Synthesis of *Meso*-Halogenated BODIPYs and Access to *Meso*-Substituted Analogues

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Boradiaza-s-indacene (*bo*ron *dipy*rrin or *bo*ron *dipy*rromethene, BODIPY) dyes have earned a prominent place as outstanding fluorophores for use in fluorescent materials, labels, and probes.¹ A rich chemistry has been developed for the decoration of the BODIPY scaffold with reactive functionalities.² Derivatization at the 8-position, or *meso*position, is a preferred method for the construction of complex BODIPY fluorophores because of the straightforward synthesis starting from aromatic aldehydes or acylium equivalents.^{2,3}

An elegant alternative to the introduction of functionality at the 8-position by linear synthesis is the use of 8-methylthioBODIPYs. Such thioethers, introduced by Bielmann and co-workers, undergo nucleophilic aromatic substitution (S_NAr via addition-elimination) with amines to form dyes with blue-shifted UV-vis absorption and fluorescence spectra in comparison to common boradiazas-indacenes.⁴ This work was followed by reports on the palladium-catalyzed derivatization of the BODIPY scaffold using the Liebeskind–Srogl cross-coupling starting from 8-methylthioBODIPYs and boronic acids.⁵ Bielmann's methodology⁴ has found application in the preparation of blue-emitting boron dipyrrin dyes.⁶

However, the scope of thioether substitutions remains limited when compared to the plethora of reactions described for aromatic halogens.⁷ To fill this void, we report the synthesis of 8-halogenated (Cl, Br, I) boron

 ^{(1) (}a) Treibs, A.; Kreuzer, F. *Liebigs Ann. Chem.* **1968**, *718*, 208.
 (b) Haugland, R. *Handbook of Fluorescent Probes and Research Chemicals*, 10th ed.; Molecular Probes: Eugene, OR, 2005. (c) Boens, N.; Leen, V.; Dehaen, W. *Chem. Soc. Rev.* **2012**, *41*, 1130.

^{(2) (}a) Loudet, A.; Burgess, K. Chem. Rev. 2007, 107, 4891.
(b) Ulrich, G.; Ziessel, R.; Harriman, A. Angew. Chem., Int. Ed. 2008, 47, 1184.

⁽³⁾ Wood, T.; Thompson, A. Chem. Rev. 2007, 107, 1831.

⁽⁴⁾ Goud, T. V.; Tutar, A.; Biellmann, J.-F. Tetrahedron 2006, 62, 5084.

^{(5) (}a) Peña-Cabrera, E.; Aguilar-Aguilar, A.; González-Domínguez, M.; Lager, E.; Zamudio-Vázquez, R.; Godoy-Vargas, J.; Villanueva-García, F. *Org. Lett.* **2007**, *9*, 3985. (b) Lager, E.; Liu, J.; Aguilar-Aguilar, A.; Tang, B. Z.; Peña-Cabrera, E. J. Org. Chem. **2009**, *74*, 2053. (c) Arroyo, I. J.; Hu, R.; Tang, B. Z.; López, F. I.; Peña-Cabrera, E. Tetrahedron **2011**, *67*, 7244.

^{(6) (}a) Gómez-Durán, C. F. A.; García-Moreno, I.; Costela, A.; Martín, V.; Sastre, R.; Bañuelos, J.; López Arbeloa, F.; López Arbeloa, I.; Peña-Cabrera, E. *Chem. Commun.* **2010**, *46*, 5103. (b) Bañuelos, J.; Martín, V.; Gómez-Durán, C. F. A.; Arroyo Córdoba, I. J.; Peña-Cabrera, E.; García-Moreno, I.; Costela, A.; Pérez-Ojeda, M. E.; Arbeloa, T.; López Arbeloa, I. *Chem.—Eur. J.* **2011**, *17*, 7261. (c) Kim, D.; Yamamoto, K.; Ahn, K. H. *Tetrahedron* **2012**, *68* (26), 5279.

dipyrromethenes and their substitution via S_NAr yielding heteroatom (N, O, S) *meso*-substituted derivatives and by palladium-catalyzed Suzuki, Stille, and Sonogashira cross-coupling.

