

New tetrasubstituted thioureas containing the 1-iminoethyl moiety*

A. N. Proshin,^{a*} T. P. Trofimova,^b and S. O. Bachurin^a

^a*Institute of Physiologically Active Compounds, Russian Academy of Sciences,
1 Severnyi proezd, 142432 Chernogolovka, Moscow Region, Russian Federation.*

Fax: +7 (496) 524 9508. E-mail: proshin@ipac.ac.ru

^b*Department of Chemistry, M. V. Lomonosov Moscow State University,
1 Leninskie Gory, 119992 Moscow, Russian Federation*

The reactions of 1,1,3-trisubstituted thioureas with acetonitrile afforded the corresponding tetrasubstituted thioureas containing the *N*-(1-iminoethyl) moiety.

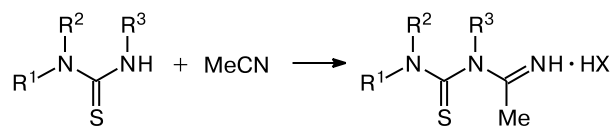
Key words: thioureas, acetonitrile, amidines.

Thioureas belong to a class of compounds that hold promise in chemical and pharmacological aspects. In medicinal chemistry, thioureas are used in the synthesis of neuroactive acyclic isothioureas^{1,2} and various sulfur-containing heterocycles bearing the pharmacophoric isothio-uronium group.³ Thioureas can suppress HIV reverse transcriptase,⁵ exhibit antifungal⁵ and antibacterial⁶ properties, and inhibit NO synthase.⁷

The aim of the present study was to search for new reactions giving tetrasubstituted thioureas. We developed a convenient one-pot method for the synthesis of the previously unknown tetrasubstituted thioureas containing the 1-iminoethyl group as one of *N*-substituents. This method is based on the reaction of 1,1,3-trisubstituted thioureas **1a–g** with acetonitrile. The reaction can be performed with the use of aqueous solutions of hydrochloric or hydrobromic acid (method *A*, products **2a–f**), as well as under milder conditions with the use of triphenylphosphine and

carbon tetrachloride (method *B*, product **2g**). The reaction produces acetimidoyl halides, which react with thiourea to give the corresponding iminoethyl derivatives **2** (Scheme 1).

Scheme 1



1a–g

2a–g

Compound	R ¹	R ²	R ³	X
a	Me	Me	Ph	Br
c	Me	Me	4-Me ₂ CHC ₆ H ₄	Cl
d	Me	Me	4-Me ₂ CHC ₆ H ₄	Br

Compound	NR ¹ R ²	R ³	X
b		Ph	Br
e		4-Me ₂ CHC ₆ H ₄	Cl
f		4-Me ₂ CHC ₆ H ₄	Br
g		4-Me ₂ CHC ₆ H ₄	Cl

Conditions: *A*, HX; *B*, Ph₃P, CCl₄.

The structures of the resulting compounds were confirmed by the X-ray diffraction data for thiourea **2d** (Fig. 1).

Experimental

The ¹H NMR spectra were recorded on a Bruker CXP-200 spectrometer operating at 200 MHz in DMSO-*d*₆ with tetrame-

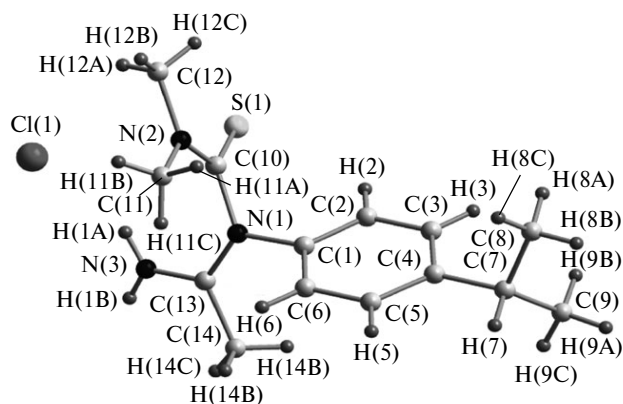


Fig. 1. Molecular structure of compound **2d** in the crystal.

* Dedicated to Academician of the Russian Academy of Sciences O. M. Nefedov on the occasion of his 80th birthday.

thylsilane as the internal standard. The electrospray ionization mass spectra were obtained on a Shimadzu LCMS-2010A quadrupole mass spectrometer. The elemental analysis was performed on a Carlo-Erba CHN analyzer.

1-(1-Iminoethyl)thioureas 2a–f (general procedure). Method A. The starting thiourea **1a–f** (1 mmol) was dissolved with stirring in acetonitrile (3–4 mL). Then an aqueous solution of hydrochloric or hydrobromic acid (1 mmol) was added. The reaction mixture was stirred for 2–4 h and left for 16 h. The crystalline precipitate that formed was filtered off and washed with diethyl ether.

