

# Supporting Information

## Heterocyclization of N-(1-chloro-2,2,2-trifluoroethylidene)carbamates with $\beta$ -enaminoesters – a novel synthetic strategy to functionalized trifluoromethylated pyrimidines

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02094, Ukraine

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### X-ray Structure determination for compound 5a.

Crystal data: C<sub>10</sub>H<sub>11</sub>B<sub>2</sub>F<sub>3</sub>N<sub>2</sub>O<sub>3</sub>, M 285.83, monoclinic, space group P2<sub>1</sub>/c,  $a = 4.8955(6)$ ,  $b = 11.0400(15)$ ,  $c = 21.316(3)$  Å,  $\beta = 93.177(4)^\circ$ ,  $V = 1150.3(3)$  Å<sup>3</sup>,  $Z = 4$ ,  $d_c = 1.56$  g·cm<sup>-3</sup>,  $\mu = 0.148$  mm<sup>-1</sup>,  $F(000) = 584$ , crystal size ca. 0.19 x 0.27 x 0.35 mm. All crystallographic measurements were performed at room temperature on a Bruker Smart Apex II diffractometer operating in the  $\omega$  and  $\varphi$  scans mode. The intensity data were collected within the  $\theta_{\max} \leq 26.38^\circ$  using Mo-K $\alpha$  radiation ( $\lambda = 0.71078$  Å). The intensities of 5640 reflections were collected (2164 unique reflections,  $R_{\text{merg}} = 0.0202$ ).

The structure was solved by direct methods and refined by the full-matrix least-squares technique in the anisotropic approximation for non-hydrogen atoms using the Bruker SHELXTL program package<sup>1</sup>. All hydrogen atoms were refined isotropically. In the refinement 2164 reflections (1539 reflections with  $I \geq 2\sigma(I)$ ) were used. Convergence was obtained at  $R1 = 0.0588$  and  $wR2 = 0.1039$ , for all reflection and  $R1 = 0.0374$  and  $wR2 = 0.0921$ ,  $GOF = 1.007$  for observed (207 parameters; observed/variable ratio 7.4; the largest and minimal peaks in the final difference map 0.18 and -0.18 e/Å<sup>3</sup>, weighting scheme is as follows:  $\omega = 1/[\sigma^2(Fo^2) + 0.0509P^2 + 0.2021P]$ , where  $P = (Fo^2 + 2Fc^2)/3$ ). Full crystallographic details have been deposited

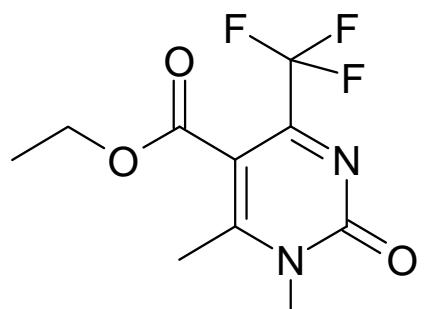
at Cambridge Crystallographic Data Centre (CCDC). Any request to the CCDC for these materials should quote the full literature citation and reference number CCDC 878801.

*X-ray Structure determination for compound 5I.*

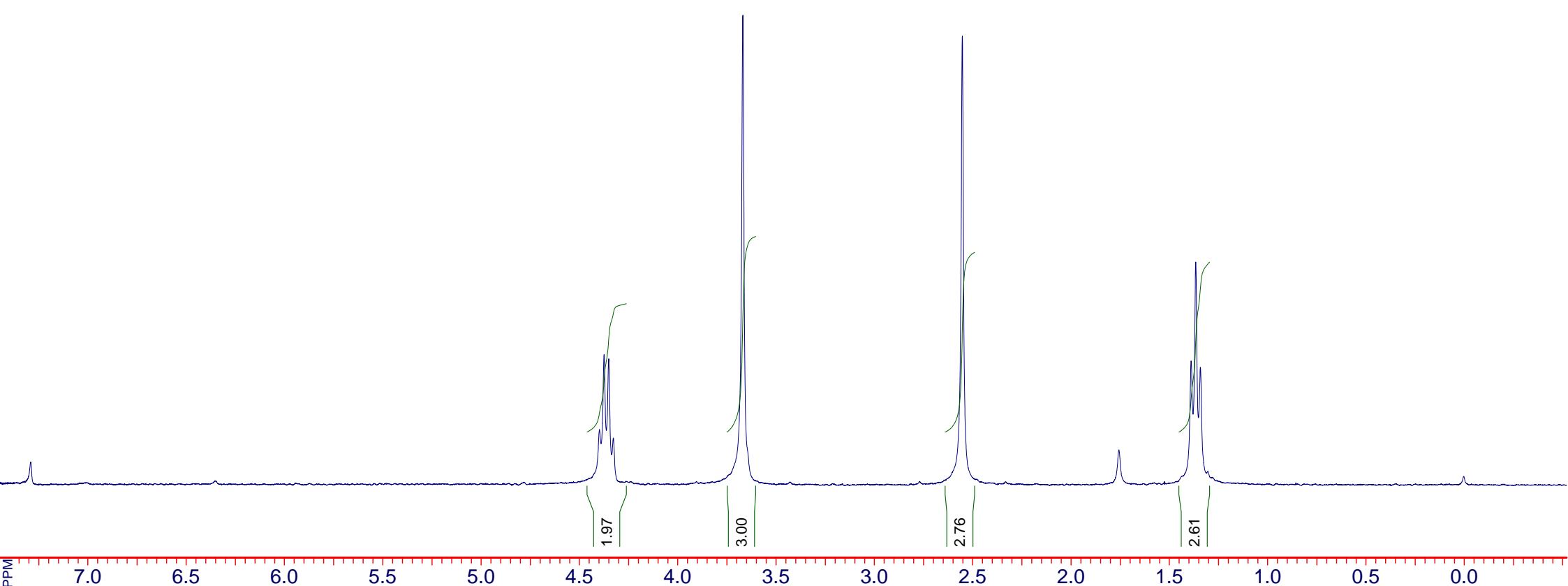
Crystal data: C<sub>35</sub>H<sub>30</sub>F<sub>6</sub>N<sub>4</sub>O<sub>6</sub>, M 716.63, monoclinic, space group P2<sub>1</sub>/c, *a* = 20.2101(4), *b* = 11.0539(2), *c* = 15.0422(3) Å, β = 100.384(1)°, V = 3305.40(11) Å<sup>3</sup>, Z = 4, d<sub>c</sub> = 1.44 g·cm<sup>-3</sup>, μ = 0.121 mm<sup>-1</sup>, F(000) = 1480, crystal size ca. 0.19 x 0.27 x 0.35 mm. All crystallographic measurements were performed at room temperature on a Bruker Smart Apex II diffractometer operating in the ω and φ scans mode. The intensity data were collected within the range θ ≤ 28.94° using Mo-K<sub>α</sub> radiation (λ = 0.71078 Å). The intensities of 35627 reflections were collected (8636 unique reflections, R<sub>merg</sub> = 0.0509).

The structure was solved by direct methods and refined by the full-matrix least-squares technique in the anisotropic approximation for non-hydrogen atoms using the Bruker SHELXTL program package<sup>1</sup>. The Et group one of the independent molecule is disordered over two position with occupancy of 50%. The hydrogen atoms that supported N atoms were found in difference Fourier maps and refined isotropically, rest ones is placed geometrically and refined as “riding” model. In the refinement 8636 reflections (4990 reflections with I ≥ 2σ(I)) were used. Convergence was obtained at R1 = 0.1126 and wR2 = 0.1924, for all reflection and R1 = 0.0565 and wR2 = 0.1544, GOF = 1.061 for observed (467 parameters; observed/variable ratio 10.7; the largest and minimal peaks in the final difference map 0.64 and -0.36 e/Å<sup>3</sup>, weighting scheme is as follows: ω = 1/[σ<sup>2</sup>(Fo<sup>2</sup>) + (0.1P)<sup>2</sup> ], where P = (Fo<sup>2</sup> + 2Fc<sup>2</sup>)/3). Full crystallographic details have been deposited at Cambridge Crystallographic Data Centre (CCDC). Any request to the CCDC for these materials should quote the full literature citation and reference number CCDC 878802.

1.Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112-122.



5a



PPM

164.09  
161.90  
158.17  
157.89  
154.61  
154.60  
154.59

120.39  
118.17

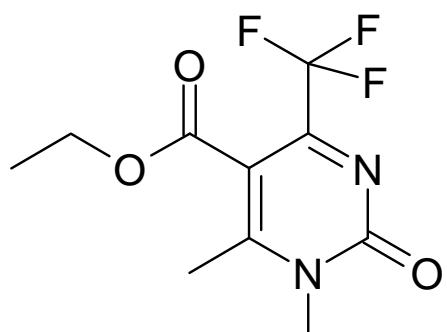
109.36

62.86

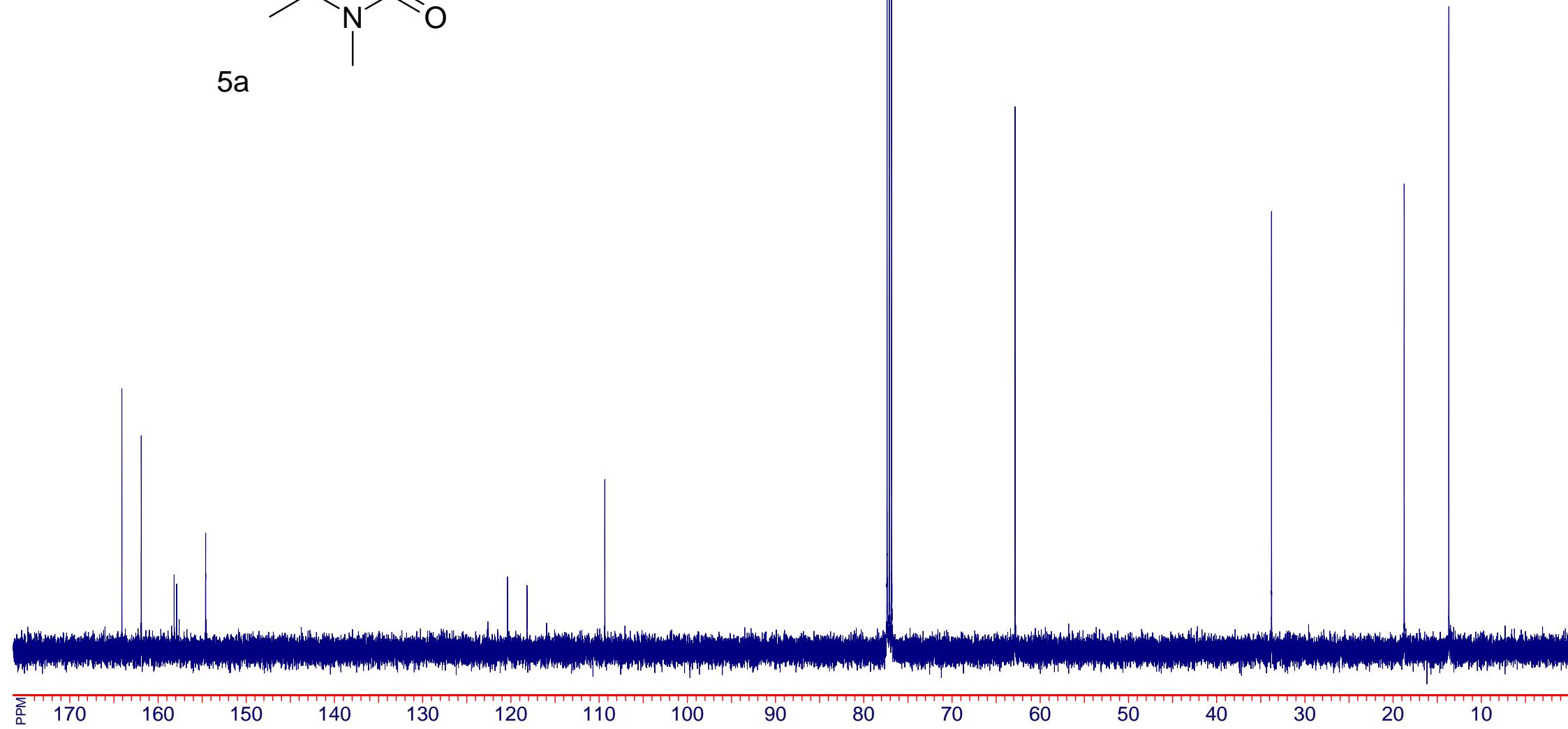
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18.76

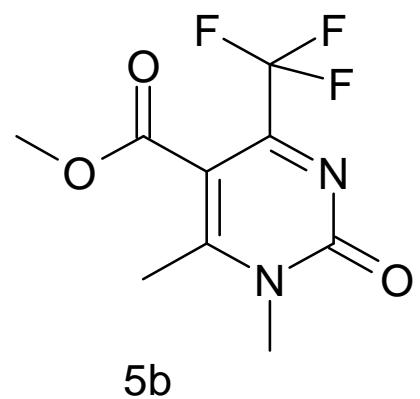
13.70



5a

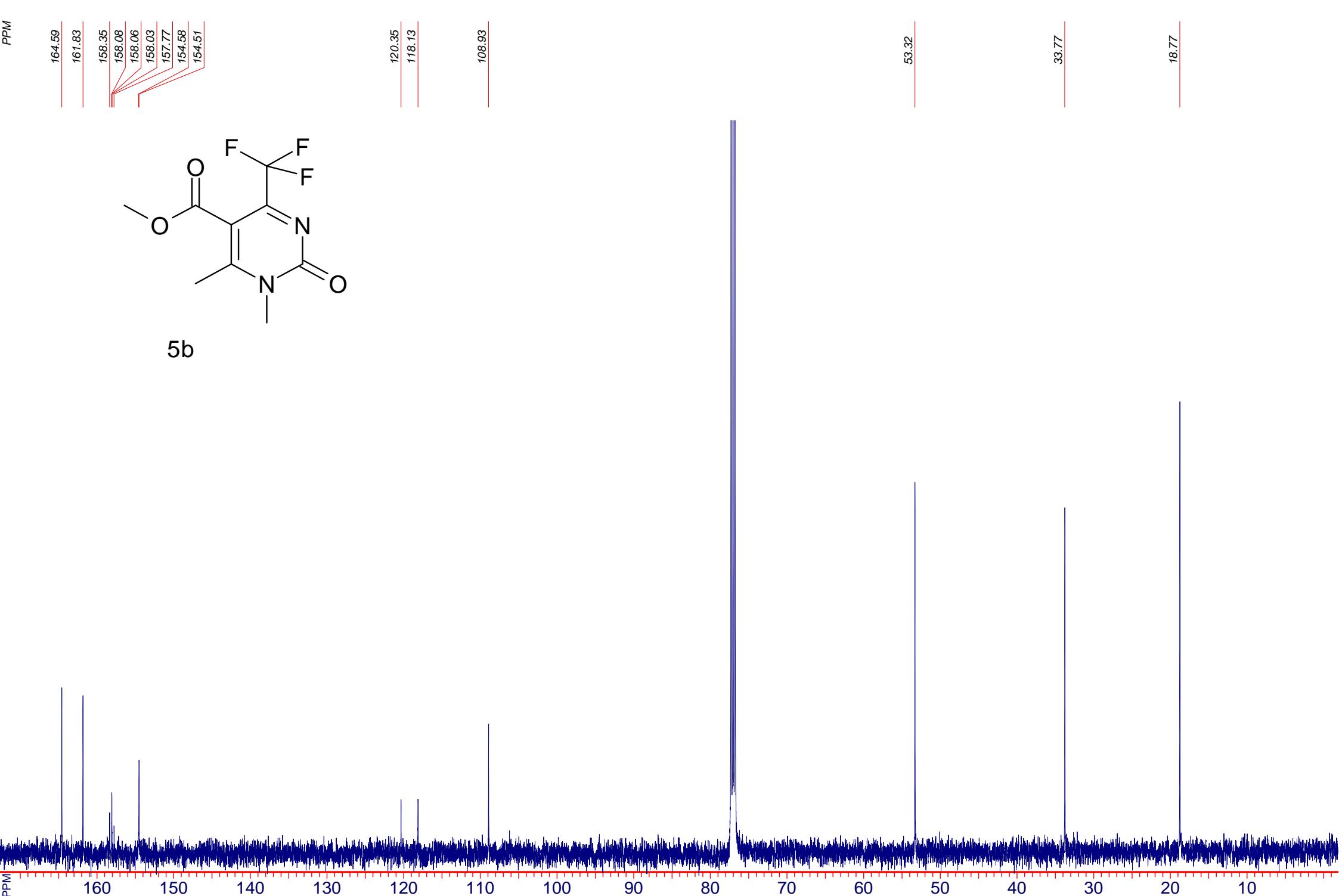


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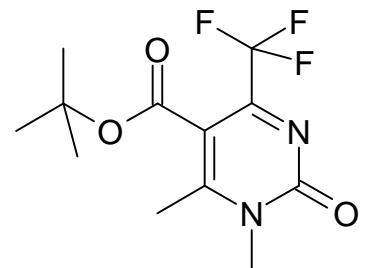


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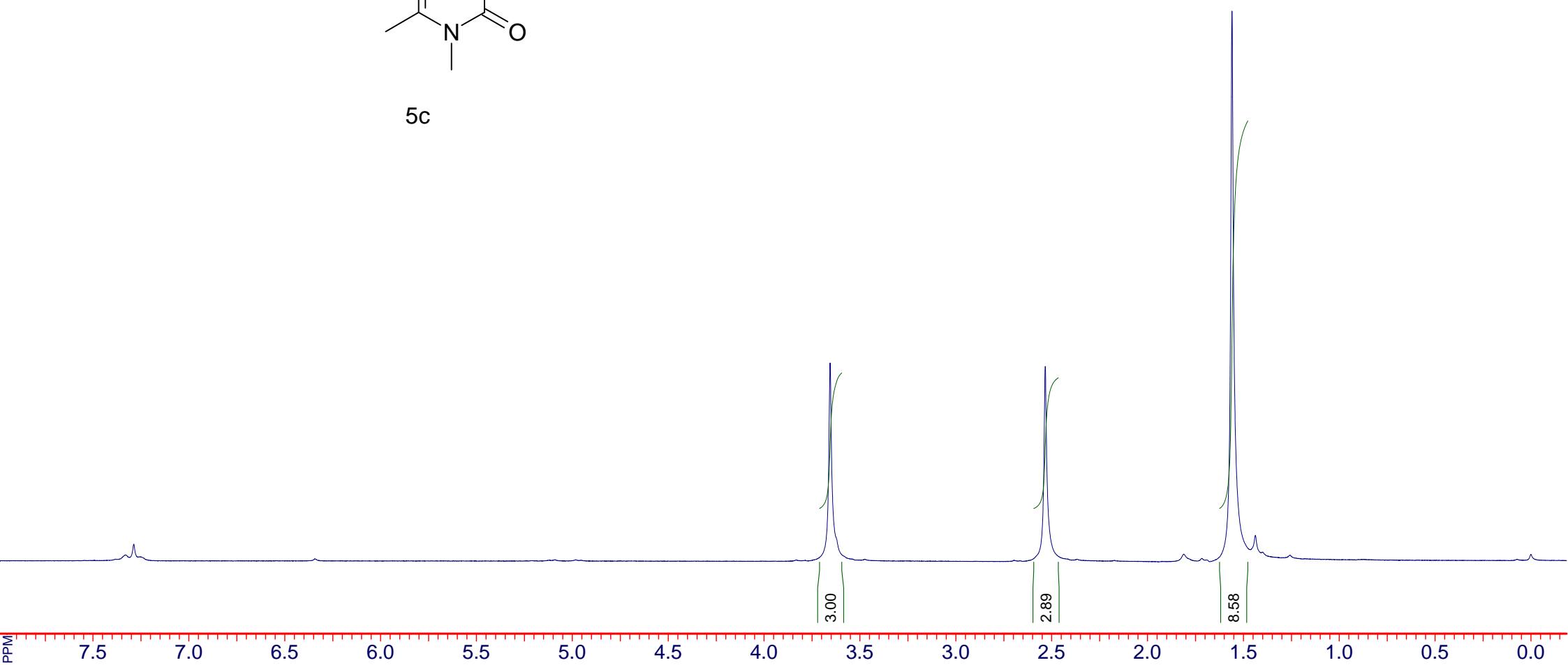