Our proposed synthesis, outlined in Scheme 1, introduces the halogen through deoxygenative substitution on a dipyrrylketone. An early mention of this reaction dates back to the work of Fisher and Ort, where a chlorinated dipyrrin was obtained by the action of phosgene.⁸ Because we wanted to avoid this highly toxic gas, other synthetic routes toward halogenated dipyrrins were investigated.

Following literature procedures,⁹ the oxidative conversion of dipyrrylthioketone to symmetric ketone 1 using hydrogen peroxide is efficient and fast, and product 1 is isolated as a crystalline solid. Initial attempts to bring about halogenation of ketone 1 used phosphorus oxychloride and showed rapid conversion to the dipyrrinium salt. In situ deprotonation and complexation subsequently resulted in a single fluorescent compound, which was identified as the target meso-chlorinated compound 2a. Similarly, phosphorus oxybromide converted di(pyrrol-2yl)methanone 1 to a *meso*-brominated dipyrrinium salt, which was deprotonated and complexed in situ to yield 2b. Additional efforts to screen for other halogenating agents, such as SOCl₂, PCl₃, and PCl₅, all resulted in lowered yields and significant side product formation. The use of PI₃ as iodinating agent was unsuccessful. The resulting mixture was contaminated with several byproducts, and a good vield of the desired borondiaza-s-indacene could not be attained. All attempts at fluorination led to a complex reaction mixture.

However, halogen exchange could be achieved by stirring chlorinated 2a in acetone in the presence of sodium iodide. In this modified Finkelstein procedure, insoluble sodium chloride precipitates and drives the reaction to the *meso*-iodinated dye 2c, which is isolated in good yield.

In the direct conversion of the ketone to sulfonates, by reaction with trifluoromethyl or nonaflyl sulfonyl anhydride, the yellow color of a dipyrrin intermediate is observed, but complexation with boron trifluoride did not result in the formation of the boron heterocycle.

Despite early worries about the stability of such halogenated compounds, compounds **2** proved to be stable at room temperature as highly crystalline solids.

8-HaloBODIPYs are interesting compounds because they are very promising starting materials for the preparation of more complex *meso*-substituted BODIPY analogues via elaboration on the reactive halogen through, e.g., S_NAr or transition-metal-catalyzed transformations (Suzuki, Stille, Heck, Negishi, Sonogashira, etc.).

The chloride substituent is an efficient leaving group in S_NAr reactions. 8-ChloroBODIPY **2a** can be used to prepare the previously reported amine **3a**⁴ and thioether fluorophores **3b** and **3c**,⁴ which are isolated in high yield after stirring with the suitable nucleophile and a base. Scheme 1. Synthesis of 8-Halogenated BODIPY Dyes



Similarly, 8-substituted ethers can be obtained by nucleophilic displacement of the halogen substituent. As such, aryl (**3d**) and alkyl (**3e**) ethers were obtained by reaction of **2a** and phenol or methanol under basic conditions (Scheme 2).

Scheme 2. Nucleophilic Displacement of Chlorine on Dye 2a



Transition-metal-catalyzed cross-coupling of 8-haloBO-DIPYs with arylboronic acids is an alternate strategy to the classic condensation—oxidation sequence of pyrroles and aromatic aldehydes.² Thus, standard Suzuki (Table 1, entries 1–4) and Stille (Table 1, entry 5 and 6) crosscoupling procedures efficiently led to the substituted dyes **4a** and **4b** (Table 1). Heteroaromatic groups can also be introduced, as exemplified by the formation of 8-(2thienyl)BODIPY **4c** (Table 1, entry 6), which is a potential building block for the preparation of novel luminescent materials.^{5a} Beneficially, our method eliminates the need for Cu⁺ reagents, which are required in the Liebeskind— Srogl cross-coupling of 8-methylthioBODIPYs with boronic acids.⁵ All halogenated (Cl, Br, I) compounds **2** undergo the Suzuki and Stille reactions with varying yields.

