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Enantioselective nucleophilic addition to *N*-(2-pyridylsulfonyl)imines by use of dynamically induced chirality

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Abstract—Enantioselective nucleophilic addition of Grignard reagents to *N*-(2-pyridylsulfonyl)imines in the presence of bis(oxazoline) afforded products with good enantioselectivity. Dynamically induced chirality on the sulfur by coordination of a chiral Lewis acid to a pyridyl nitrogen and one of the sulfonyl oxygens fixes the conformation of the complex and induces enantioselectivity. Since the 2-pyridylsulfonyl group can be easily removed after the addition reaction, it acts not only as a protecting group but also as an efficient stereocontroller.

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1. Introduction

The stereoselective nucleophilic addition reaction of organometallic reagents to C=N bonds is one of the most efficient methods for the preparation of chiral amines. For the preparation of chiral amines, diastereoselective reactions of imines having various chiral auxiliaries with organometallic reagents have been extensively studied.¹ Enantioselective addition of organometallic reagents to imines has also been developed² in the reaction of N-substituted imines,³ diphenylphosphinoylimines,⁴ N-acylimines,⁵ and sulfonyl imines⁶ since the first report of chiral ligand-mediated addition of organolithium reagents to a C=N bond.⁷ In these reactions, various Lewis acids and chiral ligands have been examined to coordinate to the imino nitrogen in a monodentate manner for the stereoselective intermolecular alkylation of imines. Exceptionally, there are only a few reports for asymmetric nucleophilic addition reaction to imines through bidentate fixation with a chiral Lewis acid;8 however, it is often found difficult in these reactions to remove the protecting group from

the products. Enantioselective reactions of imines described above have been performed by alkylation with organolithium or organozinc reagents, but there are no reports of the enantioselective reaction using Grignard reagents. It is important to develop an easily removable protecting group, which also has sufficient activating property for imines to react with Grignard reagents. Thus, we designed the 2-pyridylsulfonyl group as a new coordinative protecting and activating group as well as a stereocontroller:⁹ The 2-pyridylsulfonyl group may have the following advantages: (1) the electronwithdrawing property increases the reactivity of imines for the nucleophilic addition, (2) coordination of a Lewis acid to the pyridyl nitrogen and one of the sulfonyl oxygens controls the stereoselectivity and stereochemistry, and (3) the 2-pyridylsulfonyl group is easily removable from the products (Fig. 1). We now report a new enantioselective nucleophilic addition to 2-pyridylsulfonylimines with Grignard reagents and conversion to chiral amines.^{10,11}



Figure 1. 2-Pyridylsulfonyl group as a new coordinative protecting and activating group as well as a stereocontroller.

Keywords: 2-Pyridylsulfonyl group; Grignard reagents; Stereocontroller; Imine.

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2. Results and discussion

We examined the reaction of *N*-benzylidene-2-pyridinesulfonamide **1a** with CH_3MgI (2 equiv) using various chiral ligands (1.5 equiv) such as bis(oxazoline)s **3–6** and (–)-sparteine **7** (Scheme 1). As a result, bis(oxazoline)-Ph **3** was found to be an efficient ligand for this reaction (entries 1–5).¹² Performing the reaction at -95 °C increased enantioselectivity (entry 6). Reaction of **1a** with CH_3MgBr gave better results in comparison with CH_3MgI and CH_3MgCl to give the *S*-isomer with 83% ee in 79% yield (entry 7).¹³ The reactions of various



Table 1. Enantioselective addition of organometallic reagents to aldimines 1a-k

Scheme 1.

