

**A highly efficient and selective synthesis of lissoclinolide  
featuring hydrogen transfer hydrozirconation, *trans*-  
selective Pd-catalyzed cross coupling of alkenylzirconiums  
with 1,1-dibromoalkenes and Ag-catalyzed lactonization  
providing (*Z*)- $\gamma$ -alkylidenebutenolides**

**Caiding Xu and Ei-ichi Negishi\***

*Department of Chemistry, Purdue University, West Lafayette, Indiana 47907-1393, U.S.A.*

Received 14 October 1998; revised 30 October 1998; accepted 2 November 1998

**Abstract:**

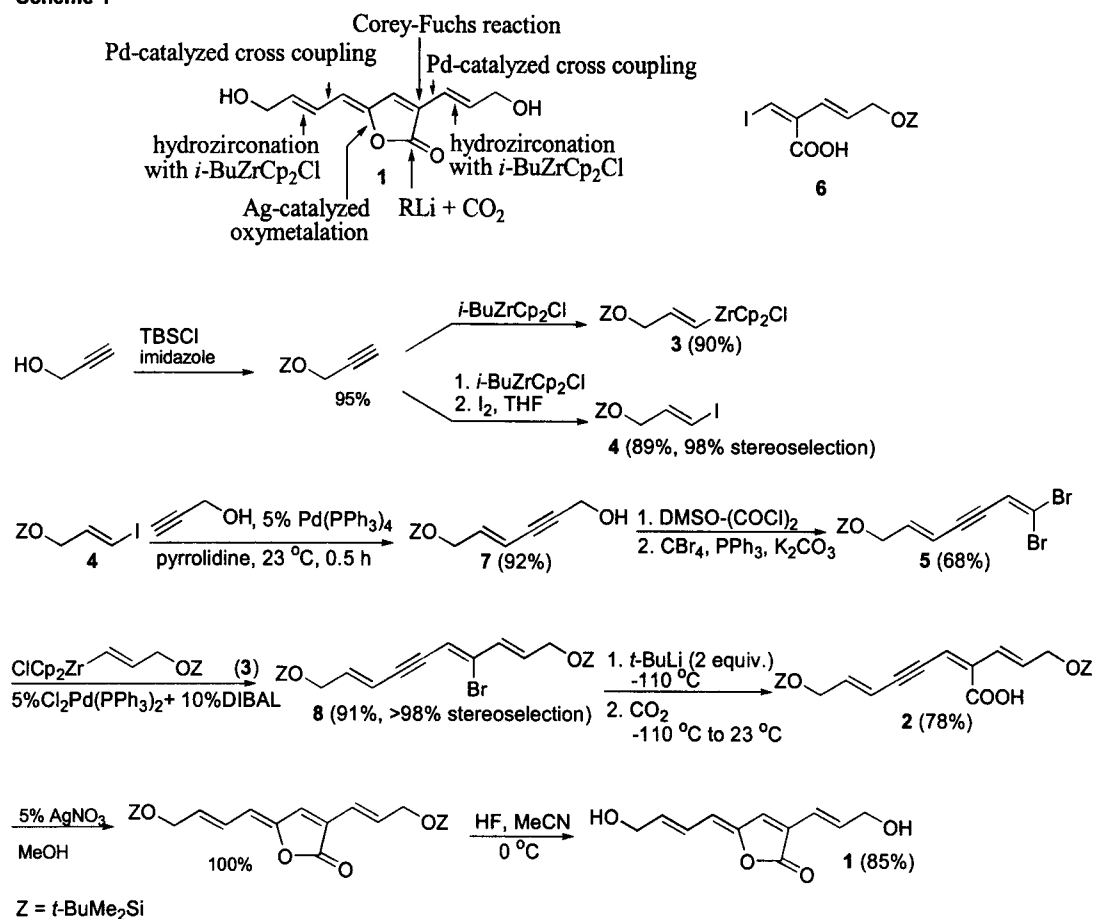
An antibiotic lissoclinolide has been synthesized from propargyl alcohol in 9 steps and in 32% overall yield *via* (i) hydrogen transfer hydrozirconation of TBS-protected propargyl alcohol with *i*-BuZrCp<sub>2</sub>Cl, (ii) Pd-catalyzed *trans*-selective cross coupling of the hydrozirconation product with a key 1,1-dibromoalkene intermediate **5** and (iii) Ag-catalyzed lactonization of a trienynoic acid precursor **2**. © 1998 Elsevier Science Ltd. All rights reserved.