Sonogashira cross-coupling with phenylacetylene with the iodinated dye 2c led to complex reaction mixtures in which the desired product could only be observed in trace amounts. The side reactions and decomposition could be circumvented by shifting to the chlorinated dye 2a, which reacted very rapidly at low temperatures (30 min at 0 °C) to provide alkyne 4d in good yield (Table 1, entry 7). Such alkynes with bathochromically shifted spectra are interesting for the development of new sensors and fluorescent materials, but their synthesis has previously only been reported from unstable propynoyl chloride.¹⁰

Expansion of the methodology to substituted pyrroles and the corresponding dipyrrylketones, such as **5**, led to the synthesis of both symmetrically and asymmetrically

⁽⁷⁾ De Meijere, A., Diederich, F. *Metal-Catalyzed Cross-Coupling Reactions*, 2nd ed.; Wiley-VCH: Weinheim, 2004.

⁽⁸⁾ Fisher, H.; Ort, H. Liebigs Ann. Chem. 1933, 502, 237.

⁽⁹⁾ Plater, M. J.; Aiken, S.; Bourhill, G. Tetrahedron 2002, 58, 2405.

⁽¹⁰⁾ Bonardi, L.; Ulrich, G.; Ziessel, R. Org. Lett. 2008, 10, 2183.

 Table 1. Palladium-Catalyzed Cross-Coupling Reactions
 (Suzuki, Stille, Sonagashira) of 8-HaloBODIPY



entry	Х	reaction type	R	yield	product 4
1	Cl	Suzuki	Ph	36	4a
2	\mathbf{Br}	Suzuki	Ph	53	4a
3	Ι	Suzuki	Ph	75	4a
4	Ι	Suzuki	4-MeOPh	58	4b
5	\mathbf{Br}	Stille	Ph	73	4a
6	\mathbf{Br}	Stille	2-thienyl	61	4c
7	Cl	Sonogashira	C≡CPh	76	4d

core-substituted 8-chloroBODIPY dyes (Scheme 3). In proofof-concept reactions, compounds such as **6a,b** displayed the same reactivity as the unsubstituted systems **2** in both S_NAr and Pd-catalyzed cross-coupling. For instance, Suzuki cross-coupling of **6a** with phenylboronic acid yields dye **7**, while the nucleophilic displacement of chlorine by phenolate yields ether **8**.

The substituted dipyrryl ketones **5** were not readily available, as the oxidation of thioketones failed and no general synthetic route has yet been described. Eventually, dipyrryl ketones **5** were prepared in moderate yield through pyrrolic Vilsmeier reagents (Supporting Information),¹¹ and the full scope of this reaction is currently under investigation. It is also noteworthy that for asymmetric dipyrryl ketone **5a** an equilibrium between two tautomeric forms could be observed in the ¹H NMR spectrum (Supporting Information).

Scheme 3. Synthesis of the Substituted 8-ChloroBODIPYs 6 and Their Subsequent Functionalization



The rich variety of the groups introducible at the 8-position of BODIPY derivatives with this method leads to a set of dyes with UV-vis absorption and fluorescence emission spectra covering a broad range of the visible spectrum (Figure 1, extended figure in the Supporting Information). Table 2 summarizes some key spectroscopic and photophysical data of selected derivatives in tetrahydrofuran (THF) solution. Full details will be described elsewhere.



Figure 1. Normalized, visible absorption spectra and corresponding fluorescence emission spectra of a selection of 8-substituted BODIPY dyes in THF.