2-pyridylsulfonylimines **1b–f** ($\mathbf{R}^2 = p$ -tolyl, 4-chlorophenyl, 1-naphthyl, cinnamyl, 2-furyl) with CH₃MgBr were examined and it was found that the corresponding sulfonamides **2b**-f were formed in good yields with good enantioselectivities (entries 9-13). To our knowledge, this reaction is the first highly enantioselective addition of Grignard reagents to imines.¹⁴ Efficiency of 2-pyridylsulfonyl group in the reaction with Grignard reagents is shown by the following results. N-(2-Pyridylmethyl)aldimine 1g did not show enough reactivity toward CH₃MgBr because of the lack of a sulfonyl group (entry 14). In addition, N-benzylidene-p-methoxyaniline 1h, which is often used in the enantioselective addition of organolithium reagents was not a good substrate for the addition with CH₃MgI (entry 15).¹⁵ Furthermore, N-benzylidene-p-toluenesulfonamide 1i and N-benzylidene-2,4,6-triisopropylbenzenesulfonamide 1j were also not proper choice as substrates and their reaction with CH₃MgI at -78 °C in the presence of bis(oxazoline)-



Scheme 2.

Entry	Imine	\mathbb{R}^1	\mathbb{R}^2	Nucleophile	Ligand	Temperature (°C)	Product	Yield (%)	ee ^a (%)
1	1a	2-PySO ₂	Ph	CH ₃ MgI	3	-78	2a	75	59
2	1a	2-PySO ₂	Ph	CH ₃ MgI	4	-78	2a	31	7
3	1a	2-PySO ₂	Ph	CH ₃ MgI	5	-78	2a	56	40
4	1a	2-PySO ₂	Ph	CH ₃ MgI	6	-78	2a	77	0
5	1a	2-PySO ₂	Ph	CH ₃ MgI	7	-78	2a	90	0
6	1a	2-PySO ₂	Ph	CH ₃ MgI	3	-95	2a	27	72
7	1a	2-PySO ₂	Ph	CH ₃ MgBr	3	-95	2a	79	83 (>99) ^b
8	1a	2-PySO ₂	Ph	CH ₃ MgCl	3	-95	2a	53	86
9	1b	2-PySO ₂	<i>p</i> -Tol	CH ₃ MgBr	3	-95	2b	67	83
10	1c	2-PySO ₂	p-ClC ₆ H ₄	CH ₃ MgBr	3	-95	2c	77	76
11 ^c	1d	2-PySO ₂	1-Naphthyl	CH ₃ MgBr	3	-95	2d	74	76 (95) ^b
12	1e	2-PySO ₂	Cinnamyl	CH ₃ MgBr	3	-95	2e	98	82
13°	1f	2-PySO ₂	2-Furyl	CH ₃ MgBr	3	-95	2f	38	87
14	1g	2-PyCH ₂	Ph	CH ₃ MgI	3	-95	2g	0	
15	1h	p-MeOC ₆ H ₄	Ph	CH ₃ MgI	3	-95	2h	0	
16	1i	p-TolSO ₂	Ph	CH ₃ MgI	3	-78	2i	92	11
17	1j	$TipSO_2^d$	Ph	CH ₃ MgI	3	-78	2j	69	20
18	1k	8-QnSO ₂ ^e	Ph	CH ₃ MgI	3	-78	2k	52	0
19	1a	2-PySO ₂	Ph	EtMgBr	3	-95	21	74	50
20	1a	2-PySO ₂	Ph	BuMgBr	3	-95	2m	69	51
21	1a	2-PySO ₂	Ph	ⁱ PrMgBr	3	-95	2n	87	6
$22^{c,f}$	1a	2-PySO ₂	Ph	^t BuMgBr	3	-78	20	15	6
23	1b	$2-PySO_2$	Ph	PhMgI	3	-95	2p	41	66
24	1b	2-PySO ₂	<i>p</i> -Tol	PhMgBr	3	-95	2p	72	61
25 ^g	1a	$2-PySO_2$	<i>p</i> -Tol	Et_2Zn	3	-40	21	49	46
26	1a	2-PySO ₂	Ph	MeLi	3	-78	2a	35	35 ^g

^a ee was determined by the HPLC analysis.

^b ee in parentheses is that obtained after single recrystallization.

^d TipSO₂ = (2,4,6-triisopropylphenyl)sulfonyl.

