ISSN 1070-3632, Russian Journal of General Chemistry, 2012, Vol. 82, No. 6, pp. 1174–1177. © Pleiades Publishing, Ltd., 2012. Original Russian Text © M.B. Gazizov, R.A. Khairullin, A.A. Minnikhanova, A.I. Perina, O.G. Sinyashin, 2012, published in Zhurnal Obshchei Khimii, 2012, Vol. 82, No. 6, pp. 1040–1043.

> LETTERS TO THE EDITOR

## Synthesis of Imines with N-Containing Substituents and Their Reactions with Dialkylphosphorous Acids

M. B. Gazizov<sup>a</sup>, R. A. Khairullin<sup>a</sup>, A. A. Minnikhanova<sup>a</sup>, A. I. Perina<sup>a</sup>, and O. G. Sinyashin<sup>b</sup>

<sup>a</sup> Kazan State Technological University, ul. K. Marksa,68, Kazan, Tatarstan, 420015 Russia e-mail: mironov@iopc.ru

<sup>b</sup> Arbuzov Institute of Organic and Physical Chemistry, Kazan Scientific Centre, Russian Academy of Sciences, Kazan, Tatarstan, Russia

Received May 26, 2011

## **DOI:** 10.1134/S1070363212060242

The addition of hydrophosphoryl compounds to the double C=N bond of imines is one of the main methods of synthesis of the phosphorus-containing amino compounds [1]. The latter, possessing such useful properties as biological activity, can be obtained via modifying carbonyl and amine components of the imine. The analysis of publications shows that imines with *N*-containing substituents were not virtually used in these reactions, since they were little known. The aim of this work is the synthesis of new substituted imines and the study of their reactions with dialkylphosphorous acids.

The triazole fragment is a part of the biologically active compounds, which are widely used in medicine, pharmacology, microbiology, agriculture, photography and textile industries [2]. So one of the selected amines was 1,2,4-triazol-4-ylamine **Ia**. In recent years, the alkylamino- and dialkylamino-substituted aldehydes have become readily available [3]. We used 2,2-dimethyl-3-(ethylamino)propanal **IIa** as the carbonyl component. Since 1,2,4-triazol-4-ylamine is readily soluble in organic solvents, it was used as a diluted solution in ethanol. Sulfuric acid was used as a catalyst. The reaction of aldehyde **IIa** with the unsubstituted propan-1-amine **Ib** was carried out in the solvent-free conditions in the presence of solid potassium hydroxide. The structure of compounds **III** was confirmed by the <sup>1</sup>H NMR spectra.

$$R^{1}NH_{2} + R^{2}CHO \xrightarrow{H_{2}SO_{4},\Delta} R^{1}N=CHR^{2} + H_{2}O$$

$$I \qquad II$$

$$I, R^{1} = \bigvee_{N=-N}^{N} N \quad (a), n-Pr \quad (b); II, R^{2} = CMe_{2}CH_{2}NHEt \quad (a), i-Pr \quad (b); III, R^{1} = \bigvee_{N=-N}^{N} N \quad , R^{2} = i-Pr \quad (a), CMe_{2}CH_{2}NHEt \quad (b),$$

$$R^{1} = n-Pr, R^{2} = CMe_{2}CH_{2}NHEt \quad (c).$$

The reaction of imine **IIIa** with dimethylphosphorous acid **IVa** was carried out under the Pudovik reaction conditions. To a mixture of the starting compounds 5–6 drops of a sodium methoxide solution in methanol were added. The reaction mixture slightly warmed. After keeping the reaction mixture at room temperature for 3 days and processing it using common technique compound **Va** was isolated as crystals.

The reactions of imines **III** with dimethyl phosphite **IVa** was carried out without a catalyst by stirring the reactants. The reaction mixture warmed. Obviously, the ethylamino group plays the role of a nucleophilic catalyst. The reaction mixture was kept at room temperature for 5 days. Dimethyl [2,2-dimethyl-1-(*n*-propylamino)-3-(ethylamino)propyl]phosphonate **Vb** was isolated by the vacuum distillation.