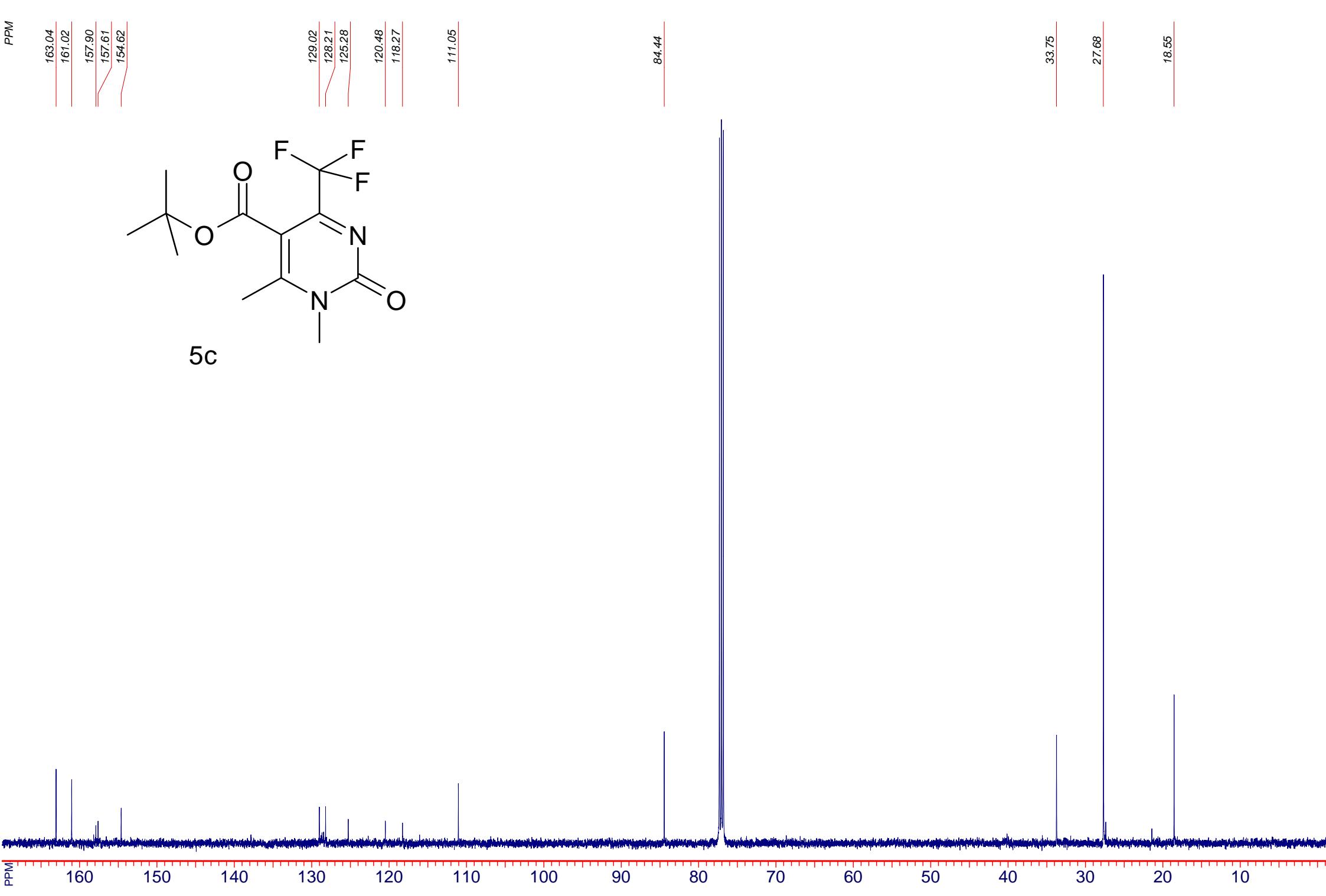


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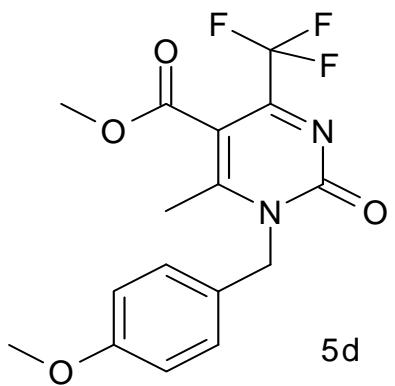
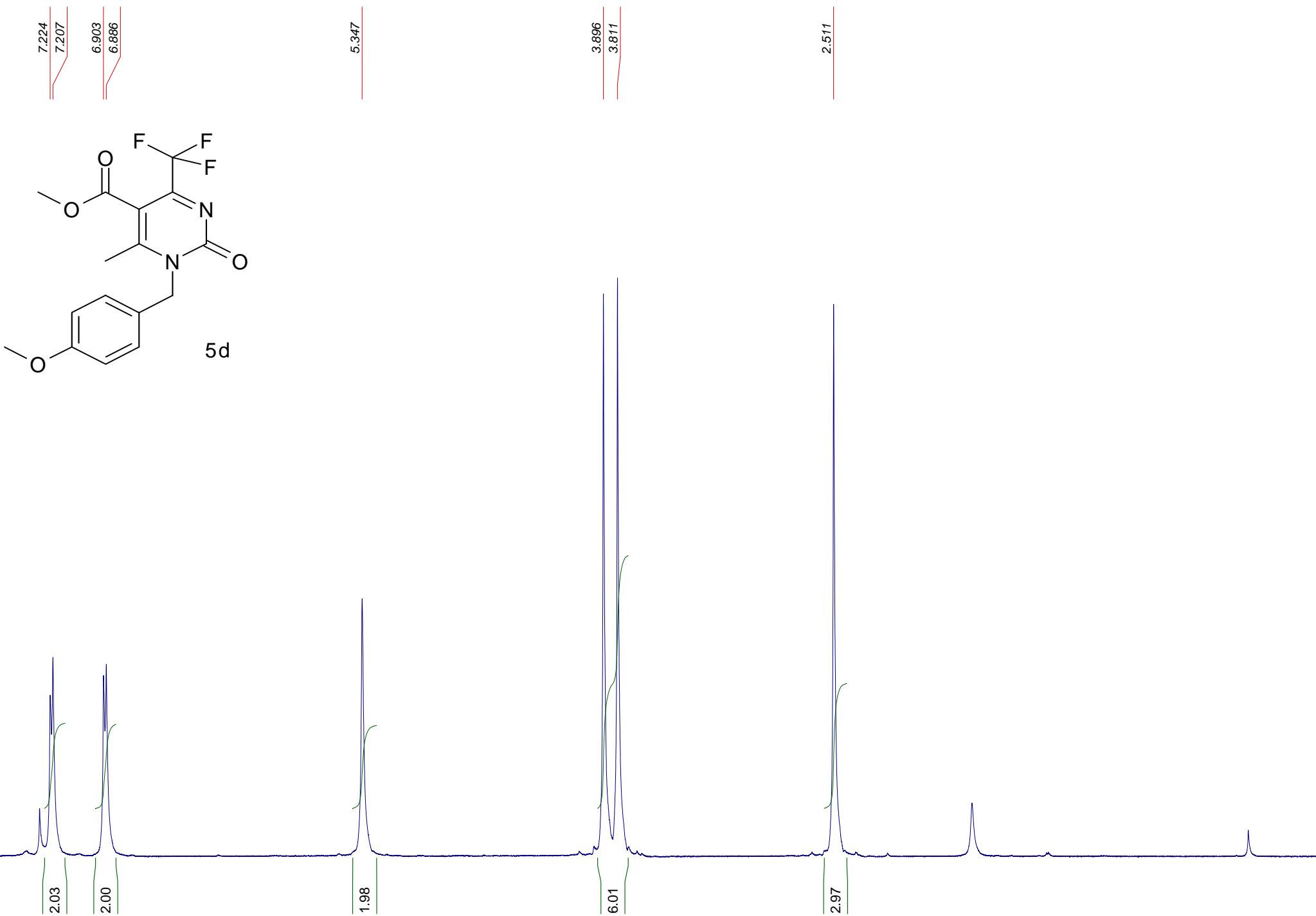


5c





PPM



PPM

164.54  
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159.72  
158.66  
158.37  
154.88

128.59

125.60

120.33

118.11

114.65

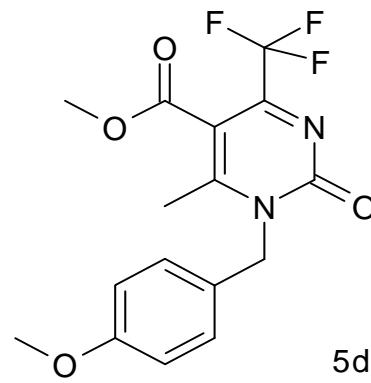
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55.35

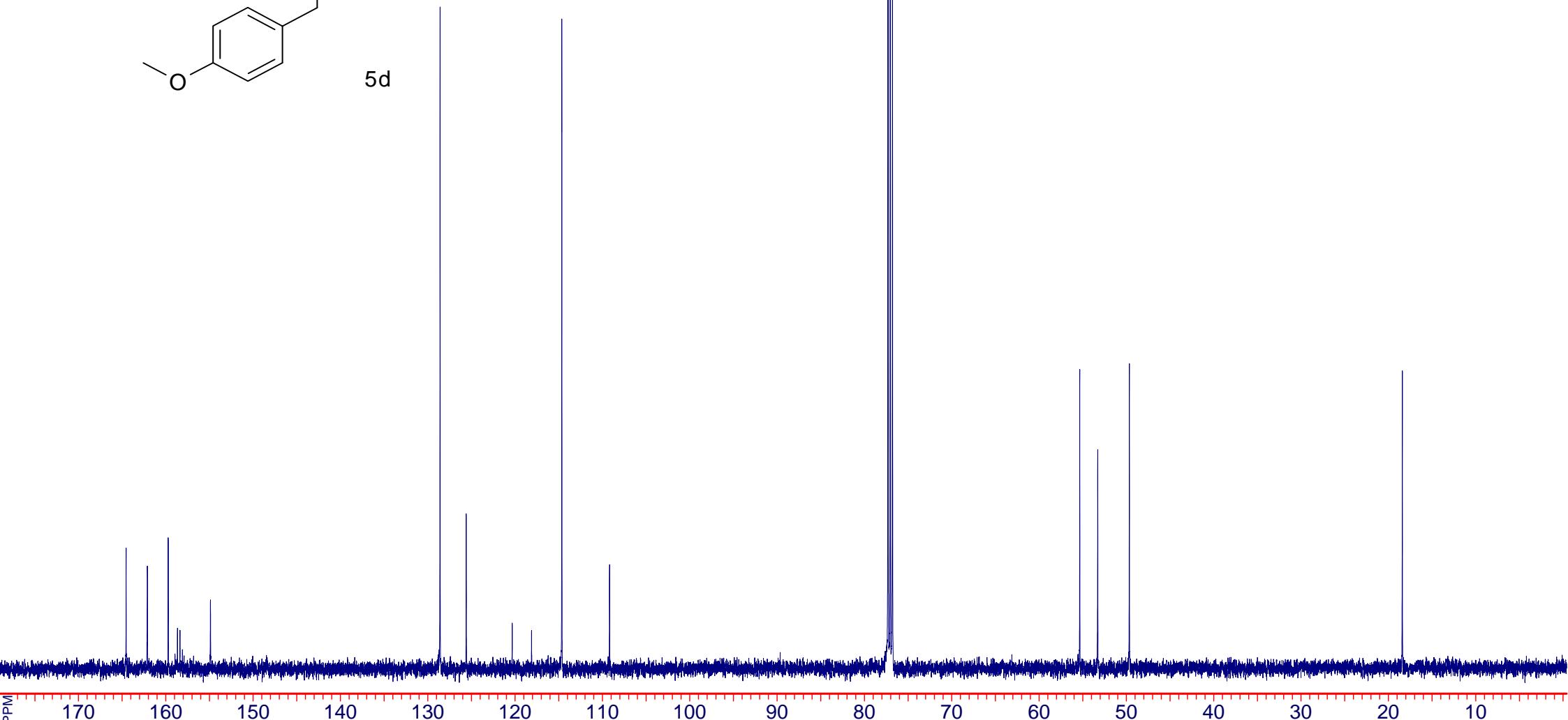
53.29

49.67

18.41



5d



PPM

PPM

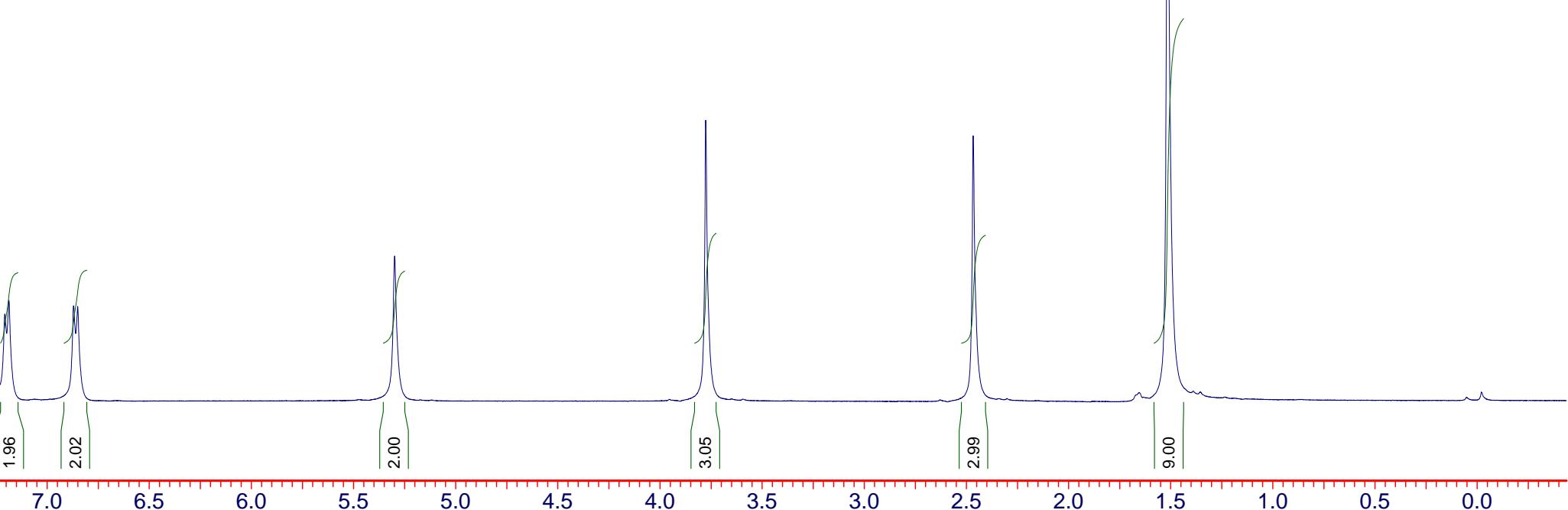
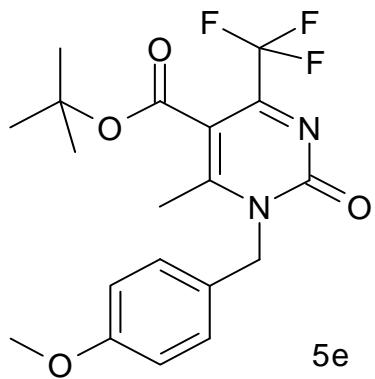
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7.187  
6.870  
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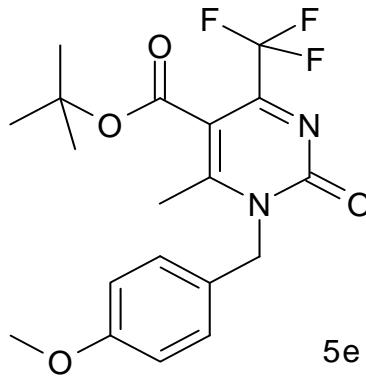
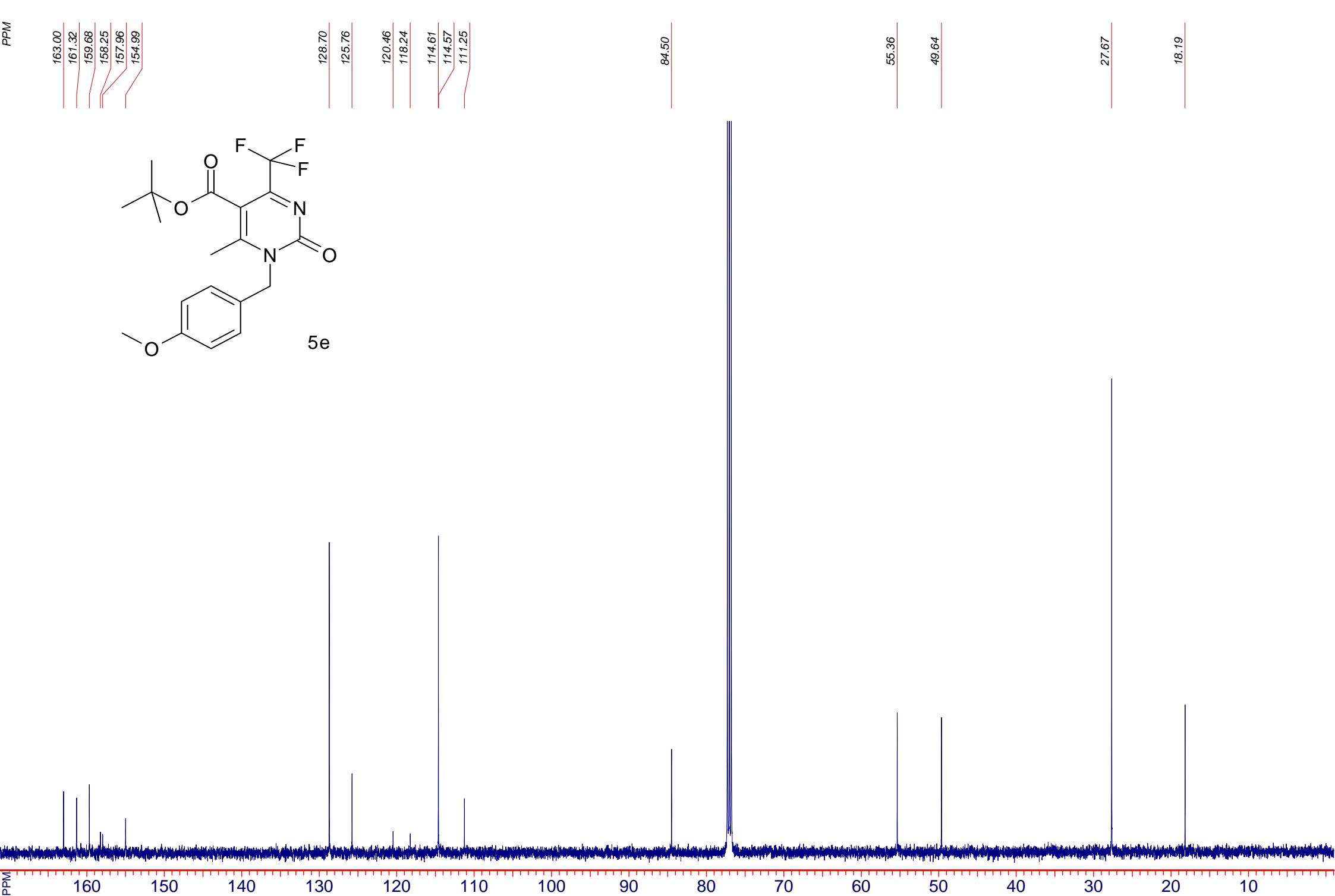
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3.776

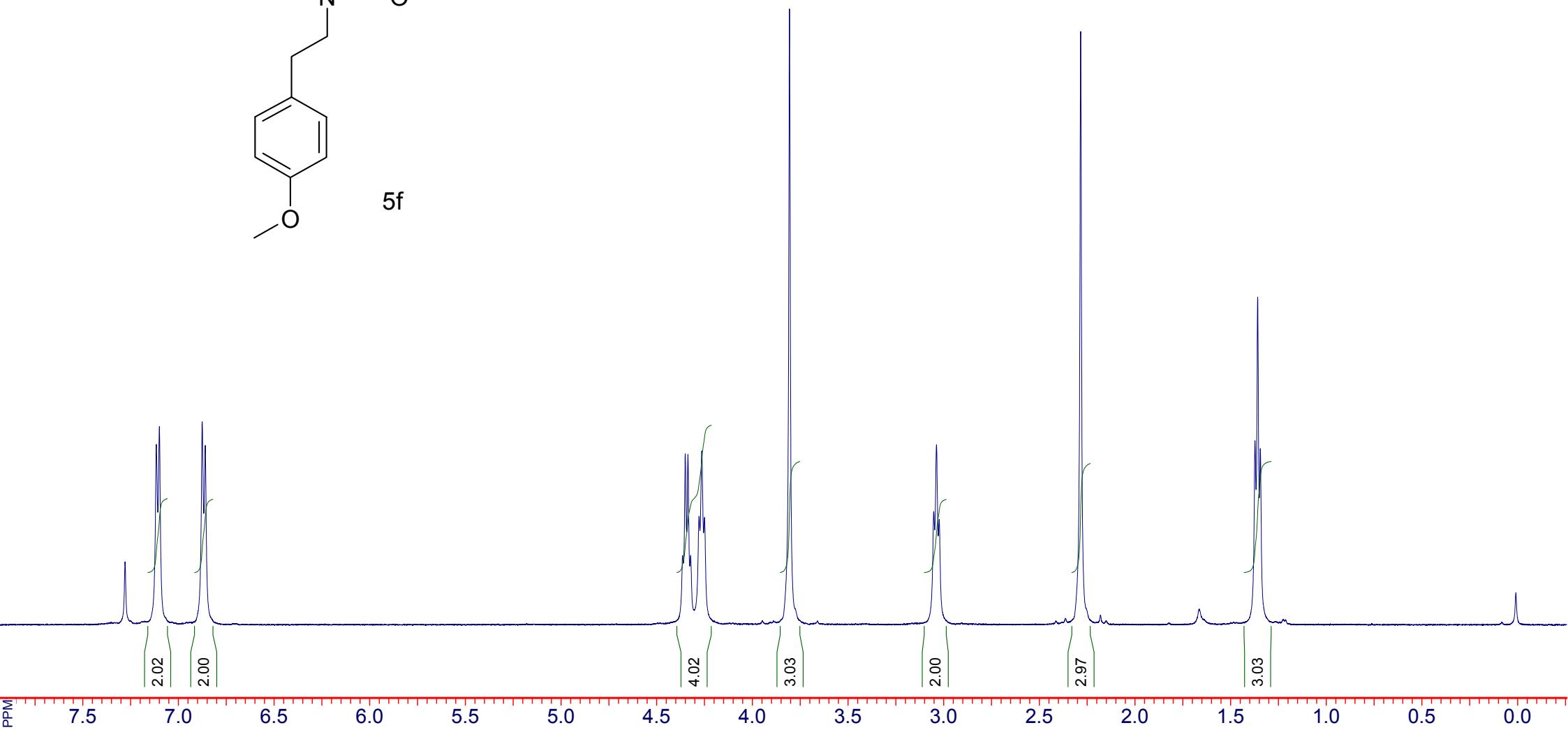
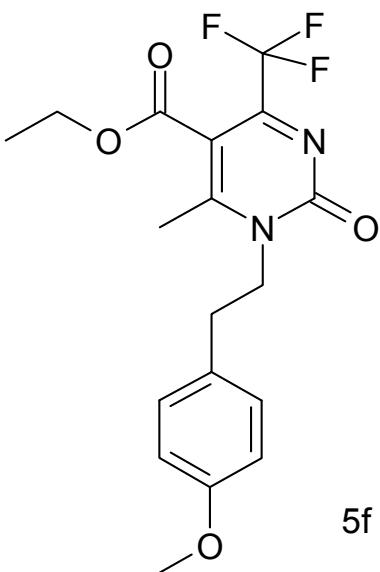
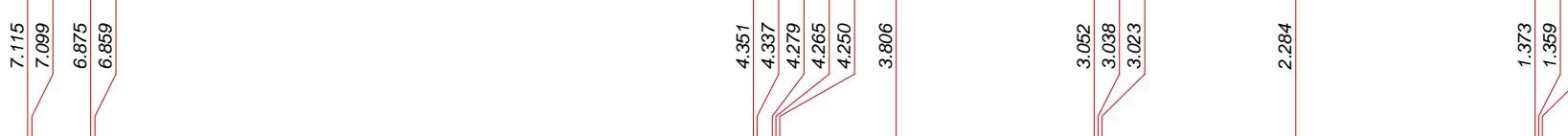
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1.516



