#### Natural Products

## **Total Synthesis of Schilancitrilactones B and C\*\***

Liang Wang, Hengtao Wang, Yihang Li, and Pingping Tang\*

**Abstract:** The first total syntheses of schilancitrilactones B and C have been accomplished in 17 steps (longest linear sequence) from commercially available materials. Key steps include an intramolecular radical cyclization to provide the seven-membered ring, late-stage iodination, and an intermolecular radical addition reaction to complete the total synthesis.

Schilancitrilactones B and C (1 and 2; Figure 1)<sup>[1]</sup> were isolated in 2012 by Sun and co-workers from the stems of *Schisandra Lancifolia*, which have been used in traditional Chinese medicine for the treatment of neurasthenia and related diseases.<sup>[2]</sup> Preliminary biological assays indicated that



Figure 1. Schilancitrilactones B and C.

schilancitrilactone C showed biological activities for inhibiting HIV-1 while schilancitrilactone B was not bioactive. The structures of these compounds were striking in that they contain a 5/7/5/5/5-fused pentacyclic ring system bearing nine stereogenic centers. In addition, the three *cis*-fused fivemembered rings (rings C–E), all with the envelope conformations, and seven contiguous chiral centers (including two quaternary centers) form a structurally rigid tricyclic ring system. Construction of these highly oxygenated unique motifs remains challenging. Herein, we present the first total synthesis of schilancitrilactones B and C. The key steps include the successful implementation of an intramolecular radical cyclization to prepare a seven-member ring, late-stage iodination, and an intermolecular radical C–C bond formation.

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Recently, the total synthesis of *Schisandraceae* triterpenoids has been of great interest to synthetic organic chemists because of the intriguing structures and diverse biological activities.<sup>[3]</sup> In 2011, Yang and co-workers reported the first total synthesis of schindilactone A.<sup>[4]</sup> Recently, the group of Li disclosed the first asymmetric total synthesis of rubriflordilactone A.<sup>[5]</sup> Herein we report our efforts on developing a new strategy to solve the chemical synthesis of 1 and 2, and a pathway for the synthesis of their analogues and derivatives for medicinal studies. Our retrosynthetic analysis is shown in Scheme 1. It was hypothesized that 1 and 2 might be



Scheme 1. Retrosynthetic analysis of schilancitrilactones B and C.

accessible by an intermolecular radical addition reaction between the alkyl iodide **3** and vinyl stannane **4**. The alkyl iodide **3** was expected to arise by late-stage iodination at C20 from the compound **5**, which in turn could be prepared from the compound **6** by a series of steps including an intramolecular radical cyclization at the C7–C8 bond to prepare the seven-membered ring.<sup>[6]</sup> The compound **6** was further deconstructed at the C10–C19 bond into the two simple building blocks **7** and **8**, which could be put together by an aldol reaction. The building blocks **4**, **7**, and **8** could be prepared from the commercially available compounds citraconic anhydride (**16**), L-carvone (**9**), and 1,3-cyclohexadiene (**19**), respectively.

Our work began with the synthesis of the alkyl iodide 7 (Scheme 2). Following the procedure by Fukuyama and co-workers,<sup>[7]</sup> L-carvone (9) was converted into the correspond-

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**Scheme 2.** Reagents and conditions: a)  $30\% H_2O_2$ , NaOH (aq), MeOH,  $0^{\circ}C$ ; b)  $H_2SO_4$ , THF/H\_2O (5:1), reflux; c) NaIO<sub>4</sub>, *i*PrOH/H\_2O (1:1),  $0^{\circ}C$  to RT; d)  $I_2$ , KI, NaHCO<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>/H<sub>2</sub>O (1:3),  $0^{\circ}C$ , 51% for 4 steps; e) NaBH<sub>4</sub>, MeOH,  $0^{\circ}C$ , 85%; f) AIBN, Bu<sub>3</sub>SnH, toluene,  $100^{\circ}C$ , 90%; g) PPh<sub>3</sub>,  $I_2$ , imidazole,  $0^{\circ}C$  to RT, THF, 84%. AIBN = 2,2'-azobis (2-methylpropionitrile), THF = tetrahydrofuran.

ing aldehyde **13** in a four-step sequence involving epoxidation, epoxide hydrolysis, oxidative cleavage of diols, and iodolactonization in 51 % overall yield (4 steps). The aldehyde **13** was selectively reduced with NaBH<sub>4</sub> to provide the alcohol **14** in 85% yield. The deiodination of **14** with AIBN and Bu<sub>3</sub>SnH afforded the compound **15**, which was converted into the corresponding **7** with I<sub>2</sub> in the presence of Ph<sub>3</sub>P and imidazole in 84% yield.<sup>[8]</sup>

