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## Estrogen receptor β ligands: Design and synthesis of new 2-phenyl-isoindole-1,3-diones

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Abstract—The design, synthesis, and biological evaluation of the 2-phenyl-isoindole-1,3-diones will be discussed. Detailed modeling studies with X-ray support were used to understand the ligand binding orientation and observed selectivity. © 2006 Elsevier Ltd. All rights reserved.

The estrogen receptor (ER) is a ligand-activated transcription factor that belongs to the nuclear hormone receptor superfamily. Other members of this family include the progestin, androgen, glucocorticoid, and mineralocorticoid receptors.<sup>1</sup> Until recently estrogenmediated events had been thought to be regulated by only one estrogen receptor which now is identified as  $ER\alpha$ <sup>2,3</sup> With the discovery in 1996 of a second estrogen receptor subtype (ER $\beta$ ), there was an intense interest in elucidating ER $\beta$  function.<sup>4,5</sup> The three dimensional structures of  $ER\alpha^6$  and  $ER\beta^7$  have been solved by co-crystallization with a variety of ligands. The primary ligand binding domain of these receptors are 58% homologous. However, the well-recognized ligand-binding domain (LBD) cavity varies by only two amino acids<sup>7</sup> (i.e., ER $\beta$  Met<sub>336</sub> is replaced by ER $\alpha$  Leu<sub>384</sub> and ER $\beta$  Ile<sub>373</sub> is replaced by ER $\alpha$  Met<sub>421</sub>). As a consequence of these conservative substitutions, it is not surprising that the many ER ligands known prior to the discovery of the ER<sup>β</sup> subtype possessed limited selectivity for that subtype. Despite the fact that the design of highly selective subtype ligands appeared to be a daunting task, there have been many reports of high affinity and functionally selective  $ER\alpha$  and  $ER\beta$  ligands.<sup>8-12</sup> Recently, we have disclosed some of our efforts toward identifying possible utilities for  $\text{ER}\beta$  selective ligands.<sup>13</sup>

The phytoestrogen, genistein, has been reported to have modest ER $\beta$  selectivity (~40-fold)<sup>14,15</sup> and has been used by our group as a starting point to design novel nonsteroidal ER $\beta$  selective ligands (Fig. 1). Some detailed work focused on substitution around simplified tricyclic A-BC scaffolds (i.e., the 6-phenylnaphthalenes<sup>15</sup> and the 2-phenyl quinolines<sup>16</sup>). Further variation on the central ring (ring-B) came about by modification of the benzofuran 1 (obtained from elaboration of an HTS screening hit),<sup>17</sup> with an indenone core (i.e., 2).<sup>18</sup> The authors hypothesized that these analogs would mimic the binding activity of the phytoestrogen genistein as well as allowing a handle to incorporate the C-5 hydroxy group of genistein into the indenone core. Maintaining the positions of the A and C ring hydroxyls led to two regioisomeric indenone forms, the 6-hydroxy series 2a and the 5-hydroxy series 2b (not unlike that of genistein). These compounds, although potent binders to ER $\beta$ , have moderate <15fold ERβ-selectivity. The core template, arylindene-1one 2 of this series, having the ability to flip 180° (around the x- and y-axes), allows for multiple binding modes. To simplify this binding paradigm, we explored substitution on a central phthalimide core which has been utilized in the design of a variety of biologically active molecules.<sup>19</sup> The 2-phenyl-isoindole-1,3-dione core (i.e., 3), bearing two carbonyls, has additional symmetry along the x-axis.

Keywords: ER $\alpha$  and ER $\beta$  receptor; Estrogen receptor; Nuclear hormone receptor.

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Figure 1. Compounds of interest.

If compound **3** rotates along the *x*-axis the 'B-ring' of this A-BC ring system will not be altered and has the advantage of allowing the template to serve as a vehicle to rapidly evaluate and design ER $\beta$  selectivity into the new structural motif.

The synthesis of the phthalimides (i.e., 3) was straightforward and is outlined in Scheme 1. It was accomplished by reacting a mixture of 4-hydroxyphthalic acid and *N*-methyl-2-pyrrolidine (NMP) in xylenes at reflux temperature with an appropriately substituted amine. The free hydroxyls were obtained by deprotection of the methyl ethers with pyridine-HCl (190 °C) to provide the desired hydroxy phenyl-phthalimides (i.e., 3). Alternatively, a copper-mediated coupling of an appropriately substituted N-H phthalimide and a suitably substituted iodoanisole may construct the core phthalimide.