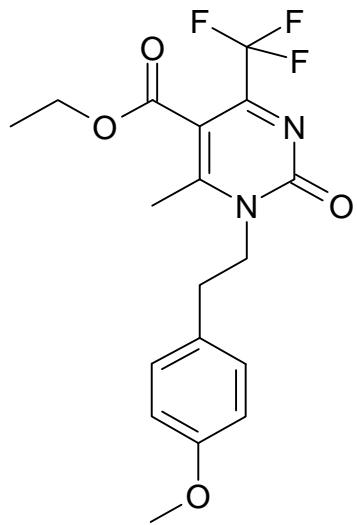
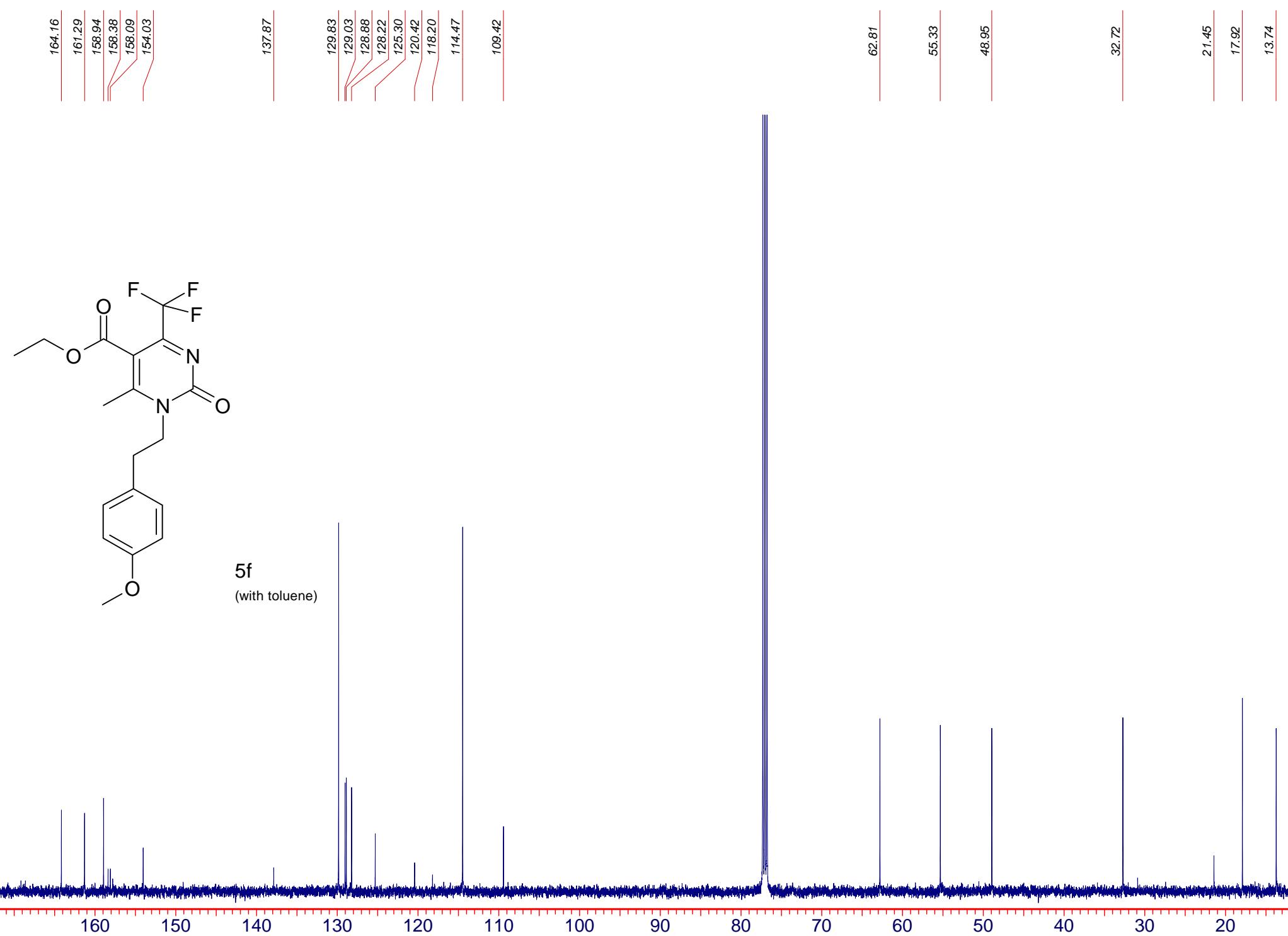


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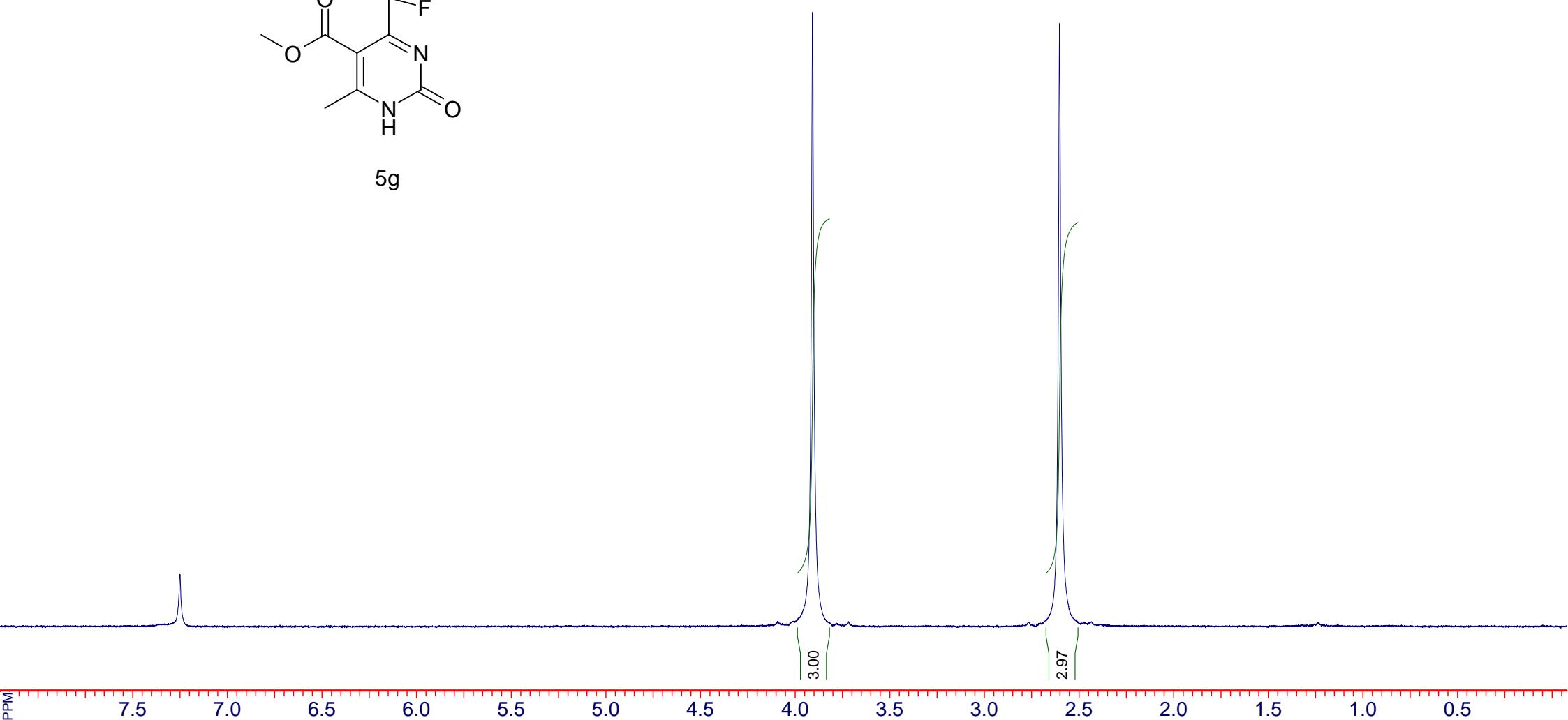
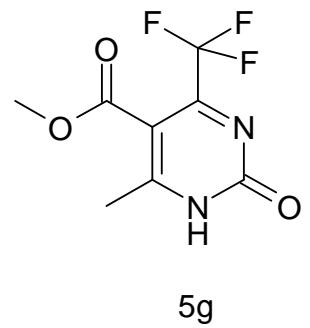


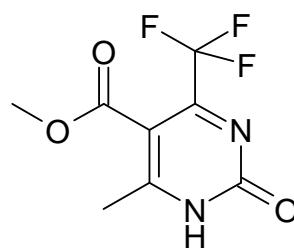
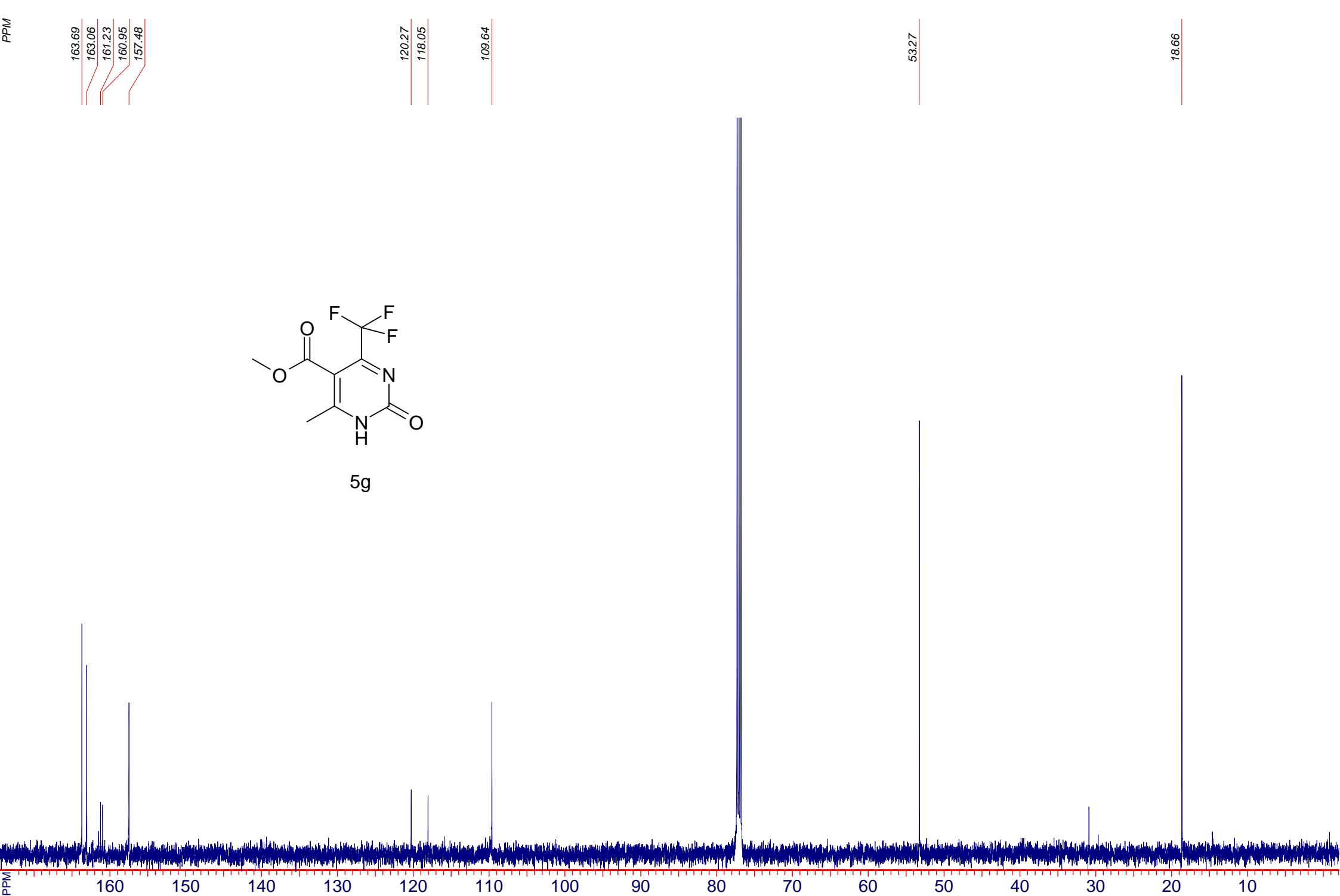
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PPM



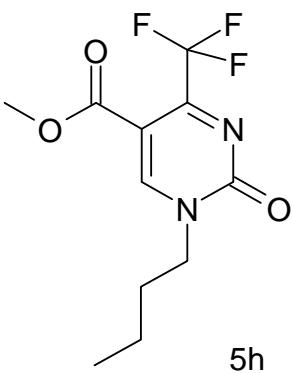
PPM





5g

8.522

4.004  
3.986  
3.8911.808  
1.7921.403  
1.3860.987  
0.969  
0.951**5h**

1.00

5.17

1.89

1.93

2.88



PPM

161.61  
160.20  
159.90  
155.06  
153.36

120.04  
117.82

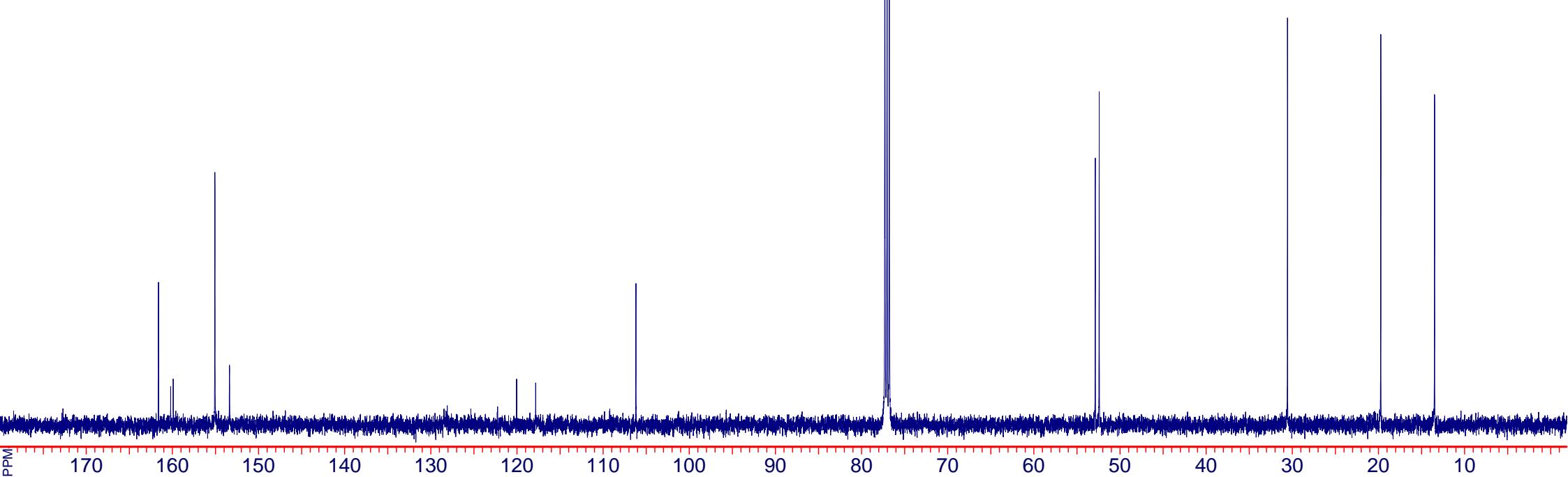
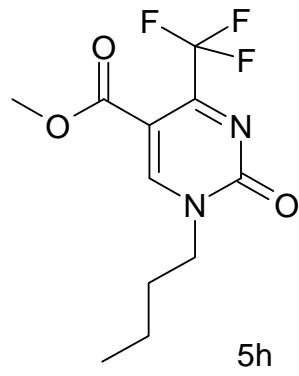
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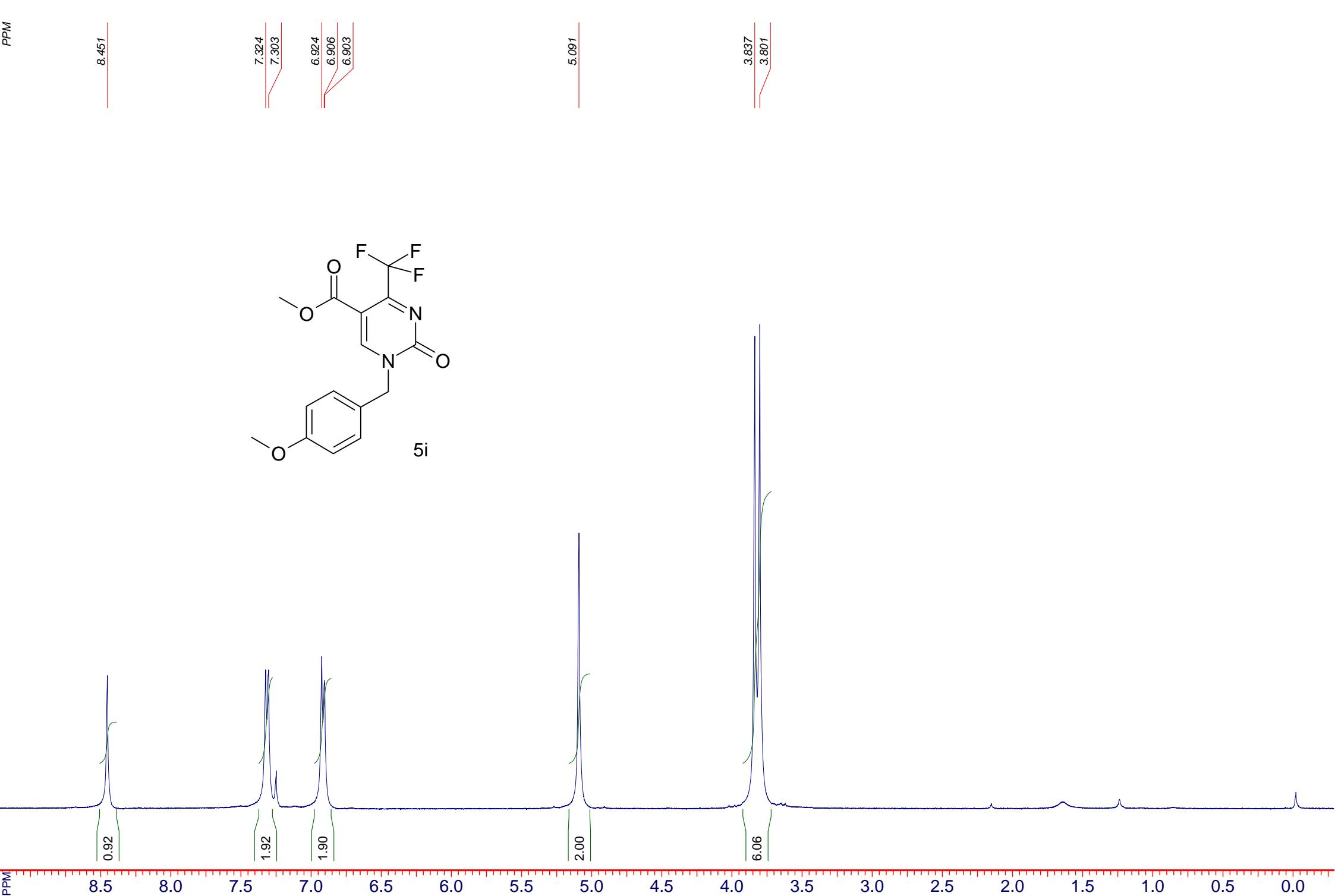
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30.56

19.74

13.50





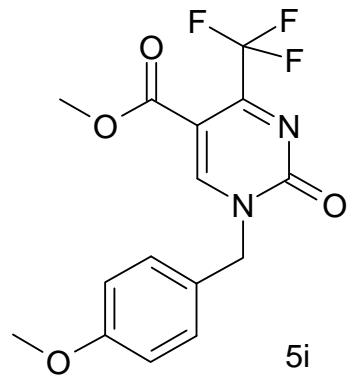
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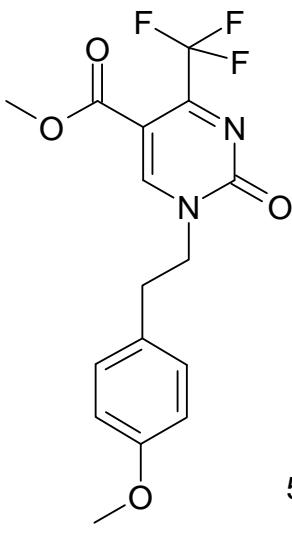
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106.54