Imine **IIIb** was the least active in the reactions with dialkylphosphorous acids. The reaction was activated by metallic sodium: To a benzene solution of acid **IVb** were added imine **IIIb** and 2–3 very small pieces of sodium

$$R^{1}N=CHR^{2} + (R^{3}O)_{2}PHO \longrightarrow R^{1}NHCHR^{2}P(O)(OR^{3})_{2}O$$
HI IV V  
IV, R<sup>3</sup> = Me (a), *i*-Pr (b); V, R<sup>1</sup> =  $N$  N, R<sup>2</sup> = *i*-Pr, R<sup>3</sup> = Me (a); R<sup>1</sup> = *n*-Pr, R<sup>2</sup> = CMe\_{2}CH\_{2}NHEt, R<sup>3</sup> = Me (b); R<sup>1</sup> =  $N$  N,  
R<sup>2</sup> = CMe\_{2}CH\_{2}NHEt, R<sup>3</sup> = *i*-Pr (c).  
Va + 2,4,6-(NO\_{2})\_{3}C\_{6}H\_{2}OH  $\longrightarrow N$  N + NH\_{2}CHP(O)(OMe)\_{2}  
2,4,6-(NO\_{2})\_{3}C\_{6}H\_{2}OH  $\longrightarrow N$  + NH\_{2}CHP(O)(OMe)\_{2}  
Vla  
Vb + 2[2,4,6-(NO\_{2})\_{3}C\_{6}H\_{2}OH]  $\longrightarrow EtN$  H<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>O(OMe)\_{2}  
+ NH<sub>2</sub>Pr-*n* 2X<sup>-</sup>  
Vlb  
Vc + HX  $\longrightarrow N$  N - NH<sub>2</sub>CHP(O)(OPr-*i*)\_{2}  
CMe\_{2}CH\_{2}NH\_{2}Et X<sup>-</sup>  
VIc, VId  
VI, X = 2,4,6-(NO\_{2})\_{3}C\_{6}H\_{2}O^{-} (b, c), CIO\_{4} (d).

metal. A slight self-heating was observed. The mixture was kept at room temperature for several days. After the removal of benzene the residue was recrystallized to give diisopropyl [2,2-diemethyl-1-(1,2,4trazol-4-ylamino)-3-(ethylamino)propyl]phosphonate Vc.

The composition and structure of compounds V were confirmed by the elemental analysis data, <sup>1</sup>H and <sup>31</sup>P NMR spectroscopy, as well as by their transformation into the corresponding picrates **VIa–VIc** and perchlorate **VId**.

Diaminoalkylphosphonate Vb reacts with picric acid to form dipicrate VIb. At the same time phosphonate Vc with two amino groups reacts with perchloric and picric acids to give monosalts VId and VIc, even when using a two-fold excess of these acids. Obviously, the 1,2,4-triazole group in compound Vc creates more steric hindrances at the N(III) atom than *n*-propyl moiety in phosphonate Vb.

*N*-(1,2,4-Triazol-4-yl)-2-methylpropanimine (IIIa). To a solution of 4.72 g (0.0561 mol) of 1,2,4-triazol-4ylamine Ia in 45 ml of ethanol was added in small portions 6.6 g (0.0561 mol) of methylpropanal IIb. The reaction mixture slightly warmed. Then one drop of sulfuric acid was added, and the mixture was heated for 5 h. After the removal of the solvent in a high vacuum the residue was recrystallized from an ether– hexane mixture. Yield 4.78 g (62%), mp 47°C. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>),  $\delta$ , ppm: 1.36 d (6H, CMe<sub>2</sub>,  ${}^{3}J_{\text{HH}}$  7.0 Hz), 2.86 m (1H, CH), 8.16 d (1H, CH=N,  ${}^{3}J_{\text{HH}}$  4.9 Hz), 8.61 s (2H, C<sub>2</sub>H<sub>2</sub>N<sub>3</sub>).

*N*-(1,2,4-Triazol-4-yl)-2,2-dimethyl-3-(ethylamino)propanimine (IIIb) was prepared similarly from 7.69 g (0.0914 mol) of 1,2,4-triazol-4-ylamine Ia in 45 ml of ethanol and of 11.81 g (0.0914 mol) of 2,2-dimethyl-3-(ethylamino)propanal IIa. Yield 11.87 g (67%), mp 83°C (diethyl ether–hexane, 1:2). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>), δ, ppm: 1.22 t (3H, NHCH<sub>2</sub><u>Me</u>, <sup>3</sup>*J*<sub>HH</sub> 7.5 Hz), 1.36 s (6H, CMe<sub>2</sub>), 1.41 s (1H, NH), 2.78 q (2H, NH<u>CH</u><sub>2</sub>Me, <sup>3</sup>*J*<sub>HH</sub> 7.5 Hz), 2.84 s (2H, <u>CH</u><sub>2</sub>NH), 8.20 s (1H, CH=N), 8.61 s (2H, C<sub>2</sub>H<sub>2</sub>N<sub>3</sub>).

