Reptilian Chemistry: Enantioselective Syntheses of Novel Components from a Crocodile Exocrine Secretion

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Abstract: Enantioselective syntheses of the two naturally-occurring diastereomers of dianeackerone (**1a** and **1b**) and of a closely related steroidal β -oxo ester (**2**), characterized from the paracloacal glandular secretion of the African dwarf crocodile, *Osteolaemus tetraspis*, are described.

Key words: enantioselective alkylation, crocodile, dianeackerone, natural products, steroids

The "ruling reptiles" (Archosauria), which dominated terrestrial animal communities during the Mesozoic era (245-65 million years ago), are now represented on earth by only about two dozen species of crocodilians.¹ All of these surviving species possess a pair of skin glands, designated the paracloacal glands, secretions of which are thought to serve a pheromonal role.² We recently elucidated the structures of the major constituents of the paracloacal gland secretion of the African dwarf crocodile (Osteolaemus tetraspis), and we have established the absolute configuration of both naturally occurring diastereomers of dianeackerone (1a and 1b), the secretion's chief volatile component, by synthesis.^{3,4} In this paper, we report in detail the enantioselective syntheses of 1a and 1b, as well as the synthesis of a steroidal β -oxo ester 2, which would appear to serve as the immediate biosynthetic precursor of the dianeackerones.



The methodology described by Enders and co-workers,^{5,6} based on the alkylation of optically active hydrazones derived from (*S*)-(–)-1-amino-2-(methoxymethyl)pyrrolidine (SAMP) and its (*R*)-(+) enantiomer provided highly stereoselective syntheses of phosphonium salts **3a** and **3b** (Scheme 1). Asymmetric alkylation of the SAMP-hydrazone (+)-**4** or (–)-**4**⁷ provided the products (+)-**5** or (–)-**5** in good yield and more than 95% de.⁸ Treatment of hydra-



Scheme 1

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zones (+)-5 or (-)-5 with CH₃I, followed by acidic hydrolysis with HCl in a two phase system readily formed the aldehydes (+)-6 or (-)-6. Reduction of (+)-6 or (-)-6 with borane-THF complex, followed by bromination with NBS in the presence of Ph_3P , yielded the bromides (+)-7 or (-)-7 in good yield. Reaction of these bromides with Ph₃P gave the desired chiral phosphonium salts, **3a** or **3b**. Similarly, the chiral precursor 8 was prepared via the SAMP-hydrazone (-)-4 (Scheme 2). Asymmetric alkylation of (-)-4 with *t*-butyl bromoacetate produced the hydrazone 9 in 60% yield and better than 95% de.⁸ Ozonolysis of 9 in CH₂Cl₂ at -78 °C gave the corresponding aldehyde 10. Reaction of the aldehyde 10 with CH₃MgI at -78 °C yielded an unstable alcohol, which was immediately oxidized by Jones' reagent to afford the ketone 11. Protection of the carbonyl group in 11 with ethylene glycol and trimethyl orthoformate in the presence of p-TsOH provided the ketal 12 in excellent yield. Reduction of the ester 12 with LiAlH₄, followed by oxidation with PDC, supplied the desired aldehyde 8.



Scheme 2

Coupling of aldehyde **8** with the Wittig reagent derived from phosphonium salt **3a** provided the alkene **13** as a mixture of *cis*- and *trans*-isomers (*cis*:*trans*, approx. 9:1) (Scheme 3). Hydrogenation of **13** with Pd/C in MeOH, followed by deprotection with PPTS provided **1a**, $[\alpha]_D$ -8.7 (*c* 0.2, CHCl₃) {Lit.³ $[\alpha]_D$ -9 (*c* 0.1, CHCl₃)}. In a completely analogous fashion, coupling of aldehyde **8** with the Wittig reagent prepared from **3b** furnished the alkene **14** as a 9:1 mixture of *cis*- and *trans*-isomers. Subsequent hydrogenation and deprotection provided the desired **1b**, $[\alpha]_D$ +3.2 (*c* 0.2, CHCl₃) {Lit.³ $[\alpha]_D$ +4 (*c* 0.8, CHCl₃)}.



Scheme 3

While **1a** and **1b** are the major volatile constituents of the *O. tetraspis* paracloacal glandular fluid, the bulk of the secretion is made up of a series of steroidal esters of aromatic acids closely related to dianeackerone. Of these compounds, the β -oxo ester **2** is especially interesting, since it would appear to be the immediate precursor (upon hydrolysis and decarboxylation) of dianeackerone itself.

A retrosynthetic analysis of β -oxo ester 2 suggested that this target compound should be obtained by the reaction of aldehyde 15 with cholesteryl diazoacetate 16⁹(Scheme 4), which is readily prepared via by the treatment of cholesterol with the acid chloride 17¹⁰ (Scheme 5).



Scheme 4





Using the methodology described earlier, aldehyde 15 was synthesized as outlined in Scheme 6. Alkylation of the SAMP-hydrazone (-)-4 afforded hydrazone 18 in excess of 95% de.⁸ Ozonolysis of the 18 yielded a crude aldehyde, which was immediately reduced with borane-THF complex to give alcohol 19. Protection of 19 with benzyl bromide in the presence of NaH, followed by the removal of the silvl ether with HCl, provided alcohol 20 in good yield. Bromination of 20 with NBS in the presence of Ph_3P provided bromide 21, which was converted into the chiral phosphonium salt 22. Coupling of the Wittig, the reagent derived from 22, with aldehyde (-)-6generated a mixture of cis- and trans-alkene 23 (cis:trans, approx. 9:1). Hydrogenation and hydrogenolysis of 23 with Pd/C in MeOH gave the alcohol 24 in good yield. Oxidation of 24 with N-morpholine N-oxide (NMO) mediated by tetrapropylammonium perruthenate (TPAP) afforded aldehyde 15. Coupling of 15 with the diazoacetate 16 in the presence of a catalytic amount of SnCl₂ afforded the desired β -oxo ester 2.

We would like to emphasize, in conclusion, is only by virtue of having synthesized the dianeackerone diastereomers by the Enders' technique of stereoselective alkylation was it possible to assign absolute configurations to the natural products. The same technique has also enabled us to confirm the structure of cholesteryl β -oxo ester 2 by synthesis.

