

Chiral Allylic Cyanohydrins as Versatile Substrates for Diastereoselective Copper(I)-mediated S_N2' Allylic Substitutions

Sylvie Perrone, Albrecht Metzger, Paul Knochel*

Department Chemie und Biochemie, Ludwig-Maximilians-Universität München, Butenandtstraße 5-13, 81377 München, Germany
Fax +49(89)218077680; E-mail: Paul.Knochel@cup.uni-muenchen.de

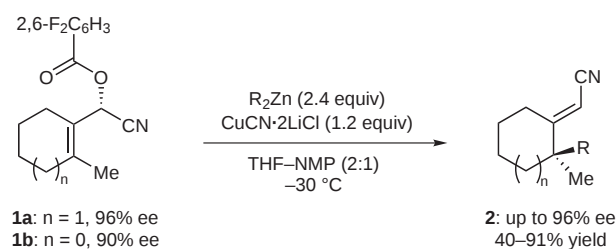
Received 17 January 2007

Abstract: 2,6-Difluorobenzoated derivatives bearing a protected cyanohydrin function undergo highly stereoselective copper(I)-mediated S_N2' allylic substitution reactions with diorganozinc reagents leading to chiral unsaturated nitriles.

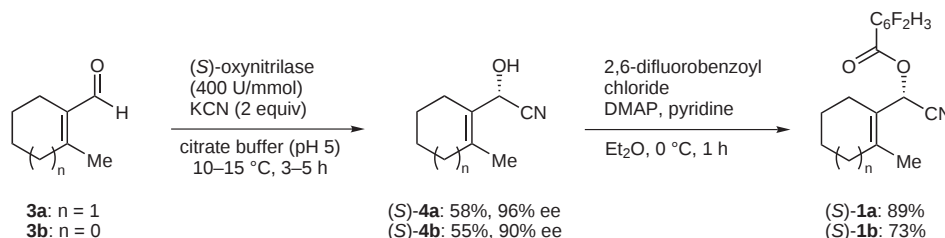
Key words: asymmetric synthesis, allylic substitution, cyanohydrins, copper, zinc, unsaturated nitriles

Cyanohydrins occupy a unique place at the interface between chemistry and biology.¹ They are readily prepared by a number of synthetic methods,² and their stereoselective synthesis can be achieved either by chemical³ or enzymatic procedures.⁴ Herein, we wish to report a highly stereoselective S_N2' allylic substitution reaction^{5–8} on chiral difluorobenzoated allylic cyanohydrins with diorganozinc reagents in the presence of $\text{CuCN}\cdot 2\text{LiCl}$ that produces α,β -unsaturated nitriles bearing a new stereogenic center in the γ -position.

Tetrasubstituted allylic difluorobenzoated cyanohydrins of type **1** bearing a methyl group in the γ -position undergo highly regio- and stereoselective S_N2' substitutions and afforded exclusively the S_N2' -substituted product of type **2** with perfect transfer of chirality (Scheme 1 and Table 1).



Scheme 1



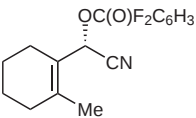
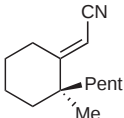
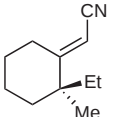
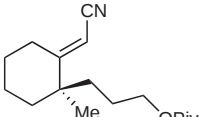
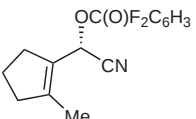
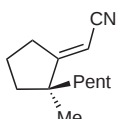
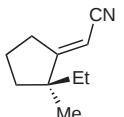
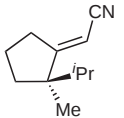
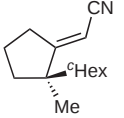
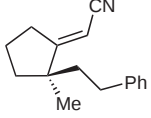
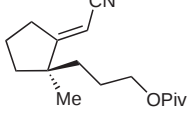
Scheme 2

Cyanohydrin derivatives like **1** were synthesized in their enantiomerically enriched form starting from the corresponding carbaldehydes **3a** and **3b** (Scheme 2). The enantioselective (S)-oxynitrilase-catalyzed addition of KCN (2 equiv) in a citrate buffer⁹ gave the chiral cyanohydrins (S)-**4a** (58%, 96% ee) and (S)-**4b** (55%, 90% ee). Subsequent esterification with 2,6-difluorobenzoyl chloride afforded (S)-**1a** in 83% and (S)-**1b** in 73% yield without racemization.