*N*-Propyl-2,2-dimethyl-3-(ethylamino)propanimine (IIIc). To 5.3 g (0.0886 mol) of propylamine Ib with stirring at -4 to -5°C was added dropwise 11.58 g (0.0886 mol) of 2,2-dimethyl-3-(ethylamino)propanal IIa. The reaction mixture slightly warmed. The mixture was stirred for 15 min with cooling and then warmed to room temperature. The solid KOH was added till separation of the mixture into two layers. The organic layer was separated, dried over KOH, and distilled. Yield 8.80 g (58%), bp 70–71°C (10 mm Hg),  $n_D^{20}$  1.4352. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub> + CCl<sub>4</sub>),  $\delta$ , ppm: 0.78 t (3H, NCH<sub>2</sub>Me, <sup>3</sup>J<sub>HH</sub> 7.5 Hz), 0.90 t (3H, NCH<sub>2</sub>CH<sub>2</sub>Me, <sup>3</sup>J<sub>HH</sub> 7.0 Hz, the signal is partially overlapped with the signals of CMe<sub>2</sub>), 0.96 s and 1.03 s (6H, CMe<sub>2</sub>), 0.96 s (1H, NH), 1.50 m (2H, Me<u>CH<sub>2</sub></u>· CH<sub>2</sub>N=), 2.46 s (2H, <u>CH<sub>2</sub>NCH<sub>2</sub>Me</u>), 2.50 q (2H, N<u>CH<sub>2</sub>Me</u>,  ${}^{3}J_{HH}$  7.5 Hz), 3.21 t (2H, MeCH<sub>2</sub><u>CH<sub>2</sub>N</u>,  ${}^{3}J_{HH}$  3.35 Hz), 7.37 s (1H, N=CH).

Dimethyl [2-methyl-1-(1,2,4-triazol-4-ylamino)propyllphosphonate (Va). To 3.15 g (0.0228 mol) of N-(1,2,4-triazol-4-yl)-2-methylpropanimine IIIa was added dropwise under argon 2.63 g (0.0239 mol) of dimethylphosphorous acid IVa. To this mixture 3-4 times was added 5-6 drops of sodium alcoholate. The reaction mixture slightly warmed. The mixture was kept at room temperature for 3 days. Then the solvent was removed, and the residue was precipitated from an ethyl acetate-toluene mixture. Yield 2.5 g (44%), mp 103–104°C. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>),  $\delta$ , ppm: 1.25 d and 1.30 d (6H, CMe<sub>2</sub>, <sup>3</sup>J<sub>HH</sub> 7.0 Hz), 2.24 m (1H, <u>CH</u>Me<sub>2</sub>), 3.29 d.d (1H, PCH,  ${}^{2}J_{PH}$  15.0,  ${}^{2}J_{HH}$  4 Hz), 3.91 d [6H, P(OMe)<sub>2</sub>,  ${}^{3}J_{PH}$  10.8 Hz], 6.00 br.s (1H, NH), 8.47 s (2H,  $C_2H_2N_3$ ). <sup>31</sup>P NMR spectrum (CDCl<sub>3</sub>): δ<sub>P</sub> 27.0 ppm. Found, %: C 38.85; H 6.65; N 22.54; P 12.25. C<sub>8</sub>H<sub>17</sub>PO<sub>3</sub>N<sub>4</sub>. Calculated, %: C 38.71; H 6.9; N 22.57; P 12.48.