GC/MS analyses were performed using a Hewlett–Packard (HP) 5890 gas chromatograph installed with a 30 m × 0.25 mm fused silica column coated with DB-5 (J & W Scientific, Folsom, CA) coupled to an HP-5970 Mass Selective Detector. MS analyses were performed using a Micromass Autospec instrument operated in electron ionization (EI), chemical ionization (CI) (CH₄ as the reagent gas), or electrospray ionization modes. ¹H and ¹³C NMR spectra were recorded on a Varian XL 400 instrument using CDCl₃ or C₆D₆ as solvent. Chemical shifts (δ) are given in ppm downfield from TMS.

Alkylation of Butyraldehyde (+)-(*R*)-1-Amino-2-(methoxymethyl)pyrrolidinehydrazone [(+)-4] with 2-Phenylethyl Iodide

To a solution of LDA (1.5 M, 5.8 mL, 8.7 mmol) in anhyd Et₂O (20 mL), hydrazone (+)-4 (1.6 g, 8.7 mmol) in anhyd Et₂O (10 mL) was added dropwise at 0 °C over 10 min. The mixture was stirred at 0 °C under N₂ for 5 h and then cooled to -110 °C using a dry-ice/liquid N₂ bath. 2-Phenylethyl iodide (1.64 mL, 11.3 mmol) in Et₂O (20 mL) was added dropwise over 30 min. The mixture was slowly warmed up to 0 °C and quenched with sat. NH₄Cl (20 mL). The organic layer was separated, the aqueous layer was extracted with EtOAc (2×50 mL). The combined organic layers were washed with brine, dried (Na₂SO₄), filtered, and concentrated in vacuo. The residue was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to afford (*R*)-2-ethyl-4-phenylbutyraldehyde (2'*R*)-1'-amino-2'-(methoxymethyl)pyrrolidinehydrazone [(+)-**5**] as a colorless oil.

Yield: 1.63 g (65%); $[\alpha]_{D}$ +64 (*c* 1.4, CHCl₃).

¹H NMR (400 MHz,CDCl₃): $\delta = 0.92$ (t, 3H, J = 7.2 Hz), 1.40–1.56 (m, 2H), 1.71–2.00 (m, 5H), 2.15–2.24 (m, 1H), 2.58–2.74 (m, 3H), 3.34–3.50 (m, 4H), 3.39 (s, 3H), 3.36 (m, 1H), 6.49 (d, 1H, J = 6.8 Hz), 7.16–7.30 (m, 5H).



Scheme 6

¹³C NMR (100 MHz, CDCl₃): δ = 143.0, 142.8, 128.4 (2C), 128.2 (2C), 125.5, 74.7, 63.5, 59.2, 50.5, 43.9, 35.1, 33.6, 26.7, 26.5, 22.0, 11.4.

GC/MS (EI): m/z = 288 (M⁺), 243, 197, 184, 139, 123, 105, 91, 70. HRMS (EI): m/z calcd for C₁₈H₂₈N₂O (M⁺): 288.2202. Found: 288.2202.

(-)-5

Prepared from (-)-4 using above procedure, colorless oil, $[\alpha]_D = -64.9$ (*c* 1.2, CHCl₃).

(-)-(*R*)-2-Ethyl-4-phenylbutyraldehyde [(-)-6]

A mixture of hydrazone (+)-5 (900 mg, 3.12 mmol) and CH₃I (4 mL) was warmed to reflux under N₂ for 4 h. The surplus CH₃I was

removed in vacuo, and the residue was treated with 10% HCl (10 mL) in pentane (20 mL) with stirring at r.t. for 1 h. The organic layer was separated, the aqueous layer was extracted with pentane (2 × 40 mL). The combined organic layers were washed with brine, dried (Na₂SO₄), filtered, and concentrated in vacuo. The residue was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to afford aldehyde (–)-6 as a colorless oil.

Yield: 440 mg (80%); $[\alpha]_D = 1.8$ (*c* 4.0, CHCl₃).

¹H NMR (400 MHz,CDCl₃): $\delta = 0.94$ (t, 3H, J = 7.2 Hz), 1.54–1.64 (m, 1H), 1.64–1.81 (m, 2H), 1.95–2.05 (m, 1H), 2.22–2.29 (m, 1H), 2.57–2.70 (m, 2H), 7.18–7.32 (m, 5H), 9.62 (d, 1H, J = 2.8 Hz).

¹³C NMR (100 MHz, CDCl₃): δ = 205.1, 141.5, 128.4 (2C), 128.3 (2C), 126, 52.6, 33.2, 30.1, 21.8, 11.3.

GC/MS (EI): *m*/*z* = 176 (M⁺), 158, 115, 105, 104, 91, 77.

HRMS (EI): m/z calcd for $C_{12}H_{16}O$ (M⁺): 176.1201. Found: 176.1201.

(+)-6

Prepared from (-)-5 using above procedure, colorless oil, $[\alpha]_{D} = +1.5$ (*c* 1.2, CHCl₃).

(-)-(R)-2-Ethyl-4-phenyl-1-butyl Bromide [(-)-7]

To a solution of aldehyde (-)-**6** (260 mg, 1.47 mmol) in THF (8 mL), borane-THF complex (1 M, 2.95 mL, 2.95 mmol) was added dropwise at 0 °C. The mixture was stirred at 0 °C under N₂ for 1 h and then quenched with 1 M HCl (5 mL) at 0 °C. The organic layer was separated, the aqueous layer was extracted with EtOAc (2 × 40 mL). The combined organic layers were washed with 3% NaHCO₃ and brine, dried (Na₂SO₄), filtered, and concentrated in vacuo to provide the crude alcohol (260 mg).

¹H NMR (400 MHz, CDCl₃): δ = 0.95 (t, 3H, *J* = 7.2 Hz), 1.61 (br s, 1H), 1.41–1.78 (m, 5H), 2.67 (t, 2H, *J* = 8.0 Hz), 3.62 (d, 2H, *J* = 5.6 Hz), 7.20–7.32 (m, 5H).