Depicted in Scheme 3 is the construction of the vinyl stannane compound 4. The vinyl bromide 18 was prepared



**Scheme 3.** Reagents and conditions: a) PPh<sub>3</sub>CHCO<sub>2</sub>tBu, toluene, RT, 54%; b) TFA,  $CH_2Cl_2$ , 0°C; c) Br<sub>2</sub>, TFA,  $CDCl_3/CCl_4$  (1:1), RT; (d) Et<sub>3</sub>N, DMF, 0°C to RT, 76% for 3 steps; (e) [{Pd(allyl)Cl}<sub>2</sub>] (5 mol%), (Bu<sub>3</sub>Sn)<sub>2</sub>, LiCl, 1,4-dioxane, RT, 49%. DMF = *N*,*N*-dimethylformamide, TFA = trifluoromethanesulfonyl.

from the commercially available compound citraconic anhydride (16) in a reported four-step process in a 41% overall yield.<sup>[9]</sup> Stannylation of 18 was achieved and afforded 4 with  $[{Pd(allyl)Cl}_2]$  and  $(Bu_3Sn)_2$  in 49% yield.<sup>[10]</sup> It is noteworthy that 4 is not stable during purification, thus resulting in a low yield.

We then moved on to construct the aldehyde compound **8** (Scheme 4). By using the reaction conditions developed by



Scheme 4. Reagents and conditions: a) TPP, O<sub>2</sub>, hv, CCl<sub>4</sub>,  $-10^{\circ}$ C then thiourea, MeOH; b) BzCl, Et<sub>3</sub>N, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, RT, 73% for 2 steps; c) [{Pd(allyl)Cl}<sub>2</sub>] (3 mol%), ligand A, 22, K<sub>2</sub>CO<sub>3</sub>, MeOH, THF, 0°C; then DIPEA, 55°C; 70%; d) NaH, CH<sub>3</sub>I, DMF, 0°C, 90%, d.r. (at C13) = 4:1; e) NaBr, DMF, 180°C, 88%, d.r. (at C13) = 1:1; f) LDA, BrCH<sub>2</sub>CO<sub>2</sub>tBu, THF,  $-78^{\circ}$ C; g) TFA, CH<sub>2</sub>Cl<sub>2</sub>, 0°C to RT, 93% for 2 steps; h) EtMgBr, THF/Et<sub>2</sub>O (1:1),  $-78^{\circ}$ C to RT, 80%; i) O<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>/MeOH (1:1),  $-78^{\circ}$ C, then Me<sub>2</sub>S,  $-78^{\circ}$ C to RT; j) (Bn<sub>2</sub>NH<sub>2</sub>) (OCOCF<sub>3</sub>), toluene, 63°C, 80% for 2 steps. Bz = benzoyl, DIPEA = diisopropylethylamine, DMAP = 4-(*N*,*N*-dimethylamino) pyridine, LDA = lithium-diisopropyl amide, TPP = 5,10,15,20-tetraphenyl-21H,23H-porphine. Thermal ellipsoids are shown at 50% probability.<sup>[21]</sup>

Trost and co-workers,<sup>[11]</sup> the lactone 23 was obtained in a reported three-step process from the commercially available 1,3-cyclohexadiene (19). The steps included asymmetric palladium-catalyzed allylic alkylation. Methylation of 23 with NaH and CH<sub>3</sub>I provided the compound 24 in 90% yield with 4:1 diastereoselectivity at C13, and was then subjected to decarboxylation mediated by NaBr to produce a 1:1 mixture of the lactone 25 in 88% yield. Alkylation of 25 with *tert*-butyl bromoacetate gave the single diastereomer 26. Deprotection of 26 was achieved using trifluoroacetic acid and gave the acid 27 in 93% yield (two steps). Addition of ethyl magnesium bromide followed by acidic workup gave rise to the tricycle 28, having an ethyl group installed stereoselectively onto the tricyclic framework.<sup>[12]</sup> The absolute configuration of 28 was determined by X-ray crystallographic analysis. The cyclohexene ring in 28 was oxidatively cleaved by ozonolysis and the resulting dialdehyde 29 was directly subjected to intramolecular aldol condensation, thus vielding the ring-closed unsaturated aldehyde 8 (80% yield for two steps).<sup>[13]</sup>

