

Synthetic Studies toward Potent Cytotoxic Agents Amphidinolides: Synthesis of the C₁-C₁₈ Moiety of Amphidinolides G, H and L

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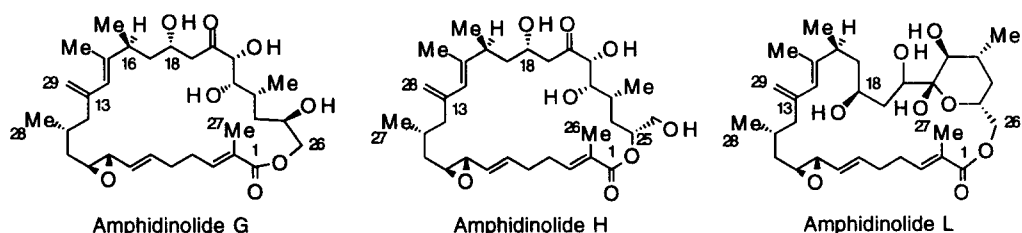
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Abstract

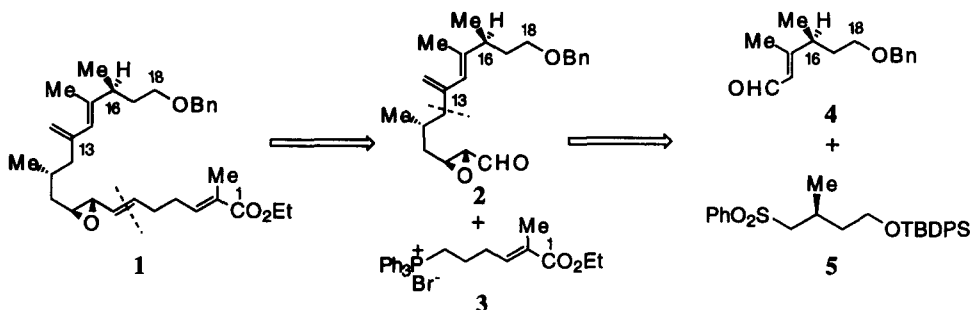
Stereoselective synthesis of the (8*S*, 9*S*, 11*R*, 16*S*)-C₁-C₁₈ segment **1** of amphidinolides G, H and L, bearing the unique trisubstituted “*s-cis*-1,3-diene” moiety (C₂₈₍₂₉₎=C₁₃-C₁₄=C₁₅), has been achieved for the first time following a highly efficient convergent strategy. © 1998 Elsevier Science Ltd. All rights reserved.

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The amphidinolides constitute a family of structurally complex macrolide molecules isolated from marine sources. Many of them have potent toxicity against various tumor cell lines [1,2]. Total synthesis of none of these compounds has so far been reported [3-15]. One of the major obstacles encountered en route to the total synthesis of some of the important members of this family, like amphidinolides B, D, G, H and L, is the construction of an uncommon trisubstituted “*s-cis*-1,3-diene” moiety (C₂₈₍₂₉₎=C₁₃-C₁₄=C₁₅), present in these molecules, in its naturally occurring configuration. We report here the first synthesis of this very important structural entity followed by its elaboration to the (8*S*, 9*S*, 11*R*, 16*S*)-C₁-C₁₈ fragment of amphidinolides G, H and L.

