## First stereoselective total synthesis of macrocarpal C: structure elucidation of macrocarpal G

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The first stereoselective total synthesis of macrocarpal C is achieved *via* a coupling reaction of a silyl dienol ether with a novel hexasubstituted benzene chromium tricarbonyl complex as an optically active benzyl cation equivalent, thereby clarifying the identity of macrocarpal C and G.

Macrocarpals<sup>1,2</sup> are structurally characterized by fusion of isopentylphloroglucinol dialdehyde to various sesquiterpene skeletons and are known to exhibit various interesting biological activities, such as antibacterial activity, 2a,c,e,g inhibitory activity of HIV reverse transcriptase, 2b aldose reductase 2d,e and glucosyl transferase. 2e,g Since the original isolation of macrocarpal A 1 from Eucalyptus macrocarpa in 1990,<sup>2a</sup> a number of macrocarpals have been isolated from Eucalyptus species.<sup>2b-g</sup> Complete structures have been elucidated for macrocarpals A (1), B (2) and C (3) using X-ray diffraction studies and spectral and chemical investigations, and the absolute stereochemistries of 1 and 3 have been determined by a modified Mosher's method.<sup>2b</sup> Macrocarpal G 4,<sup>2c</sup> however, has been assigned the same planar structure as 3,† and quite a few macrocarpals have not been designated relative stereochemistries. In this communication, we report the first and highly stereoselective total synthesis of macrocarpal C 3, thereby clarifying its identity with respect to macrocarpal G 4.

Ar = 2,4,6-trihydroxy-3,5-diformylphenyl

As illustrated in the retrosynthetic analysis of macrocarpal C 3 (Scheme 1), our synthetic approach began from the previously reported tricyclic enone 6,3 which we prepared from the commercially available and inexpensive (+)-3-carene 5. The coupling reaction of the sesquiterpene moiety 6 with the isopentylphloroglucinol dialdehyde part is a crucial step requiring high stereoselectivity at both the benzylic and C11 positions. To this end, we planned the Lewis acid mediated coupling reaction of silyldienol ether 7 with a novel hexasubstituted benzene chromium tricarbonyl complex 8 as an optically active benzyl cation equivalent. Our previous studies suggested that an electrophile would be introduced stereoselectively at the C11 position from the less hindered  $\beta$ -side of 7.3 In connection with stereochemical control at the benzylic position, it is known that, with some nucleophiles, S<sub>N</sub>1-type carbon-carbon bond formation via the Cr(CO)<sub>3</sub>-stabilized carbonium ion proceeds with stereochemical retention at the benzylic position.<sup>4</sup> Hence, the desired coupling product 9,

which possesses the requisite functionalities and desired stereochemistry for the subsequent manipulations toward macrocarpal C 3, would be obtained selectively after com-

Scheme 1

plexation. Scheme 2 outlines the synthesis of the chiral chromium complex **8** starting from commercially available 1,3,5-trimethoxybenzene **10**. Friedel–Crafts acylation of **10** with isovaleryl chloride, followed by LAH reduction led to alcohol **12** *via* ketone **11**. After complexation of **12** with Cr(CO)<sub>6</sub>, stepwise introduction of two methoxycarbonyl groups *via* direct nuclear lithiation<sup>5</sup> to the aromatic ring furnished diester *rac-***13**. The resolution of this racemic benzyl alcohol was then examined. Diastereomeric carbamates derived from the CuCl assisted<sup>6</sup>

OMe 
$$R^2$$
  $R^1$   $R^1$   $R^1$   $R^2$   $R^2$ 

Scheme 2 Reagents and conditions: i, isovaleryl chloride,  $AlCl_3$ ,  $CH_2Cl_2$ , 0 °C, 81%; ii, LAH,  $Et_2O$ , 0 °C, 91%; iii,  $Cr(CO)_6$ ,  $Bu^n_2O-1$ ,4-dioxane-n-heptane (5:5:1), 120 °C, 43% (92% based on conversion); iv,  $Bu^nLi$ , TMEDA, THF, -78 °C, then  $CO_2$ , -78 °C, then TMSCHN $_2$ ,  $C_6H_6$ -MeOH (4:1); v, LDA, TMEDA, THF, -50 °C, then  $CO_2$ , -78 °C, then TMSCHN $_2$ ,  $C_6H_6$ -MeOH (4:1), 75% overall; vi, (R)-(-)-1-(1-naphthyl)ethyl isocyanate, CuCl, DMF, 99%, then separation; vii, CuCl-CuCl