**Dimethyl [2,2-dimethyl-1-(propylamino)-3-(ethylamino)propyl]phosphonate (Vb).** To 3.68 g (0.0216 mol) of *N*-propyl-2,2-dimethyl-3-(ethylamino)propanimine **IIIc** was added dropwise under argon 2.38 g (0.0216 mol) of dimethylphosphorous acid **IVa**. The reaction mixture slightly warmed. The mixture was kept at room temperature for 5 days and distilled in a vacuum. Yield 3.23 g (55%), bp 84–85°C (0.05 mm Hg),  $n_D^{20}$ 1.4582. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub> + CCl<sub>4</sub>),  $\delta$ , ppm: 1.20–1.70 m (6H, CMe<sub>2</sub>, 3H, NCH<sub>2</sub><u>Me</u>, 3H, CH<sub>2</sub>CH<sub>2</sub><u>Me</u> and <u>NH</u>Et), 1.75–2.10 m (2H, NCH<sub>2</sub><u>CH<sub>2</sub></u>Me and 1H, <u>NH</u>Pr), 2.60–3.30 m. (1H, PCH, 2H, NH<u>CH<sub>2</sub>Et, 2H, CCH<sub>2</sub>NH, 2H, NH<u>CH<sub>2</sub></u>Me), 3.75 m (6H, MeOP). <sup>31</sup>P NMR spectrum (CCl<sub>4</sub>):  $\delta_P$  30.9 ppm.</u>

**Diisopropyl** [2,2-dimethyl-1-(1,2,4-triazol-4-ylamino)-3-(ethylamino)propyl]phosphonate (Vc). To a solution of 6.98 g (0.0421 mol) of isopropylphosphorous acid **IVb** in 40 ml of benzene was added under argon with stirring 8.21 g (0.0421 mol) of *N*-(1,2,4-triazol-4-yl)-2,2-dimethyl-3-(ethylamino)propanimine **IIIb**. To this mixture were added few small pieces of metallic sodium. The reaction mixture slightly warmed. The mixture was kept at room temperature for 2–3 days. The reaction mixture was treated with ether and filtered from the small amount of crystalline product. After the removal of the solvent in a high vacuum the residue was recrystallized from ether. Yield 9.12 g (60%), mp 88°C. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>),  $\delta$ , ppm: 1.10–1.60 m (12H, <u>Me</u><sub>2</sub>CHO, 6H, Me<sub>2</sub>C, 3H, the signals of <u>Me</u>CH<sub>2</sub>N are partially overlapped), 2.64–3.08 m (4H, <u>CH</u><sub>2</sub>NH<u>CH</u><sub>2</sub>Me), 3.44 d.d (1H, PCH, <sup>2</sup>*J*<sub>PH</sub> 13.7, <sup>3</sup>*J*<sub>HH</sub> 4.0 Hz), 4.89 heptet (2H, PO<u>CH</u><sup>3</sup>Me<sup>1</sup>Me<sup>2</sup>, <sup>3</sup>*J*<sub>PH</sub><sup>3</sup> = <sup>3</sup>*J*<sub>H<sup>1</sup>H<sup>3</sup></sub> = <sup>3</sup>*J*<sub>H<sup>2</sup>H<sup>3</sup></sub> = 6.3 Hz), 6.97 d.d (1H, <u>NH</u>N, <sup>3</sup>*J*<sub>PH</sub> 17.0, <sup>3</sup>*J*<sub>HH</sub> 4.0 Hz), 8.47 s (2H, C<sub>2</sub>N<sub>3</sub>H<sub>2</sub>). <sup>31</sup>P NMR spectrum (CDCl<sub>3</sub>):  $\delta_P$  22.1 ppm.

Dimethyl [2-methyl-1-(1,2,4-triazol-4-ylamino) propyl]phosphonate picrate (VIa). To a solution of 0.65 g (0.0026 mol) of dimethyl [2-methyl-1-(1,2,4triazol-4-ylamino)propyl]phosphonate Va in 5.5 ml of ethanol was added in small portions a solution of 0.6 g (0.0026 mol) of picric acid in 5.13 ml of ethanol. The mixture was kept at room temperature for 24 h. The precipitated crystals were filtered off and dried in a vacuum. Yield 0.86 g (70%), mp 140–142°C. <sup>1</sup>H NMR spectrum (acetone- $d_6$ ),  $\delta$ , ppm: 1.23 d and 1.28 d (6H, CMe<sub>2</sub>, <sup>3</sup>J<sub>HH</sub> 7.0 Hz), 2.34 m (1H, <u>CH</u>Me<sub>2</sub>), 3.83 d and 3.85 d [6H, P(OMe)<sub>2</sub>, <sup>3</sup>J<sub>PH</sub> 10.7 Hz], 3.86 d [1H, PCH,  $^{2}J_{\rm PH}$  13.6 Hz, the signal is partially overlapped with the signals of P(OMe)<sub>2</sub>], 7.41 br.s (2H, <sup>+</sup>NH<sub>2</sub>), 8.87 s (2H,  $C_2H_2N_3$ , 9.56 s (2H,  $C_6H_2$ ). <sup>31</sup>P NMR spectrum (acetone- $d_6$ ):  $\delta_P$  25.3 ppm. Found, %: C 35.77; H 4.11; N 20.83; P 6.75. C<sub>14</sub>H<sub>20</sub>PO<sub>10</sub>N<sub>7</sub>. Calculated, %: C 35.38; H 4.02; N 20.59; P 6.5.

