) ₂ /C, H ₂ , CH ₃ CO ₂ C ₂ H ₅ , rt, 22 h	11a	62
2	5a	B	1. BCl ₃ /CH ₂ Cl ₂ , -78 °C, 4 h; 2. MeOH/CH ₂ Cl ₂ (v/v 1 : 1), -78 °C-rt, 30 min	11a	90
3	5b	A	20% Pd(OH) ₂ /C, H ₂ , CH ₃ CO ₂ C ₂ H ₅ , rt, 17 h	11b	81
4	6a	B	1. BCl ₃ /CH ₂ Cl ₂ , -78 °C, 2.5 h; 0 °C, 30 min; 2. MeOH, 0 °C-rt, 12 h	12	99
5 ^c	6b	A	20% Pd(OH) ₂ /C, H ₂ , CH ₃ CO ₂ C ₂ H ₅ , rt, 6 h	—	—
6 ^d	8	A	20% Pd(OH) ₂ /C, H ₂ , CH ₃ CO ₂ C ₂ H ₅ , rt, 15 h	—	—
7 ^e	8	B	1. BCl ₃ /CH ₂ Cl ₂ , -78 °C, 1 h; 2. MeOH/CH ₂ Cl ₂ (v/v 1 : 1), -78 °C-rt, 30 min	—	—
8	8	B	1. BCl ₃ /CH ₂ Cl ₂ , -78 °C, 1 h; 2. Et ₃ N/MeOH/CH ₂ Cl ₂ , 78 °C-rt, 30 min	13	47
9	8	B	1. BCl ₃ /CH ₂ Cl ₂ , -78 °C, 1 h; 2. NaHCO ₃ (aq.), 78 °C-rt, 30 min	13	52
10	10	A	20% Pd(OH) ₂ /C, H ₂ , CH ₃ CO ₂ C ₂ H ₅ , rt, 15 h	14	49
11	10	A	5% Pd/C, H ₂ , CH ₃ CO ₂ C ₂ H ₅ , rt, 15 h	14	49
12	10	B	1. BCl ₃ /CH ₂ Cl ₂ , -78 °C, 1 h; 2. Et ₃ N/MeOH/CH ₂ Cl ₂ , 78 °C-rt, 30 min	14	19

^a S: Starting materials (substrates). ^b P: Products. ^c Cytosine ring also reduced. ^d The results are complicated. ^e Product is decomposed.

derivatives (**6a** and **6b**), Method A in addition to the debenzoylation, reduced the 5,6-double bond of the cytosine ring (entry 5). In contrast, Method B removes the benzyl and acetyl groups from cytidine derivative **6a** without affecting the 5,6-double bond to generate **12** in 99% yield (entry 4).

Attempts to remove the benzyl groups from adenosine derivative **8** by Method A gave complex results (entry 6). However, for Method B, if a base such as triethylamine or aqueous sodium bicarbonate was added to quench the reaction (entries 8, 9), the adenosine derivative **8** undergoes selective debenzoylation to give compound **13** in 47–52% yield. Without the base, the product



Scheme 3 Synthesis of 2'- α -aminomethyl-2'-deoxynucleosides.

decomposes (entry 7). For the guanosine derivative **10** under the same conditions, Method B gives the corresponding debenzylated product **14** in only 19% yield (entry 12). In contrast, Method A converts the guanosine derivative **10** to compound **14** in 49% yield (entries 10, 11). To complete the synthesis, **11a**, **11b**, **12**, **13** and **14** were treated with *n*-butylamine in ethanol at 55 °C to give the final 2'-*C*- α -aminomethyl-2'-deoxynucleosides **15** (A, C, G, U) in 71–97% yields (Scheme 3).

NOESY experiments confirmed the structures of β -anomers. For the uridine derivative **5a**, we observed strong NOE between 2'-H (δ 3.02) and 3'-H (δ 4.22), between 1'-H (δ 6.14) and CH₂N (δ 4.00) but no NOE between 1'-H and 2'-H. For cytidine derivative **6a**, we observed strong NOE between 1'-H (δ 6.04) and CH₂N (δ 4.00) but no NOE between 1'-H and 2'-H (δ 2.97). For adenosine derivative **8**, we also observed strong NOE between 1'-H (δ 6.34) and CH₂N (δ 4.16) but only weak NOE between 1'-H and 2'-H (δ 3.84). For 2'-*C*- α -aminomethyl-2'-deoxyguanosine (**15d**), we observed strong NOE between 1'-H (δ 5.81) and 4'-H (δ 4.08), but no NOE between 1'-H (δ 5.81) and 3'-H (δ 4.48); we also observed stronger NOE between 1'-H (δ 5.81) and CH₂N (δ 3.30) but weak NOE between 1'-H (δ 5.81) and 2'-H (δ 3.12). These data suggest that 1'-H and 2'-CH₂N reside on the same face of the ribose ring and therefore the glycosylation products have the β -configuration.