^e 8-QnSO₂ = (8-quinolyl)sulfonyl.

 $^{\rm f}$ N-(2-Pyridyl sulfonyl)benzylamine was obtained in 35% yield.

^g The *R*-isomer was obtained.

^c Bis(oxazoline) 3 (2 equiv) was used.

Ph 3 afforded the products 2i and 2j, respectively, in poor yields with low enantioselectivities (entries 16 and 17). The reaction of N-benzylidene-8-quinolinesulfonamide 1k afforded the racemic product 2k (entry 18). The reaction of 1a with EtMgBr, BuMgBr, and PrMgBr gave products 21-n in good yields but with enantioselectivity lower than that with CH₃MgBr (entries 19–21 vs entry 7). The reaction using ^tBuMgBr resulted in a low yield of the product 20 besides the formation of an abnormal Grignard product¹⁶ (entry 22). The reaction of 1a with PhMgI and PhMgBr gave products 2p in good yields and enantioselectivity (entries 23 and 24). The nucleophilic addition of Et₂Zn to N-benzylidene-2-pyridinesulfonamide 1a did not proceed with high enantioselectivity (entry 25), whereas the nucleophilic addition of MeLi mainly gave the R-isomer (entry 26).

Since most of the products were crystalline, enantiomerically pure sulfonamides were easily obtainable by recrystallization. For example, recrystallization of **2a** (83% ee) from hexane/CH₂Cl₂ afforded enantiomerically pure (*S*)-**2a**. To realize the synthetic potential of this stereoselective preparation of chiral amines, we confirmed the easy removal of the 2-pyridylsulfonyl group. Although removal of arylsulfonyl groups generally needs drastic reaction conditions,¹⁷ the pyridylsulfonyl group could be removed from the optically active (*S*)-**2a** on treatment with magnesium in MeOH at 0 °C¹⁸ and chiral amine (*S*)-**8** was found to be formed without loss of optical purity (Scheme 2).¹⁹

The reaction of *p*-tolylsulfonylimine 1i and 2,4,6-triisopropylphenylsulfonylimine 1j with CH₃MgI afforded products 2i and 2j with low enantioselectivities, whereas the reaction of 2-pyridylsulfonylimine 1a showed high enantioselectivity (Table 1, entries 16 and 17 vs entry 1), showing that 2-pyridylsulfonyl group acts not only as an activating group but also as an efficient stereocontroller. Assuming that Mg(II) forms a tetrahedral bidentate-coordinating complex with the substrate **1a**,²⁰ there would be two types of coordination for the complex to be considered:²¹ One is the N,N-type complex, in which Mg(II) coordinates to the imino and pyridyl nitrogens and the other is the N,O-type complex, in which Mg(II) coordinates to a pyridyl nitrogen and one of the sulfonyl oxygens. Energies of the Mg(II) complexes of 2-pyridylsulfonylimine 1a and bis(oxazoline)-Ph 3, were calculated by the MOPAC 93/PM3 method,²² showing the stable complex in which Mg(II) coordinates to a pyridyl



Figure 2. Geometry optimization of 1a–3 complex by MOPAC 93/ PM3.

nitrogen and a *pro-S* sulfonyl oxygen (Fig. 2). CH₃MgX approaches the *Si*-face of this imine complex to form (*S*)-**2a**. Dynamically induced chirality on the sulfur indeed plays a definitive role in induction of enantioselectivity: It can be categorized as a new type of chiral relay.²³

3. Conclusion

The first highly enantioselective reaction of imines with Grignard reagents was achieved in the presence of bis(oxazoline)s by using a pyridylsulfonyl group as a protective group. The reaction was suggested by the MOPAC 93/PM3 calculation to proceed through a complex in which one of the sulfonyl oxygens is preferably coordinated to Mg(II). The 2-pyridylsulfonyl group was shown to be an easily removable, efficient protective group, which has notable properties of high chiral inducibility and activation of the imino group toward the addition of Grignard reagents.²⁴

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