The substitution of the allylic cyanohydrin derivative **1a** with Pent_2Zn (2.4 equiv) in the presence of $\text{CuCN}\cdot 2\text{LiCl}$ [1.2 equiv, THF–NMP (2:1), -30°C to 0°C , 5 h) afforded the γ -substituted unsaturated nitrile **2a** as a single regioisomer in 75% yield and 96% ee (entry 1 of Table 1). No traces of the S_N2 product was detected and the substitution afforded only the *E*-isomer.¹⁰

The substitution with Et_2Zn led to the expected product **2b** (65%, 96% ee, entry 2). Similarly, a functionalized zinc reagent like $[\text{PivO}(\text{CH}_2)_3]_2\text{Zn}$ gave the chiral pivalate **5c** in 65% yield and 96% ee (entry 3). Although the substitution occurred in excellent diastereoselectivity, it was limited to primary alkyl substituents. Indeed, reactions with more sterically demanding groups like *i*- Pr_2Zn or *c*- Hex_2Zn were not selective. However, the cyclopentyl cyanobenzoate derivative **1b** did not suffer from these limitations and reacted with a wide range of diorganozincs. Thus, the substitution of **1b** with Pent_2Zn as well as with Et_2Zn in the presence of $\text{CuCN}\cdot 2\text{LiCl}$ afforded the corresponding α,β -unsaturated nitriles **2d** (91%, 90% ee, entry 4) and **2e** (80%, 90% ee, entry 5) with an optimum transfer of chirality. The substitution proceeded with excellent diastereoselectivity with secondary diorganozincs like *i*- Pr_2Zn and *c*- Hex_2Zn to furnish the sterically hindered substituted cyclopentane derivatives **2f** (75%, 90%

Table 1 Copper(I)-Mediated S_N2' Allylic Substitution of 2,6-Difluorobenzoated Allylic Cyanohydrins **1a,b**

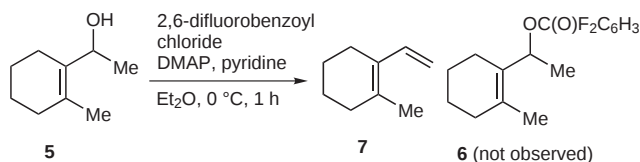
Entry	Allyl substrate (ee, %) ^a	R ² Zn (R)	α,β -Unsaturated nitrile of type 1	Yield (%) ^b	ee (%) ^a
1	 1a (96)	Pent	 2a	75	96
2	1a	Et	 2b	65	96
3	1a	PivO(CH ₂) ₃	 2c	65	96
4	 1b (90)	Pent	 2d	91	90
5	1b	Et	 2e	80	90
6	1b	<i>i</i> -Pr	 2f	75	90
7	1b	<i>c</i> -Hex	 2g	60	90
8	1b	Ph(CH ₂) ₂	 2h	40	90
9	1b	PivO(CH ₂) ₃	 2i	71	90

^a The ee was determined by GC analysis of the pure product on Chirasil-Dex CB column. In each case the racemic product was used for calibration.