**Scheme 3** Reagents and conditions: i, Bu'Me<sub>2</sub>SiOTf, Et<sub>3</sub>N, Et<sub>2</sub>O-CH<sub>2</sub>Cl<sub>2</sub> (1:1), 98%; ii, **8**, ZnCl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>; iii, CAN, MeOH, 0 °C, 61% from **8** 

reaction of rac-13 with (R)-(-)-1-(1-naphthyl)ethyl isocyanate<sup>7</sup> were separated by simple chromatography on silica gel. Hydrolysis of the polar carbamate 14 was achieved by treatment with BF<sub>3</sub>·Et<sub>2</sub>O in hydrous MeCN to provide the alcohol (S)-13,  $[\alpha]_{20}^{26}$  -21.9 (>98% ee) without racemization.‡ Finally, (S)-13 was converted to the chloroacetate 8, which was expected to have the required reactivity.<sup>8</sup>

With the desired chiral complex **8** as an aromatic side-chain unit in hand, we subjected **8** to the coupling reaction (Scheme 3). For this purpose, the enone **6** was converted to the *tert*-butyldimethylsilyl dienol ether **7**. A 1.5-fold excess of this intermediate was then coupled with the chiral complex **8** in the presence of ZnCl<sub>2</sub><sup>9</sup> to afford **9** stereoselectively§ after decomplexation with ceric ammonium nitrate (CAN).

Since the stereoselective route to the promising precursor of macrocarpal C 9 had already been developed, we further pursued the total synthesis (Scheme 4). Catalytic hydrogenation of the enone 9 afforded desilylated ketone 15 with the desired stereochemistry.3 NaBH<sub>4</sub> reduction of 15 followed by acetylation of the primary hydroxy group led to monoacetate 17¶ via diol 16. The application of modified Grieco's protocol<sup>10</sup> allowed dehydration of the C10 secondary hydroxy group of 17. Deacetylation of the resulting product followed by catalytic hydrogenation afforded alcohol 18, which was subjected to dehydration again, furnishing exo-olefin 19. DIBAL-H reduction of diester 19 followed by oxidation afforded trimethyl macrocarpal C 21 via diol 20. Finally, the cleavage of all three methyl ethers was fully achieved by our original method (10 equiv. of 4-MeC<sub>6</sub>H<sub>4</sub>SLi, 50 equiv. of HMPA, toluene, reflux) to furnish macrocarpal C 3, which was identical to a natural authentic sample (1H and 13C NMR).\*\* Moreover, the synthetic sample was found to be identical to natural macrocarpal G 4.††

Ar' = 2,4,6-trimethoxy-3,5-bis(methoxycarbonyl)phenyl

**Scheme 4** Reagents and conditions: i,  $H_2$  (5 atm), 10% Pd–C, MeOH, 88%; ii, NaBH<sub>4</sub>, MeOH, 96%; iii, Ac<sub>2</sub>O, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, 100%; iv, 2-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>-SeCN, Bu<sup>n</sup><sub>3</sub>P, THF, sealed tube, 75 °C, then 30% aq.  $H_2O_2$ 0 °C; v, NaOMe, MeOH, 64% from **17**; vi,  $H_2$  (5 atm), 10% Pd–C, MeOH, 100%; vii, 2-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>SeCN, Bu<sup>n</sup><sub>3</sub>P, THF, 50 °C, then 30% aq.  $H_2O_2$ , 0 °C, 77%; viii, DIBAL-H, toluene, -78 °C, 100%; ix,  $Pr^a_4NRuO_4$ , 4-methylmorpholine *N*-oxide, molecular sieves 4 Å, MeCN, 87%; x, 4-MeC<sub>6</sub>H<sub>4</sub>SLi, HMPA, toluene, reflux, 58%

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## **Footnotes and References**

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† The <sup>1</sup>H and <sup>13</sup>C NMR spectra of macrocarpals C **3** and G **4** were measured in different solvents, which made comparison difficult. In ref. 1, however, it has been considered that these were diastereomers due to the difference between their physicochemical properties.

‡ Enantiomeric purity was determined by ¹H NMR analysis of Mosher ester derivatives.¹¹ Absolute stereochemistry was assigned by Mosher's method.¹²

§ A small amount (2%) of the benzylic epimer of **9** was obtained, but neither the C11 epimer nor the regioisomer was isolated.

