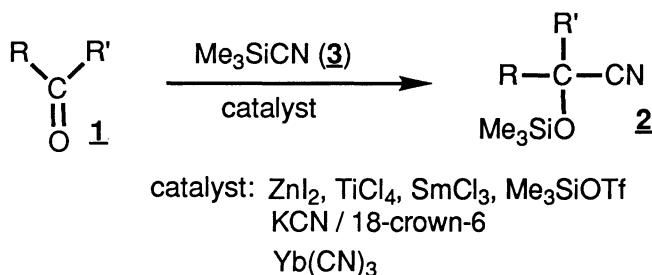


Ytterbium Tricyanide, a Highly Efficient Catalyst for  
the Addition of Cyanotrimethylsilane to Carbonyl Compounds

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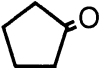
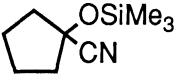

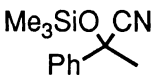
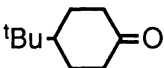
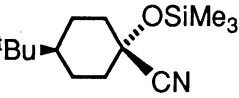
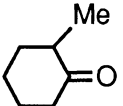
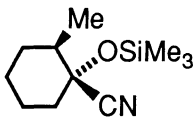
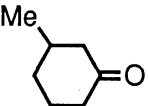
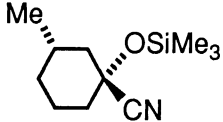
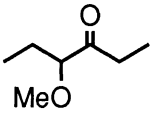
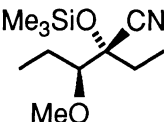
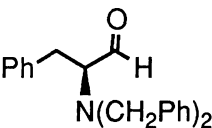
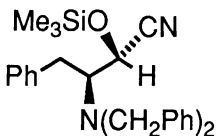
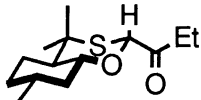
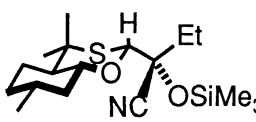
Reaction of cyanotrimethylsilane with various carbonyl compounds was effectively catalyzed by  $\text{Yb}(\text{CN})_3$  to give the adducts in excellent yields; reaction with substituted cyclohexanones proceeded in a highly stereoselective manner.

$\alpha$ -Trimethylsiloxy nitriles **2**, trimethylsilyl ether of cyanohydrins, are useful intermediates in organic synthesis and have been prepared by the reaction of a carbonyl compounds **1** with cyanotrimethylsilane (**3**).<sup>1-8)</sup> Various types of catalysts have been employed: Lewis acids, such as  $\text{ZnI}_2$  and  $\text{TiCl}_4$ , are generally used,<sup>1,2,4)</sup> and lanthanoid salts have been applied.<sup>5)</sup> Trimethylsilyl triflate also showed a high catalytic activity.<sup>6)</sup> On the other hand, KCN-crown ether system generates highly reactive cyanating species and can be employed to the reaction with easily enolizable ketones.<sup>7-9)</sup> We have reported that lanthanoid cyanides are highly effective catalysts for the reaction of **3** with oxiranes<sup>10)</sup> or aziridines.<sup>11)</sup> In these reactions,  $\text{Yb}(\text{CN})_3$  showed little Lewis acidity, although it worked as powerful cyanating reagent.<sup>10)</sup> These observations prompted us to anticipate that  $\text{Yb}(\text{CN})_3$  might possess high catalytic activity for the above described cyanohydrin formation. This paper describes the successful application of  $\text{Yb}(\text{CN})_3$  as a catalyst for the reaction of **3** with various types of carbonyl compounds **1**.



Reaction of **3** with ketones proceeded smoothly under  $\text{Yb}(\text{CN})_3$  catalysis. To a mixture of  $\text{Yb}(\text{CN})_3$  (50 mg, 0.2 mmol) and **3** (2.4 mmol) in THF (5 mL), acetophenone (**1b**, 2.0 mmol) was added at room temperature and the reaction mixture was stirred for 1 h. Hydrolytic workup gives the adduct **2b** in 99% yield.<sup>12)</sup> Various ketones reacted similarly with **3** to give the corresponding **2** in excellent yields and in high diastereoselectivity. The combinations of ketone **1**, reaction temperature, reaction time, yield of the product **2**, as well as the

