

## Synthesis of 3-( $\omega$ -Phenylalkyl)catechols, Phenolic Lipids found in Sap of the Burmese Lac Tree, *via* Directed Metallation

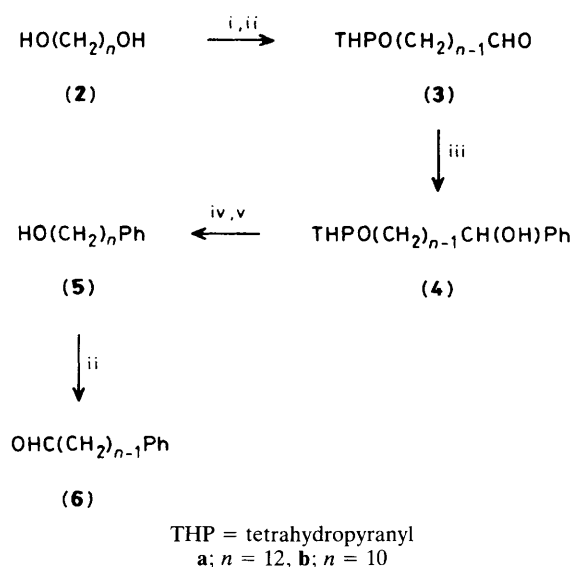
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3-( $\omega$ -Phenylalkyl)catechols were synthesized *via* directed metallation.

The phenolic lipids 3-(12-phenyldodecyl)catechol (**1a**) and 3-(10-phenyldecyl)catechol (**1b**) were recently isolated from the sap of the Burmese lac tree, *Melanorrhoea usiata*,<sup>1</sup> but neither their biological activities nor chemical properties are known. We tried to synthesize these lipids to carry out further studies on them. In an earlier work on synthesizing 3-alkyl-

catechols, *e.g.* urushiol, Yukawa *et al.*<sup>2</sup> alkylated catechol bistetrahydropyranyl ether *via* directed metallation but the yield was not very good. In this communication, we describe a new method of synthesizing these phenolic lipids *via* directed metallation using methoxymethyl ether as the directing group.<sup>3</sup>

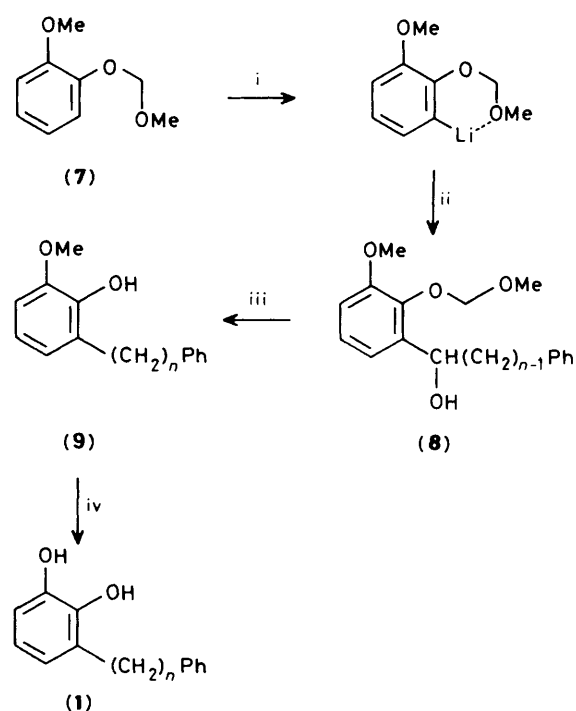


**Scheme 1.** Reagents: *i*, dihydropyran, *p*-MeC<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>H, CH<sub>2</sub>Cl<sub>2</sub>; *ii*, (COCl)<sub>2</sub>, Me<sub>2</sub>SO, CH<sub>2</sub>Cl<sub>2</sub>, Et<sub>3</sub>N; *iii*, PhMgBr, Et<sub>2</sub>O; *iv*, Et<sub>3</sub>SiH, CF<sub>3</sub>CO<sub>2</sub>H; *v*, NaOH, MeOH.

Monoprotection of dodecane-1,12-diol (**2a**) with dihydropyran and a catalytic amount of *p*-MeC<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>H followed by oxidation by Swern's procedure<sup>4</sup> gave the protected aldehyde (**3a**). Treatment of (**3a**) with phenylmagnesium bromide gave the alcohol (**4a**) [30% from (**2a**)]. Hydrogenolysis of (**4a**) with triethylsilane and trifluoroacetic acid<sup>5</sup> followed by hydrolysis with alkali gave the alcohol (**5a**) (59%). Oxidation of (**5a**) by Swern's procedure<sup>4</sup> gave the aldehyde (**6a**) (87%).

When the guaiacol methoxymethyl ether (**7**) was treated with *n*-butyl-lithium in diethyl ether at room temperature, a white precipitate formed and warming the mixture at 30 °C for 30 minutes resulted in an orange suspension. Addition of the aldehyde (**6a**) to this suspension gave the alcohol (**8a**) (49%). Ionic hydrogenation by Kursanov's procedure<sup>5</sup> gave the phenol (**9a**) (30%). Treatment of (**9a**) with boron tribromide gave 3-(12-phenyldecyl)catechol (**1a**) (84%), which had physical data identical with those reported by Oshima,<sup>1</sup> thus confirming the structure of the natural product.

3-(10-Phenyldecyl)catechol (**1b**) was also prepared by a similar method from decane-1,10-diol (**2b**) in 2.3% overall yield.



a;  $n = 12$ , b;  $n = 10$

**Scheme 2.** Reagents: *i*, Bu<sup>n</sup>Li, Et<sub>2</sub>O, 30 °C; *ii*, (**6**); *iii*, Et<sub>3</sub>SiH, CF<sub>3</sub>CO<sub>2</sub>H; *iv*, BBr<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>.

We thank Dr. Oshima for the spectral data of the natural products, and the Ministry of Education, Science and Culture, and Kinki University for the financial support of this work.

Received, 6th May 1986; Com. 597

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