¶ The structural assignment of **17** was confirmed by X-ray crystallographic analysis of its C10 carbonylimidazolide derivative **22**. *Crystal data* for **22**:  $C_{39}H_{54}N_2O_{11}$ , M=726.86, orthorhombic, space group  $P2_12_12_1$ , a=16.723(2), b=19.979(2), c=11.547(2) Å, V=3858.0(9) ų, Z=4,  $D_c=1.251$  g cm<sup>-3</sup>,  $\mu$ (Cu-K $\alpha$ ) = 7.13 cm<sup>-1</sup>, F(000)=1560, T=293 K. The structure was solved by direct methods and refined by full-matrix least-squares to R=0.052 and  $R_w=0.067$  using 2185 reflections with  $F_o>3\sigma(F_o)$ . CCDC 182/601.

 $\parallel$  A preliminary study suggested that only bis-demethylation would occur by the use of sodium salt (4-MeC<sub>6</sub>H<sub>4</sub>SNa), as previously reported.<sup>13</sup>

\*\* Synthetic 3 displayed <sup>1</sup>H and <sup>13</sup>C NMR spectra that were indistinguishable from those of the natural isolate and showed the following optical properties:  $[\alpha]_D^{25} - 22.1$  (c 0.150, EtOH). A small rotation,  $[\alpha]_D^{24} - 3.0$  (c 0.92, EtOH), was originally reported for 3 that had been isolated from E.  $globulus.^{2b}$  This rotation, however, is believed to be erroneous due to contamination of the natural sample (M. Nishizawa, personal communication, April 15, 1997).

†† Synthetic 3 exhibited spectroscopic data ( $^{1}$ H,  $^{13}$ C, IR) identical to those for natural macrocarpal G **4**. The rotation for synthetic 3,  $[\alpha]_{\rm D}^{25}$  –24.7 (c 0.135, MeOH), corresponded closely to the rotation reported for **4**,  $[\alpha]_{\rm D}$  –27.1 (c 0.59, MeOH). $^{2c}$ 

- 1 For a review of bioactive acylphloroglucinol derivatives from *Eucalyptus* species, see: E. L. Ghisalberti, *Phytochemistry*, 1996, **41**, 7.
- 2 (a) M. Murata, Y. Yamakoshi, S. Homma, K. Aida, K. Hori and Y. Ohashi, Agric. Biol. Chem., 1990, 54, 3221; (b) M. Nishizawa, M. Emura, Y. Kan, H. Yamada, K. Ogawa and N. Hamanaka, Tetrahedron Lett., 1992, 33, 2983; (c) Y. Yamakoshi, M. Murata, A. Shimizu and S. Homma, Biosci. Biotech. Biochem., 1992, 56, 1570; (d) M. Murata, Y. Yamakoshi, S. Homma, K. Arai and Y. Nakamura, Biosci. Biotech.Biochem., 1992, 56, 2062; (e) K. Osawa, H. Yasuda, H. Morita, K. Takeya and H. Itokawa, Phytochemistry, 1995, 40, 183; (f) I. P. Singh and H. Etoh, Biosci. Biotech. Biochem., 1995, 59, 2330; (g) K. Osawa, H. Yasuda, H. Morita, K. Takeya and H. Itokawa, J. Nat. Prod., 1996, 59, 823.
- 3 T. Tanaka, K. Maeda, H. Mikamiyama, Y. Funakoshi, K. Uenaka and C. Iwata, *Tetrahedron*, 1996, **52**, 4257.
- 4 M. Uemura, T. Kobayashi, K. Isobe, T. Minami and Y. Hayashi, *J. Org. Chem.*, 1986, **51**, 2859 S. G. Davies and T. J. Donohoe, *Synlett*, 1993, 323
- 5 M. F. Semmelhack, J. Bisaha and M. Czarny, J. Am. Chem. Soc., 1979, 101, 768; H.-G. Schmalz, T. Volk, D. Bernicke and S. Huneck, Tetrahedron, 1997, 53, 9219.
- 6 M. E. Duggan and J. S. Imagire, Synthesis, 1989, 131.
- 7 W. H. Pirkle and J. R. Hauske, J. Org. Chem., 1977, 42, 1839.
- 8 W. R. Roush and C. K. Wada, Tetrahedron Lett., 1994, 35, 7347.
- 9 M. T. Reetz and M. Sauerwald, Tetrahedron Lett., 1983, 24, 2837.
- P. A. Grieco, S. Gilman and M. Nishizawa, J. Org. Chem., 1976, 41, 1485.
- 11 J. A. Dale, D. L. Dull and H. S. Mosher, J. Org. Chem., 1969, 34, 2543.
- 12 J. A. Dale and H. S. Mosher, J. Am. Chem. Soc., 1973, 95, 512.
- 13 C. Hansson and B. Wickberg, Synthesis, 1976, 191.

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