A competitive radioligand binding assay was used to assess the relative binding affinity (IC<sub>50</sub>) of compounds for the human ligand binding domains (LBD) of ER $\beta$  and ER $\alpha$ .<sup>20</sup> Table 1 shows the binding affinities for analogs with variations of the hydroxyl group on the *N*-phenyl ring. The C-4' placement of the hydroxyl is optimal for binding to the ER.

Table 2 shows structural variations on the appended *N*-phenyl ring where we replaced the aryl ring with a

Table 1. Binding affinity for 2-phenyl-isoindole-1,3-dione core

## R 4' A 3' 2' O 7 OH

Compound	R	A ring	$\frac{\text{ER}\beta \text{ IC}_{50}{}^{a}}{(nM)}$	ERa IC <sub>50</sub> <sup>a</sup> (nM)
E2	_		$3.6 \pm 1.6$	$3.2 \pm 1.0$
Genistein			$9.7 \pm 4.3$	$395 \pm 181$
3	OH	4'-OH	84 ± 31 (7)	1980 ± 719 (7)
4	OH	3'-OH	2900 ± 346 (3)	>5000 (3)
5	Н	4'-OH	>10000	>10000
6	OH	Н	2700	>5000
7	OH	4'-OMe	3300	>5000
8	OH	2′,4′-OH	2470	>5000
9	OH	3′,5′ <b>-</b> OH	>5000	>5000

<sup>a</sup>Values are means of independent determinations (number in parentheses)  $\pm$  SD.

saturated six-membered ring (i.e., **10**) and observed a significant decrease in affinity. In addition, a fused aryl system (i.e., **11** and **12**) also decreased affinity. Introduction of a C-2' electron-withdrawing substituent enhanced binding affinity (i.e., **14** and **16**) which has been previously observed in the literature for ER $\beta$  selective ligands.<sup>8,9,12,18</sup> Although an increase in affinity was



Scheme 1. Reagents and conditions: (a) NMP, xylene, 150 °C; (b) Py-HCl, 190 °C; (c) Cu<sub>2</sub>O, collidine.

Table 2. Structural variations on 2-phenyl-isoindole-1,3-dione



Compound	R	$\mathbf{R}^1$	$ER\beta IC_{50}{}^{a} (nM)$	ERa $IC_{50}^{a}$ (nM)	ERα/ERβ ratio
3	OH	4'-OH-phenyl	84 ± 31 (7)	1980 ± 719 (7)	24
10	OH	4'-OH-C <sub>6</sub> H <sub>11</sub> (trans)	>5000	>5000	n/a
11	OH	5'-OH-1-naphthyl	802	2810	4
12	OH	4'-OH-1-naphthyl	2640	560	0.2
13	OH	2'-NO <sub>2</sub> -4'-OMe-phenyl	>5000	>5000	n/a
14	OH	2'-NO <sub>2</sub> -4'-OH-phenyl	21	92	4
15	OH	2'-Me 4'-OH-phenyl	802	2810	4
16	OH	2'-F 4'-OH-phenyl	53 ± 13 (2)	339 ± 92 (2)	6
17	OMe, 7-Br	4'-OH-phenyl	>5000	>5000	n/a
18	OH, 7-Br	4'-OH-phenyl	36 ± 2.1 (2)	1625 ± 7.1 (2)	45

<sup>a</sup> Values are means of independent determinations (number in parentheses)  $\pm$  SD.

observed with substitution at the C-2' position, the compounds did not exhibit increased ER $\beta$ -selectivity when compared to the core template, 3.