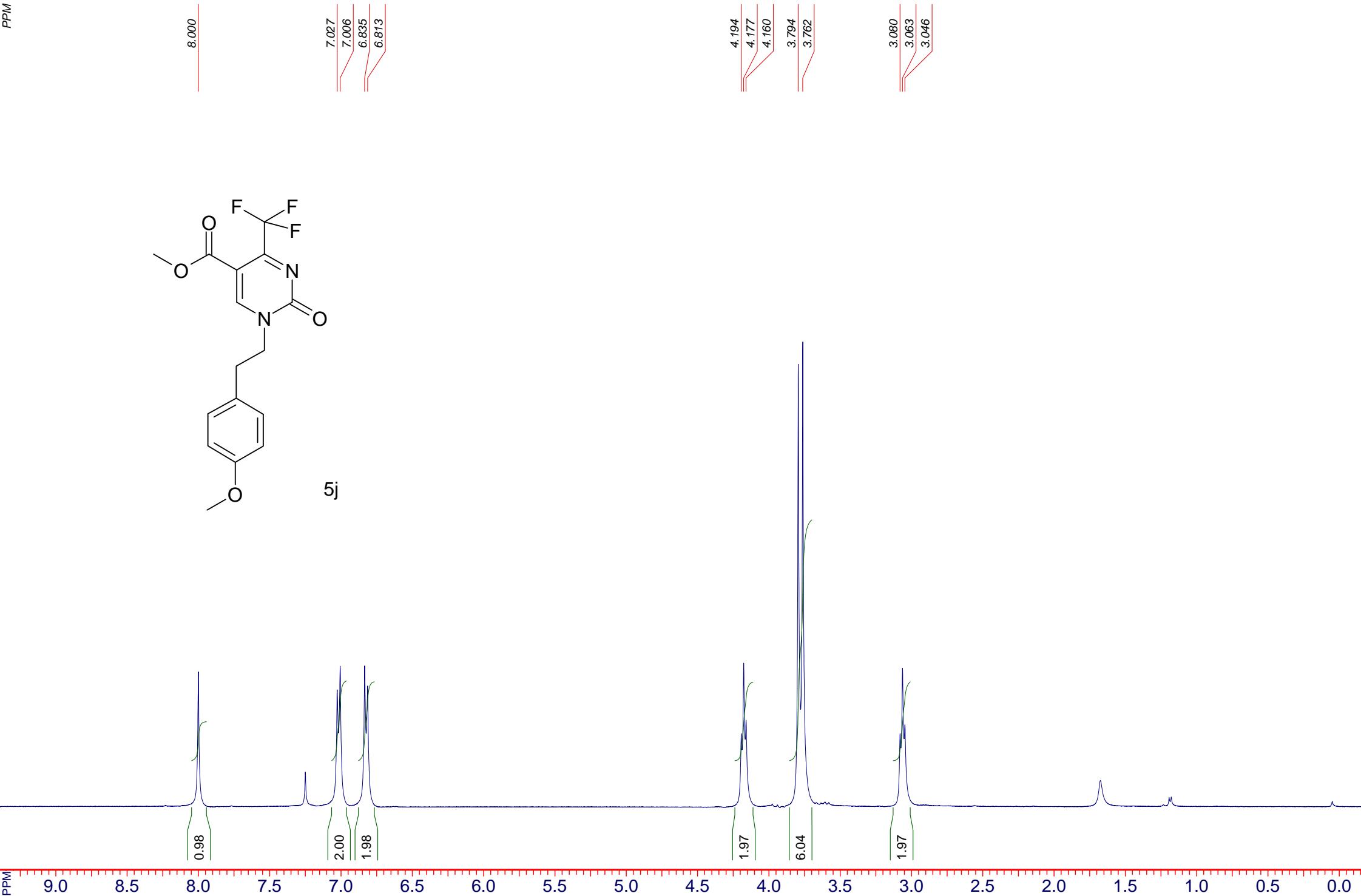
55.38  
54.39  
52.85



PPM



5j



161.32  
160.32  
160.03  
159.04  
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153.29

129.99

127.93

122.23

120.02

117.80

115.59

114.56

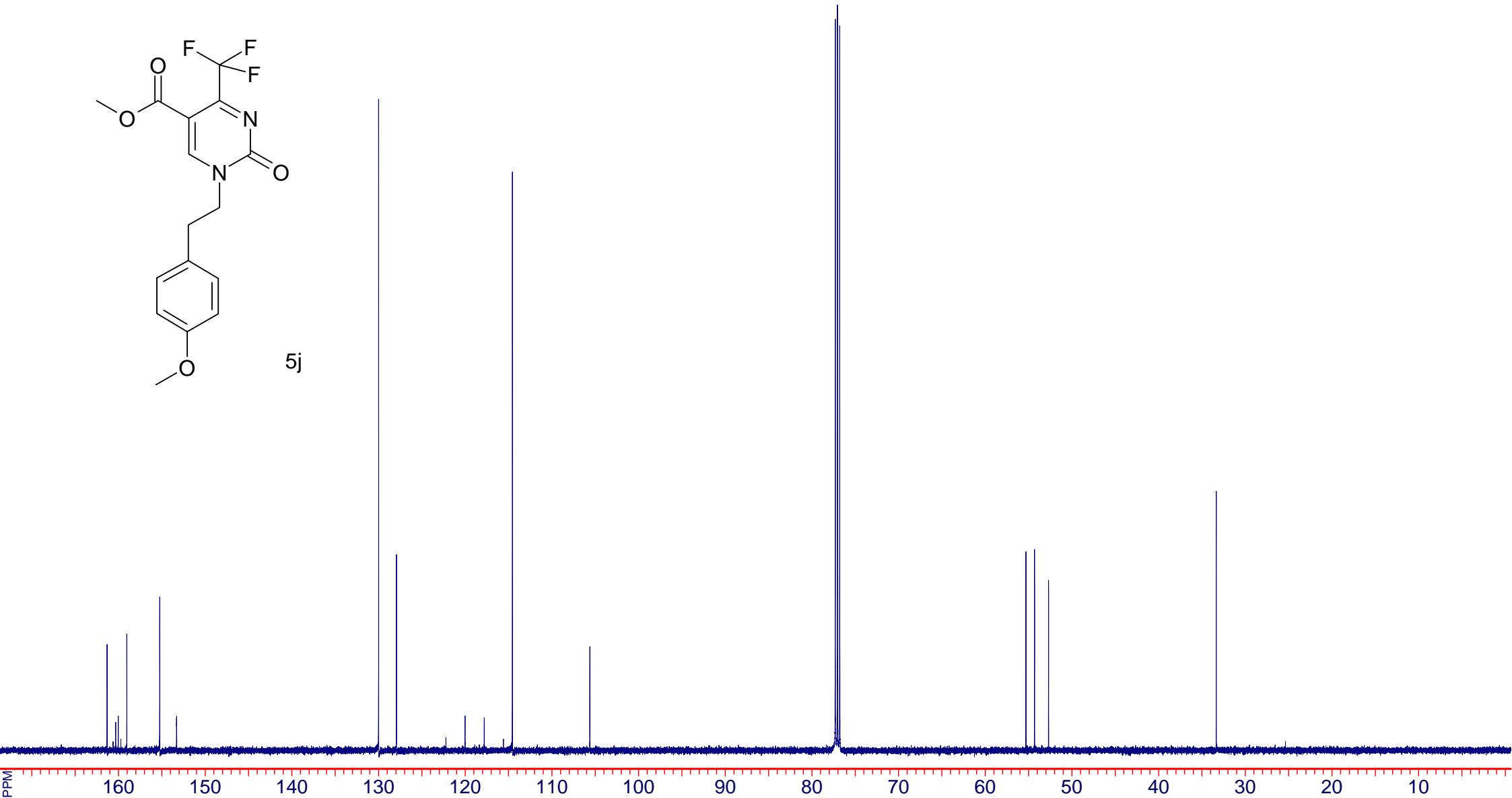
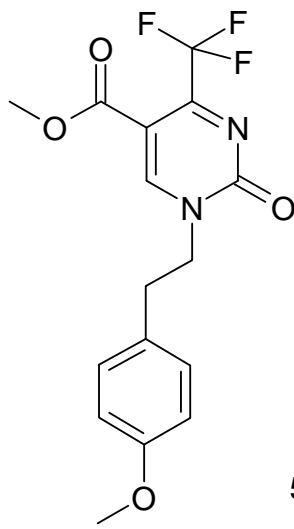
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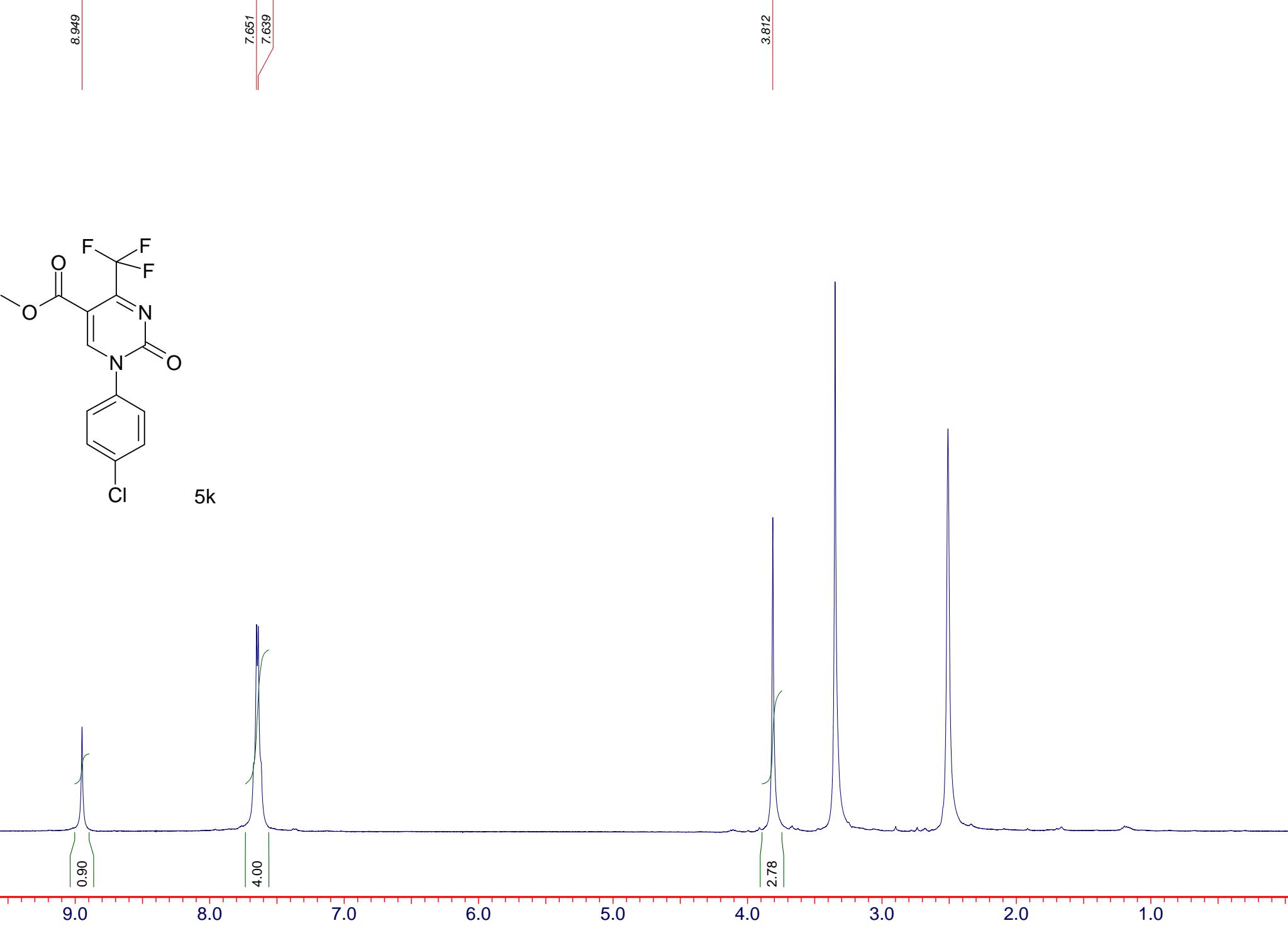
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54.32

52.70

33.36



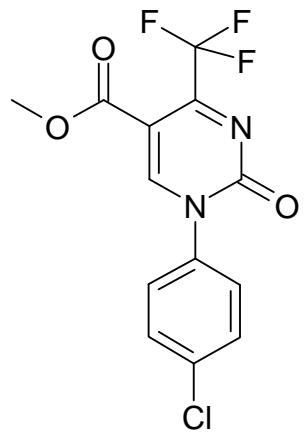


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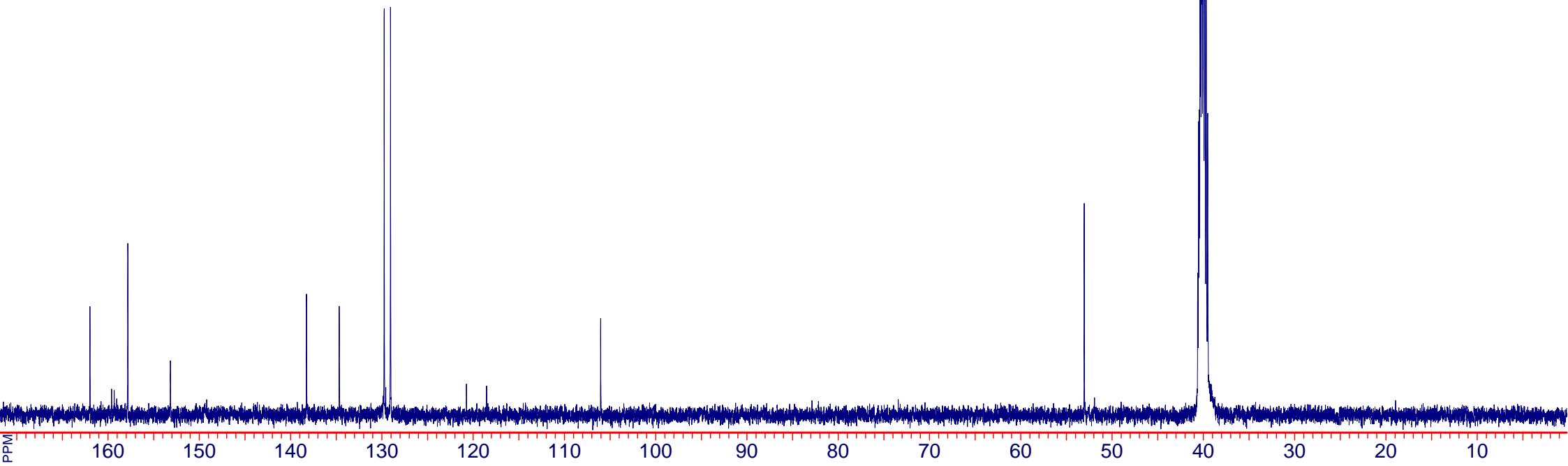
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138.25  
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118.52

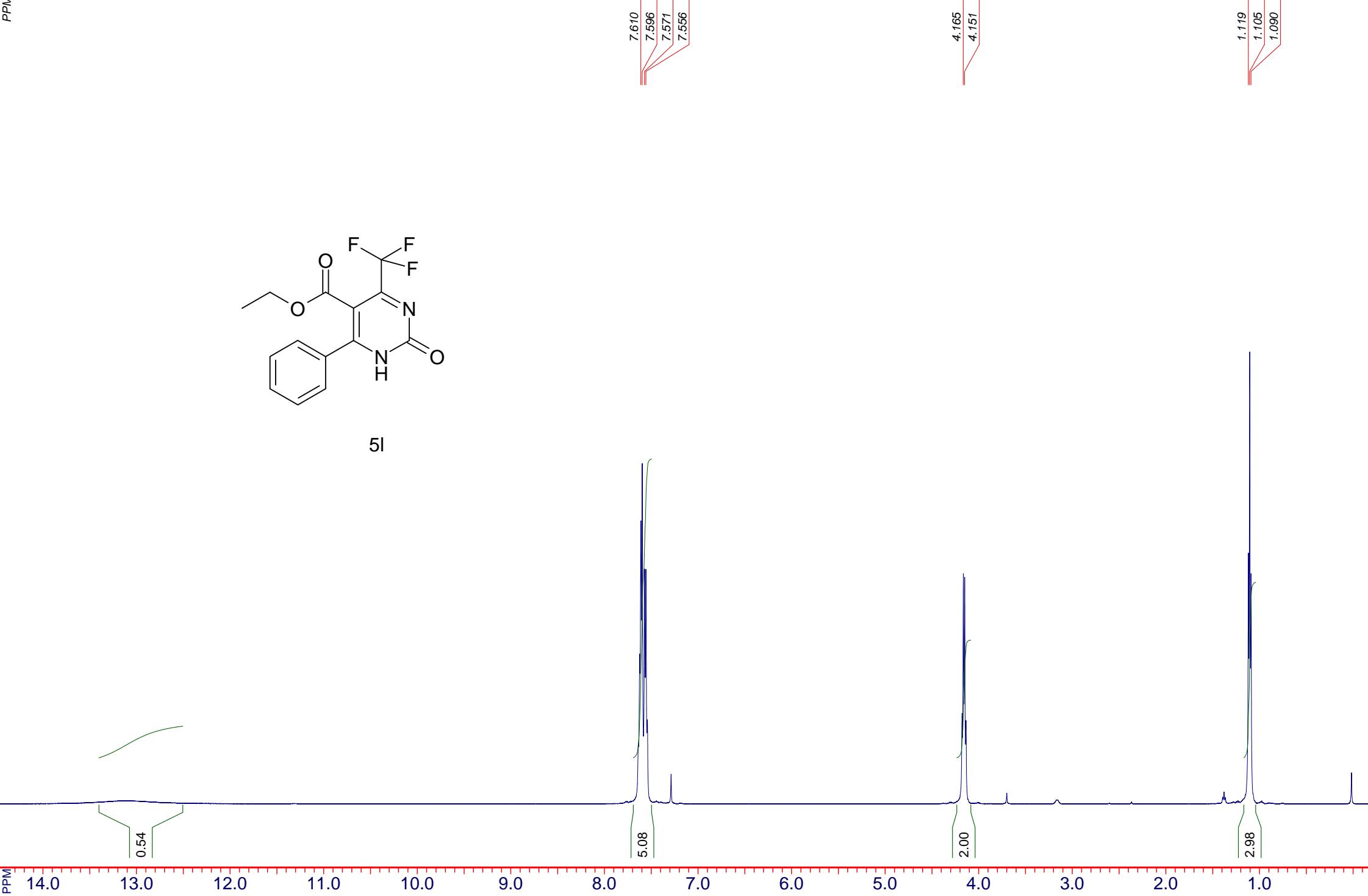
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5k



PPM



PPM

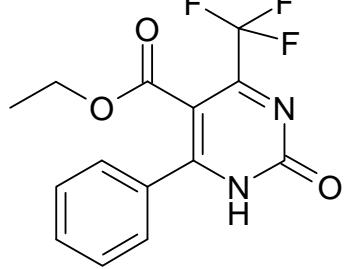
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160.32  
157.15

132.52  
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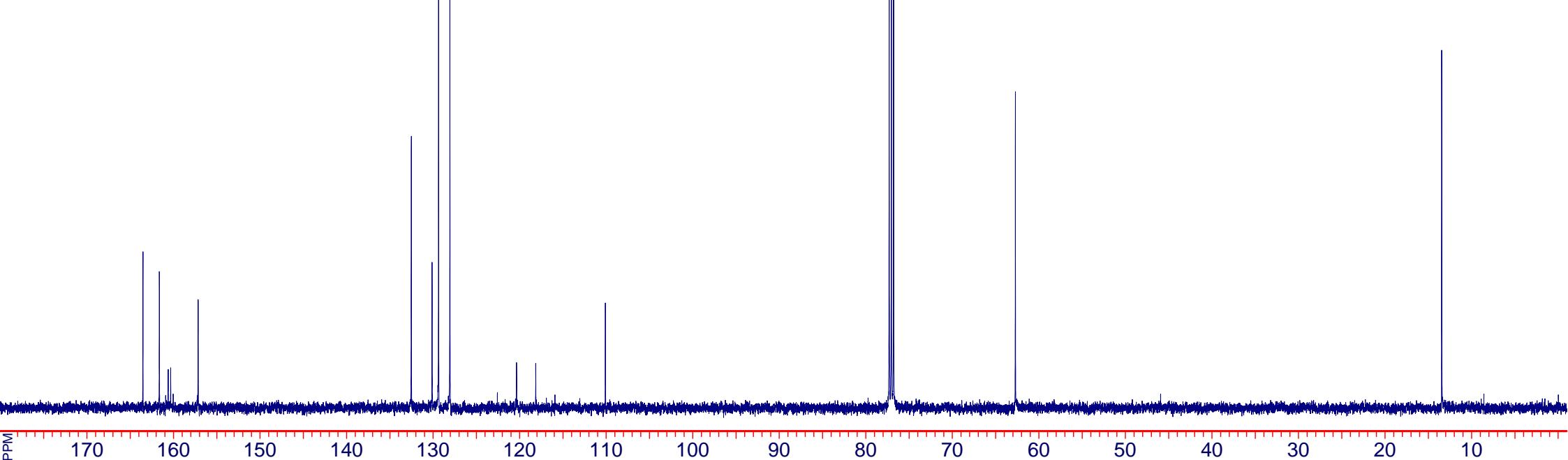
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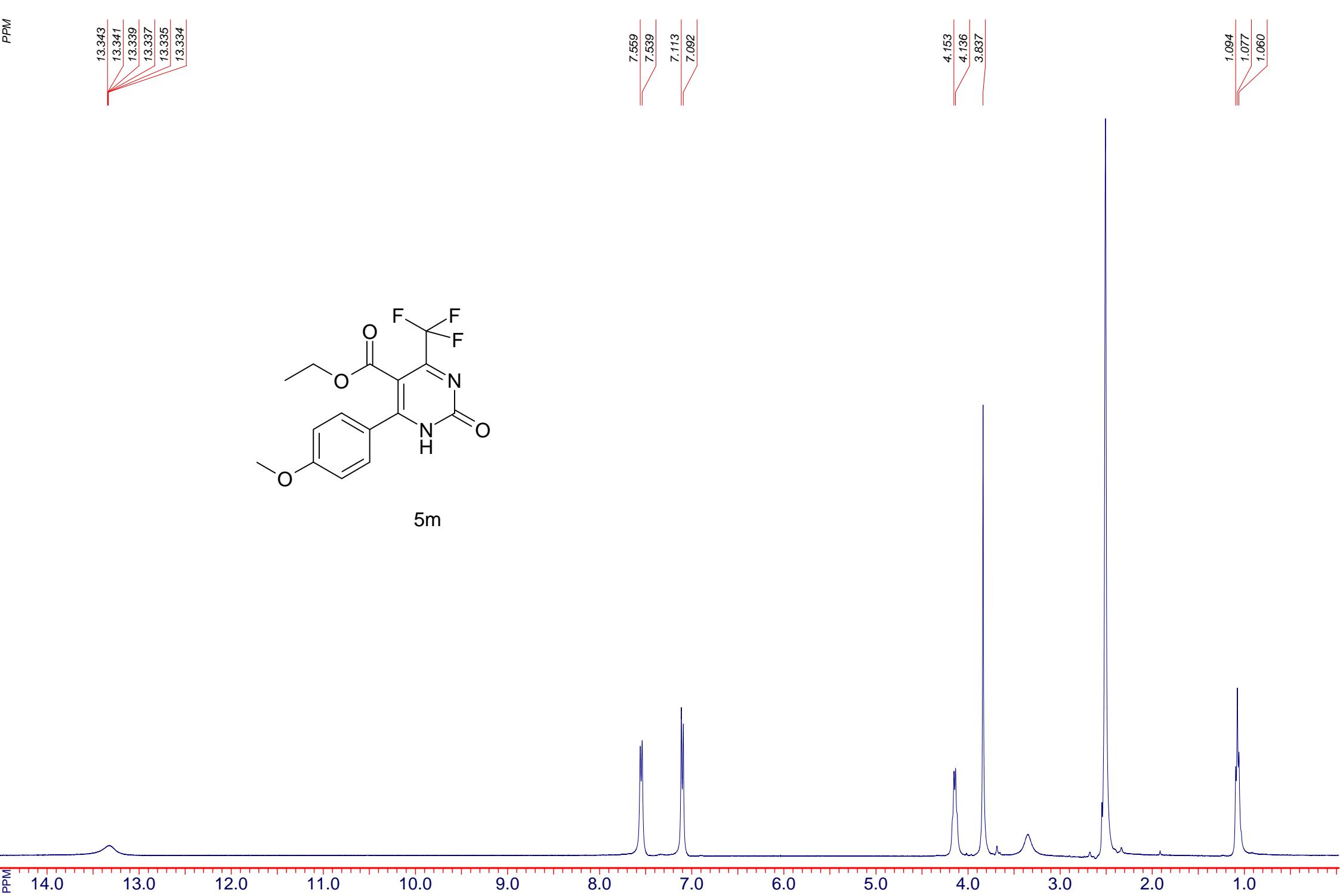
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13.46