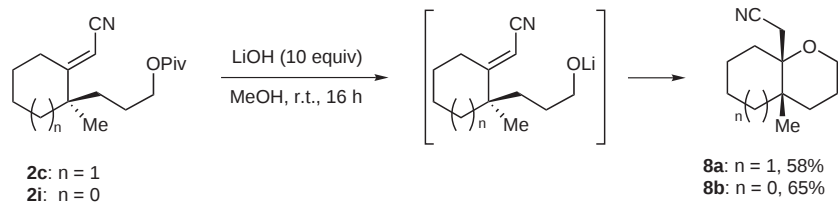
^b Isolated yield of analytically pure product.

ee, entry 6) and **2g** (60%, 90% ee, entry 7). Diorganozincs like [Ph(CH₂)₂]₂Zn (entry 8) or [PivO(CH₂)₃]₂Zn (entry 9) reacted similarly with **1b** and afforded, respectively, **2h** in 40% yield and 90% ee and **2i** (71% and 90% ee).

Interestingly, we could demonstrate that in this sequence the presence of the cyano group was essential. Indeed, any attempts to derivatize the cyclic allylic alcohol **5**, in which the cyano group is replaced by a methyl group, into the corresponding benzoated ester **6** provided only the elimination product **7** (Scheme 3).



Scheme 3



Scheme 4

Acknowledgment

S.P. thanks the DFG for a fellowship. We thank the Fonds der Chemischen Industrie, Chemetall GmbH (Frankfurt), Degussa AG (Frankfurt), BASF AG (Ludwigshafen) for the generous gift of chemicals.