Dimethyl [2,2-dimethyl-1-(propylamino)-3-(ethylamino)propyl|phosphonate dipicrate (VIb). To a solution of 2.62 g (0.0114 mol) of pieric acid in 15 ml of ethyl acetate was added in small portions 1.42 g (0.0057 mol) dimethyl [2,2-dimethyl-1-(propylamino)-3-(ethylamino)propyl]phosphonate Vb in 8 ml of ethyl acetate. The reaction mixture slightly warmed. The mixture was kept at room temperature for 24 h. The precipitated crystals were filtered off, recrystallized from ethanol, and dried in a vacuum. Yield 3.65 g (87%), mp 125°C. <sup>1</sup>H NMR spectrum (acetone- $d_6$ ),  $\delta_7$ , ppm: 0.94 t (3H, <sup>+</sup>NCH<sub>2</sub>CH<sub>2</sub><u>Me</u>, <sup>3</sup> $J_{HH}$  7.0 Hz), 1.35 t (3H, <sup>+</sup>NCH<sub>2</sub><u>Me</u>, <sup>3</sup> $J_{HH}$  7.0 Hz), 1.37 s and 1.40 s (6H, Me<sub>2</sub>), 1.69 m (2H, <sup>+</sup>NCH<sub>2</sub>CH<sub>2</sub>Me), 3.03 sextet and 3.23 м (2H, <sup>+</sup>N<u>CH</u><sub>2</sub>CH<sub>2</sub>Me), 3.30 q (2H, <sup>+</sup>N<u>CH</u><sub>2</sub>Me,  ${}^{3}J_{\rm HH}$  7.0 Hz), 3.40 d and 3.54 d (2H, CH<sub>2</sub>N<sup>+</sup>Et,  ${}^{2}J_{\rm HH}$ 13.2 Hz), 3.62 d (1H, PCH,  ${}^{2}J_{PH}$  17.7 Hz), 3.90 d (6H, 2MeOP,  ${}^{3}J_{PH}$  11 Hz), 4.70–6.10 br.s (4H, 2<sup>+</sup>NH<sub>2</sub>), 8.80 s (4H, 2ArO). <sup>31</sup>P NMR spectrum (C<sub>4</sub>H<sub>8</sub>O<sub>2</sub>):  $\delta_P$  29.2 ppm. Found, %: C 38.70; H 4.72; N 15.09; P 4.45. C<sub>24</sub>H<sub>35</sub>. PO<sub>17</sub>N<sub>8</sub>. Calculated, %: C 39.03; H 4.77; N 15.17; P 4.19.

**Diisopropyl** [2,2-dimethyl-1-(1,2,4-triazol-4-ylamino)-3-(ethylamino)propyl]phosphonate picrate (VIc). To a solution of 1.34 g (0.0037 mol) diisopropyl

[2,2-dimethyl-1-(1,2,4-triazol-4-ylamino)-3-(ethylamino)propyl]phosphonate Vc in 10 ml of dioxane was added in small portions with stirring a solution of 1.70 g (0.0074 mol) of picric acid in 10 ml of dioxane at 13-16°C. The reaction mixture slightly warmed. The mixture was kept at room temperature for 24 h. The precipitated crystals were filtered off and dried in a vacuum. Yield 1.97 g (65%), mp 114°C. <sup>1</sup>H NMR spectrum (CD<sub>3</sub>CN), δ, ppm: 1.25–1.70 m (12H, Me<sub>2</sub>CHO, 6H, Me<sub>2</sub>C, 3H, Me<sub>2</sub>CH<sub>2</sub>N, the signals are overlapped), 3.10-3.50 m (4H. partially <u>CH<sub>2</sub></u><sup>+</sup>NH<sub>2</sub><u>CH<sub>2</sub></u>Me), 3.7 d.d (1H, PCH,  ${}^{2}J_{PH}$  17.8,  ${}^{3}J_{HH}$ 5.0 Hz), 5.02 heptet (2H, POCH<sup>3</sup>Me<sup>1</sup>Me<sup>2</sup>,  ${}^{3}J_{PH^{3}} = {}^{3}J_{H^{1}H^{3}} = {}^{3}J_{H^{2}H^{3}} = 6.3$  Hz), 6.07 d.d (1H, NH,  ${}^{3}J_{PH}$  12.6,  ${}^{3}J_{\text{HH}}$  5.0 Hz), 8.56 s (2H, C<sub>6</sub>H<sub>2</sub>), 8.97 s (2H, C<sub>2</sub>H<sub>2</sub>N<sub>3</sub>), 9.24 br.s and 9.5 br.s (2H, <sup>+</sup>NH<sub>2</sub>). <sup>31</sup>P NMR spectrum (CD<sub>3</sub>CN): δ<sub>P</sub> 21.4 ppm. Found, %: C 42.21; H 5.65; N 18.15; P 5.4. C<sub>21</sub>H<sub>35</sub>PO<sub>10</sub>N<sub>8</sub>. Calculated, %: C 42.73; H 5.98; N 18.98; P 5.25.