Table 1. Yb(CN)<sub>3</sub> Catalyzed Reaction of Ketones with Me<sub>3</sub>SiCN<sup>a)</sup>

Ketone (1)	Conditions		Product (2)	Yield <sup>b)</sup> % (%de)	ZnI <sub>2</sub> <sup>c)</sup>	Me <sub>3</sub> SiOTf <sup>d)</sup>	KCN/ crown <sup>e)</sup>
	Time / h	Temp / °C					
a 	1.5	0		93	94	>99	99
b 	12	25		>99	91	88	>99
c 	2	0		88 (88) [>99 (94)] <sup>f)</sup>	97 (80)	99 (82)	99 (56)
d 	3.5	0		87 (86) [92 (90)] <sup>f)</sup>	94 (32)	>99 (31)	90 (46)
e 	2	0		94 (54) [90 (94)] <sup>f)</sup>	94 (2)	>99 (4)	90 (16)
f 	5	0		81 (56)	82 (64)	85 (58)	80 (10) <sup>g)</sup>
g 	1	0		91 (44)	89 (82) <sup>h)</sup>	99 (82)	92 (72)
h 	1.5	0		>99 (70) [99 (74)] <sup>i)</sup>	99 (44)	74 (44)	98 (26) <sup>g)</sup>

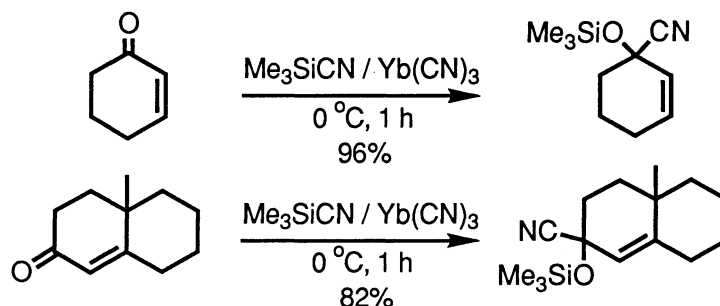
a) To a slurry of Yb(CN)<sub>3</sub> (0.1 mmol) in THF (4 mL), **3** (1.2 mmol) and a ketone (1.0 mmol) was added. The mixture was stirred at the indicated temperature. b) Isolated yields. c) ZnI<sub>2</sub>, Ref.1 and 2. d) Me<sub>3</sub>SiOTf, Ref.6.

e) KCN/18-crown-6, Ref.8. f) Reaction at -89 °C. g) Major product is the diastereomer of the isomer shown in Table.

h) See Ref.4. i) Reaction at -23 °C.

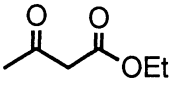
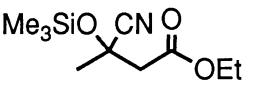
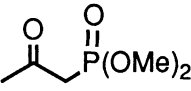
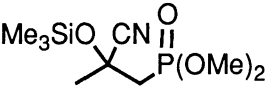
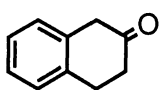
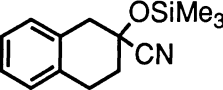
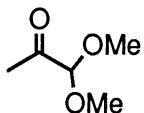
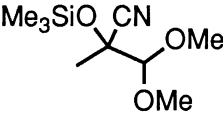
yield of the reported methods<sup>13)</sup> are summarized in Table 1. Reaction of **3** with 4-*t*-butylcyclohexanone (**1c**) completes at  $-89^{\circ}\text{C}$  within 6 min in highly diastereoselective manner to give **2c** in >99% (94% d.e.).

Reaction of  $\alpha,\beta$ -unsaturated ketones with **3** afforded 1,2-adducts selectively at  $0^{\circ}\text{C}$ ; the reaction in refluxing THF gave the same 1,2-adducts.



Reaction with easily enolizable ketones (**1i** - **1k**) or a ketone (**1l**) containing acid-sensitive group afforded the corresponding adducts in good yields (Table 2). As can be seen from Table 2,  $\text{Yb}(\text{CN})_3$  catalyst gave the corresponding adducts in good yields as analogous to the KCN-crown ether catalyst system, whereas  $\text{ZnI}_2$  could not be employed in these cases.