¹³C NMR (100 MHz, CDCl₃): δ = 142.7, 128.3 (4C), 125.7, 65.0, 41.5, 33.2, 32.3, 23.2, 11.0.

GC/MS (EI): m/z = 178 (M⁺).

The above crude alcohol (260 mg, 1.46 mmol) and Ph₃P (765 mg, 2.92 mmol) were dissolved in DMF (4 mL). NBS (520 mg, 2.92 mmol) was added to the solution in small portions at r.t over 20 min. The reaction mixture was warmed to 55 °C under N₂ overnight and then cooled to r.t. The mixture was diluted with Et₂O (100 mL), washed with 3% NaHCO₃ and brine, dried (Na₂SO₄), filtered, and concentrated in vacuo. The residue was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to give bromide (–)-7 as a colorless oil.

Yield: 240 mg (68%); $[\alpha]_D$ –10.5 (*c* 2.0, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.93$ (t, 3H, J = 7.2 Hz), 1.45–1.79 (m, 5H), 2.57–2.72 (m, 2H), 3.52 (d, 2H, J = 4.8 Hz), 7.20–7.32 (m, 5H).

¹³C NMR (100 MHz, CDCl₃): δ = 142.1, 128.4 (2C), 128.3 (2C), 125.8, 40.4, 38.7, 34.1, 32.9, 25.1, 10.8.

GC/MS (EI): *m*/*z* = 240 (M⁺), 160, 145, 131, 105, 91, 77.

HRMS (EI): m/z calcd for $C_{12}H_{17}^{-79}Br$ (M⁺): 240.0513. Found: 240.0511.

(+)-7

Prepared from (+)-6 using above procedure, colorless oil, $[\alpha]_D = +10.8 (c \ 1.5, CHCl_3).$

(*R*)-2-Ethyl-4-phenylbutyltriphenylphosphonium Bromide (3a) The mixture of bromide (-)-7 (152 mg, 0.63 mmol) and Ph₃P (165 mg, 0.63 mmol) was heated at 120 °C under Ar for 18 h, and then cooled to r.t. The resulting solid was washed twice with H₂O and dried in vacuum overnight to give phosphonium salt **3a** (207 mg, 65%) as a white solid.

3b

Prepared from (+)-7 using above procedure, white solid.

Alkylation of Hydrazone (-)-4 with tert-Butyl Bromoacetate

To a solution of hydrazone (-)-4 (1.23 g, 6.68 mmol) in anhyd Et₂O (20 mL), LDA (1.5 M, 4.5 mL, 6.75 mmol) was added dropwise at 0 °C over 10 min. The mixture was stirred at 0 °C under Ar for 5 h and then cooled to -110 °C using a liquid N₂/95% EtOH bath. *t*-Bu bromoacetate (1.5 mL, 10 mmol) in Et₂O (20 mL) was added dropwise over 30 min. The mixture was slowly warmed up to 0 °C and quenched with sat. NH₄Cl (20 mL). The organic layer was separated, the aqueous layer was extracted with EtOAc (2 × 50 mL). The combined organic layers were washed with brine, dried (Na₂SO₄), filtered, and concentrated in vacuo. The residue was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to afford hydrazone **9** as a colorless oil.

Yield: 1.2 g (60%); [α]_D -74.3 (*c* 0.7, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.90$ (t, 3H, J = 7.2 Hz), 1.42 (s, 9H), 1.48 (m, 2H), 1.72–1.97 (m, 5H), 2.29 (dd, 1H, J = 14.8, 6.4 Hz), 2.39 (dd, 1H, J = 14.8, 8.0 Hz), 2.60 (m, 1H), 2.69 (dd, 1H, J = 16.8, 8.0 Hz), 3.35 (s, 3H), 3.38 (m, 2H), 3.56 (m, 1H), 6.51 (d, 1H, J = 6.0 Hz).

¹³C NMR (100 MHz, CDCl₃): δ = 172.0, 140.1, 80.0, 74.7, 63.4, 59.1, 50.2, 40.6, 39.1, 28.1 (3C), 26.5, 26.2, 22.0, 11.3.

GC/MS (EI): *m*/*z* = 298 (M⁺), 253, 242, 225, 197, 183, 128, 110, 82, 70, 57.

HRMS (EI): m/z calcd for $C_{16}H_{30}N_2O_3$ (M⁺): 298.2256. Found: 298.2258.

(-)-tert-Butyl (3S)-3-Ethyl-4-oxobutyrate (10)

Ozone was bubbled into a solution of hydrazone **9** (2.6 g, 8.7 mmol) in CH₂Cl₂ (20 mL) at -78 °C until no starting material was detected by TLC. The reaction mixture was bubbled with N₂ while warming up to r.t. The mixture was concentrated in vacuo, and the residue was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to afford aldehyde **10** as a colorless oil.

Yield: 1.33 g (82%); $[\alpha]_D$ –39.4 (*c* 1.5, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.97$ (t, 3H, J = 7.2 Hz), 1.44 (s, 9H), 1.55 (m, 1H), 1.77 (m, 1H), 2.35 (dd, 1H, J = 16.0, 6.0 Hz), 2.62 (dd, 1H, J = 16.0, 8.0 Hz), 2.70 (m, 1H), 9.71 (d, 1H, J = 1.2 Hz).

¹³C NMR (100 MHz, CDCl₃): δ = 203.2, 171.2, 80.9, 49.3, 34.1, 28.0 (3C), 21.6, 11.2.

GC/MS (EI): *m*/*z* 171 (M⁺-Me), 158, 129, 113, 101,85, 73, 57.

(-)-tert-Butyl (3S)-3-Ethyl-4-oxopentanoate (11)

To a solution of aldehyde **10** (633 mg, 3.4 mmol) in Et₂O (10 mL), CH₃MgCl (3.0 M, 1.24 mL, 3.7 mmol) was added dropwise at -78 °C. The mixture was stirred at -78 °C for 30 min and then quenched with sat. NH₄Cl at -78 °C. The organic layer was separated, the aqueous layer was extracted with Et₂O (2 × 20 mL). The combined organic layers were washed with brine, dried (Na₂SO₄) for 5 min, filtered, and concentrated in vacuo. The residue was dissolved in acetone (10 mL), and Jones' reagent was added dropwise at 0 °C within 2 min (a red color was retained). The reaction was quenched at 0 °C with H₂O (20 mL), and extracted with pentane (3 × 40 mL). The

combined organic layers were washed with brine, dried (Na_2SO_4) , filtered, and concentrated in vacuo. The residue was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to afford ketone **11** as colorless oil.