In summary we have developed a novel synthetic approach to a new class of nucleoside analogues: all four 2'-*C*- α -aminomethyl-2'-deoxynucleosides (A, C, G, U) efficiently and

convergently synthesized by the glycosylation of persilylated nucleobases with 2- α -phthalimidomethyl ribofuranoside **3** or 2- α -trifluoroacetylaminoethyl ribofuranoside **4**. As glycosylating reagent, 2- α -phthalimidomethyl ribofuranoside **3** gives better results (yield and selectivity) than 2- α -trifluoroacetylaminoethyl ribofuranoside **4**. Debenzylation of the glycosylation products was accomplished either by hydrogenation (Method A) for guanosine derivatives or by boron trichloride induced benzyl ether cleavage (Method B) for uridine, cytidine and adenosine derivatives. These analogues will enable new strategies for investigating structure-function relationships in RNA.

This work was supported by an N.I.H. grant to J.A.P. (1R01AI081987).

Notes and references

- S. R. Das, R. Fong and J. A. Piccirilli, *Curr. Opin. Chem. Biol.*, 2005, **9**, 585–593.
- S. Shukla, C. S. Sumaria and P. I. Pradeepkumar, *ChemMedChem*, 2010, **5**, 328–349.
- F. S. Santiago and L. M. Khachigian, *J. Mol. Med. (Berlin, Ger.)*, 2001, **79**, 695–706.
- G. Asif, S. J. Hurwitz, J. Shi, B. I. Hernandez-Santiago and R. F. Schinazi, *Antimicrob. Agents Chemother.*, 2007, **51**, 2877–2882.
- S. Benzaria, D. Bardiot, T. Bouisset, C. Counor, C. Rabeson, C. Pierra, R. Storer, A. G. Loi, A. Cadeddu, M. Mura, C. Musiu, M. Liuzzi, R. Loddo, S. Bergelson, V. Bichko, E. Bridges, E. Cretton-Scott, J. Mao, J. P. Sommadossi, M. Seifer, D. Standring, M. Tausek, G. Gosselin and P. La Colla, *Antiviral Chem. Chemother.*, 2007, **18**, 225–242.
- N. Goris, A. De Palma, J. F. Toussaint, I. Musch, J. Neyts and K. De Clercq, *Antiviral Res.*, 2007, **73**, 161–168.
- E. Murakami, C. Niu, H. Bao, H. M. Micolochick Steuer, T. Whitaker, T. Nachman, M. A. Sofia, P. Wang, M. J. Otto and P. A. Furman, *Antimicrob. Agents Chemother.*, 2008, **52**, 458–464.
- K. Yamagami, A. Fujii, M. Arita, T. Okumoto, S. Sakata, A. Matsuda, T. Ueda and T. Sasaki, *Cancer Res.*, 1991, **51**, 2319–2323.
- J. T. Hwang and M. M. Greenberg, *J. Org. Chem.*, 2001, **66**, 363–369.
- C. Kessler, *J. Biotechnol.*, 1994, **35**, 165–189.
- N.-S. Li, J. Lu and J. A. Piccirilli, *Org. Lett.*, 2007, **9**, 3009–3012.
- O. Mitsunobu, *Synthesis*, 1981, 1–28.
- M. S. Motawia, J. Wengel, A. E. S. Abdelmegid and E. B. Pedersen, *Synthesis*, 1989, 384–387.
- N.-S. Li and J. A. Piccirilli, *Synthesis*, 2005, 2865–2870.
- N.-S. Li and J. A. Piccirilli, *J. Org. Chem.*, 2006, **71**, 4018–4020.
- Y. Furukawa and M. Honjo, *Chem. Pharm. Bull.*, 1968, **16**, 1076–1080.
- M. Imazawa and F. Eckstein, *J. Org. Chem.*, 1979, **44**, 2039–2041.
- R. Zou and M. J. Robins, *Can. J. Chem.*, 1987, **65**, 1436–1437.
- J.-D. Ye, X.-M. Liao and J. A. Piccirilli, *J. Org. Chem.*, 2005, **70**, 7902–7910.
- J. M. Chong and K. K. Sokoll, *Org. Prep. Proced. Int.*, 1993, **25**, 639–647.
- W. Bourgeois and F. Seela, *J. Chem. Soc., Perkin Trans. 1*, 1991, 279–283.
- E. A. Meade, L. L. Wotring, J. C. Drach and L. B. Townsend, *J. Med. Chem.*, 1992, **35**, 526–533.
- D. R. Williams, D. L. Brown and J. W. Benbow, *J. Am. Chem. Soc.*, 1989, **111**, 1923–1925.