References and Notes

- (1) (a) Hickel, A.; Hasslacher, M.; Griengl, H. *Physiol. Plant.* **1996**, *98*, 891. (b) Wajant, H.; Effenberger, F. *Biol. Chem.* **1996**, *377*, 611.
- (2) (a) *Synthesis and Applications of Non-Racemic Cyanohydrins and α -Amino Acids*, In *Tetrahedron Symposia-in-Print*, North, M., Ed. *Tetrahedron* **2004**, *60*, 10371. (b) Brunel, J.-M.; Holmes, I. P. *Angew. Chem. Int. Ed.* **2004**, *116*, 2810. (c) North, M. *Tetrahedron: Asymmetry* **2003**, *14*, 147. (d) Vachal, P.; Jacobsen, E. N. In *Comprehensive Asymmetric Catalysis*, Suppl. 1; Jacobsen, E. N.; Pfaltz, A.; Yamamoto, H., Eds.; Springer: Berlin, **2004**, 117–129. (e) Gregory, R. J. H. *Chem. Rev.* **1999**, *99*, 3649. (f) North, M. *Synlett* **1993**, 807. (g) Effenberger, F. *Angew. Chem., Int. Ed. Engl.* **1994**, *33*, 1555.
- (3) (a) García Ruano, J. L.; Martín Castro, A. M.; Rodríguez, J. H. *J. Org. Chem.* **1992**, *57*, 7235. (b) Deng, H.; Ister, M. P.; Snapper, M. L.; Hoveyda, H. A. *Angew. Chem. Int. Ed.* **2002**, *41*, 1009. (c) Krueger, C. A.; Kuntz, K. W.; Dzierba, C. D.; Wirschun, W. D.; Gleason, J. D.; Snapper, M. L.; Hoveyda, H. A. *J. Am. Chem. Soc.* **1999**, *121*, 4284. (d) Sigman, M. S.; Jacobsen, E. N. *J. Am. Chem. Soc.* **1998**, *120*, 5315. (e) Kim, S. S.; Rajagopal, G.; Song, D. H. *J. Organomet. Chem.* **2004**, *689*, 1734. (f) Tian, S. K.; Hong, R.; Deng, L. *J. Am. Chem. Soc.* **2003**, *125*, 9900. (g) Hayashi, M.; Miyamoto, Y.; Inoue, S.; Oguni, N. *J. Org. Chem.* **1993**, *58*, 1515. (h) Casas, J.; Baeza, A.; Nájera, C.; Sansano, J. M.; Saá, J. M. *Eur. J. Org. Chem.* **2006**, 1949. (i) Baeza, A.; Nájera, C.; Sansano, J. M. *Eur. J. Org. Chem.* **2007**, 1101.
- (4) (a) van Langen, L. M.; Selassa, R. P.; van Rantwijk, F.; Sheldon, R. A. *Org. Lett.* **2005**, *7*, 327. (b) Griengl, H.; Klempier, N.; Pöchlauer, P.; Schmidt, M.; Shi, N.; Zabelinskaja-Mackova, A. A. *Tetrahedron* **1998**, *54*, 14477. (c) Effenberger, F.; Ziegler, T.; Förster, S. *Angew. Chem., Int. Ed. Engl.* **1987**, *26*, 458. (d) Effenberger, F.; Gutterer, B.; Ziegler, T.; Eckhardt, E.; Aichholz, R. *Liebigs Ann. Chem.* **1991**, *47*, 54. (e) Santaniello, E.; Ferraboschi, P.; Grisenti, P.; Manzocchi, A. *Chem. Rev.* **1992**, *92*, 1071. (f) Liu, X.; Qin, B.; Zhou, X.; He, B.; Feng, X. *J. Am. Chem. Soc.* **2005**, *127*, 12224. (g) Ooi, T.; Miura, T.; Takaya, K.; Ichikawa, H.; Maruoka, K. *Tetrahedron* **2001**, *57*, 867.
- (5) Soorukram, D.; Knochel, P. *Angew. Chem. Int. Ed.* **2006**, *45*, 3686.
- (6) (a) Calaza, M. I.; Hupe, E.; Knochel, P. *Org. Lett.* **2003**, *5*, 1059. (b) Harrington-Frost, N.; Leuser, H.; Calaza, M. I.; Kneisel, F. F.; Knochel, P. *Org. Lett.* **2003**, *5*, 2111. (c) Soorukram, D.; Knochel, P. *Org. Lett.* **2004**, *6*, 2409.
- (7) For other examples of this type, see: (a) Demel, P.; Keller, M.; Breit, B. *Chem. Eur. J.* **2006**, *12*, 6669. (b) Demel, P.; Keller, M.; Breit, B. *Chem. Eur. J.* **2006**, *12*, 6684. (c) Breit, B.; Demel, P. *Adv. Synth. Catal.* **2001**, *343*, 429. (d) Breit, B.; Demel, P. *Tetrahedron* **2000**, *56*, 2833. (e) Breit, B. *Chem. Eur. J.* **2000**, *6*, 1519. (f) Breit, B.; Demel, P.; Studte, C. *Angew. Chem. Int. Ed.* **2004**, *43*, 3785. (g) Herber, C.; Breit, B. *Angew. Chem. Int. Ed.* **2005**, *44*, 5267. (h) Breit, B.; Herber, C. *Angew. Chem. Int. Ed.* **2004**, *43*, 3790. (i) Breit, B. *Angew. Chem. Int. Ed.* **1998**, *37*, 525. (j) Smitrovich, J. H.; Woerpel, K. A. *J. Org. Chem.* **2000**, *65*, 1601. (k) Spino, C.; Beaulieu, C.; Lafreniere, J. *J. Org. Chem.* **2002**, *65*, 7091. (l) Belelie, J. L.; Chong, J. M. *J. Org. Chem.* **2001**, *66*, 5552. (m) Ibuka, T.; Habashita, H.; Otaka, A.; Fujii, N.; Oguchi, Y.; Uyehara, T.; Yamamoto, Y. *J. Org. Chem.* **1991**, *56*, 4370. (n) Yamamoto, Y.; Tanaka, M.; Ibuka, T.; Chounan, Y. *J. Org. Chem.* **1992**, *57*, 1024. (o) Marino, J. P.; Viso, A.; Lee, J.-D.; Fernandez de la Pradilla, R.; Fernandez, P.; Rubio, M. B. *J. Org. Chem.* **1997**, *62*, 645.

In addition, we explored the transformation of the resulting α,β -unsaturated nitriles of type **2**. We found that **2c** and **2i** underwent a diastereoselective cyclization when treated with LiOH leading to the fused bicycles **8a** and **8b** (Scheme 4).

