



Synthetic Communications

An International Journal for Rapid Communication of Synthetic Organic Chemistry

ISSN: 0039-7911 (Print) 1532-2432 (Online) Journal homepage: <http://www.tandfonline.com/loi/lcyc20>

Synthesis of (E) α ,2,4-Trinitrostilbenes from (E) 2,4-Dinitrostilbenes

S. Saravanan & P. C. Srinivasan

To cite this article: S. Saravanan & P. C. Srinivasan (2003) Synthesis of (E) α ,2,4-Trinitrostilbenes from (E) 2,4-Dinitrostilbenes, Synthetic Communications, 33:8, 1261-1266, DOI: [10.1081/SCC-120018685](https://doi.org/10.1081/SCC-120018685)

To link to this article: <http://dx.doi.org/10.1081/SCC-120018685>



Published online: 15 Aug 2006.



Submit your article to this journal [↗](#)



Article views: 25



View related articles [↗](#)



Citing articles: 2 View citing articles [↗](#)



SYNTHETIC COMMUNICATIONS®

Vol. 33, No. 8, pp. 1261–1266, 2003

Synthesis of (*E*) α ,2,4-Trinitrostilbenes from (*E*) 2,4-Dinitrostilbenes

S. Saravanan and P. C. Srinivasan*

Department of Organic Chemistry, University of
Madras, Guindy Campus, Chennai, India

ABSTRACT

A convenient synthesis of (*E*) α ,2,4-trinitrostilbenes **2a–d** from (*E*) 2,4-dinitrostilbenes **1a–e** by direct nitration is reported. Isomerization of the *trans* geometry of the reactant stilbenes to *cis* in the products was found. The structures of stilbenes **2a**, **2c** and **2d** were confirmed by XRD.

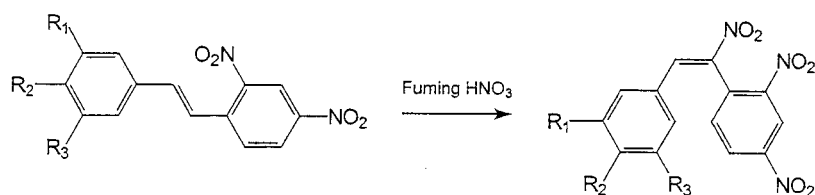
*Correspondence: P. C. Srinivasan, Department of Organic Chemistry, University of Madras, Guindy Campus, Chennai 600 025, India; E-mail: pancol2000@yahoo.co.in.



Aromatic nitro compounds are important synthetic targets due to their uses as explosives, dye intermediates and battery cathodes.^[1] α -Nitrostilbenes^[2] belong to a class of important nitro compounds as potential precursors for biologically active β -phenylethylamines and Michael acceptors.^[3] A survey of literature^[4] shows that only a few styrenes undergo direct nitration to give β -nitrostyrenes, while there are no reports on side chain nitration of stilbenes. α -Nitrostilbenes are synthesized by condensing α -nitrotoluenes with aromatic aldehydes. Recent reports on the synthesis of α -2,4-trinitrotoluene^[5] and α -nitroalkylbenzenes^[6] also prompted us to report our synthesis of (*E*) α ,2,4-trinitrostilbenes here.

We report here the first synthesis of (*E*) α ,2,4-trinitrostilbenes **2a–d** by direct nitration of easily available^[7] (*E*) 2,4-dinitrostilbenes **1a–e** in 55–60% yield, using fuming nitric acid^[8] (Sch. 1). The nitration was found to be temperature independent in case of stilbenes **1a**, **1b**, and **1e**. Whereas in the case of **1c** it gives exclusively **2b** at 25°C and **2c** at 90°C. In the case of **1d** only side chain nitrated product **2b** at 25°C and side chain as well as nuclear nitrated product **2c** at 90°C was obtained (Table 1).

It was found that the *trans* geometry of the reactant stilbenes **1a–e** (confirmed by ¹H NMR (**1a–e**) and XRD in case of **1e** alone) isomerized to *cis* in the products with respect to the aryl rings in the reactant. The *cis* geometry of the products was confirmed by XRD in the case of stilbenes **2a**, **2c** and **2d** (Fig. 1). This isomerization may be due to the favorable π - π interaction between the two aryl rings during nitration.



1a R₁=R₂=R₃=H

1b R₁=NO₂; R₂=R₃=H

1c R₁=R₃=H; R₂=CH₃

1d R₁=NO₂; R₂=CH₃; R₃=H

1e R₁=R₃=H; R₂=OCH₃

2a R₁=NO₂; R₂=R₃=H

2b R₁=NO₂; R₂=CH₃; R₃=H

2c R₁=R₃=NO₂; R₂=CH₃

2d R₁=NO₂; R₂=OCH₃; R₃=H

Scheme 1.