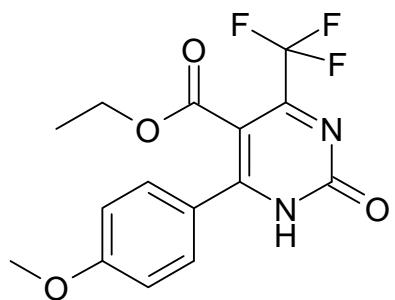


5l

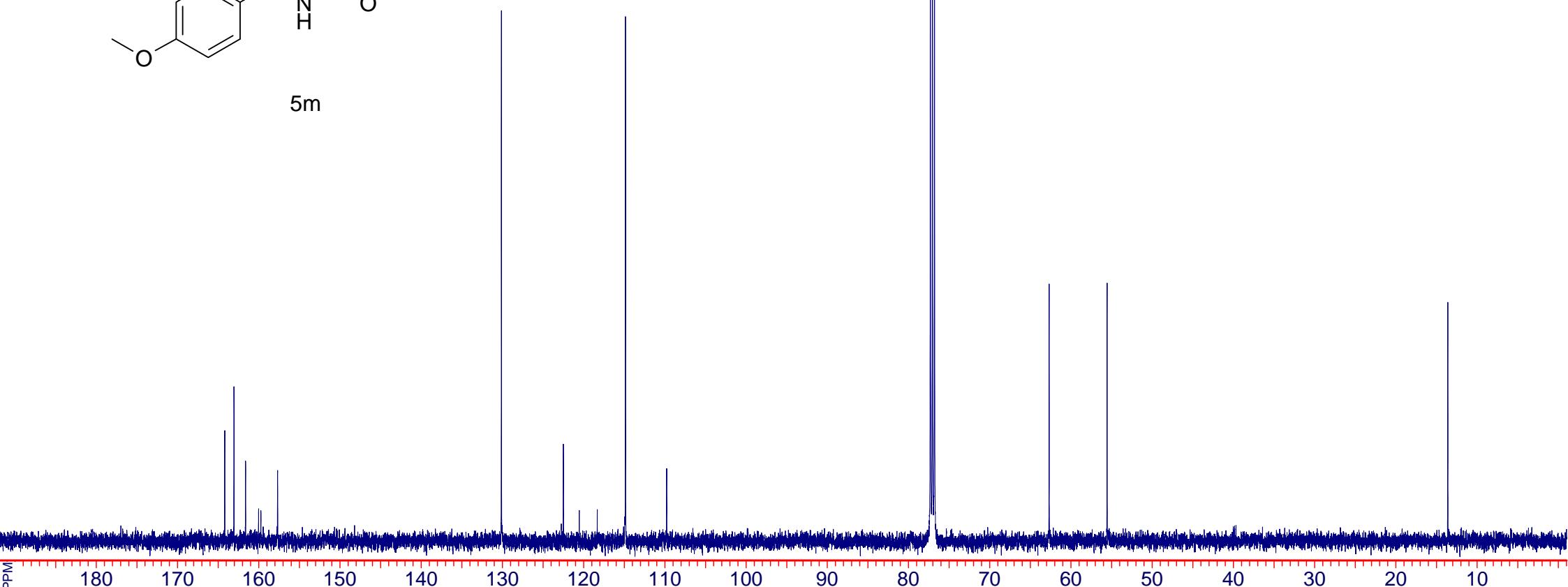




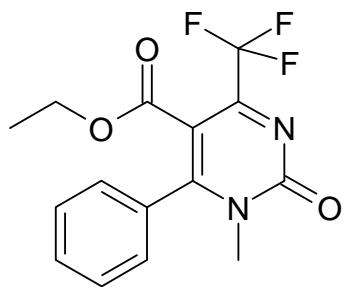
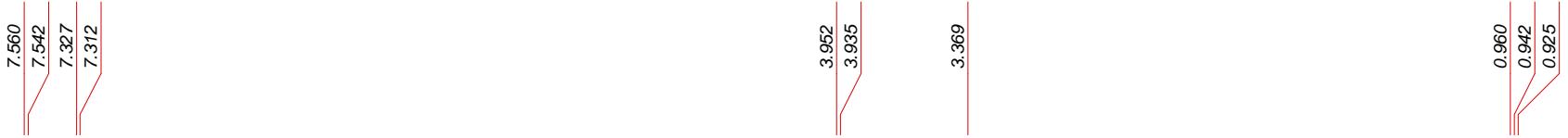
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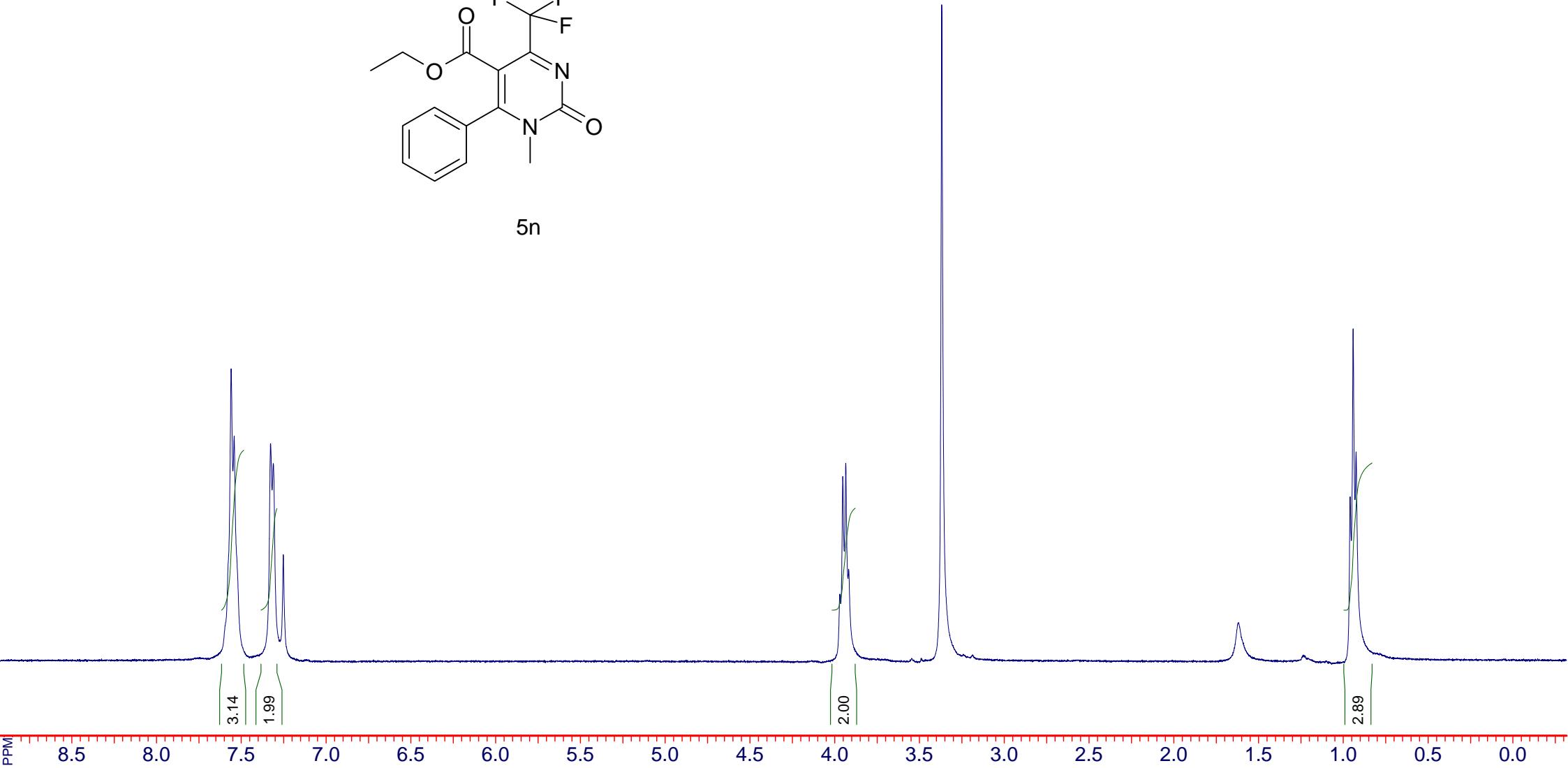
5m



PPM



5n



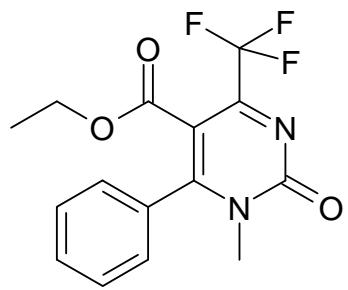
PPM

163.06  
162.42  
158.46  
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118.13  
110.31

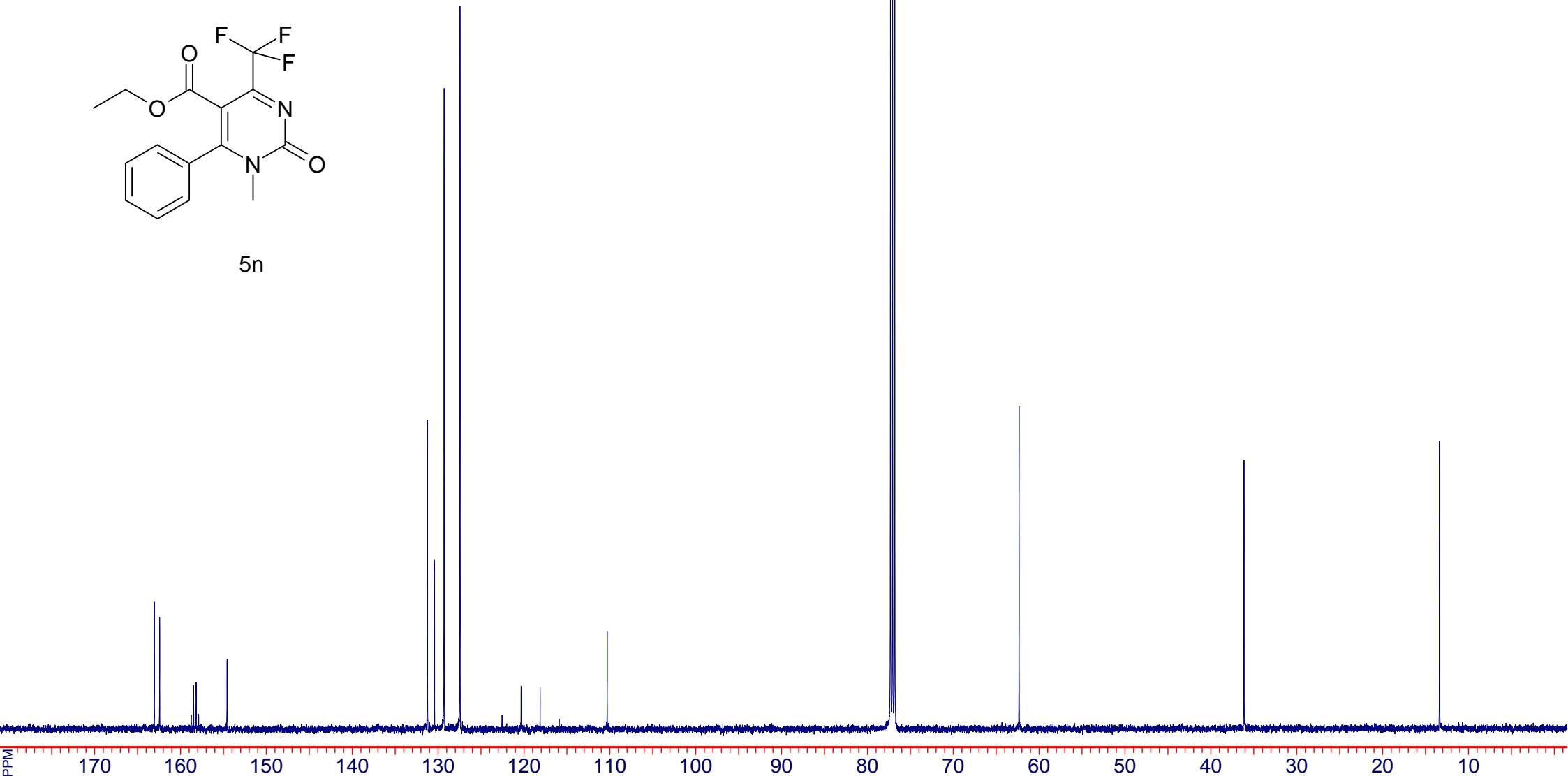
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36.14

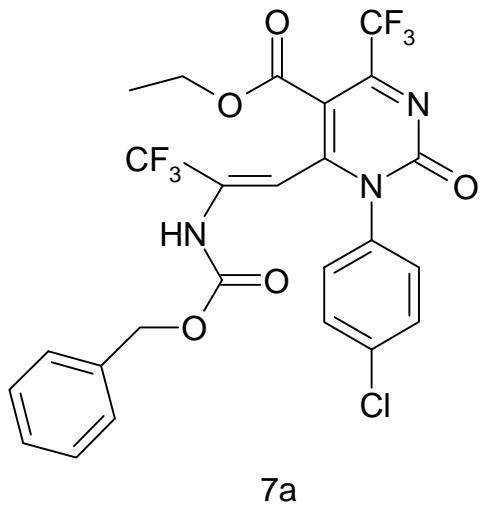
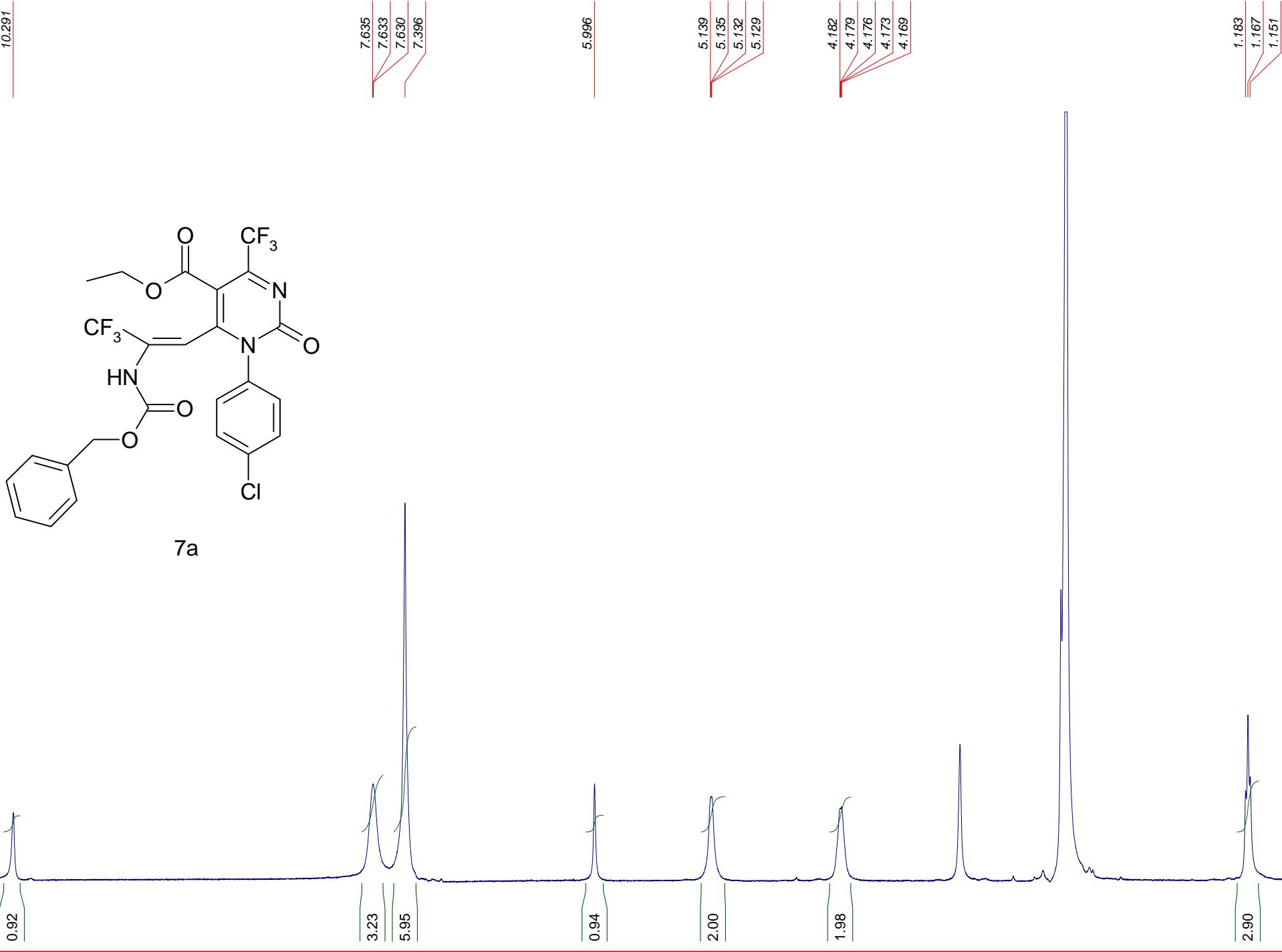
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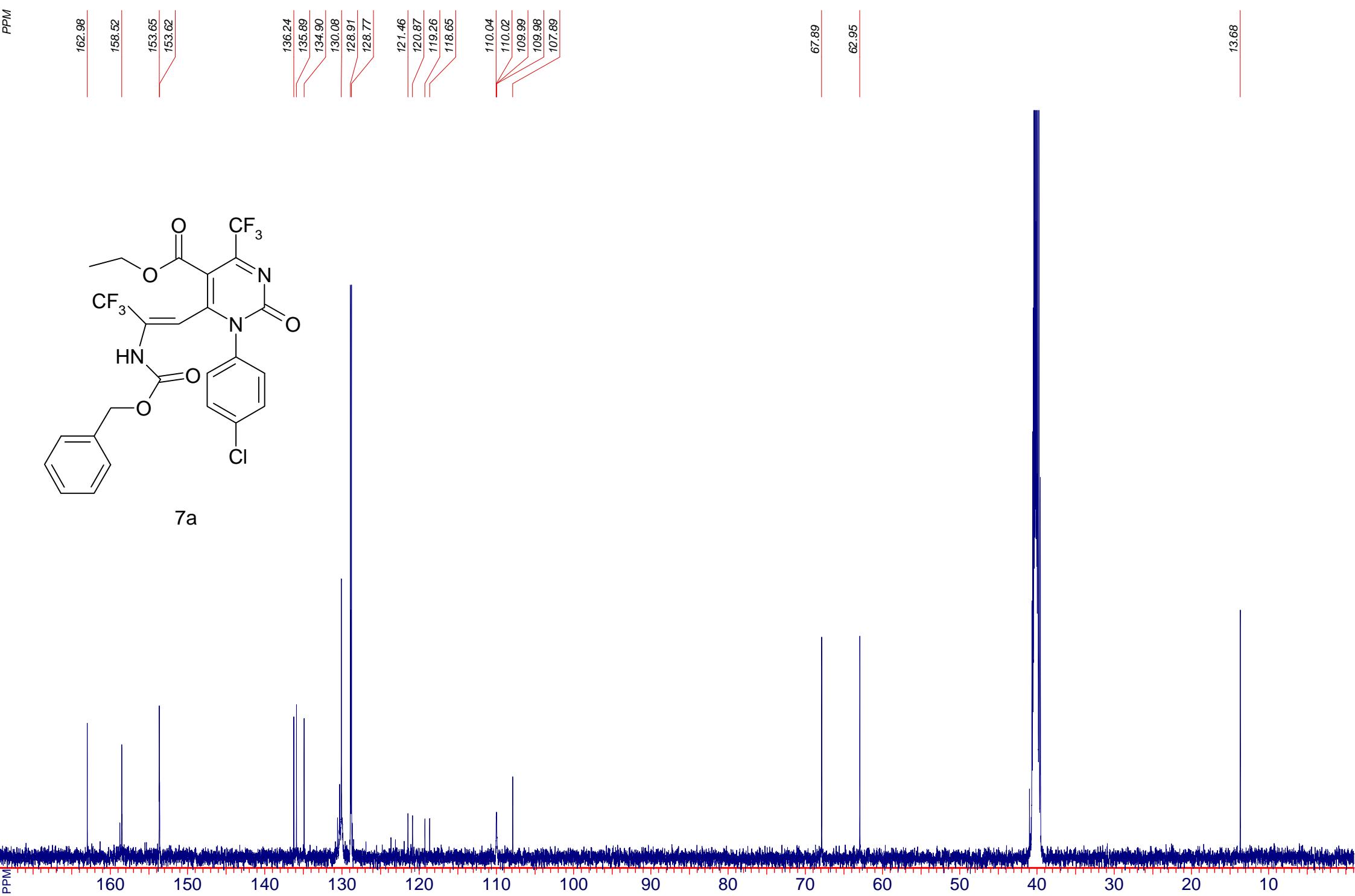