In summary, we have shown that enantiomerically enriched difluorobenzoated cyanohydrin derivatives constituted substrates of choice for the copper(I)-mediated S_N2' allylic substitution. They reacted with diorganozinc reagents in the presence of CuCN·2LiCl leading to α,β -unsaturated nitriles bearing a new stereogenic center in the γ position with an excellent transfer of chirality.¹¹

- (8) For copper-catalyzed reactions using chiral ligands, see: (a) van Klaveren, M.; Persson, E. S. M.; del Villar, A.; Grove, D. M.; Bäckvall, J.-E.; van Koten, G. *Tetrahedron Lett.* **1995**, *36*, 3059. (b) Karlström, A. S. E.; Huerta, F. F.; Meuzelaar, G. J.; Bäckvall, J.-E. *Synlett* **2001**, 923. (c) Meuzelaar, G. J.; Karlström, A. S. E.; van Klaveren, M.; Persson, E. S. M.; del Villar, A.; van Koten, G.; Bäckvall, J.-E. *Tetrahedron* **2000**, *56*, 2895. (d) Dübner, F.; Knochel, P. *Angew. Chem. Int. Ed.* **1999**, *38*, 379. (e) Dübner, F.; Knochel, P. *Tetrahedron Lett.* **2000**, *41*, 9233. (f) Alexakis, A.; Malan, C.; Lea, L.; Benhaim, C.; Fournioux, X. *Synlett* **2001**, 927. (g) Alexakis, A.; Croset, K. *Org. Lett.* **2002**, *4*, 4147. (h) Tissot-Croset, K.; Polet, D.; Alexakis, A. *Angew. Chem. Int. Ed.* **2004**, *43*, 2426. (i) Tissot-Croset, K.; Alexakis, A. *Tetrahedron Lett.* **2004**, *45*, 7375. (j) Falcicola, C.; Tissot-Croset, K.; Alexakis, A. *Angew. Chem. Int. Ed.* **2006**, *45*, 5995. (k) Alexakis, A.; Tomassini, A.; Andrey, O.; Bernardinelli, G. *Eur. J. Org. Chem.* **2005**, 1332. (l) Polet, D.; Alexakis, A. *Org. Lett.* **2005**, *7*, 1621.
- (9) Klempier, N.; Pichler, U.; Griengl, H. *Tetrahedron: Asymmetry* **1995**, *6*, 845.
- (10) The stereochemistry of the double bond was confirmed by 2D ^1H NMR spectroscopy.
- (11) **Typical Procedure for the $\text{S}_{\text{N}}2'$ Substitution: Preparation of (2*S*,*E*)-2-(2-Methyl-2-pentylcyclopentylidene)acetonitrile (2d)**
A flame-dried flask equipped with a magnetic stirring bar, an argon inlet, and a septum was charged with $\text{CuCN}\cdot 2\text{LiCl}$ solution (2.6 mL, 1.0 M in THF, 2.64 mmol, 1.2 equiv), anhyd NMP (2.6 mL, overall ratio of solvents THF–NMP = 2:1). The mixture was cooled at $-30\text{ }^\circ\text{C}$. To this solution was added dropwise dipentylzinc solution (1.1 mL, 4.8 M in THF, 5.28 mmol, 2.4 equiv). The resulting mixture was stirred at $-30\text{ }^\circ\text{C}$ for 45 min, and then (2*S*)-cyano(2-methylcyclopent-1-enyl)methyl 2,6-difluorobenzoate (**1b**, 610 mg, 2.2 mmol, 90% ee) was added dropwise as a solution in THF (1.5 mL). The reaction mixture was stirred at $-30\text{ }^\circ\text{C}$ to $0\text{ }^\circ\text{C}$ for 2 h and sat. aq NH_4Cl solution (5 mL) was added. The quenched reaction mixture was poured into 25% aq NH_3 (2 mL), aq sat. NH_4Cl (100 mL) and Et_2O (100

mL) and stirred at $25\text{ }^\circ\text{C}$ until the copper salts had dissolved then extracted with Et_2O ($3 \times 100\text{ mL}$). The combined extracts were washed with H_2O , brine and dried over Mg_2SO_4 . Evaporation of the solvents and purification by column chromatography (silica gel, pentane– Et_2O , 9:1) afforded the unsaturated nitrile **2d** (382 mg, 2.0 mmol, 91%, 90% ee) as a pale yellow oil.