Yield: 611 mg (89%); $[\alpha]_D$ –45.7 (*c* 0.9, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.91$ (t, 3H, J = 7.2 Hz), 1.43 (s, 9H), 1.42–1.53 (m, 1H), 1.58–1.69 (m, 1H), 2.22 (s, 3H), 2.28 (dd, 1H, J = 16.8, 4.8 Hz), 2.65 (dd, 1H, J = 16.8, 9.6 Hz), 2.85–2.92 (m, 1H).

¹³C NMR (100 MHz, CDCl₃): δ = 210.9, 171.7, 80.6, 49.4, 36.1, 29.4, 28.0 (3C), 24.3, 11.3.

GC/MS (EI): *m*/*z* = 200 (M⁺), 185, 172, 144, 127, 116, 109, 98, 85, 73, 71, 57.

HRMS (EI): m/z calcd for $C_{11}H_{20}O_3$ (M⁺): 200.1412. Found: 200.1411.

(+)-tert-Butyl (3S)-3-Ethyl-4,4'-dioxoethylenepentanoate (12)

A mixture of ketone **11** (610 mg, 3.05 mmol), trimethyl orthoformate (1.62 mL, 12.2 mmol), *p*-TsOH (6 mg, 0.03 mmol) in ethylene glycol (4 mL) was stirred at r.t. under Ar overnight. The mixture was diluted with sat. NaHCO₃, extracted with pentane (3×40 mL). The combined organic layers were washed with brine, dried (Na₂SO₄), filtered, and concentrated in vacuo. The residue was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to afford ketal **12** as a colorless oil.

Yield: 616 mg (82%); $[\alpha]_D$ +11.4 (*c* 2.2, CHCl₃).

¹H NMR (400 MHz, CDCl₃): δ = 0.94 (t, 3H, *J* = 7.2 Hz), 1.21 (m, 1H), 1.25 (s, 3H), 1.44 (s, 9H), 1.65 (m, 1H), 2.09 (m, 1H), 2.10 (dd, 1H, *J* = 16.8, 6.4 Hz), 2.31 (dd, 1H, 16.8, 8.4 Hz), 3.91 (m, 4H).

¹³C NMR (100 MHz, CDCl₃): δ = 173.0, 111.8, 79.8, 64.5, 64.4, 44.9, 36.1, 28.0 (3C), 23.5, 20.8, 12.3.

MS (CI): m/z = 245 (MH⁺).

HRMS (CI): m/z calcd for $C_{13}H_{25}O_4$ (MH⁺): 245.1753. Found: 245.1754.

(-)-(S)-3-Ethyl-4,4'-dioxoethylenepentanal (8)

To a solution of ester **12** (616 mg, 2.52 mmol) in H_2O (20 mL), LiAl H_4 (190 mg, 5.0 mmol) was added in portions at r.t. The mixture was stirred at r.t. for 2 h and then treated with H_2O (0.2 mL), 15% NaOH (0.2 mL) and H_2O (0.6 mL). The mixture was filtered and concentrated in vacuo to provide the crude alcohol (341 mg).

GC/MS (EI): m/z = 174 (M⁺).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.94$ (t, 3H, J = 7.6 Hz), 1.12–1.22 (m, 1H), 1.28 (s, 3H), 1.53–1.58 (m, 1H), 1.63–1.72 (m, 4H), 3.58–3.76 (m, 2H), 3.92–3.99 (m, 4H).

¹³C NMR (100 MHz, CDCl₃): δ = 112.5, 64.4, 64.2, 62.2, 46.6, 32.6, 23.8, 20.2, 12.5.

The crude alcohol (341 mg, 1.96 mmol) was dissolved in CH_2Cl_2 (10 mL), and PDC (1.47 g, 3.92 mmol) was added. The mixture was stirred at r.t. under Ar for 24 h and then filtered through a pad of silica gel. The filtrate was concentrated in vacuo, and the residue was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to afford aldehyde **8** as a colorless oil.

Yield: 275 mg (62%); [α]_D -8.4 (*c* 1.5, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.94$ (t, 3H, J = 7.6 Hz), 1.25 (m, 1H), 1.26 (s, 3H), 1.69 (m, 1H), 2.20 (m, 1H), 2.25 (ddd, 1H, J = 16.0, 4.0, 1.2 Hz), 2.37 (ddd, 1H, 16.0, 8.4, 4.0 Hz), 3.88 (m, 4H), 9.63 (dd, 1H, J = 4.0, 1.2 Hz).

¹³C NMR (100 MHz, CDCl₃): δ = 201.9, 111.4, 64.0, 44.4, 43.5, 23.2, 21.1, 12.3.

GC/MS (EI): *m*/*z* = 172 (M⁺), 157, 144, 130, 113, 99, 87, 71, 55.

HRMS (CI): m/z calcd for $C_9H_{17}O_3$ (MH⁺): 173.1178. Found: 173.1177.

(3*S*,7*R*)-3,7-Diethyl-9-phenylnon-5-en-2-one Ethylene Ketal (13)

To a suspension of phosphonium salt **3a** (50 mg, 0.1 mmol) in THF (2 mL), *n*-BuLi (1.5 M, 0.073 mL, 0.11 mmol) was added dropwise at r.t. The mixture was stirred at r.t. under Ar until the solid dissolved and a red solution appeared. Aldehyde **8** (19 mg, 0.11 mmol) in THF (1 mL) was added at r.t., and the mixture was stirred overnight at r.t. under Ar. The mixture was filtered through a pad of silica gel and the white precipitate was washed with Et₂O (20 mL). The filtrate was concentrated in vacuo, and the residue was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to give alkeen **13** as a colorless oil.