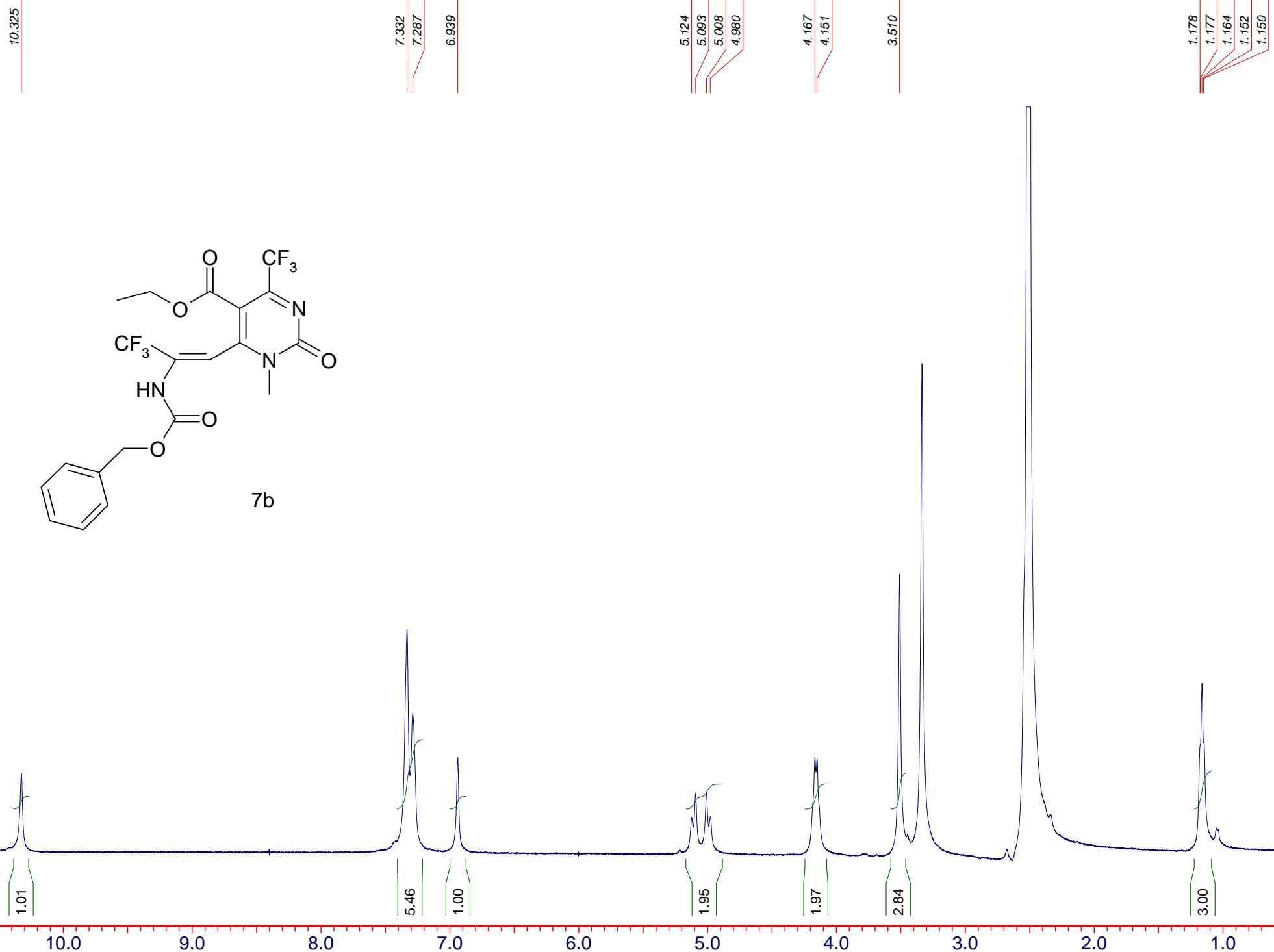
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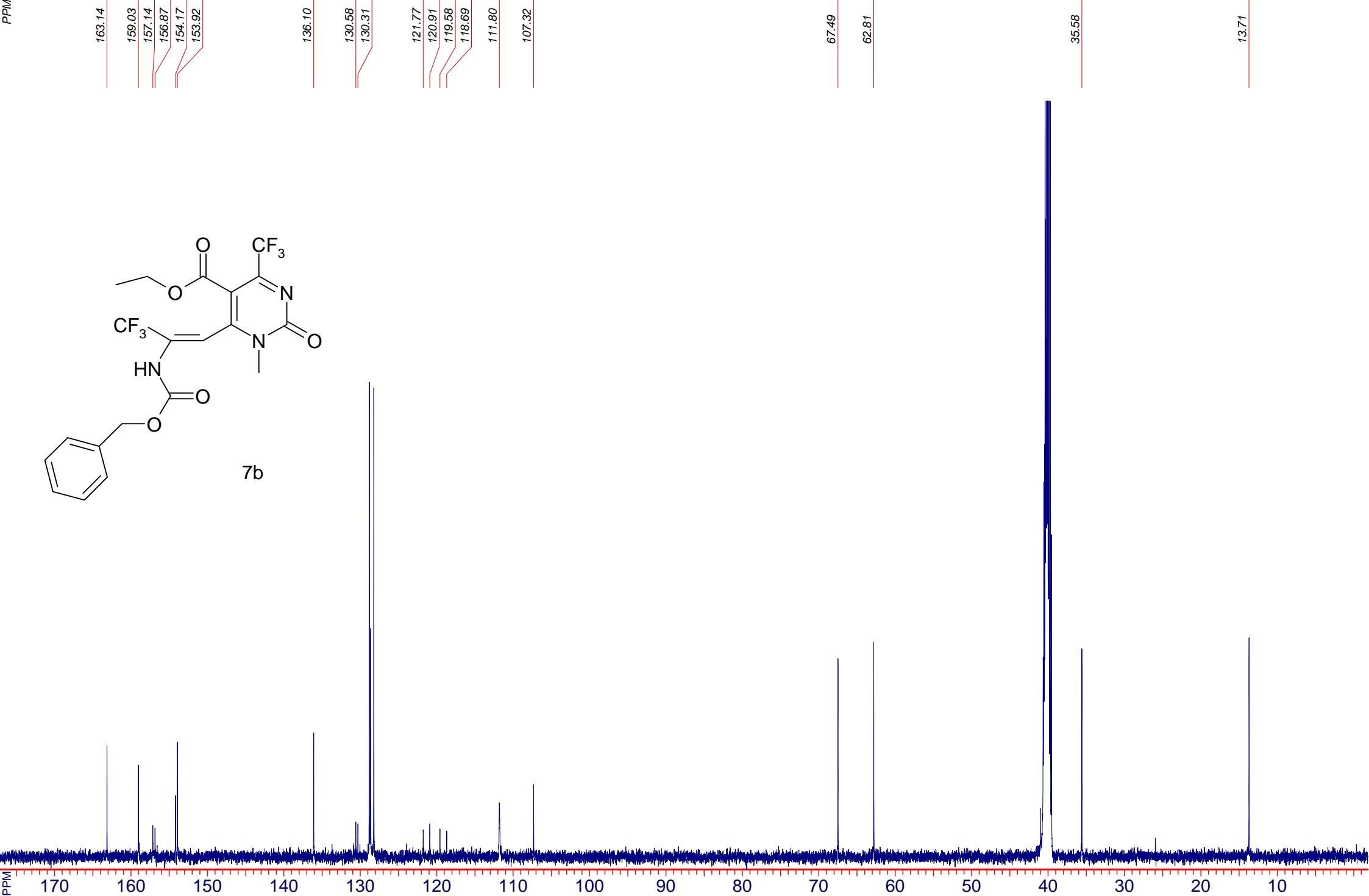


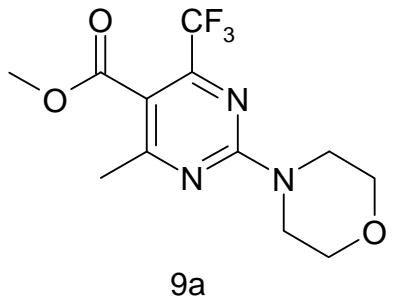
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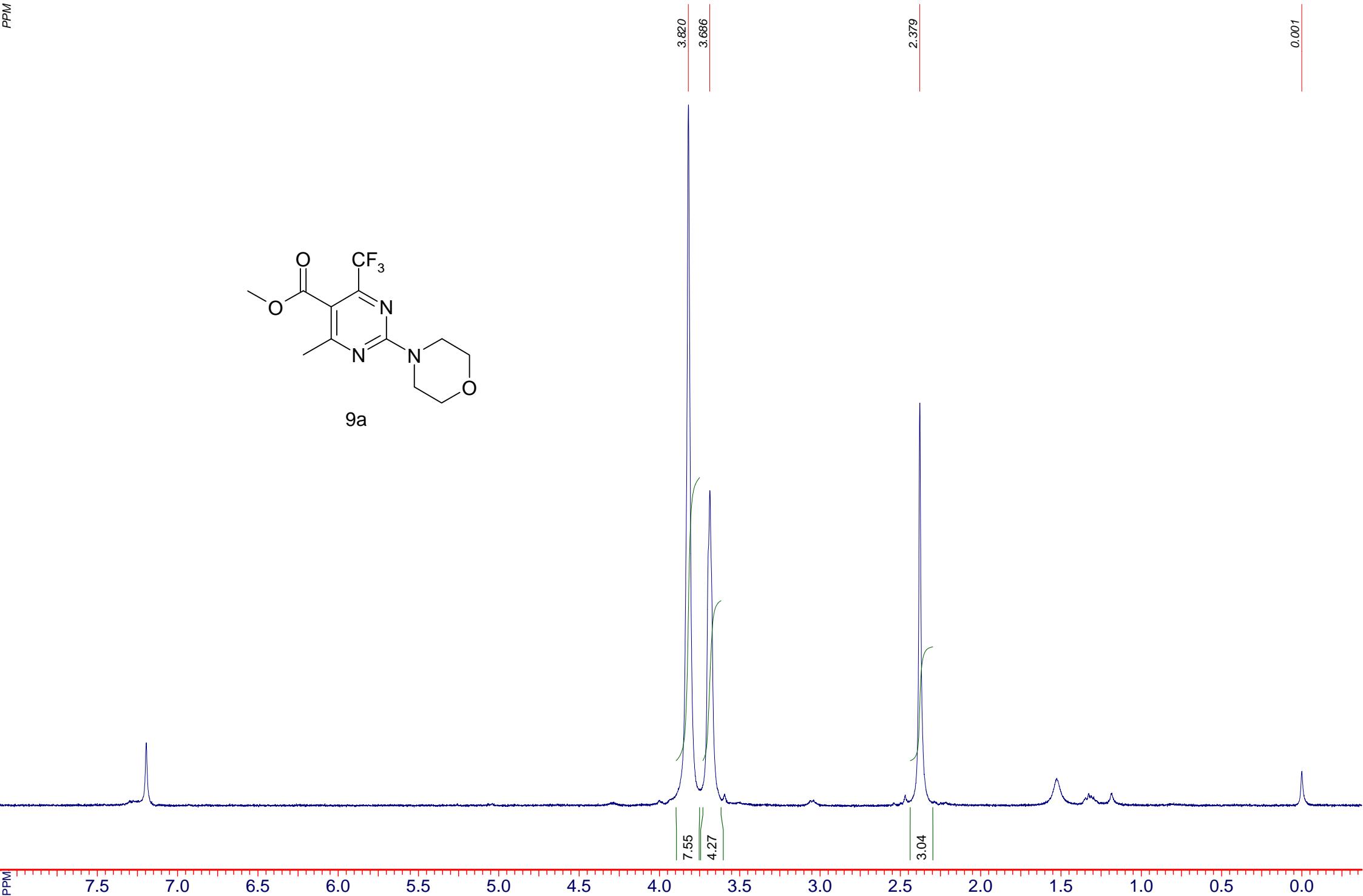








9a



PPM

PPM

168.59  
166.83

160.00

153.42  
153.40  
153.13  
153.12121.57  
119.37

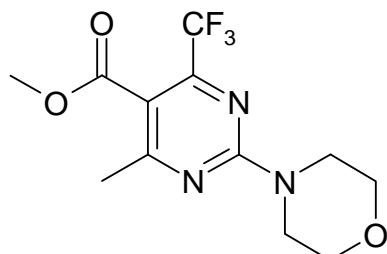
112.65

66.70

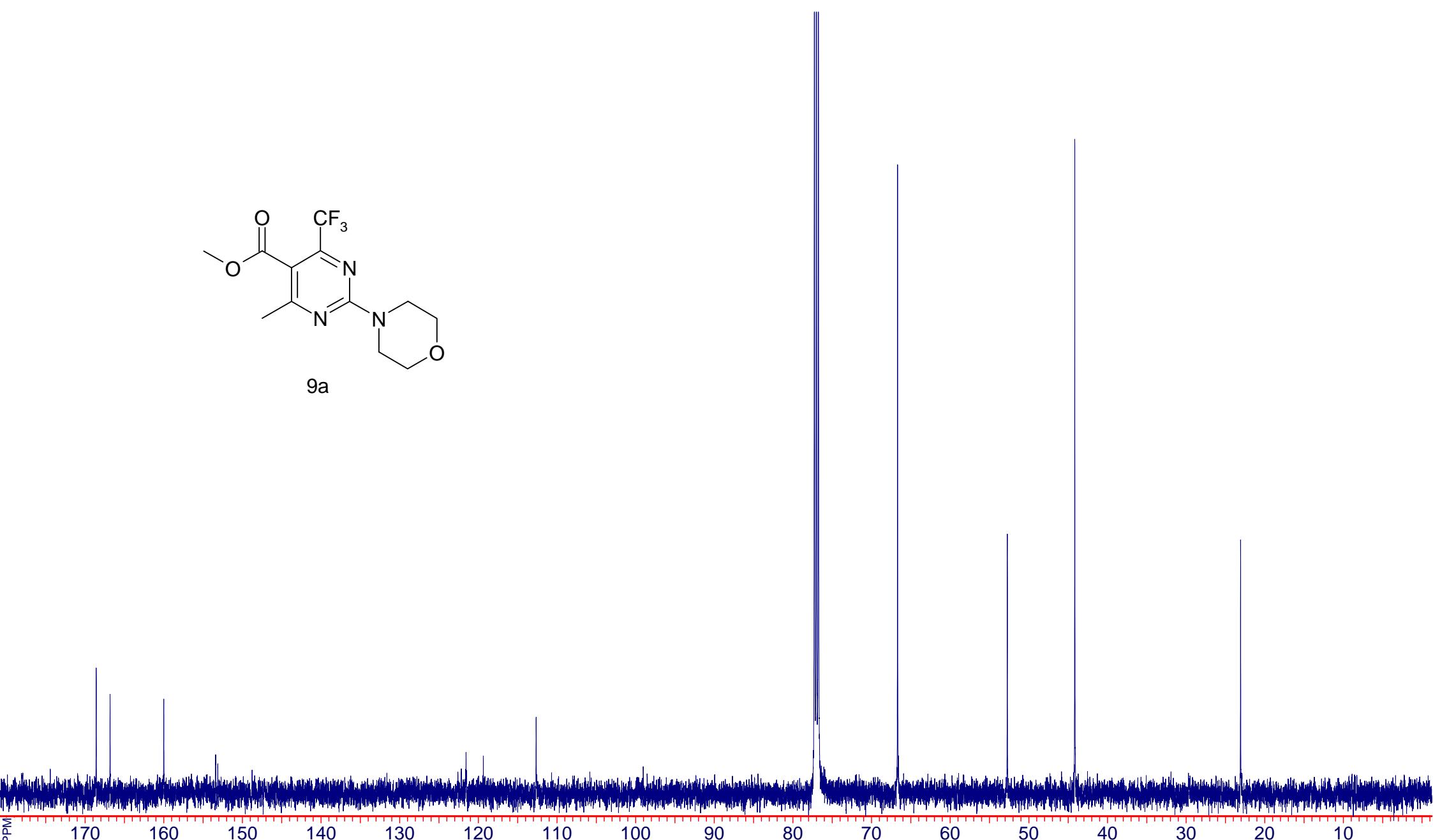
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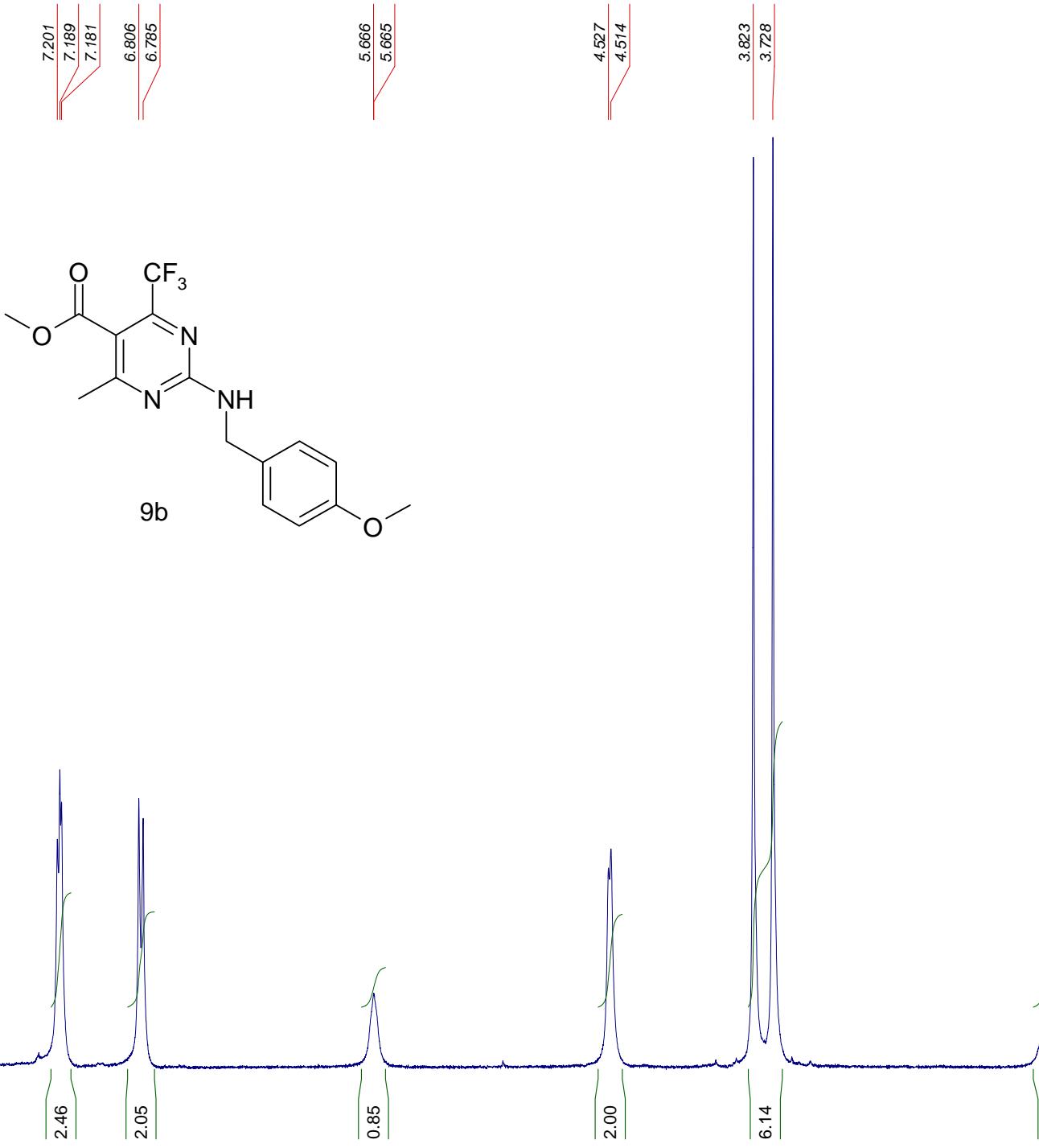
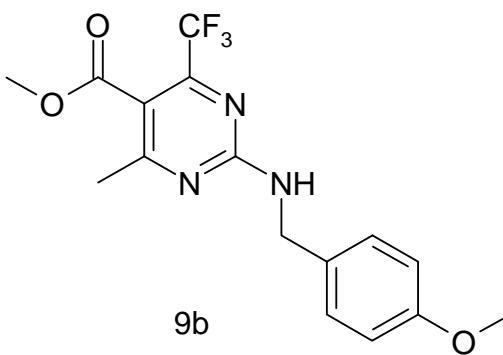
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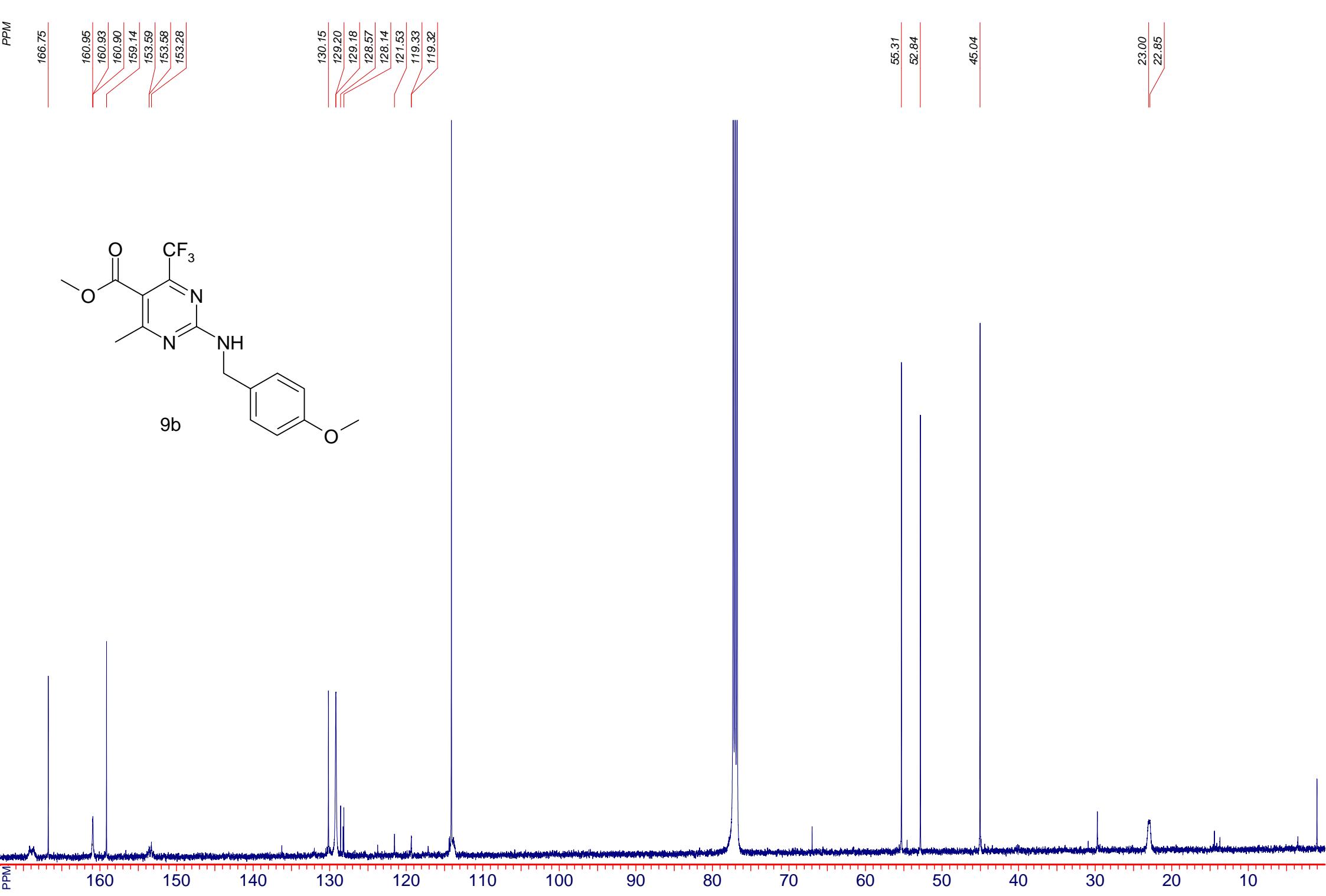
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9a