^1H NMR (300 MHz, CDCl_3): δ = 5.02 (t, 3J = 2.55 Hz, 1 H), 2.81–2.56 (m, 2 H), 1.78–1.66 (m, 2 H), 1.66–1.53 (m, 2 H), 1.34–1.16 (m, 8 H), 1.04 (s, 3 H), 0.87 (t, 3J = 6.75 Hz, 3 H).

^{13}C NMR (75 MHz, CDCl_3): δ = 181.9, 117.9, 89.9, 47.6, 40.4, 38.9, 34.0, 32.6, 26.1, 24.4, 22.7, 22.1, 14.3. MS (EI, 70 eV): m/z (%) = 191 (8) [M^+], 176 (11), 162 (8), 148 (12), 121 (89), 120 (100), 106 (12), 93 (23), 79 (25). HRMS: m/z calcd: 191.1674; found: 191.1651. $[\alpha]_{\text{D}}^{20}$ –15.9 (c 1.49, CHCl_3). GC (Chirasil-Dex CB), $100\text{ }^\circ\text{C}$ (5 min), ramp of $2\text{ }^\circ\text{C}/\text{min}$ to $140\text{ }^\circ\text{C}$; $t_{\text{R}}(\text{min})$ = 23.45 (R), 23.93 (S).

(2*R*,*E*)-(-2-Methyl-2-phenethylcyclopentylidene)acetonitrile (2h)

^1H NMR (400 MHz, CDCl_3): δ = 7.31–7.14 (m, 5 H), 5.09 (t, 3J = 2.55 Hz, 1 H), 2.91–2.46 (m, 4 H), 1.88–1.58 (m, 6 H), 1.14 (s, 3 H). ^{13}C NMR (100 MHz, CDCl_3): δ = 181.4, 142.2, 128.7, 128.4, 126.2, 117.8, 90.3, 47.7, 42.5, 38.9, 34.0, 31.3, 26.1, 22.2. MS (EI, 70 eV): m/z (%) = 225, 121 (16), 134 (9), 105 (100), 104 (63), 91 (64), 79 (18), 77 (18), 65 (15). HRMS: m/z calculated: 225.1517; found: 225.1487. $[\alpha]_{\text{D}}^{20}$ –5.8 (c 0.69, CHCl_3). GC (Chirasil-Dex CB), $100\text{ }^\circ\text{C}$ (5 min), ramp of $2\text{ }^\circ\text{C}/\text{min}$ to $160\text{ }^\circ\text{C}$; $t_{\text{R}}(\text{min})$ = 49.21 (R), 50.24 (S).

{4*a*-Methyl-hexahydro-cyclopenta[b]pyran-7*a*-yl}acetonitrile (8b)

^1H NMR (300 MHz, CDCl_3): δ = 3.81–3.74 (m, 1 H), 3.53–3.43 (m, 1 H), 2.85 (d, 2J = 16.8 Hz, 1 H), 2.35 (d, 2J = 16.8 Hz, 1 H), 2.19–2.00 (m, 2 H), 1.94–1.58 (m, 5 H), 1.45–1.17 (m, 3 H), 0.85 (s, 3 H). ^{13}C NMR (75 MHz, CDCl_3): δ = 117.8, 82.8, 61.8, 42.5, 36.5, 34.8, 30.2, 25.5, 21.3, 20.8, 18.9. MS (EI, 70 eV): m/z (%) = 180, 162 (2), 150 (2), 139 (100), 111 (15), 93 (15), 81 (12), 68 (36). HRMS: m/z calcd: 180.1388; found: 180.1390.

Copyright of Synlett is the property of Georg Thieme Verlag Stuttgart and its content may not be copied or emailed to multiple sites or posted to a listserv without the copyright holder's express written permission. However, users may print, download, or email articles for individual use.