Yield: 21 mg (66%).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.87$ (t, 3H, J = 7.2 Hz), 0.96 (t, 3H, J = 7.2 Hz), 1.25 (m, 1H), 1.27 (s, 3H), 1.33 (m, 1H), 1.46 (m, 1H), 1.56 (m, 2H), 1.65 (m, 1H), 1.73 (m, 1H), 2.05 (m, 1H), 2.27 (m, 1H), 2.33 (m, 1H), 2.51 (ddd, 1H, J = 13.6, 11.2, 5.2 Hz), 2.64 (ddd, 1H, J = 13.6, 11.2, 5.2 Hz), 3.93 (m, 4H), 5.15 (m, 1H, J = 10.8, 1.0 Hz), 5.55 (m, 1H, J = 10.8, 7.2 Hz), 7.14–7.29 (m, 5H).

¹³C NMR (100 MHz, CDCl₃): δ = 143.1, 134.6, 129.7, 128.3 (2C), 128.2 (2C), 125.5, 112.6, 64.4 (2C), 48.4, 38.8, 37.5, 33.8, 28.6, 28.2, 22.9, 20.9, 13.0, 11.8.

MS (CI): m/z = 317 (MH⁺).

HRMS (CI): m/z calcd for $C_{21}H_{33}O_2$ (MH⁺): 317.2483. Found: 317.2481.

Dianeackerone (1a)

The *cis/trans* mixture of alkene **13** (10 mg, 0.031 mmol) and 5% Pd/ C (5 mg) in MeOH (1 mL) were stirred overnight at r.t. under H₂ (1 atm). The mixture was filtered through a pad of silica gel and the solid was washed with MeOH (20 mL). The filtrate was concentrated in vacuo to afford a crude product as a light yellow oil (8 mg).

 $[\alpha]_{\rm D}$ –10 (*c* 0.5, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.87$ (t, 3H, J = 7.2 Hz), 0.96 (t, 3H, J = 7.2 Hz), 1.26 (s, 3H), 1.20–1.42 (m, 10H), 1.52–1.62 (m, 4H), 2.59 (t, 2H, J = 8.8, 7.6 Hz), 3.90–3.95 (m, 4H), 7.17–7.30 (m, 5H).

¹³C NMR (100 MHz, CDCl₃): δ = 143.3, 128.3 (2C), 128.2 (2C), 125.5, 112.9, 64.4, 64.3, 47.9, 38.5, 35.3, 33.5, 33.1, 30.5, 25.7, 25.6, 23.3, 20.7, 13.0, 10.7.

MS (CI): m/z = 319 (MH⁺).

HRMS (CI): m/z calcd for $C_{21}H_{35}O_2$ (MH⁺): 319.2637. Found: 319.2633.

The above crude product was dissolved in acetone and H_2O (9:1, 4 mL), and PPTS (6 mg, 0.025 mmol) was added. The mixture was refluxed for 1 h under N_2 and then cooled to r.t., diluted with Et_2O (20 mL), washed with brine, dried (Na_2SO_4), filtered, and concentrated in vacuo. The residue was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to afford dianeackerone **1a** as a colorless oil.

Yield: 6 mg (70%); $[\alpha]_D$ +3.2 (*c* 0.2, CHCl₃).

¹H NMR (400 MHz, C₆D₆): δ = 0.75 (t, 3H, *J* = 7.2 Hz), 0.83 (t, 3H, *J* = 7.2 Hz), 1.16–1.52 (m, 13H), 1.76 (s, 3H), 2.14 (m, 1H), 2.55 (m, 2H), 7.15–7.25 (m, 5H).

¹³C NMR (100 MHz, C_6D_6): δ = 210.34, 143.62, 129.06 (2C), 129.04 (2C), 126.36, 54.94, 38.90, 35.81, 33.84, 33.83, 32.15, 28.85, 26.36, 25.16, 25.12, 12.23, 11.27.

GC/MS (EI): *m*/*z* = 274 (M⁺), 227, 186, 172, 157, 145, 123, 105, 99, 91, 86, 71, 55.

HRMS (EI): m/z calcd for $C_{19}H_{30}O$ (M⁺): 274.2297. Found: 274.2298.

(35,75)-3,7-Diethyl-9-phenylnon-5-en-2-one Ethylene Ketal (14)

The phosphonium salt **3b** (62 mg, 0.12 mmol) in THF (2 mL) was converted to the ylide with *n*-BuLi (1.5 M, 0.088 mL, 0.13 mmol) by a procedure similar to that described for the ketal **13**. The ylide was coupled to aldehyde **8** (23 mg, 0.13 mmol) in THF (1 mL), and the product was purified to give alkene **14** as a colorless oil.

Yield: 34 mg (82%).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.86$ (t, 3H, J = 7.2 Hz), 0.97 (t, 3H, J = 7.2 Hz), 1.24 (m, 1H), 1.27 (s, 3H), 1.35 (m, 1H), 1.42– 1.60 (m, 4H), 1.73 (m, 1H), 2.05 (m, 1H), 2.26 (m, 1H), 2.32 (m, 1H), 2.52 (ddd, 1H, J = 13.6, 11.2, 5.6 Hz), 2.64 (ddd, 1H, J = 13.6, 11.2, 5.6 Hz), 3.94 (m, 4H), 5.15 (m, 1H), 5.55 (ddd, 1H, J = 10.8, 7.2, 4.0 Hz), 7.14–7.29 (m, 5H).

¹³C NMR (100 MHz, CDCl₃): δ = 143.1, 134.7, 129.7, 128.3 (2C), 128.2 (2C), 125.5, 112.6, 64.4 (2C), 48.5, 38.8, 37.7, 33.9, 28.6, 28.2, 22.9, 20.9, 13.1, 11.8.

MS (CI): m/z = 317 (MH⁺).

HRMS (CI): m/z calcd for $C_{21}H_{33}O_2$ (MH⁺): 317.2483. Found: 317.2482.

Dianeackerone (1b)

The *cis/trans* mixture of alkene **14** (12 mg, 0.038 mmol) was hydrogenated by a procedure similar to that described for dianeackerone **1a** to afford a light yellow oil (11 mg).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.88$ (t, 3H, J = 7.2 Hz), 0.96 (t, 3H, J = 7.2 Hz), 1.26 (s, 3H), 1.20–1.42 (m, 10H), 1.52–1.62 (m, 4H), 2.59 (t, 2H, J = 8.4, 7.6 Hz), 3.90–3.95 (m, 4H), 7.17–7.30 (m, 5H).