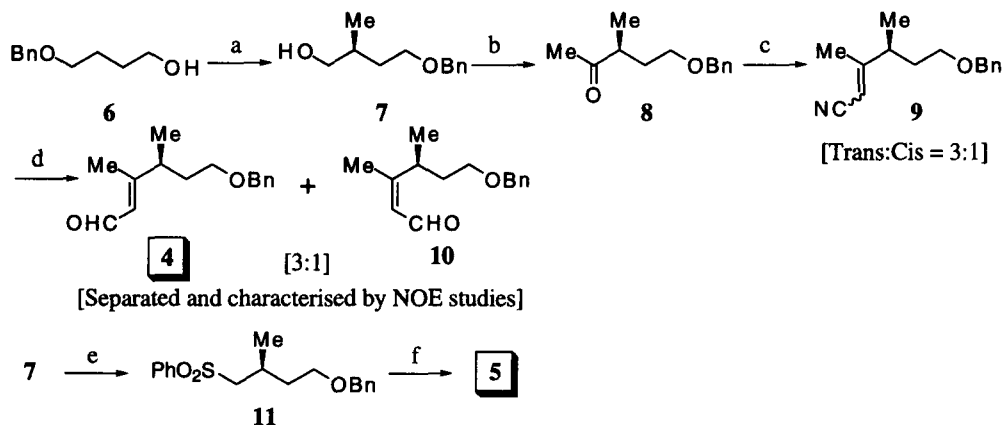


Retrosynthetically, the C₁-C₁₈ fragment **1** can be divided, as shown in Scheme 1, into two halves: the C₇-C₁₈ fragment **2** bearing the diene and the epoxyaldehyde functionalities and the C₁-C₆ Wittig component **3**. The important diene moiety of **2** was planned to be constructed by coupling the *E*- α,β -unsaturated aldehyde **4** and the functionalized sulfone unit **5**.



Scheme 1. Retrosynthetic analysis of **1**.

Scheme 2 outlines the syntheses of fragments **4** and **5**. Use of a common chiral precursor, (*S*)-4-benzyloxy-2-methylbutan-1-ol (**7**), for the syntheses of both the fragments is the salient feature of this scheme. The monobenzyl-protected butane-1,4-diol was transformed into **7** by Evans asymmetric alkylation method following reported procedures [16–18]. The chiral alcohol **7** was then converted to the methylketo intermediate **8** in 3 steps in 75% overall yield. Horner-Wadsworth-Emmons olefination of **8** with diethyl cyanomethylphosphonate gave a mixture of acrylonitriles **9** (3:1) in 98% yield. Reduction of this mixture of nitriles with DIBAL afforded the isomeric aldehydes **4** and **10** (3:1), in 93% yield, which could be separated easily at this stage by silica gel column chromatography. That the major isomer was the required *E*-olefin **4** was confirmed by ¹H NOE difference spectroscopic studies. Irradiation of the olefinic C₁₄-H signal of **10** at δ 5.87 caused significant enhancement of the 29-Me resonance at δ 1.87, indicating a *cis*-relationship between them. As expected, there was no NOE observed between the C₁₄-H and the 29-Me in the *E*-olefin **4**.