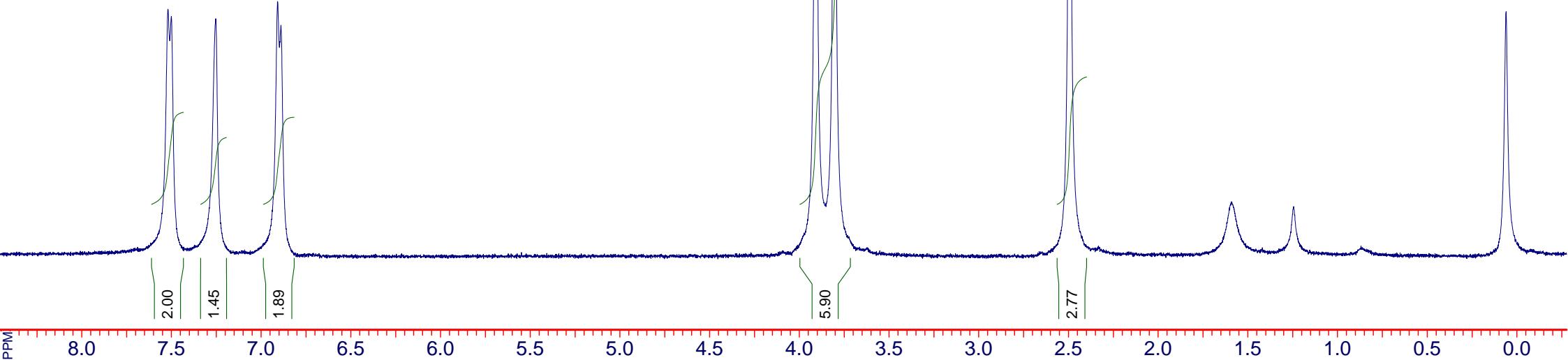
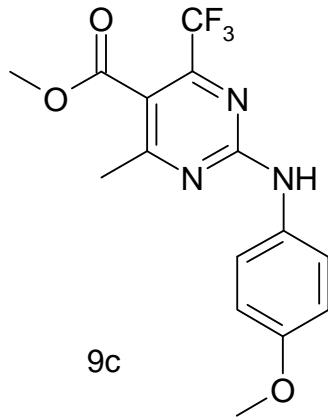
7.520  
7.501  
7.253

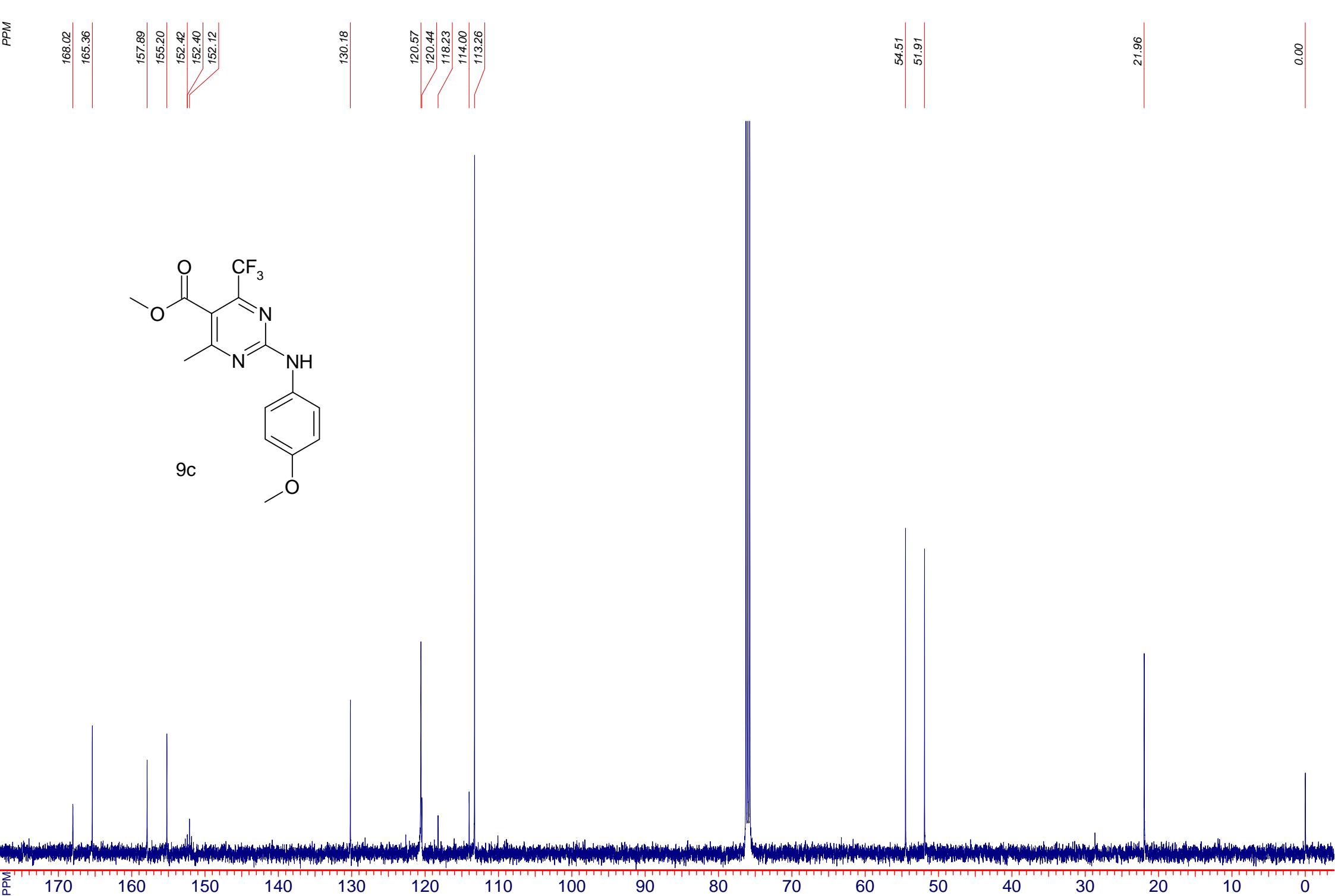
6.908  
6.889

3.908

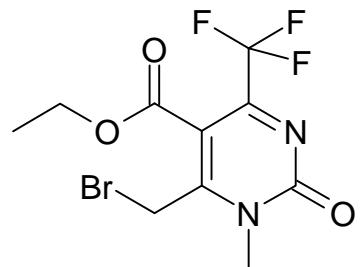
3.804

2.493

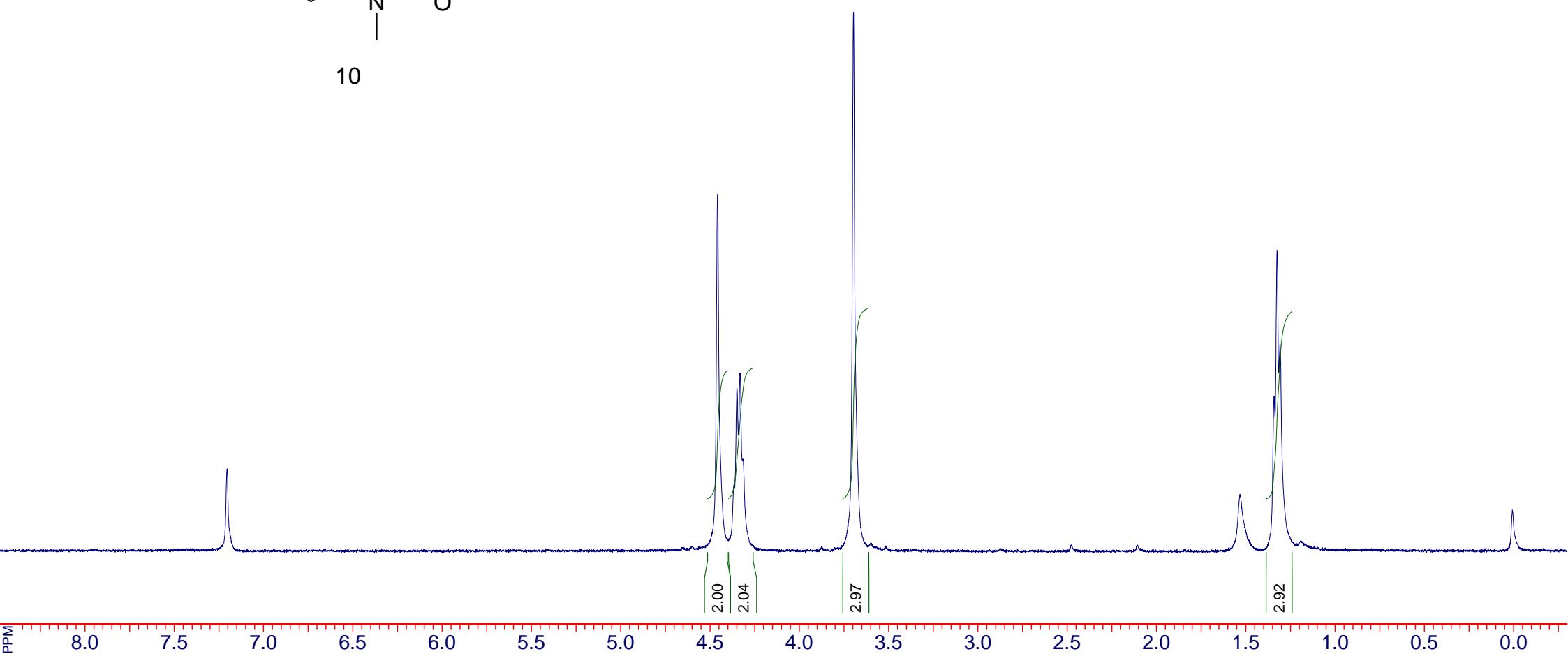




PPM



10



PPM

163.17  
159.50  
159.21  
158.35  
154.17120.14  
117.92

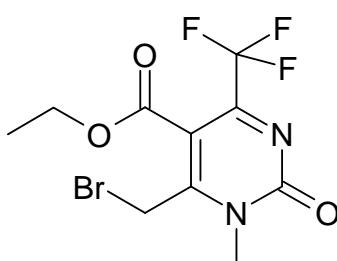
108.70

63.31

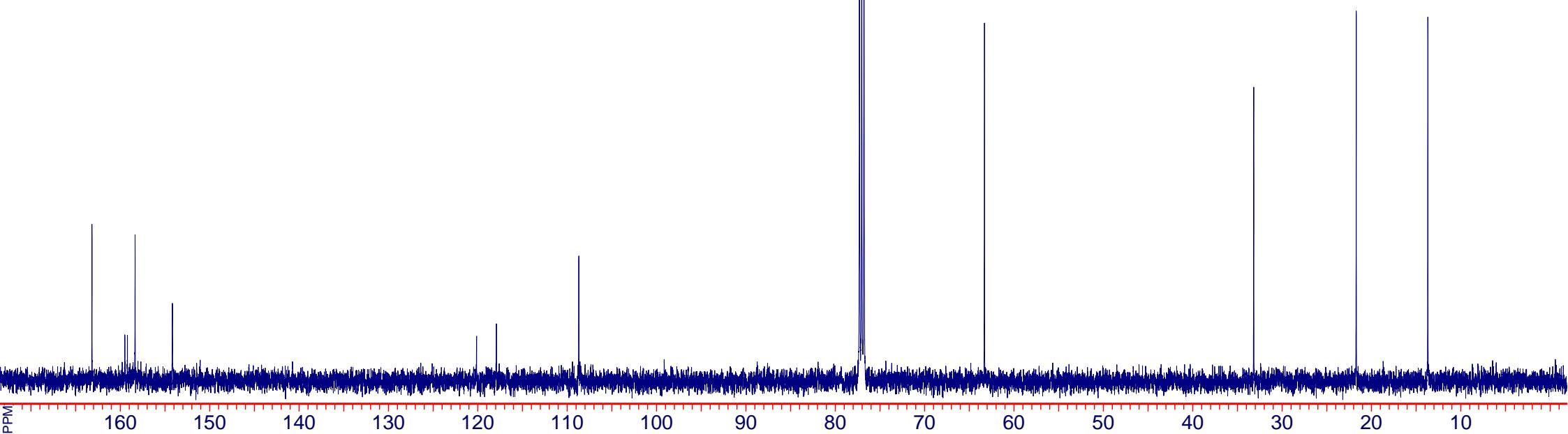
33.18

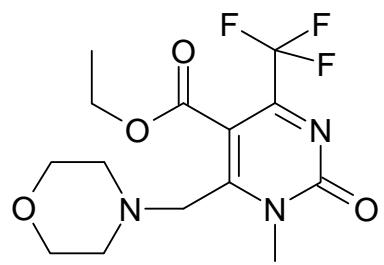
21.71

13.70

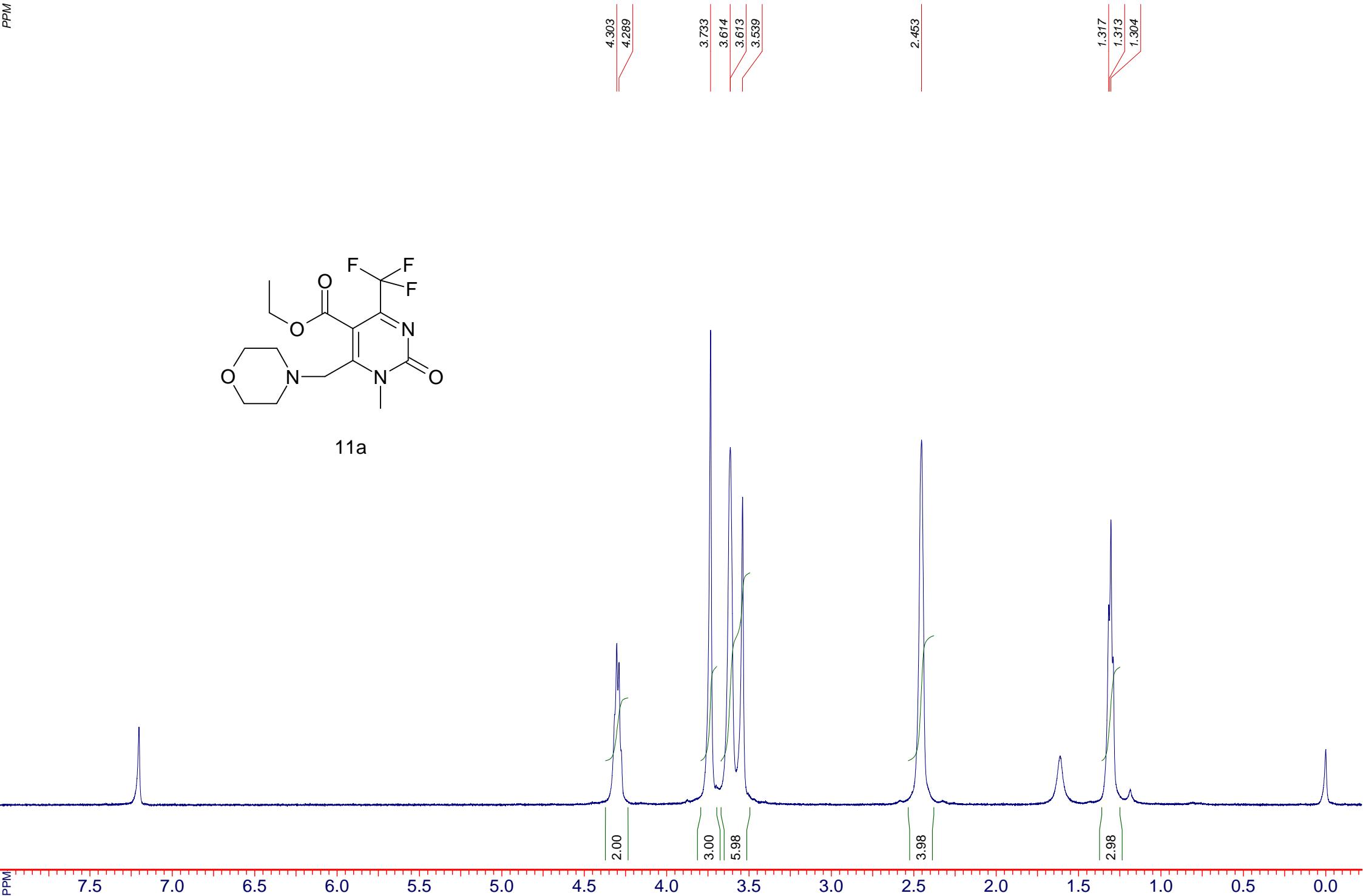


10





11a



PPM

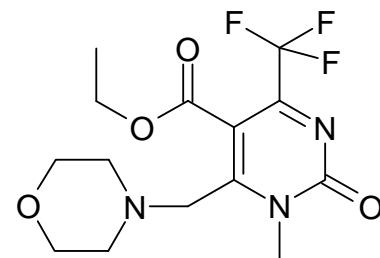
163.75  
159.58  
158.48  
158.19  
158.19  
154.90