¹³C NMR (100 MHz, CDCl₃): δ = 143.3, 128.3 (2C), 128.2 (2C), 125.5, 112.9, 64.4, 64.3, 47.9, 38.5, 35.2, 33.5, 33.2, 30.5, 25.8, 25.6, 23.3, 20.7, 13.0, 10.8.

MS (CI): m/z = 319 (MH⁺).

The above crude product was deprotected with PPTS (4 mg, 0.016 mmol) and purified by procedures similar to those described for dianeackerone **1a** to afford dianeackerone, **1b** as a colorless oil.

Yield: 8 mg (78%); $[\alpha]_D = 8.7$ (*c* 0.2, CHCl₃).

¹H NMR (400 MHz, C_6D_6): $\delta = 0.75$ (t, 3H, J = 7.2 Hz), 0.83 (t, 3H, J = 7.2 Hz), 1.16–1.52 (m, 13H), 1.76 (s, 3H), 2.14 (m, 1H), 2.55 (m, 2H), 7.15–7.25 (m, 5H).

¹³C NMR (100 MHz, C_6D_6): δ = 210.34, 143.62, 129.06 (2C), 129.04 (2C), 126.36, 54.94, 38.89, 35.89, 33.88, 33.83, 32.16, 28.84, 26.27, 25.16, 25.10, 12.23, 11.22.

GC/MS (EI): *m*/*z* = 274 (M⁺), 227, 186, 172, 157, 145, 123, 105, 99, 91, 86, 71, 55.

HRMS (EI): m/z calcd for $C_{19}H_{30}O$ (M⁺): 274.2297. Found: 274.2297.

Cholesterol Diazoacetate (16)

To a solution of acid chloride **17** (1.39 g, 5.0 mmol) and cholesterol (1.93 g, 5.0 mmol) in CH_2Cl_2 (25 mL), Et_3N (1.4 mL, 10 mmol) was added dropwise at 0 °C over 20 min. The mixture was stirred at 0

°C under Ar for 3 h and the solvent was removed in vacuo. The residue was dissolved in benzene (30 mL) and mixed with Florisil (15 g). The mixture was filtered and washed with benzene (30 mL). The filtrate was concentrated in vacuo at 30 °C and the residue was purified by chromatography on Florisil (Et₂O/pentane, 1:4) to give the diazoacetate **16** as a yellow solid.

Yield: 1.5 g (66%); $[\alpha]_D$ –23.6 (*c* 1.8, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.68$ (s, 3H), 0.86 (d, 3H, J = 6.4 Hz), 0.87 (d, 3H, J = 6.4 Hz), 0.92 (d, 3H, J = 6.4 Hz), 1.02 (s, 3H), 1.11–1.65 (m, 20H), 1.79–2.03 (m, 6H), 2.29–2.39 (m, 2H), 4.69 (m, 1H), 4.71 (s, 1H), 5.39 (d, 1H, J = 5.2 Hz).

¹³C NMR (100 MHz, CDCl₃): δ = 166.4, 139.6, 122.8, 74.6, 56.7, 56.2, 50.0, 42.3, 39.7, 39.5, 38.3, 36.9, 36.6, 36.2, 35.8, 31.9, 31.8, 28.2, 28.0 (2C), 24.3, 23.8, 22.8, 22.5, 21.0, 19.3, 18.7, 11.8.

Alkylation of Hydrazone (-)-4 with 2-Iodoethanol *tert*-Butyldimethylsilyl Ether

To a solution of hydrazone (-)-4 (3.0 g, 15.3 mmol) in anhyd Et₂O (40 mL), LDA (1.5 M, 11 mL, 16.5 mmol) was added dropwise at 0 °C over 10 min. The mixture was stirred at 0 °C under Ar for 5 h and cooled to -78 °C. 2-Iodoethanol *tert*-butyldimethylsilyl ether (5.5 g, 19.2 mmol) in Et₂O (20 mL) was added dropwise at -78 °C over 30 min. The reaction mixture was slowly warmed up to 0 °C and quenched with sat. NH₄Cl (30 mL). The organic layer was separated, the aqueous layer was extracted with EtOAc (2 × 50 mL). The combined organic layers were washed with brine, dried (Na₂SO₄), filtered, and concentrated in vacuo. The residue was purified by chromatography on silica gel (hexanes/EtOAc/Et₃N, 95:5:2) to afford hydrazone **18** as a colorless oil

Yield: 3.65 g (65%) [α]_D -47.9 (*c* 1.9, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.03$ (s, 6H), 0.88 (s, 9H), 0.89 (t, 3H, J = 7.2 Hz), 1.44 (m, 2H), 1.65 (m, 2H), 1.76–1.95 (m, 4H), 2.20 (m, 1H), 2.69 (dd, 1H, J = 16.4, 8.0 Hz), 3.36 (s, 3H), 3.31–3.43 (m, 3H), 3.54–3.67 (m, 3H), 6.46 (d, 1H, J = 6.8 Hz).

¹³C NMR (100 MHz, CDCl₃): δ = 142.7, 74.7, 63.5, 61.5, 59.1, 50.4, 40.7, 36.1, 26.8, 26.5, 25.9 (3C), 22.0, 18.3, 11.4, -5.2 (2C).

GC/MS (EI): *m*/*z* = 342 (M⁺), 327, 297, 285, 253, 228, 184, 170, 139, 120, 96, 89, 73, 70.

HRMS (ESI): m/z calcd for $C_{18}H_{39}N_2O_2Si$ (MH⁺): 343.2780. Found: 343.2764.