1-(1-Iminoethyl)-3,3-dimethyl-1-phenylthiourea hydrobromide (2a). The yield was 60%, m.p. 143–144 °C. Found (%): C, 44.02; H, 5.13; N, 14.07; S, 10.43. $C_{11}H_{15}N_3S \cdot HBr$. Calculated (%): C, 43.71; H, 5.34; N, 13.90; S, 10.61. 1H NMR, δ : 2.30 (s, 3 H, CH_3); 3.33 (s, 3 H, CH_3); 3.50 (s, 3 H, CH_3); 7.43–7.95 (m, 5 H, Ar); 9.75 (br.s, 1 H, NH); 10.85 (br.s, 1 H, N^+H).

N-(1-Iminoethyl)-N-phenylpyrrolidine-1-carbothioamide hydrobromide (2b). The yield was 75%, m.p. 137–138 °C. Found (%): C, 47.28; H, 5.11; N, 13.06; S, 9.38. $C_{13}H_{17}N_3S \cdot HBr$. Calculated (%): C, 47.57; H, 5.53; N, 12.80; S, 9.77. 1H NMR, δ : 1.80–2.20 (m, 4 H, CH_2CH_2); 2.32 (s, 3 H, CH_3); 3.40–4.30 (m, 4 H, CH_2NCH_2); 7.52–7.83 (m, 5 H, Ar); 9.73 (br.s, 1 H, NH); 10.86 (br.s, 1 H, N^+H).

1-(1-Iminoethyl)-1-(4-isopropylphenyl)-3,3-dimethylthiourea hydrochloride (2c). The yield was 65%, m.p. 168–169 °C. Found (%): C, 55.81; H, 7.12; N, 13.72; S, 10.24. $C_{14}H_{21}N_3S \cdot HCl$. Calculated (%): C, 56.08; H, 7.40; N, 14.01; S, 10.69. 1H NMR, δ : 1.23 (d, 6 H, $CH(CH_3)_2$, $J = 7.3$ Hz); 2.25 (s, 3 H, CH_3); 2.95 (m, 1 H, $CH(CH_3)_2$); 3.42 (s, 3 H, CH_3); 3.50 (s, 3 H, CH_3); 7.37 (d, 2 H, Ar, $J = 8.6$ Hz); 7.75 (d, 2 H, Ar, $J = 8.6$ Hz); 9.71 (br.s, 1 H, NH); 11.30 (br.s, 1 H, N^+H).

1-(1-Iminoethyl)-1-(4-isopropylphenyl)-3,3-dimethylthiourea hydrobromide (2d). The yield was 85%, m.p. 159–160 °C. Found (%): C, 48.52; H, 6.14; N, 12.44; S, 9.12. $C_{14}H_{21}N_3S \cdot HBr$. Calculated (%): C, 48.84; H, 6.404; N, 12.20; S, 9.31. 1H NMR, δ : 1.28 (d, 6 H, $CH(CH_3)_2$, $J = 7.3$ Hz); 2.25 (s, 3 H, CH_3); 3.02 (m, 1 H, $CH(CH_3)_2$); 3.40 (s, 3 H, CH_3); 3.50 (s, 3 H, CH_3); 7.40 (d, 2 H, Ar, $J = 8.6$ Hz); 7.75 (d, 2 H, Ar, $J = 8.6$ Hz); 8.50–11.20 (br.s, 2 H, N^+H+NH).

N-(1-Iminoethyl)-N-(4-isopropylphenyl)piperidine-1-carbothioamide hydrochloride (2e). The yield was 74%, m.p. 178–180 °C. Found (%): C, 60.32; H, 7.94; N, 12.59; S, 9.08. $C_{17}H_{25}N_3S \cdot HCl$. Calculated (%): C, 60.07; H, 7.71; N, 12.36; S, 9.43. 1H NMR, δ : 1.20 (d, 6 H, $CH(CH_3)_2$, $J = 7.3$ Hz); 1.30–1.80 (m, 6 H, $CH_2CH_2CH_2$); 2.25 (s, 3 H, CH_3); 2.95 (m, 1 H, $CH(CH_3)_2$); 3.60–4.50 (m, 4 H, CH_2NCH_2); 7.42 (d, 2 H, Ar, $J = 8.6$ Hz); 7.70 (d, 2 H, Ar, $J = 8.6$ Hz); 9.80 (br.s, 1 H, NH); 11.20 (br.s, 1 H, N^+H).

N-(1-Iminoethyl)-N-(4-isopropylphenyl)piperidine-1-carbothioamide hydrobromide (2f). The yield was 86%, m.p. 175–176 °C. Found (%): C, 53.44; H, 6.68; N, 10.75; S, 8.12. $C_{17}H_{25}N_3S \cdot HBr$. Calculated (%): C, 53.12; H, 6.82; N, 10.93; S, 8.34. 1H NMR, δ : 1.18 (d, 6 H, $CH(CH_3)_2$, $J = 7.3$ Hz); 1.45–1.95 (m, 6 H, $CH_2CH_2CH_2$); 2.15 (s, 3 H, CH_3); 2.95 (m, 1 H, $CH(CH_3)_2$); 3.60–4.60 (m, 4 H, CH_2NCH_2); 7.35 (d, 2 H, Ar, $J = 8.6$ Hz); 7.65 (d, 2 H, Ar, $J = 8.6$ Hz); 9.55 (br.s, 1 H, NH); 10.60 (br.s, 1 H, N^+H).