We have recently [1] reported that the Pd-catalyzed cross coupling-lactonization tandem process [2] based on Sonogashira coupling [3] can be optimized to give (*Z*)- $\gamma$ -alkylidenebutenolides in high yields, one of the key findings being the desirability of the use of a Pd-PPh<sub>3</sub> mixture, in which the PPh<sub>3</sub>/Pd ratio is  $\geq 4$  [4]. This tandem process has been successfully applied to the syntheses of natural products, such as rubrolides [1a], (+)-goniobutenolide [1b], and frelingyne [1c], which provided, for the first time, examples of natural products syntheses *via* Pd-catalyzed lactonization of ynoic acids [1d]. We have also reported [1a] that, despite the lack of opportunity for exploiting the highly efficient cross coupling-lactonization tandem process, lactonization of (*Z*)-2-en-4-ynoic acids catalyzed by Ag salts [5] can provide (*Z*)- $\gamma$ -alkylidenebutenolides in excellent yields under dilute conditions.

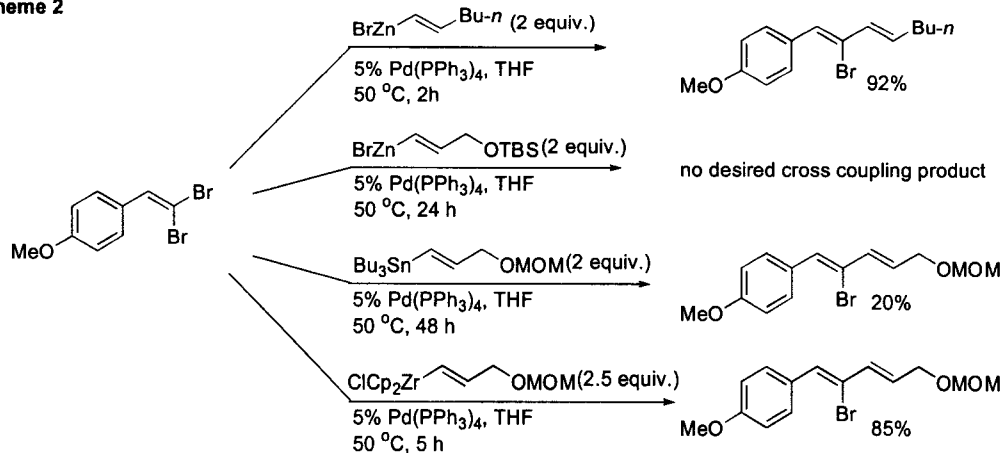
We now report that the Ag-catalyzed lactonization is significantly superior to the Pd-catalyzed procedure in the synthesis of lissoclinolide (**1**) [6] from its precursor **2**. Coupled with hydrogen transfer hydrozirconation [7] of TBS-protected propargyl alcohol, where TBS is *t*-BuMe<sub>2</sub>Si, to give **3** and Pd-catalyzed *trans*-selective cross coupling [8] of **3** with 1,1-dibromoalkene **5**, lissoclinolide has been synthesized in 9 steps and in 32% overall yield from propargyl alcohol with nearly complete (>98%) regio- and stereo-control (Scheme 1).

\* Department of Chemistry, Purdue University, 1393 Brown Building, West Lafayette, Indiana 47907-1393, U.S.A.

Scheme 1



Scheme 2



Lissoclinolide (**1**) [6], isolated from *Lissoclinum patella*, has been shown to be active against Gram negative bacteria. Interestingly, its (*5E*)-isomer, tetrenolin, is known to be active against Gram positive bacteria [9]. Despite their structural simplicity, their synthesis has not been reported.