(-)-(S)-2-Ethyl-4-*tert*-butyldimethylsiloxybutan-1-ol (19)

Ozone was bubbled into a solution of hydrazone **18** (910 mg, 2.66 mmol) in CH₂Cl₂(10 mL) at -78 °C until no starting material was detected by TLC. The reaction mixture was purged with N₂ while warming up to r.t. and then concentrated in vacuo. The residue was dissolved in Et₂O (10 mL), and borane-THF complex (1.0 M, 6 mL, 6 mmol) was added dropwise at 0 °C over 10 min. The mixture was stirred at 0 °C under Ar for 1 h and quenched with H₂O at 0 °C. The organic layer was separated and the aqueous layer was extracted with Et₂O (2 × 20 mL). The combined organic layers were washed with brine, dried (Na₂SO₄), filtered, and concentrated in vacuo. The residue was purified by chromatography on silica gel (hexanes/EtOAc, 5:1) to afford the alcohol **19** as a colorless oil.

Yield: 420 mg, (67%); $[\alpha]_D$ –6.8 (*c* 2.6, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.08$ (s, 3H), 0.09 (s, 3H), 0.91 (s, 9H), 0.91 (t, 3H, J = 7.2 Hz), 1.28 (m, 1H, J = 7.2 Hz), 1.37 (m, 1H, J = 7.2 Hz), 1.53 (m, 2H), 1.68 (m, 1H), 3.24 (br s, 1H), 3.46 (dd, 1H, J = 10.8, 6.4 Hz), 3.60 (dd, 1H, J = 10.8, 3.6 Hz), 3.65 (m, 1H), 3.77 (m, 1H).

¹³C NMR (100 MHz, CDCl₃): δ = 65.8, 61.9, 41.3, 35.2, 25.8 (3C), 24.4, 18.2, 11.5, -5.5, -5.5.

GC/MS (EI): *m*/*z* = 232 (M⁺), 202, 191, 175, 157, 132, 105, 101, 83, 75.

(-)-(S)-4-Benzyloxy-3-ethylbutan-1-ol (20)

To a solution of alcohol **19** (380 mg, 1.63 mmol) in THF (10 mL), NaH (98 mg, 60% in oil, 2.45 mmol) was added at r.t. The mixture was stirred at r.t. under Ar for 30 min, and benzyl bromide (0.29 mL, 2.45 mmol) was added dropwise. The mixture was stirred at r.t. under Ar overnight and then cooled to 0 °C. H₂O (2 mL) was added dropwise at 0 °C, followed by the addition of 3 M HCl (5 mL, 15 mmol). The mixture was stirred at r.t. for 1 h. The organic layer was separated and the aqueous layer was extracted with EtOAc (2 × 20 mL). The combined organic layers were washed with 5% NaHCO₃ and brine, dried (Na₂SO₄), filtered, and concentrated in vacuo. The residue was purified by chromatography on silica gel (hexanes/ EtOAc, 4:1) to afford alcohol **20** as a colorless oil.

Yield: 290 mg (85%); $[\alpha]_D$ –9.0 (*c* 2.3, CHCl₃).

¹H NMR (400 MHz, $CDCl_3$): $\delta = 0.90$ (t, 3H, J = 7.2 Hz), 1.36 (m, 2H, J = 7.2 Hz), 1.58 (m, 1H), 1.70 (m, 2H), 2.72 (br s, 1H), 3.36 (dd, 1H, J = 9.2, 7.2 Hz), 3.48 (dd, 1H, J = 9.2, 4.0 Hz), 3.63 (m, 1H), 3.70 (m, 1H), 4.53 (t, 2H, J = 13.2 Hz), 7.27–7.38 (m, 5H).

¹³C NMR (100 MHz, CDCl₃): δ = 137.9, 128.4 (2C), 127.7 (2C), 127.6, 73.9, 73.3, 61.2, 38.1, 35.8, 24.8, 11.4.

GC/MS (EI): *m*/*z* = 208 (M⁺), 190, 178, 161, 137, 117, 107, 99, 91, 79, 69.

HRMS (ESI): m/z calcd for $C_{13}H_{21}O_2$ (MH⁺): 209.1541. Found: 209.1521.

(+)-(S)-1-Benzyloxy-4-bromo-2-ethylbutane (21)

To a solution of alcohol **20** (230 mg, 1.1 mmol), Ph_3P (576 mg, 2.2 mmol) in DMF (4 mL), NBS (390 mg, 2.2 mmol) was added in portions at r.t. The reaction mixture was stirred at 50 °C under Ar for 2 h and cooled to r.t. The mixture was diluted with Et₂O (40 mL), washed with 5% NaHCO₃ and brine, dried (Na₂SO₄), filtered, and concentrated in vacuo. The residue was purified by chromatography on silica gel (hexanes/EtOAc, 9:1) to afford bromide **21** as a colorless oil.

Yield: 238 mg (79%); $[\alpha]_D$ +8.4 (*c* 1.0, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.91$ (t, 3H, J = 7.2 Hz), 1.32– 1.49 (m, 2H), 1.76 (m, 1H), 1.91 (m, 1H), 1.98 (m, 1H), 3.37 (dd, 1H, J = 9.2, 5.6 Hz), 3.43 (dd, 1H, J = 9.2, 4.8 Hz), 3.47 (m, 2H), 4.50 (s, 2H), 7.29–7.38 (m, 5H).

¹³C NMR (100 MHz, CDCl₃): δ = 138.5, 128.3 (2C), 127.5 (2C), 127.4, 73.1, 72.1, 38.8, 35.1, 32.2, 23.8, 11.2.

GC/MS (EI): *m*/*z* = 270 (M⁺), 242, 223, 311, 190, 172, 163, 149, 121, 107, 92, 91, 69, 65.

HRMS (EI): m/z calcd for $C_{13}H_{19}^{-79}BrO (M^+)$: 270.0619. Found: 270.0619.

(S)-3-Ethyl-4-benzyloxybutyltriphenylphosphonium Bromide (22)

The mixture of bromide **21** (215 mg, 0.793 mmol) and Ph_3P (208 mg, 0.793 mmol) was heated at 120 °C under Ar for 2 h and then cooled to r.t. The resulting solid was washed with Et₂O, and dried in vacuum at 60 °C for 8 h to provide the phosphonium salt **22** (355 mg, 84%) as a white solid.

