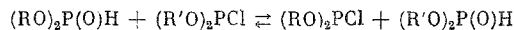


REDISTRIBUTION REACTION IN THE SYSTEM: DIALKYL  
PHOSPHITE - DIALKYL CHLOROPHOSPHITE

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We found that the following redistribution reaction exists in the system: dialkyl phosphite - dialkyl chlorophosphite



Thus, when a mixture of 7 g of diethyl phosphite and 11.2 g of di-(2-chloroethyl)chlorophosphite was heated at 100-120°C for 1 h we isolated 2.5 g (32%) of diethyl chlorophosphite with bp 34-35° (10 mm);  $n_D^{20}$  1.4370;  $\delta_{P31}$  -165 ppm (see [1, 2]), and 4 g (38%) of di-(2-chloroethyl)phosphite with bp 90-92° (0.08 mm);  $d_4^{20}$  1.4015;  $n_D^{20}$  1.4708;  $\delta_{P31}$  -8 ppm;  $J_{P-H}$  714 Hz. Found: C 23.28; H 4.43; P 14.60%.  $C_4H_9Cl_2O_3P$ . Calculated: C 23.19; H 4.34; P 14.97% (see [3]).

As a result, a reaction was found where compounds with a tetracoordinated phosphorus atom (dialkyl phosphites) are converted to derivatives with a tricoordinated phosphorus atom (dialkyl chlorophosphites).

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