**(E) α ,2,4-Trinitrostilbenes****1263****Table 1.**

Reactant	Product		M.p. (°C)	Yield (%)
	25°C	90°C		
1a	2a	2a	200	55
1b	2a	2a	200	55
1c	2b	2c	2b , 183	58
1d	2b	2c	2c , 210	60
1e	2d	2d	215	60

Further work is in progress to study the electrochemical properties of stilbenes **2a–d**.

EXPERIMENTAL**Synthesis of 4'-Methyl-2,3',4-trinitrostilbene (1d)**

A mixture of 2,4-dinitrotoluene^[7] (5 mM, 0.9 g) and 4-methyl-3-nitrobenzaldehyde^[9,10] (5 mM, 0.8 g) and pyrrolidine (0.106 g, 1.5 mM) was stirred for 5 h in dry DMF for 5 h at 25–30°C. Then it was poured over crushed ice. The crude product was filtered, washed with water and then with methanol and dried over anhydrous calcium chloride. Yield 80%; m.p. 200°C. IR (KBr): 3085, 1598, 1526, 1343 cm⁻¹. ¹H NMR (CDCl₃) 200 MHz: δ 8.8 (d, 1H, J = 2.3 Hz), 8.4 (d, 1H, J = 8.0 Hz), 8.1 (d, 1H, J = 2.5 Hz), 7.9 (d, 1H, J = 8.1 Hz), 7.7–7.5 (m, 3H, vinylic and aromatic H), 7.3 (d, 1H, J = 8 Hz), 2.6 (s, 3H). MS: m/z 329 (24% M⁺), 312 (23%), 164 (90%), 55 (100%).

Representative Procedure for the Synthesis of (E) 4'-Methyl- α ,2,3',4,5'-pentanitrostilbene (2c)

To 5 mM (2.095 g) of (E) 4'-methyl-2,3',4-trinitrostilbene 5 mL of fuming HNO₃ was added slowly with stirring. Then the reaction mixture was kept at 90°C for 5 min. After cooling to 25–30°C, the reaction mixture was poured over crushed ice, the crude product was filtered, washed with water and dessicated over anhydrous CaCl₂ and recrystallized from acetone. Yield 60%. IR (KBr): 3130, 1649, 1602, 1517, 1326 cm⁻¹. ¹H NMR (DMSO-*d*₆) 400 MHz: δ 2.4 (s, 3H), 7.96 (d, 1H, J = 8.5 Hz),

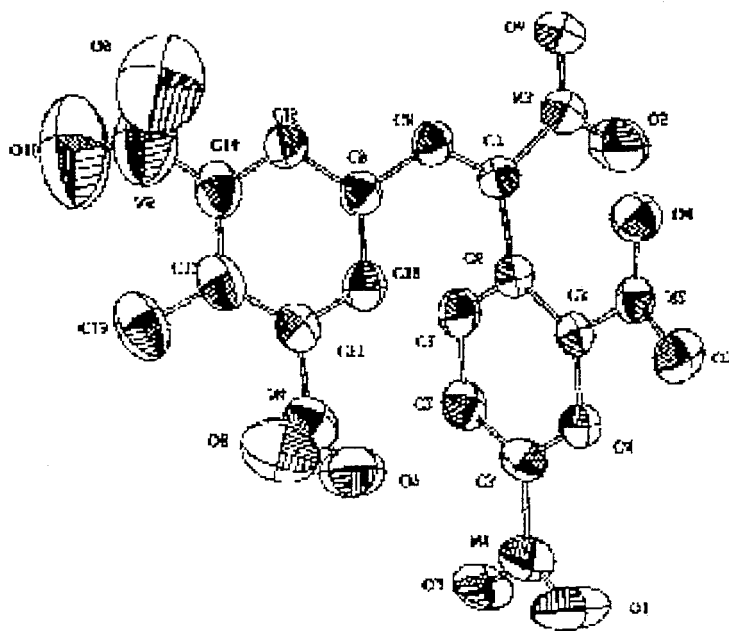
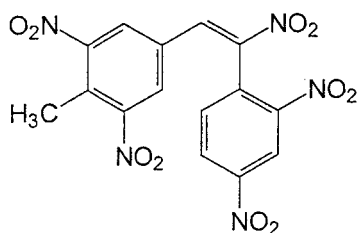


Figure 1. XRD structure of stilbene **2c**.