Docking calculations<sup>21</sup> were used to predict the orientation of our novel core phthalimide **3** in the ER $\beta$  binding site.<sup>7</sup> Figure 2 shows that the 4'-OH group makes a critical interaction with Arg<sub>346</sub> and Glu<sub>305</sub>, while the 5-OH group makes an interaction with the His<sub>475</sub> residue. This binding mode is consistent with that of other A-B/C ring systems, such as benzofuran and indenone for which ER $\beta$  X-ray structures are known.<sup>17,18</sup>

The B ring of this template lies in close proximity to the ER $\alpha$  Leu<sub>384</sub>/ER $\beta$  Met<sub>336</sub> residue mutation and it appears that the core ER $\beta$  selectivity (~24-fold) of this scaffold is due to a favorable interaction of this aromatic ring with ER $\beta$  Met<sub>336</sub>, similar to what has been described for genistein<sup>22</sup> and other compounds.<sup>23</sup>



**Figure 2.** Docked structure of compound **3** in the ER $\beta$ /genistein binding site. Only key residues are shown for simplicity. Residue differences within the binding pocket between ER $\alpha$  and ER $\beta$  are highlighted. Hydrogen bonds are shown by a yellow dotted line. Arrow indicates opportunity for increasing selectivity by targeting region of residue difference.

Looking closely at the simplified core, there appeared to be an opportunity to increase selectivity by substituting at the 7-position of the phthalimide core. Previous studies in our laboratories<sup>23,24</sup> suggested that substitution at the 7-position may satisfy the binding requirements as well as probe the Met/Ile selectivity pocket. Figure 3 shows the 7-bromo phthalimide  $18^{25}$  docked into the ERβ/genistein pocket.

As discussed earlier, only two residues are different within the ligand binding pockets of ER $\alpha$  and ER $\beta$ .



Figure 3. Compound 18 docked into the ER $\beta$ /genistein pocket. Only key residues are shown for simplicity. Residue differences within the binding pocket between ER $\alpha$  and ER $\beta$  are highlighted. Hydrogen bonds are shown by a yellow dotted line.



**Figure 4.** Compound **18** (colored by atom type) docked to the ER $\beta$ / genistein pocket (genistein ligand colored by magenta) and overlaid with ER $\alpha$ /DES<sup>26</sup> key residues. Only key residues and a Connolly surface of the ER $\beta$  binding site are shown for simplicity. Hydrogen bonds are shown by a yellow dotted line. Distance monitor shows that the 7-Br would be in close proximity to ER $\alpha$  Met<sub>421</sub> residue.

Docking calculations show that the binding mode of compound 18 remains the same and the 7-bromo group is in close proximity to the residue change (i.e., ER $\alpha$  Met<sub>421</sub> and ER $\beta$  Ile<sub>373</sub>). To further understand the origin of the ER $\beta$  selectivity (~45-fold) of this scaffold, the X-ray structure of ERa/DES<sup>26</sup> was superimposed on the docked structure (see Fig. 4). Only key residues (i.e., Leu<sub>384</sub> Met<sub>421</sub>), which are different within the ligand binding pocket, are shown from the ERa/DES structure in magenta. Of particular interest is the 7-bromo group that is in close proximity (2.1 Å) to the sulfur of the ER $\alpha$  Met<sub>421</sub> residue which would represent a repulsive interaction between  $ER\alpha$  Met<sub>421</sub> and the bromo group. A similar interaction has been suggested between the 5-OH group and the sulfur atom to explain the observed ER $\beta$  selectivity of genistein.<sup>22</sup> In contrast, a distance of 3.7 Å between the 7-bromo group and ER $\beta$  Ile<sub>373</sub> (C<sup> $\gamma$ 1</sup>) is not expected to have significant electronic or steric repulsion. The slight increase in  $ER\beta$  potency for this compound ( $\sim$ 2.3-fold) could be attributed to the increased nature of the lipophilic group (i.e., bromine occupying a hydrophobic pocket). This differential interaction may explain the overall increase in  $ER\beta$ selectivity (~45-fold) observed for the bromo-substituted phthalimide 18.

In summary, a novel series of 4-hydroxy-*N*-phenylsubstituted phthalimides was prepared. This dihydroxy-substituted core offers a symmetric B-ring that mimics genistein and binds to ER $\beta$  with modest ER $\beta$ -selectivity (~24-fold) compared to genistein. The binding mode for this series of compounds was consistent with that of other previously reported A-B/C ring systems, such as the benzofuran and indenone series. Based on our previous SAR in this area, the C-7 bromo analog **18** was prepared and found to be slightly more ER $\beta$ selective than genistein.

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