120.34  
118.12  
110.57

66.65  
62.93  
56.68  
53.25

33.95

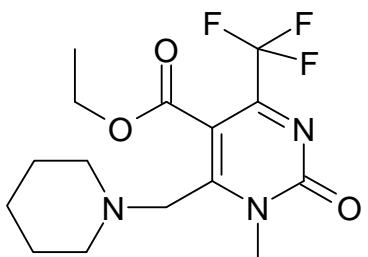
13.78



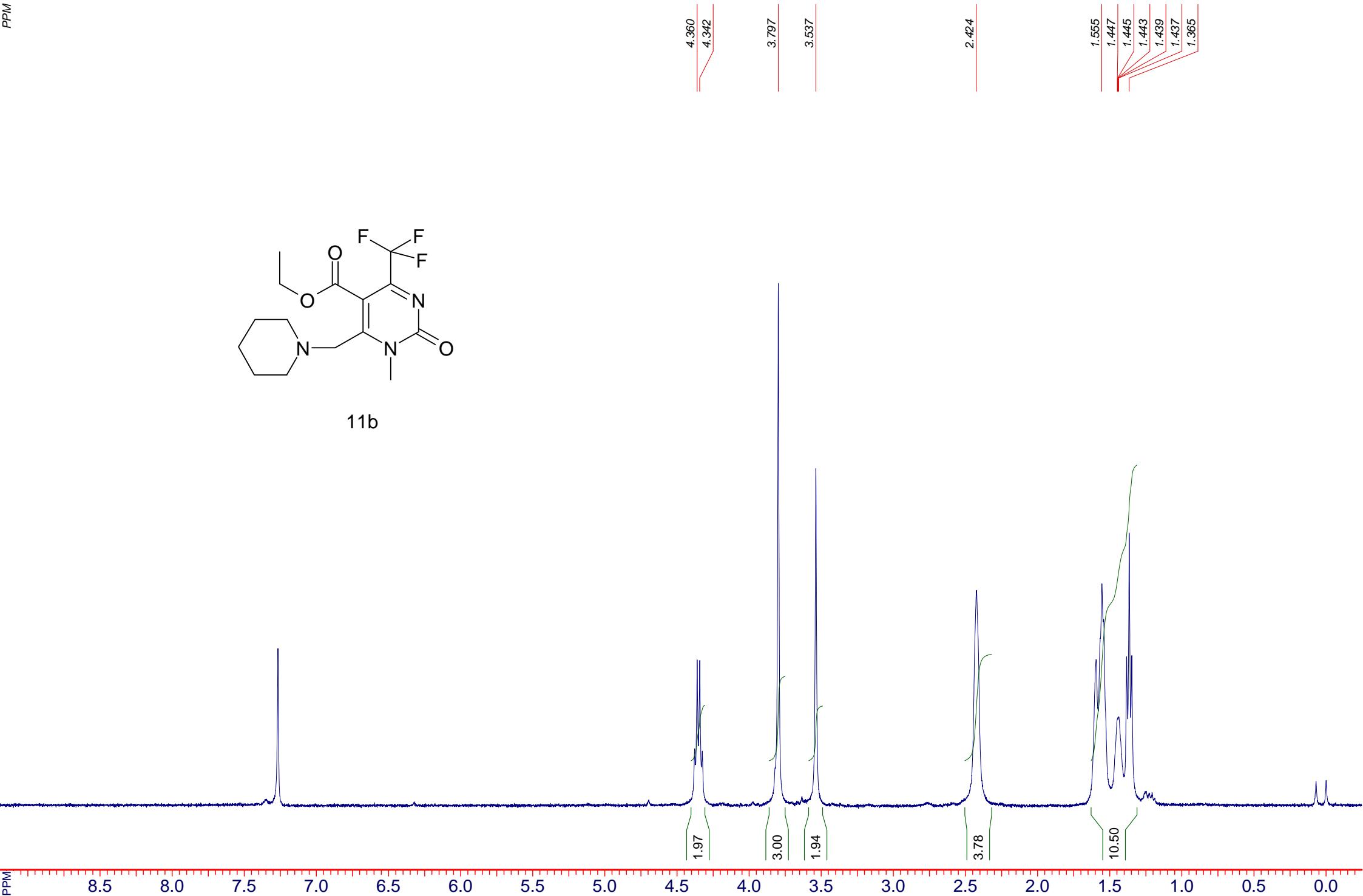
11a

PPM

160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0



11b



PPM

163.86  
160.87  
158.35  
158.07  
155.09

120.41  
118.19

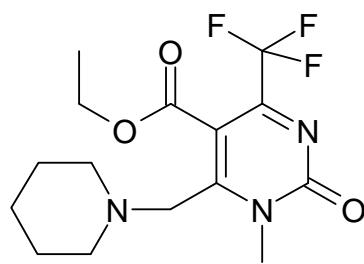
110.50

62.74  
57.24  
54.43

33.97

25.85  
23.68

13.80

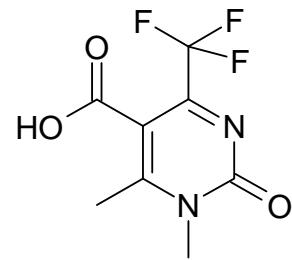


11b

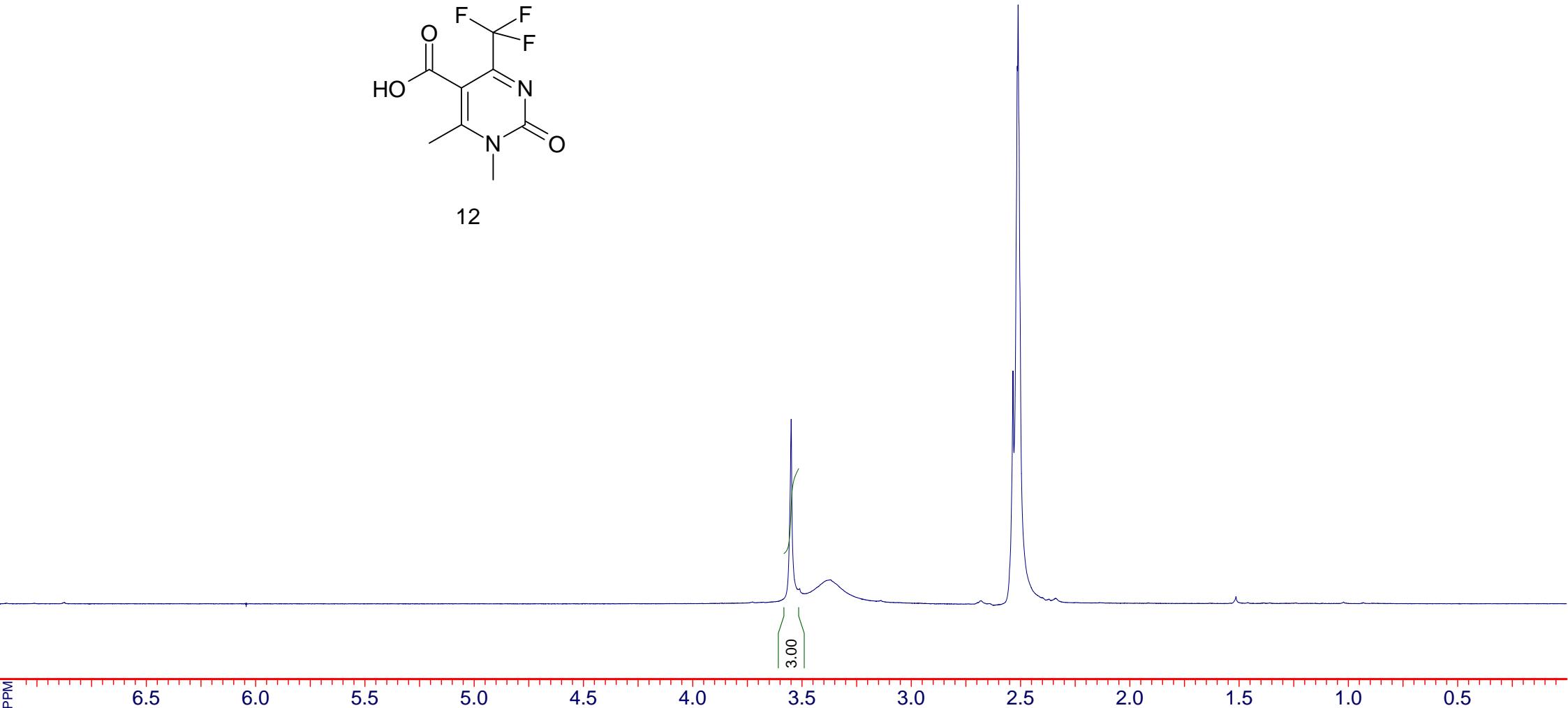
PPM

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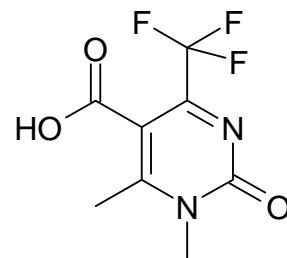
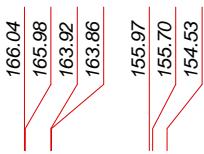
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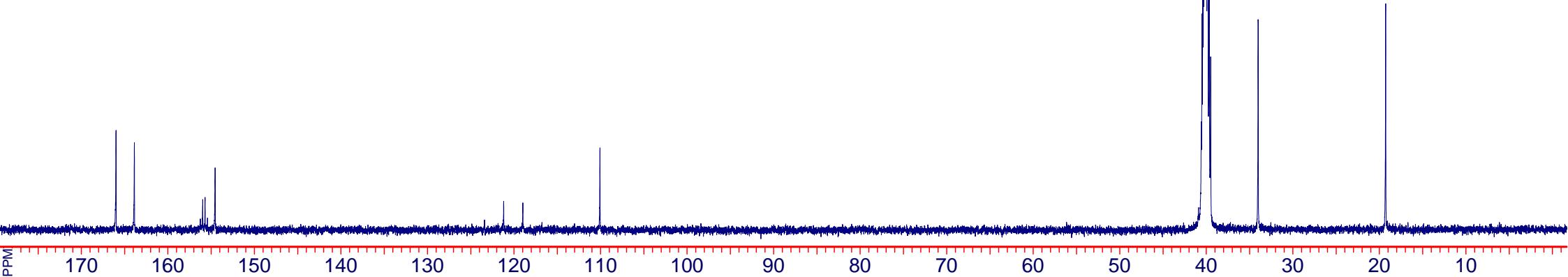
12

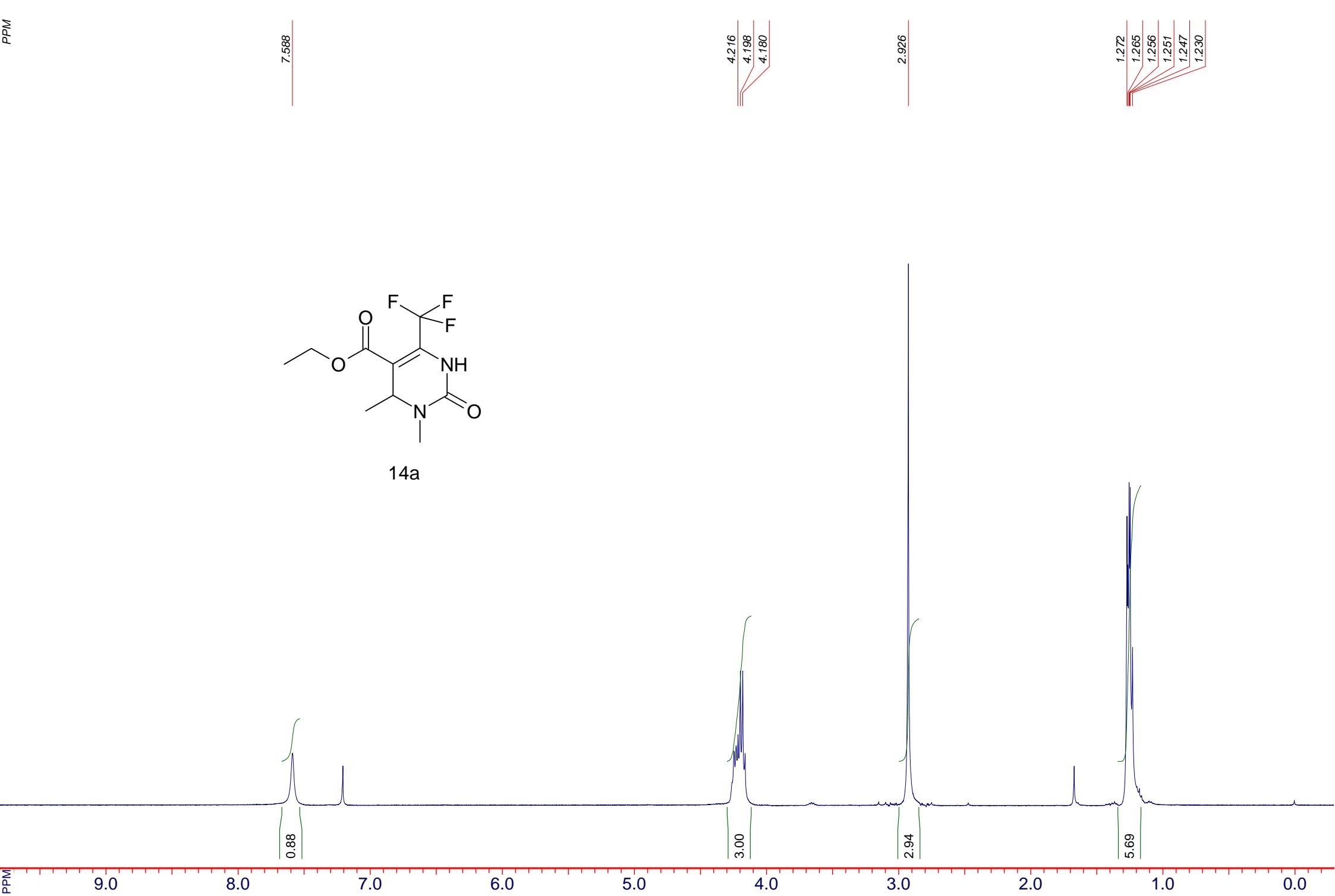


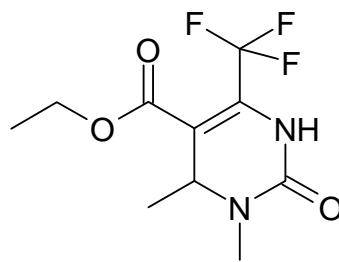
PPM



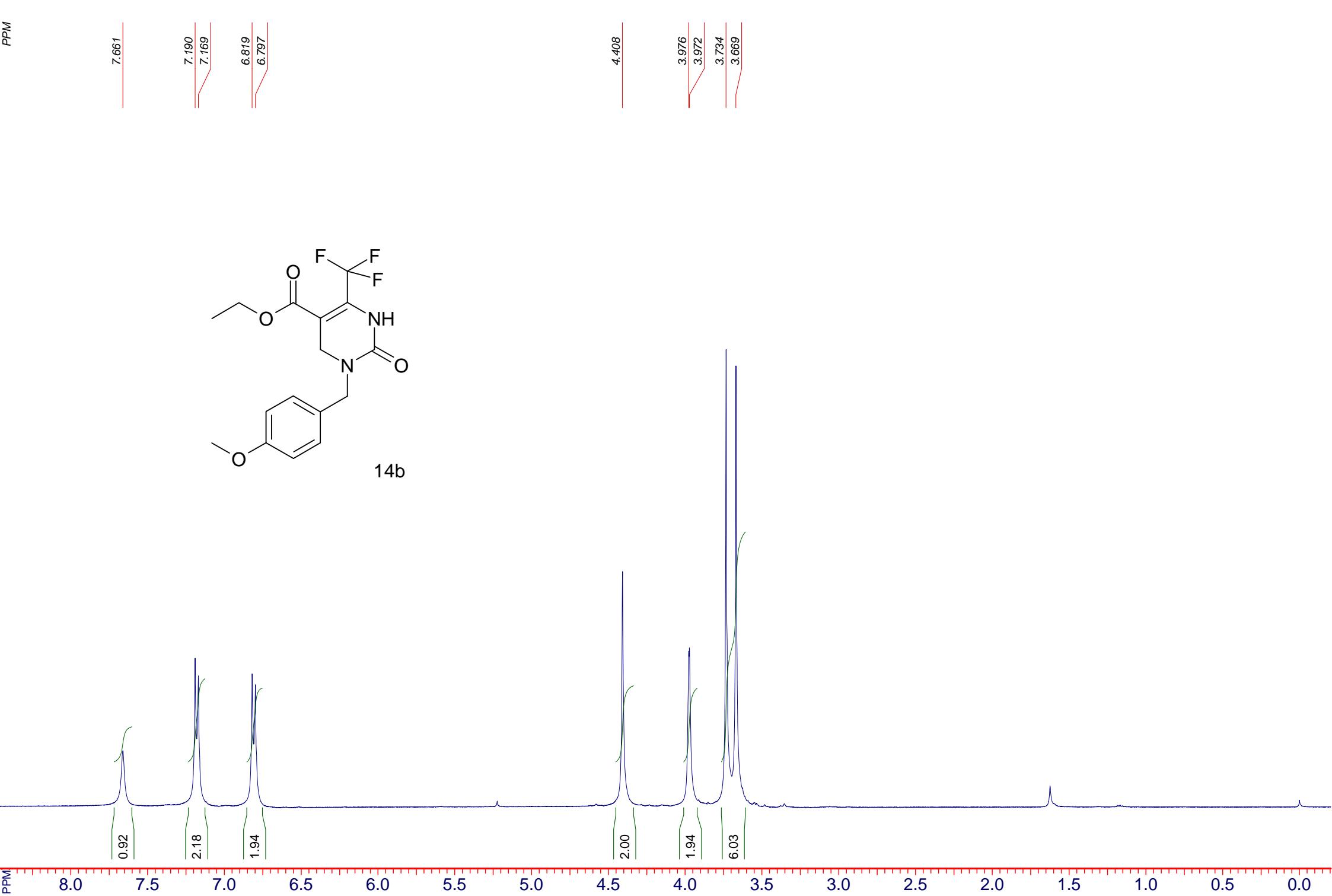
12







14a



PPM

163.25

159.49

134.01

133.72

129.81

127.28

120.19

118.00

114.20

102.28

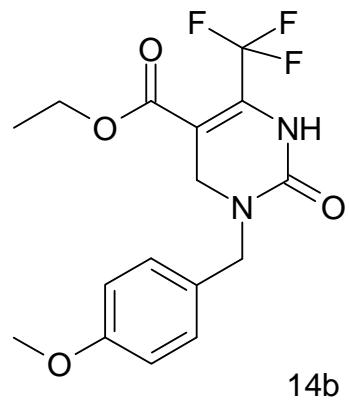
102.26

55.30

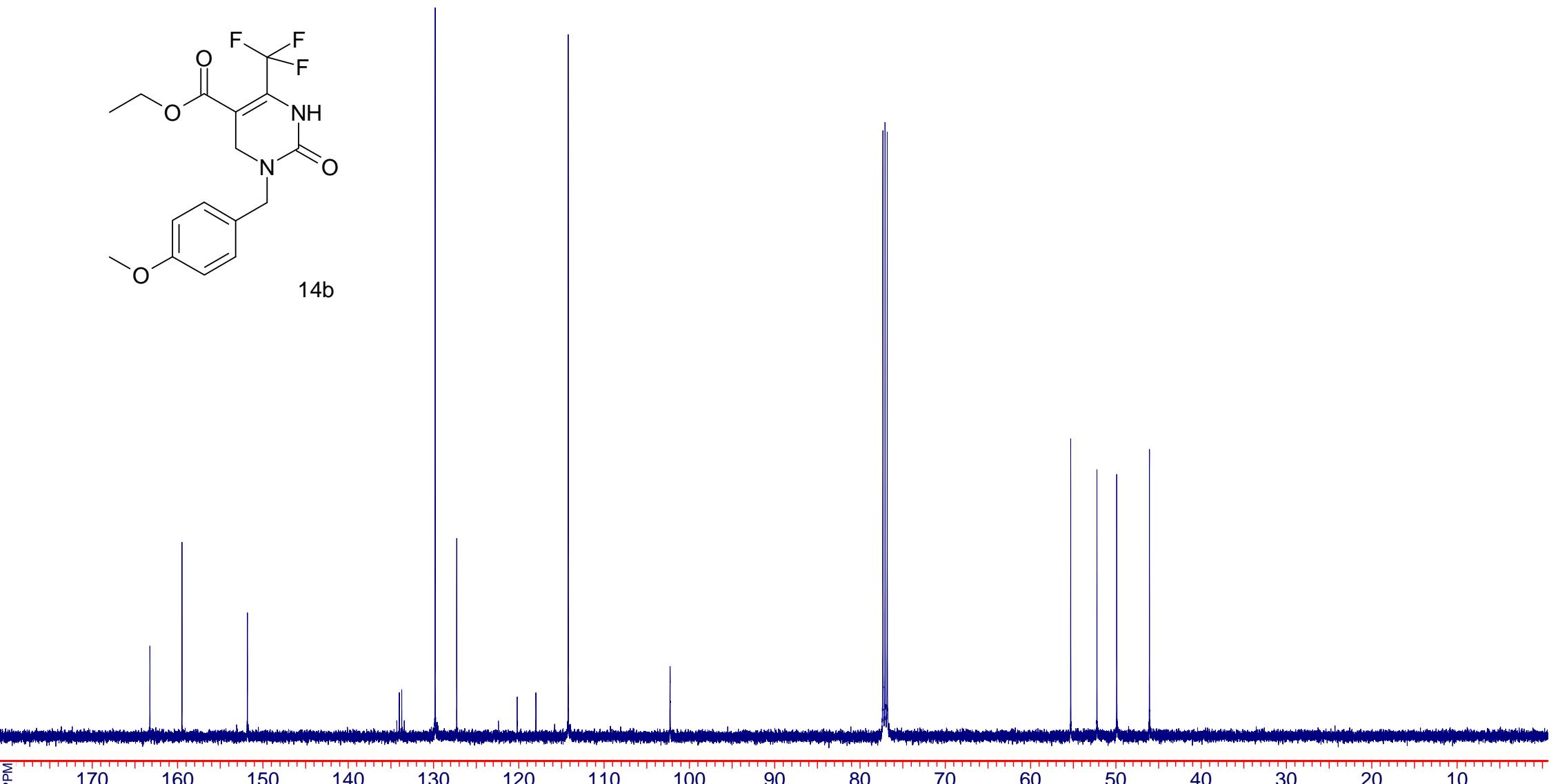
52.23

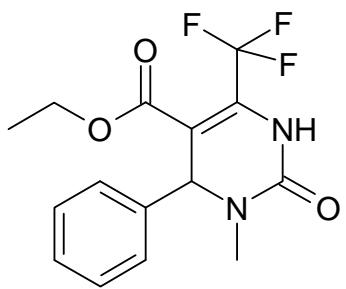
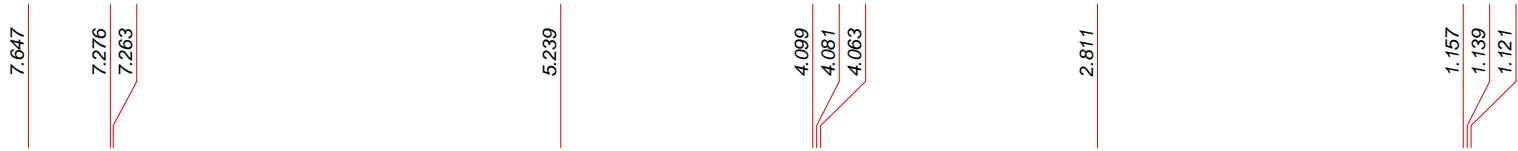
49.91

46.07

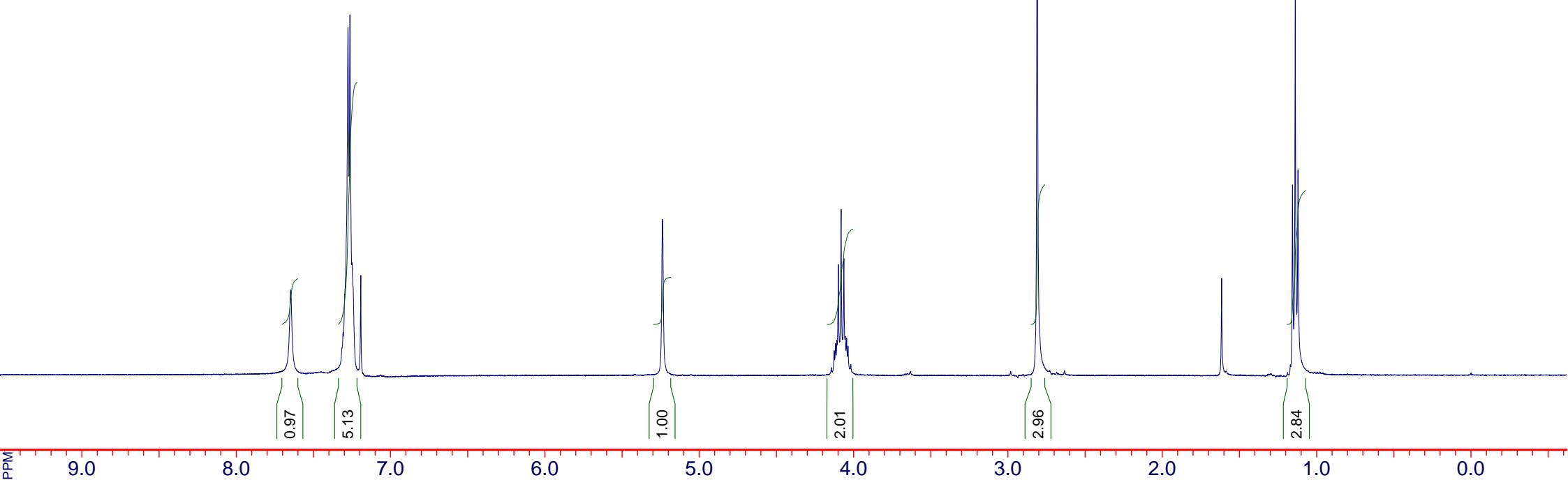


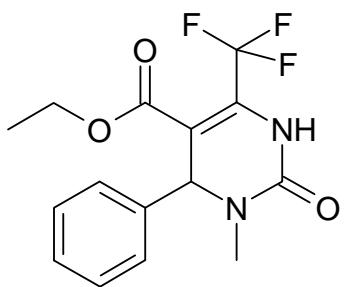
14b





14c





14c