We initially envisioned the synthesis of **1** *via* Pd-catalyzed cross coupling-lactonization tandem process [1,2] but encountered difficulties in the synthesis of the requisite (*Z*)- $\beta$ -iodo acid (**6**). We therefore opted for the synthesis of **2** as a precursor to **1**. This option gave us the flexibility of using either Pd-catalyzed or Ag-catalyzed lactonization, which eventually proved to be crucial in the synthesis of **1**. Hydrogen transfer hydrozirconation of TBS-protected propargyl alcohol at 50 °C in benzene [7] provided **3** (>98% *E*) in 90% yield, while its iodinolysis gave **4** in 89% yield. Since *i*-BuZrCp<sub>2</sub>Cl is readily generated *in situ* by treatment of commercially available and relatively stable Cp<sub>2</sub>ZrCl<sub>2</sub> with one equiv of *t*-BuMgCl, this represents a convenient alternative to conventional hydrozirconation [10] and its modifications using various metal hydrides [11]. The reaction of **4** with propargyl alcohol in the presence of pyrrolidine and 5 mol% of Pd(PPh<sub>3</sub>)<sub>4</sub> afforded **7** in 92% yield, which was oxidized with (COCl)<sub>2</sub> and DMSO [12] and then treated with CBr<sub>4</sub> (3 equiv) and PPh<sub>3</sub> (6 equiv) [13] to give **5** in 68% yield from **7**. Conversion of **5** into **8** *via* Pd-catalyzed cross coupling was initially attempted with the alkenylzinc derivative generated from **4** *via* lithiation with *t*-BuLi (2 equiv) and zincation with ZnBr<sub>2</sub>. To our surprise, **8** was not at all formed. The results were puzzling, since a model experiment led to very satisfactory *trans*-selective cross coupling, as shown in Scheme 2. Moreover, the use of a  $\gamma$ -benzyloxyzinc derivative in a similar Pd-catalyzed reaction with a 1,1-dibromoalkene was recently reported [14]. Although not clear, inactivation of the alkenylzinc derivative *via* *E*-to-*Z* isomerization-chelation may be suspected. Fortunately, direct use of **3** was found to be highly satisfactory (>98% *trans*-selective) for the Pd-catalyzed cross coupling. The use of organozirconiums in the Pd-catalyzed *trans*-selective cross coupling of 1,1-dihaloalkenes appears to be unprecedented.

Treatment of **8** with 2 equiv of *t*-BuLi at -110 °C followed by quenching with CO<sub>2</sub> produced **2** in 78% yield. It is essential to maintain the reaction temperature at -110 °C. Our attempts to achieve Pd-catalyzed carboxylation of **8** with CO and H<sub>2</sub>O, which, in principle, could be accompanied by Pd-catalyzed lactonization, have been unsuccessful. With **2** in hand, we had an opportunity to further compare the Pd-catalyzed and Ag-catalyzed lactonization procedures [1]. In the conversion of **2** to **9**, the maximum yield observed with the Pd-catalyzed lactonization was 35%, and the reaction was complicated by some side reactions which were not investigated. On the other hand, the Ag-catalyzed reaction at the concentration of 0.01 mol/L cleanly produced **9** in essentially quantitative yield. The crude product isolated by mere extractive workup and evaporation was  $\geq$ 98% pure by <sup>1</sup>H and <sup>13</sup>C NMR spectroscopy, which showed no extraneous signals for byproducts.

The following two procedures describe the two critical steps of the synthesis. **Ag-Catalyzed Lactonization: Conversion of 2 to 9.** To a solution of **2** (109 mg, 0.25 mmol) in MeOH (25 mL)

was added AgNO<sub>3</sub> (2 mg, 0.012 mmol). After stirring the mixture at 23 °C for 1 h, analysis by TLC indicated completion of the reaction. After concentration at room temperature and reduced pressure, yellow crystals thus obtained were dissolved in CDCl<sub>3</sub>, and filtered through a short-path silica gel column to remove AgNO<sub>3</sub>. Analysis by <sup>1</sup>H NMR spectroscopy using methylene bromide as a standard indicated the formation of **9** in quantitative yield. Evaporation of the solvents provided **9** in 100% yield. **Pd-Catalyzed Cross Coupling of 3 with 5.** To Cp<sub>2</sub>ZrCl<sub>2</sub> (2.32 g, 7.95 mmol) in 16 mL of benzene was added at 0 °C 2 M *t*-BuMgCl in Et<sub>2</sub>O (4.0 mL, 8.0 mmol), and the reaction mixture was heated to 50 °C for 1 h. The formation of *i*-BuZrCp<sub>2</sub>Cl in 94% yield was observed by <sup>1</sup>H NMR spectroscopy. To the solution containing *i*-BuZrCp<sub>2</sub>Cl was added TBS-protected propargyl alcohol (1.35 g, 7.95 mmol), and the reaction mixture was stirred at 50 °C for 5 h. Analysis by <sup>1</sup>H NMR spectroscopy indicated the formation of **3** in 90% yield. After evaporation of the solvents under reduced pressure, THF (20 mL), **5** (1.21 g, 3.18 mol) dissolved in 5 mL of THF, Cl<sub>2</sub>Pd(PPh<sub>3</sub>)<sub>2</sub> (0.11 g, 0.16 mmol), and DIBAL-H (0.32 mL of 1 M solution in THF, 0.32 mmol) were sequentially added. The reaction mixture was heated to 50 °C for 5 h. After the usual extractive workup, concentration and chromatography using 3% Et<sub>2</sub>O in hexane provided 1.36 g (91%) of **8**.