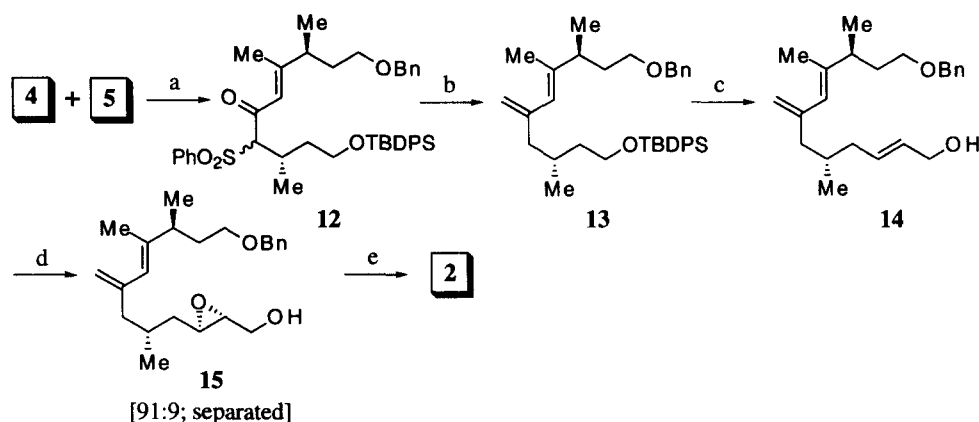


Scheme 2. Reagents and conditions. a) Ref. 16–18; b) (i) (COCl)₂ (1.5 eq.), DMSO (3.2 eq.), Et₃N (5 eq.), CH₂Cl₂, -78 to 0 °C, 1.5 h; (ii) MeMgI (2.0 M, 2 eq.), Et₂O, 0 to 25 °C, 1 h; (iii) same as step (i), 75% from **7**; c) (Et₂O)₂P(O)CH₂CN (1.5 eq.), NaH (1.5 eq.), DME, 25 °C, 1 h, then **8** in DME, 25 °C, 3 h, 98%; d) DIBAL (1.1 eq.), toluene, -78 °C, 1 h, 93%; e) (i) MsCl (1.2 eq.), pyridine, 25 °C, 3 h; (ii) PhSH (1.2 eq.), K₂CO₃ (1.5 eq.), DMF, 25 °C, 0.5 h; (iii) *m*CPBA (4 eq.), CH₂Cl₂, 0 to 25 °C, 3 h; 89% from **7**; f) (i) H₂, Pd/C, MeOH, 25 °C, 1 h; (ii) TBDPSCl (1.2 eq.), Et₃N (2 eq.), DMAP (0.1 eq.), CH₂Cl₂, 25 °C, 3 h; 95% from **11**.

For the synthesis of **5**, the common chiral precursor **7** (Scheme 2) was converted to the phenylsulfone **11** in 3 steps in 89% overall yield. This was followed by a change in the protective group, which was necessary to differentiate the two protective groups at a later stage, furnishing the requisite TBDPS-protected sulfone **5**.

Coupling of fragments **4** and **5** and further elaboration of the resulting coupled product to the advanced stage intermediate **2** is delineated in Scheme 3. Addition of the anion generated from **5** to the aldehyde **4** gave a diastereomeric mixture of β -hydroxysulfones which were oxidized using *O*-iodoxybenzoic acid (IBX) [19] to ketones **12** in 80% overall yield. Removal of the phenylsulfone appendage using lithium naphthalenide (LN) (60% yield) [20] and subsequent one-carbon Wittig olefination furnished the intermediate **13**^{1,2} in 88% yield, thus, completing successfully the first synthesis of the targeted “*s-cis*-1,3-diene” moiety.

Routine functional group manipulations converted **13** to the allylic alcohol **14** in 4 steps in 80% overall yield. Sharpless asymmetric epoxidation [21] of **14** with natural (+)-diethyl L-tartrate gave the expected (8*S*,9*S*)-epoxy alcohol **15** as the major product (in 91:9 ratio). The minor diastereomer could be easily separated by silica gel column chromatography. The epoxyalcohol **15** was subsequently oxidized to get the intermediate **2**.



Scheme 3. Reagents and conditions. a) (i) **5** (1 eq.), ⁿBuLi (1 eq.), THF, -78 °C, 20 min., then **4** in THF, -78 to 0 °C, 1 h; (ii) IBX (2 eq.), DMSO, 25 °C, 1 h, 80% in 2 steps; (b) (i) LN (excess), THF, -78 °C, 1 h, 60%; (ii) Ph₃P=CH₂ (2 eq.), Et₃O, 0 °C, 0.5 h, 88%; c) (i) TBAF (1.5 eq.), THF, 25 °C, 5 h; (ii) (COCl)₂ (1.5 eq.), DMSO (3.2 eq.), Et₃N (5 eq.), CH₂Cl₂, -78 to 0 °C, 1.5 h; (iii) Ph₃P=CHCO₂Et (2 eq.), C₆H₆, 25 °C, 1 h; (iv) DIBAL (2.2 eq.), CH₂Cl₂, -78 °C, 1 h, 80% from **13**; d) Ti(PrO)₄ (0.2 eq.), (+)-DET (0.22 eq.), TBHP (2 eq.), CH₂Cl₂, -10 °C, 12 h, 92% (based on 40% recovered starting material); e) same as step c(ii).