Diisopropyl [2,2-dimethyl-1-(1,2,4-triazol-4-ylamino)-3-(ethylamino)propyl]phosphonate perchlorate (VId). To a solution of 1.30 g (0.0036 mol) [2,2dimethyl-1-(1,2,4-triazol-4-ylamino)-3-(ethylamino)propyl]phosphonate Vc in 15 ml of dioxane was added at 13°C in small portions with stirring a solution of 1.7 g (0.0072 mol) of perchloric acid. The reaction mixture slightly warmed. The mixture was kept at room temperature for 24 h. The precipitated crystals were filtered off, washed with dioxane and dried in a vacuum. Yield 1.68 g (83%), mp 174–175°C. <sup>1</sup>H NMR spectrum (CD<sub>3</sub>CN),  $\delta$ , ppm: 1.24–1.66 m (12H, <u>Me</u><sub>2</sub>CHO, 6H, Me<sub>2</sub>C, 3H, <u>Me</u>CH<sub>2</sub>N, the signals are partially overlapped), 3.12–3.48 m (4H, <u>CH</u><sub>2</sub><sup>+</sup>NH<sub>2</sub><u>CH</u><sub>2</sub>Me), 3.64 d.d (1H, PCH, <sup>2</sup>J<sub>PH</sub> 14.8, <sup>3</sup>J<sub>HH</sub> 7.1 Hz), 5.02 heptet (2H, POCH<sup>3</sup>Me<sup>1</sup>Me<sup>2</sup>, <sup>3</sup>J<sub>PH<sup>3</sup></sub> = <sup>3</sup>J<sub>H<sup>1</sup>H<sup>3</sup></sub> = <sup>3</sup>J<sub>H<sup>2</sup>H<sup>3</sup></sub> = 7.1 Hz), 6.59 d.d (1H, NH, <sup>3</sup>J<sub>PH</sub> 12.4, <sup>3</sup>J<sub>HH</sub> 7.1 Hz), 7.03 br.s and 8.06 br.s (2H, <sup>+</sup>NH<sub>2</sub>), 9.02 s (2H, C<sub>2</sub>H<sub>2</sub>N<sub>3</sub>). <sup>31</sup>P NMR spectrum (CD<sub>3</sub>CN):  $\delta_P$  20.6 ppm.

The <sup>1</sup>H NMR spectra were recorded on a Tesla BS-567A and Bruker Avance-600 instruments operating at 100 and 600 MHz, respectively, relative to internal TMS. The <sup>31</sup>P NMR spectra were recorded on a Bruker Avance-600 instrument operating at 242.88 MHz relative to external 85%  $H_3PO_4$ .

## ACKNOWLEDGMENTS

This work was supported by the Federal Program "Scientific and scientific-pedagogical staff of innovative Russia" for 2009–2013 (contract no. P-1108).

## REFERENCES

- 1. Konovalova, I.V., *Reaktsiya Pudovika* (The Pudovik Reaction), Kazan: Kazan. Gos. Univ., 1991.
- Obshchaya organicheskaya khimiya. Azotsoderzhashchie geterotsykly (General Organic Chemistry. Nitrogen-containing Heterocycles), Kochetkova, N.K., Ed., Moscow: Khimiya, 1985, vol. 8, p. 632.
- Mannich, C., Lesser, B., and Silte, F., *Chem Ber.*, 1932, vol. 65, p. 378.