**Acknowledgments:** We thank the National Institutes of Health (GM 36792) and Purdue University for support of this research. We also thank Johnson Matthey for palladium compounds.

## References and Notes

- [1] (a) Kotori M, Negishi E. *Synthesis* 1997; 121. (b) Kotori M, Negishi E. *Tetrahedron Lett.* 1996; 37: 9041. (c) Liu F, Negishi E. *J. Org. Chem.* 1997; 62: 8591. (d) Negishi E, Kotori M. *Tetrahedron* 1997; 53: 6707.
- [2] Lu X, Huang X, Ma S. *Tetrahedron* 1993; 34: 5963.
- [3] Sonogashira K, Tohda Y, Hagihara N. *Tetrahedron Lett.* 1975; 4467.
- [4] In cases where either the PPh<sub>3</sub>/Pd ratio was 2 [2] or no PPh<sub>3</sub> was used [Lambert C, Utimoto K, Nozaki H. *Tetrahedron Lett.* 1984; 25: 5323], the reaction was significantly more complicated by some side reactions, such as competitive formation of pyranones, and conjugate substitution *via* Heck reaction. Even under the optimized conditions, however, homodimerization of alkynes may still be a significant side reaction [1b,1c].
- [5] (a) Ogawa Y, Maruno M, Wakamatsu T. *Synlett* 1995; 871. (b) Ogawa Y, Maruno M, Wakamatsu T. *Heterocycles* 1995; 41: 2587.
- [6] Davidson BS, Ireland CM. *J. Nat. Prod.* 1990; 53: 1036.
- [7] (a) Swanson, DR, Nguyen T, Noda Y, Negishi E. *J. Org. Chem.* 1991; 56: 2590. (b) Makabe H, Negishi E. unpublished results.
- [8] (a) Minato A, Suzuki K, Tamao K. *J. Am. Chem. Soc.* 1987; 109: 1257. (b) Roush, WR, Moriarty KJ, Brown BB. *Tetrahedron Lett.* 1990; 31: 6509.
- [9] Gallo GG, Coronelli C, Vigevani A, Lancini GC. *Tetrahedron* 1969; 25: 5677.
- [10] (a) Hart, DW, Schwartz J. *J. Am. Chem. Soc.* 1974; 96: 8115. (b) Schwartz J, Labinger JA. *Angew. Chem., Int. Ed. Engl.* 1976; 15: 333.
- [11] (a) Negishi E, Miller JA, Yosida T. *Tetrahedron Lett.* 1984; 25: 3407. (b) Buchwald SL, LaMadre SJ, Nielsen RB, Watson BT, King SM. *Tetrahedron Lett.* 1987; 28: 3895. (c) Lipshutz BH, Keil R, Ellsworth EL. *Tetrahedron Lett.* 1990; 31: 7257.
- [12] Omura K, Swern D. *Tetrahedron* 1978; 34: 1651.
- [13] Corey EJ, Fuchs PL. *Tetrahedron Lett.* 1972; 3769.
- [14] Panek JS; Hu T. *J. Org. Chem.* 1997; 62: 4912.

**Note added in proof:** After submission of this manuscript a paper reporting a different but related synthesis of lissoclinolide was brought to our attention [Rossi R, Bellina F, Biagetti M, Mannina L. *Tetrahedron Lett.* 1998; 39: 7799].