Table 2.  $\text{Yb}(\text{CN})_3$  Catalyzed Reaction of Enolizable Ketones with  $\text{Me}_3\text{SiCN}$ <sup>a)</sup>

Ketone ( <b>1</b> )	Condition		Product ( <b>2</b> )	Yield <sup>b)</sup> %	$\text{ZnI}_2$ <sup>c)</sup>	$\text{Me}_3\text{SiOTf}$ <sup>d)</sup>	KCN/ crown <sup>e)</sup>
	Time / h	Temp / $^{\circ}\text{C}$					
i 	4	25		90	45 <sup>f)</sup>	42 <sup>f)</sup>	73
j 	5.5	25		84	0	0	96
k 	2	0		88	68	94	84
l 	1	0		91	0	30	83

a) To a slurry of  $\text{Yb}(\text{CN})_3$  (0.1 mmol) in THF (4 mL), **3** (1.2 mmol) and a ketone (1.0 mmol) was added. The mixture was stirred at the indicated temperature. b) Isolated yields. c)  $\text{ZnI}_2$ , Ref.1 and 2. d)  $\text{Me}_3\text{SiOTf}$ , Ref.6.

e) KCN/18-crown-6, Ref.8. f) Silyl enol ether was isolated in 50% yield ( $\text{ZnI}_2$ ) and 48% yield ( $\text{Me}_3\text{SiOTf}$ ).

The results shown in Table 1 and Table 2 indicates that  $\text{Yb}(\text{CN})_3$  can be widely employed as the catalyst for the addition of **3** with various carbonyl compounds.

Some remarks on  $\text{Yb}(\text{CN})_3$  are added lastly. Ytterbium tricyanide could be obtained by the following two methods: (1)  $\text{Yb}(\text{CN})_3$  was prepared by the reaction of **3** with  $n\text{-Bu}_3\text{Yb}$  which was produced from  $\text{YbCl}_3$  and 3 equiv. of  $n\text{-BuLi}$ . IR showed two prominent peaks at 2210 (medium) and 2110 (strong).  $\text{Yb}(\text{CN})_3$  thus obtained showed a high catalytic activity in the reaction of carbonyl compounds **1** with **3**. (2)  $\text{Yb}(\text{CN})_3$  was also obtained by the reaction of  $\text{Yb}(\text{O}^i\text{Pr})_3$  with 3 equiv. of **3**. IR showed a strong peak at 2085.  $\text{Yb}(\text{CN})_3$  thus obtained showed a high activity in the reaction of **3** with **1** but much less active in the reaction of **3** with aziridines; the reactivity of  $\text{Yb}(\text{CN})_3$  was enhanced by the addition of  $\text{LiCl}$  or  $\text{LiCN}$ .<sup>14,15)</sup> Structure of the individual sample as well as the relationships between the absorptions and reactivities are still open to be answered.<sup>16)</sup>

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- 11) S. Matsubara, T. Kodama, and K. Utimoto, *Tetrahedron Lett.*, **31**, 6379 (1990).
- 12) Reaction of  $\text{PhCOMe}$  (**1b**) with **3** under  $\text{SmCl}_3$  catalysis afforded the corresponding adduct **2b** in 50% yield, under  $\text{Eu}(\text{fod})_3$  catalysis gave a mixture of **2b** (25%) and the corresponding protodesilylated nitrile (29%); reaction of  $\text{PhCHO}$  with **3** afforded the adduct in 98% yield.<sup>5)</sup>
- 13) Reactions were carried out under the conditions described in references:  $\text{ZnI}_2$ -catalyzed reaction;<sup>1)</sup>  $\text{Me}_3\text{SiOTf}$  catalyzed reaction;<sup>6)</sup> reactions catalyzed by  $\text{KCN}/18\text{-crown-6}$ .<sup>7,8)</sup>
- 14) S. Matsubara, T. Kodama, and K. Utimoto, in preparation.
- 15)  $\text{Yb}(\text{CN})_3$  was prepared by the reaction of  $\text{YbBr}_3$  with  $\text{LiCN}$ ; K. Rossmann, *Monatsh. Chem.*, **97**, 1698 (1966); IR absorptions, 2185 (weak) and 2100 (medium).
- 16) The authors thank to the Japan Society for the Promotion of Science for a Grant-in-Aid for International Exchange Program.

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