3-Hydroxymethyl-N-(1-iminoethyl)-N-(4-isopropylphenyl)piperidine-1-carbothioamide hydrochloride (2g). Method B. A mixture of 3-hydroxymethyl-N-(4-isopropylphenyl)piperidine-1-carbothioamide (0.1 g, 0.34 mmol) and triphenylphos-

phine (0.1 g, 0.38 mmol) was dissolved in acetonitrile (2.6 mL). Then CCl_4 (0.5 mL) was added, and the mixture was kept for 16 h. The crystalline precipitate that formed was filtered off and re-crystallized from acetone. Product **2g** was obtained in a yield of 0.07 g (63%). M.p. 125–126 °C. Found (%): C, 58.65; H, 7.43; N, 11.30; S, 8.61. $C_{18}H_{27}N_3OS \cdot HCl$. Calculated (%): C, 58.44; H, 7.63; N, 11.36; S, 8.67. 1H NMR, δ : 1.25 (d, 6 H, $C(CH_3)_2$); 1.3–1.9 (m, 5 H, CH_2CH_2CH); 2.25 (br.s, 3 H, $HN=CCH_3$); 2.80 (m, 1 H, $CH(CH_3)_2$); 3.0–3.4 (m, 4 H, $CHNCH$, CH_2O); 4.30–4.60 (m, 3 H, $CHNCH$, OH); 7.05–7.50 (m, 4 H, Ar); 9.50–10.30 (br.s, 2 H, $NH+N^+H$). MS, m/z : 333 $[M]^+$, 292 $[M - CH_3=NH]^+$, 249 $[M - CH_3=NH-C(CH_3)_3]^+$, 232 $[M - CH_3=NH-C(CH_3)_3-OH]^+$, 219 $[M - CH_3=NH-C(CH_3)_3-CH_2OH]^+$.

Principal crystallographic data for compound 2d. The unit cell parameters: $a = 14.9503(5)$ Å, $b = 9.2691(2)$ Å, $c = 12.1751(4)$ Å, $\alpha = 90.00^\circ$, $\beta = 110.553(2)^\circ$, $\gamma = 90.00^\circ$, $V = 1579.78(8)$ Å³, space group $P2_1/c$, $Z = 2$. The X-ray diffraction data were collected on a KAPPA APEX II single-crystal diffractometer (Mo-K α radiation, $\lambda = 0.71073$ Å). Single crystals were attached to a glass fiber and mounted on a goniometer. The crystals were cooled with liquid nitrogen at 100 K. The unit cell parameters were refined based on the total X-ray diffraction data set. The experimental intensities were corrected for absorption.⁸ The structure was solved by direct methods with the use of the SHELXTL program package⁹ and refined by the full-matrix least-squares method based on F^2 using the total X-ray data set with anisotropic displacement parameters for all nonhydrogen atoms. The structure was deposited to the Cambridge Crystallographic Data Centre (CCDC 763757).

This study was financially supported by the Russian Foundation for Basic Research (Project No. 11-03-00863-a) and the Ministry of Education and Science of the Russian Federation (State Contract No. 14.740.11.0810).

References

1. S. E. Tkachenko, A. N. Proshin, M. E. Dmitriev, N. N. Lermontova, V. V. Grigor'ev, S. O. Bachurin, N. S. Zefirov, Pat. RF No. 2252936, 2005.
2. G. L. Perlovich, A. N. Proshin, T. V. Volkova, S. V. Kurkov, V. V. Grigor'ev, L. N. Petrova, S. O. Bachurin, *J. Med. Chem.*, 2009, **2** (7), 1845.
3. J. Li, Z. Tan, S. Tang, I. Hewlett, R. Pang, M. He, S. He, B. Tian, K. Chen, M. Yang, *Bioorg. Med. Chem.*, 2009, **17**, 3177.
4. A. R. Katritzky, S. Ledoux, R. M. Witek, S. K. Nair, *J. Org. Chem.*, 2004, **69**, 2976.
5. M. Eweis, S. S. Elkholy, M. Z. Elsabee, *Int. J. Biol. Macromol.*, 2006, **38**, 1.
6. H. Arslan, N. Duran, G. Borekci, C. Koray Ozer, C. Akbay, *Molecules*, 2009, **14**, 519.
7. C. L. M. Goodyer, E. C. Chinje, M. Jaffar, I. J. Stratford, M. D. Threadgill, *Bioorg. Med. Chem.*, 2003, **11**, 4189.
8. Bruker, *SADABS*. Madison (WI, USA): Bruker AXS Inc., 2004.
9. G. M. Sheldrick, *SHELXL-97, Program for the Refinement of Crystal Structures*, University of Göttingen, Göttingen, Germany, 1997.

Received September 21, 2011;
in revised form November 7, 2011