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**Scheme 5.** Reagents and conditions: a) LDA, THF, -78 °C, then **8**, 86%, d.r. (at C19) = 17:1; b) CuCl<sub>2</sub>, EDC, toluene, 80 °C, 83%; c) CuI, Zn, Pyr/H<sub>2</sub>O (1:4), ultrasound, RT, 55% for **31**, 4% for **31**'; d) *m*CPBA, NaHCO<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -15 °C, 51%; e) NaOMe, MeOH, RT; then NiCl<sub>2</sub>·6 H<sub>2</sub>O, NaBH<sub>4</sub>, MeOH/THF (1:5), -15 °C, 73%; (f) ICl, THF, RT, 63%, d.r. (at C20) = 1.5:1; (g) **4**, AIBN, Bu<sub>3</sub>SnH, toluene, 4 Å M.S., 100 °C, 9% for **1**, 36% for **2**. EDC = 1-(3-dimethylaminopropyl)-3-ethylcarbodiimidem *m*CPBA = *m*-chloroperbenzoic acid. Thermal ellipsoids are shown at 50% probability.<sup>[21]</sup>

With the key intermediates 4, 7, and 8 in hand, we finished the total synthesis of schilancitrilactones B and C as shown in Scheme 5. The iodo compound 7 was converted into the lithium enolate with LDA at -78 °C and then reacted with 8 to give the aldol adduct 30 in 86% yield (d.r. = 17:1 at C19). Dehydration of 30 with 2 equivalents of EDC and a catalytic amount of CuCl<sub>2</sub> provided a 2:1 mixture of the inseparable diene lactone **6** in 83 % yield.<sup>[14]</sup> The structure of the  $\vec{E}$  isomer was confirmed by X-ray crystallographic analysis. Then we investigated the intramolecular radical cyclization to form the seven-membered ring. Initially, the conventional radical conditions (AIBN, Bu<sub>3</sub>SnH) led to rapid decomposition of 6 and trace amounts of cyclization product was observed. Photoredox catalysis<sup>[15]</sup> was also evaluated and no desired product was found. By using the method (CuI, Zn under ultrasound) for conjugate additions in aqueous media discovered by Luche et al.,<sup>[16]</sup> we were pleased to observe the cyclization product 31 in 55% yield, together with the isomer 31' in 4% yield. The structure of 31' was confirmed by X-ray crystallographic analysis. We reasoned that the conformation of 6 was suited for cyclization to give the seven-membered ring over the five-membered ring.<sup>[17]</sup> Epoxidation of **31** with mCPBA provided the epoxide 32 in 51% yield, and underwent ring opening with NaOMe/NiCl<sub>2</sub>·6H<sub>2</sub>O/NaBH<sub>4</sub> to give the alcohol 5 in 73%.<sup>[18]</sup> During this transformation, the epoxide 32 was converted into the intermediate 33 with NaOMe and then further reduced to give the desired product 5 with NiCl<sub>2</sub>·6H<sub>2</sub>O and NaBH<sub>4</sub>. Finally, we investigated the late-stage iodination and intermolecular radical addition reaction. It was found that treatment of 5 with ICl delivered the iodo compound **3** as a mixture of diastereomers (d.r. = 1.5:1 at C20) in 63 % yield,<sup>[18]</sup> and when **3** was heated with the vinyl stannane **4**, AIBN, and Bu<sub>3</sub>SnH provided the schilanci-trilactones B (**1**, 9%) and C (**2**, 36%) in 45% total yield. Around 25% yield of other isomers were observed based on the<sup>1</sup>H NMR analysis of the crude reaction mixture.<sup>[20]</sup> The characterization data obtained for synthetic **1** and **2** were in accord with the reported data for the natural products.

In summary, the first total synthesis of schilancitrilactones B and C has been accomplished by employing an intramolecular radical cyclization, late-stage iodination, and intermolecular radical addition as key steps in the 17 step synthesis (longest linear sequence) from commercially available materials. This strategy opens a pathway for the syntheses of other compounds related to schilancitrilactones B and C, as well as their derivatives and analogues.

**Keywords:** cyclizations · natural products · radical chemistry · total synthesis · terpenoids

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## **Communications**

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Total Synthesis of Schilancitrilactones  ${\sf B}$  and  ${\sf C}$ 



R = (R)-CH<sub>3</sub> Schilancitrilactones B R = (S)-CH<sub>3</sub> Schilancitrilactones C **In step**: The first total syntheses of schilancitrilactones B and C have been accomplished by using an intramolecular radical cyclization to provide the sevenmembered ring, late-stage iodination, and an intermolecular radical addition reaction as key steps. The approach provides a sequence for the syntheses of compounds related to the schilancitrilactones, as well as their derivatives and analogues.

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