The spectra display the typical narrow absorption and fluorescence emission bands of classic difluoroboron dipyrrins. The broadest spectral bands are found for the 8-N-aniline derivative 3a and to a lesser degree for the meso-O-substituted dves 3d and 3e. The Stokes shifts are generally quite small with exceptions for 3d and 3e. The dyes with 8-N (3a) or 8-O (3d, 3e) substituents have blueshifted absorption and fluorescence emission spectra with respect to unsubstituted BODIPY.¹² This hypsochromic shift is related to the electron-donating character of the heteroatom and is markedly larger for N than for O. In contrast, 8-halogens (Cl in 2a, Br in 2b, I in 2c) have a negligible effect on the spectral maxima. Conversely, the 8-ethynylphenyl group in 4d leads to red-shifted absorption and fluorescence emission spectra compared to those of unsubstituted and classic boron dipyrrin dyes. This indicates that this 8-substituent extends the π -conjugation. The meso-O derivatives 3d and 3e have very high fluorescence quantum yields Φ , whereas the 8-aniline (3a) and 8-thiophenol (3b) analogues are nearly nonfluorescent. The heavy atom effect on Φ is clearly seen in the series of 8-halo dyes 2a−c.

To conclude, an efficient synthesis of 8-halogenated boradiaza-s-indacenes **2** has been described starting from dipyrrylketones. N-, O-, and S-centered nucleophiles reacted smoothly with **2** via S_NAr , yielding a range of new fluorophores with hypsochromically shifted spectra compared to classic boron dipyrrin dyes. Compounds **2** are also useful scaffolds for the preparation of 8-aryl-, 8-heteroaryl-, and 8-alkynyl-substituted BODIPYs via palladium-catalyzed Suzuki, Stille, and Sonogashira cross-coupling reactions. Current synthetic efforts are focused on the

⁽¹¹⁾ White, J.; McGillivray, G. J. Org. Chem. 1977, 42, 4248.

^{(12) (}a) Arroyo, I. J.; Hu, R.; Merino, G.; Tang, B. Z.; Peña-Cabrera, E. J. Org. Chem. **2009**, 74, 5719. (b) Tram, K.; Yan, H.; Jenkins, H. A.; Vassiliev, S.; Bruce, D. Dyes Pigm. **2009**, 82, 392.

Table 2. Spectroscopic and	l Photophysical Data	of BODIPY Dyes in THF
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product	$\lambda_{abs}(max)^a (nm)$	$\lambda_{\rm em}({\rm max})^b ({\rm nm})$	$\Delta \overline{\nu}^c (\mathrm{cm}^{-1})$	${\rm fwhm_{abs}}^d({\rm cm}^{-1})$	$\mathrm{fwhm_{em}}^{e}(\mathrm{cm}^{-1})$	Φ^{f}
2a	503	515	463	921	1388	0.72 ± 0.01
2b	504	517	499	880	1424	0.496 ± 0.008
2c	507	519	456	860	1530	0.159 ± 0.006
3a	413	420	404	2233	2059	0.002 ± 0.001
3d	456	495	1728	1685	1504	0.96 ± 0.04
3e	443	486	1997	1670	1575	0.90 ± 0.05
4d	543	561	591	961	1426	0.66 ± 0.03
6a	495	510	594	1155	1416	0.737 ± 0.005
6b	502	513	427	774	1386	0.416 ± 0.008

^{*a*} Absorption maximum. ^{*b*} Fluorescence emission maximum. ^{*c*} Stokes shift [= $1/\lambda_{abs}(max) - 1/\lambda_{em}(max)$]. ^{*d*} Full width at half height of the maximum of the absorption band. ^{*e*} Full width at half height of the maximum of the fluorescence emission band. ^{*f*} Fluorescence quantum yield \pm one standard uncertainty.

optimized synthesis of symmetric and asymmetric 8-haloBODIPYs.

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The authors declare no competing financial interest.