**(E) α ,2,4-Trinitrostilbenes****1265**

8.0 (s, 1H), 8.7 (m, 2H, Ar-H and vinylic-H), 9.1 (d, 1H, $J=3.05$ Hz). Anal. calcd. for $C_{15}H_9N_3O_{10}$; C 42.9, H 2.16, N 16.7; Found C 43.04, H 1.9, N 17.1. M.p. 210°C.

(E) α ,2,3',4-Tetranitrostilbene (2a)

Yield 55%. IR (KBr): 1660, 1606, 1533, 1332 cm^{-1} . 1H NMR ($CDCl_3$) 300 MHz: δ 9.22 (d, 1H, $J=2.1$ Hz), 8.55 (d, 1H, $J=8$ Hz), 8.48 (s, 1H), 8.32 (d, 1H, $J=8$ Hz), 8.1 (s, 1H). Anal. calcd. for $C_{14}H_8N_4O_8$; C 46.67, H 2.23, N 15.55; Found C 46.67, H 2.23, N 15.93. M.p. 200°C.

(E) 4'-Methyl- α ,2,3',4-tetranitrostilbene (2b)

Yield 58%. IR (KBr): 3125, 1650, 1600, 1526, 1340 cm^{-1} . 1H NMR ($CDCl_3$) 400 MHz: δ 9.22 (d, 1H, $J=1.96$ Hz), 8.56 (d, 1H, $J=8.32$ Hz), 8.42 (s, 1H), 7.70 (s, 1H), 7.58 (d, 1H, $J=8.28$ Hz), 7.29 (d, 1H, $J=8.32$ Hz), 7.14 (d, 1H, $J=8.3$ Hz), 2.7 (s, 3H). Anal. calcd. for $C_{15}H_{10}N_4O_8$; C 48.13, H 2.69, N 15.0; Found C 48.0, H 2.09, N 15.5. M.p. 183°C.

(E) 4'-Methoxy- α ,2,3',4-tetranitrostilbene (2d)

Yield 60%. IR (KBr): 3124, 1650, 1614, 1542, 1325 cm^{-1} . 1H NMR ($CDCl_3$) 400 MHz: δ 9.21 (d, 1H, $J=1.96$ Hz), 8.58 (d, 1H, $J=8.52$ Hz), 8.38 (s, 1H), 7.62 (d, 1H, $J=8.32$ Hz), 7.58 (d, 1H, $J=8.44$ Hz), 7.22 (d, 1H, $J=8.8$ Hz), 7.03 (d, 1H, $J=8.8$ Hz), 3.97 (s, 3H). Anal. calcd. for $C_{15}H_{10}N_4O_9$; C 46.16, H 2.58, N 14.35; Found C 46.38, H 2.46, N 14.59. M.p. 215°C.

ACKNOWLEDGMENT

The authors thank Dr. Paul Ratnasamy, Director; NCL, Pune for providing CHN data for the stilbenes 2a–d, Prof. D. Velmurugan, Dept. of Biophysics and Crystallography, University of Madras, for providing XRD data for stilbenes 1e, 2a, 2c and 2d, CSIR, New Delhi for financial support and the UGC-SAP Government of India for instrument grants to this department.



REFERENCES

1. (a) Sivasamy, K.; Rajeswari, S.; Dakashinamurthi, K. J. *Electrochem. Soc. India* **1988**, 37 (4), 343–347; (b) Renuka, R.; John Fredrick, V.; Saravanan, S.; Srinivasan, P.C. *Ind. J. Chem.* **2001**, 40A, 175–181.
2. Robertson, D.N. *J. Org. Chem.* **1960**, 25, 47–49.
3. Flintoft, R.J.; Buzby, J.C.; Tucker, J.A. *Tetrahedron Lett.* **1999**, 40, 4485–4488.
4. *Nitro Group in Organic Synthesis*; Noboru Ono; Wiley-VCH Pubn, 2001.
5. Kawakami, T.; Suzuki, H. *Tetrahedron Lett.* **1999**, 40, 1157–1160.
6. Vogel, E.M.; Buchwald, S.L. *J. Org. Chem.* **2002**, 67, 106–111.
7. Saravanan, S.; Srinivasan, P.C. *Syn. Commun.* **2001**, 31 (6), 24–28.
8. Fuming nitric acid was prepared by distilling 1:1 mixture (v/v) of 70% nitric acid and conc. sulphuric acid.
9. *Beilsteins Hand Book of Organic Chemistry*; Syst. No. 640, 1925; Vol. 7, 299.
10. Rashid, I.; Hasan, N.; Muhammad Zia Ul, H.; Farag Ali, M. *J. Chem. Soc. Pak.* **1997**, 19, 141–144.

Received in the Netherlands April 18, 2002