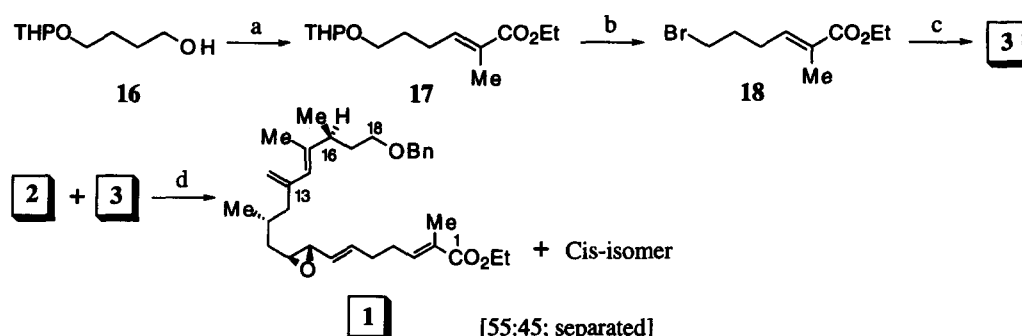
Scheme 4 describes the preparation of the phosphonium salt **3** and its use in the olefination of epoxy aldehyde **2**. Mono-THP-protected butane-1,4-diol **16** was oxidized and olefination with stabilized ylide gave the α,β -unsaturated ester **17**. Deprotection of the THP-ether, bromination of the hydroxyl group and finally, treatment with Ph₃P gave the Wittig salt **3**.

Finally, the ylide generated from **3** was reacted with the aldehyde **2**, following the procedure reported by Kobayashi *et al* [13], giving a mixture of olefins (*trans*:*cis* = 55:45) which were separated by preparative TLC to furnish the desired C₁-C₁₈ fragment **1**.^{1,3}

¹ Satisfactory NMR, IR and mass spectra were obtained for this compound.

² **13**: [α]_D²² = -10.4 (c 0.5, CHCl₃); ¹H NMR (CDCl₃, 200 MHz): δ 7.7-7.2 (m, 15 H, aromatic), 5.54 (s, 1 H, C₁₄-H), 4.91 and 4.75 (two s, 2 H, C=CH₂), 4.42 (s, 2 H, CH₂Ph), 3.67 (t, *J* = 6.8 Hz, 2 H, CH₂OTBDPS), 3.35 (t, *J* = 6.8 Hz, 2 H, CH₂OBn), 2.32 (m, 1 H, C₁₆-H), 2.06 (dd, *J* = 14.8, 5.6 Hz, 1 H, C₁₂-H), 1.88-1.52 (m, 4 H, C₁₁-H, C₁₂-H, C₁₇-H₂), 1.68 (s, 3 H, C₁₅-CH₃), 1.3 (m, 2 H, C₁₀-H₂), 1.04 (s, 9 H, SiPh₂Bu), 1.0 (d, *J* = 6.7 Hz, 3 H, C₁₆-CH₃), 0.78 (d, *J* = 6.3 Hz, 3 H, C₁₁-CH₃).

³ **1**: [α]_D²² = 14.5 (c 0.2, CHCl₃); ¹H NMR (CDCl₃, 400 MHz): δ 7.32 (m, 5 H, aromatic), 6.74 (t, *J* = 7 Hz, 1 H, C₅-H), 5.69 (dt, *J* = 15.2, 7.7 Hz, 1 H, C₆-H), 5.57 (s, 1 H, C₁₄-H), 5.08 (dd, *J* = 15.2, 8.1 Hz, 1 H, C₇-H), 4.96 and 4.81 (two s, 2 H, C₁₃=CH₂), 4.48 (s, 2 H, CH₂Ph), 4.19 (q, *J* = 7 Hz, 2 H, CO₂CH₂CH₃), 3.4 (t, *J* = 7 Hz, 2 H, CH₂OBn), 3.29 (dd, *J* = 8.1, 2.1 Hz, 1 H, C₈-H), 2.8 (dt, *J* = 7, 2.1 Hz, 1 H, C₉-H), 2.45-1.88 (m, 8 H, allylic, C₁₁-H, C₁₆-H), 1.84 (s, 3 H, C₂-CH₃), 1.8-1.58 (m, 4 H, C₁₀-H₂, C₁₇-H₂), 1.7 (s, 3 H, C₁₅-CH₃), 1.28 (t, *J* = 7 Hz, CO₂CH₂CH₃), 1.03 (d, *J* = 6.8 Hz, 3 H, C₁₆-CH₃), 0.9 (d, *J* = 6.7 Hz, C₁₁-CH₃).



Scheme 4. Reagents and conditions. a) (i) $(\text{COCl})_2$ (1.5 eq.), DMSO (3.2 eq.), Et_3N (5 eq.), CH_2Cl_2 , -78 to 0°C , 1.5 h; (ii) $\text{Ph}_3\text{P}=\text{C}(\text{CH}_3)\text{CO}_2\text{Et}$ (2 eq.), C_6H_6 , 25°C , 1 h, 90% from **16**; b) (i) PTSA (cat.), MeOH, 25°C , 2 h; (ii) CBr_4 (2.5 eq.), Ph_3P (2.5 eq.), CH_2Cl_2 , 0 to 25°C , 2 h, 92% from **17**; c) Ph_3P (1.2 eq.), CH_3CN , reflux, 12 h, 85%; d) **3** (2 eq.), $^t\text{BuLi}$ (2 eq.), THF, -78 to 25°C , 12 h, 80%.

In conclusion, an efficient convergent route presented here led to the first stereoselective synthesis of the (8*S*, 9*S*, 11*R*, 16*S*)- C_1 - C_{18} segment of amphidinolides G, H, and L which will help to achieve the total synthesis of these molecules. Further work is under progress.

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References

- [1] Ishibashi M, Kobayashi J. *Heterocycles* 1997;44:543-572.
- [2] Kobayashi J, Ishibashi M. *Chem. Rev.* 1993;93:1753-1789.
- [3] For previous synthetic studies toward various fragments of amphidinolides see references 4-15.
- [4] O'Connor SJ, Williard PG. *Tetrahedron Lett.* 1989;30:4637-4640.
- [5] Boden C, Pattenden G. *Synlett* 1994;181-182.
- [6] Tsuda M, Sasaki T, Kobayashi J. *J. Org. Chem.* 1994;59:3734-3737.
- [7] Ishibashi M, Ishiyama H, Kobayashi J. *Tetrahedron Lett.* 1994;35:8241-8242.
- [8] Eng HM, Myles DC. Abstracts of the papers of the American Chemical Society 1995;209:No. Pt 2, 405 ORGN1.
- [9] Chakraborty TK, Thippeswamy D, Suresh VR, Jayaprakash S. *Chemistry Lett.* 1997;563-564.
- [10] Chakraborty TK, Suresh VR. *Chemistry Lett.* 1997;565-566.
- [11] Lee D-H, Lee S-W. *Tetrahedron Lett.* 1997;38:7909-7910.
- [12] Tsuda M, Hatakeyama A, Kobayashi J. *J. Chem. Soc. Perkin Trans. 1* 1998;149-155.
- [13] Kobayashi J, Hatakeyama A, Tsuda M. *Tetrahedron* 1998;54:697-704.
- [14] Hollingworth GJ, Pattenden G. *Tetrahedron Lett.* 1998;39:703-706.
- [15] Cid MB, Pattenden G. *Synlett* 1998;540-542.
- [16] Evans DA, Ennis MD, Mathre DJ. *J. Am. Chem. Soc.* 1982;104:1737-1738.
- [17] Smith III AB, Hale KJ. *Tetrahedron Lett.* 1989;30:1037-1040.
- [18] Boeckman Jr. RK, Barta TE, Nelson SG. *Tetrahedron Lett.* 1991;32:4091-4094.
- [19] Frigerio M, Santagostino M. *Tetrahedron Lett.* 1994;35:8019-8022.
- [20] Jones AB, Villalobos A, Linde II RG, Danishefsky SJ. *J. Org. Chem.* 1990;55:2786-2797.
- [21] Gao Y, Hanson RM, Klunder JM, Ko SY, Masamune H, Sharpless KB. *J. Am. Chem. Soc.* 1987;109:5765-5780.