(2S,6R)-1-Benzyloxy-2,6-diethyl-8-phenyloct-4-ene (23)

The phosphonium salt **22** (266 mg, 0.5 mmol) in THF (5 mL) was converted to the ylide with BuLi (1.5 M, 0.37 mL, 0.55 mmol) by a procedure similar to that described for the ketal **13**. Aldehyde (-)-**6** (105 mg, 0.6 mmol) in THF (2 mL) was coupled to the ylide and

the product was purified by chromatography on silica gel (EtOAc/ hexanes, 1:14) to give alkene **23** as a colorless oil.

Yield: 115 mg (66%).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.86$ (t, 3H, J = 7.2 Hz), 0.89 (t, 3H, J = 7.2 Hz), 1.23 (m, 1H), 1.34–1.52 (m, 4H), 1.61–1.76 (m, 2H), 2.09 (m, 2H), 2.30 (m, 1H), 2.49 (ddd, 1H, J = 13.6, 10.4, 5.6 Hz), 2.63 (ddd, 1H, J = 13.6, 10.4, 5.6 Hz), 3.38 (m, 2H), 4.50 (s, 2H), 5.18 (tt, 1H, J = 10.8, 1.6 Hz), 5.48 (dt, 1H, J = 10.8, 7.6 Hz), 7.15–7.34 (m, 10H).

¹³C NMR (100 MHz, CDCl₃): δ = 143.1, 138.8, 135.5, 128.5, 128.3 (2C), 128.3 (2C), 128.2 (2C), 127.5 (2C), 127.4, 125.5, 73.0, 72.9, 40.6, 38.8, 37.6, 33.9, 29.2, 28.6, 23.8, 11.8, 11.3.

GC/MS (EI): *m*/*z* = 350 (M⁺), 332, 304, 282, 259, 242, 213, 199, 171, 145, 117, 104, 91, 67.

HRMS (EI): m/z calcd for $C_{25}H_{34}O$ (M⁺): 350.2610. Found: 350.2609.

(+)-(25,65)-2,6-Diethyl-8-phenyloctan-1-ol (24)

The *cis/trans* mixture of alkene **23** (92 mg, 0.26 mmol) was hydrogenated by a procedure similar to that described for dianeackerone **1a** and the product was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to give the alcohol **24** as a colorless oil.

Yield: 58 mg (84%); $[\alpha]_D$ +2.7 (*c* 1.6, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.87$ (t, 3H, J = 7.2 Hz), 0.91 (t, 3H, J = 7.2 Hz), 1.22–1.43 (m, 12H), 1.54–1.59 (m, 2H), 1.61 (br s, 1H), 2.58 (dd, 2H, J = 10.4, 7.6 Hz), 3.55 (d, 2H, J = 5.2 Hz), 7.15–7.30 (m, 5H).

¹³C NMR (100 MHz, CDCl₃): δ = 143.2, 128.3 (2C), 128.2 (2C), 125.5, 65.3, 41.9, 38.5, 35.2, 33.4, 33.2, 30.8, 25.7, 23.9, 23.3, 11.1, 10.7.

GC/MS (EI): *m*/*z* = 262 (M⁺), 244, 225, 215, 180, 174, 160, 145, 131, 117, 105, 104, 92, 91, 83, 69, 55.

HRMS (EI): m/z calcd for $C_{18}H_{30}O$ (M⁺): 262.2297. Found: 262.2296.

(+)-(2*S*,6*S*)-2,6-Diethyl-8-phenyloctanal (15)

To a mixture of alcohol **24** (20 mg, 0.076 mmol), NMO (13 mg, 0.11 mmol), 4 Å molecular sieves (30 mg) in $CH_2Cl_2(1 \text{ mL})$, TPAP (2 mg, 0.005 mmol) was added at r.t. The mixture was stirred at r.t. under Ar for 30 min, then filtered through a pad of silica gel, and washed with CH_2Cl_2 . The filtrate was concentrated in vacuo, and the residue was purified by chromatography on silica gel (EtOAc/hexanes, 1:9) to give the aldehyde **15** as a colorless oil.

Yield: 17 mg (85%); $[\alpha]_D$ +5.9 (*c* 0.9, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.87$ (t, 3H, J = 7.2 Hz), 0.92 (t, 3H, J = 7.2 Hz), 1.27–1.72 (m, 13H), 2.18 (m, 1H), 2.58 (dd, 2H, J = 9.6, 5.6 Hz), 7.18–7.30 (m, 5H), 9.58 (d, 1H, J = 3.2 Hz).

¹³C NMR (100 MHz, CDCl₃): δ = 205.6, 143.1, 128.3 (2C), 128.2 (2C), 125.5, 53.4, 38.4, 35.1, 33.1, 28.9, 25.5, 24.1, 21.9, 11.5, 10.7.

GC/MS (EI): *m*/*z* = 260 (M⁺), 242, 231, 213, 185, 171, 145, 138, 117, 105, 104, 92, 91, 65, 55.

HRMS (EI): m/z calcd for $C_{18}H_{28}O$ (M⁺): 260.2140. Found: 260.2140.

(-)-Cholesteryl (4*S*,8*S*)-4,8-diethyl-3-oxo-10-phenyldecanoate (2)

To a solution of aldehyde **15** (5 mg, 0.019 mmol) and diazoacetate **16** (10 mg, 0.022 mmol) in $CH_2Cl_2(1 \text{ mL})$, $SnCl_2$ (5 mg, 0.026 mmol) was added in portions at r.t. The mixture was stirred overnight at r.t. under Ar, then diluted with CH_2Cl_2 (10 mL), washed with 0.1 M HCl and brine, dried (Na₂SO₄), filtered and concentrated

in vacuo. The residue was purified by chromatography on silica gel (hexanes/EtOAc, 9:1) to afford β -oxo ester **2** as a colorless oil.

Yield: 8 mg (61%); [α]_D -11.0 (*c* 0.2, CHCl₃).

¹H NMR (400 MHz, CDCl₃): $\delta = 0.68$ (s, 3H), 0.86 (t, 3H, J = 8.0 Hz), 0.87 (d, 3H, J = 6.4 Hz), 0.88 (d, 3H, J = 6.4 Hz), 0.88 (t, 3H, J = 8.0 Hz), 0.92 (d, 3H, J = 6.4 Hz), 1.00 (s, 3H), 1.06–1.68 (m, 35H), 1.81–1.91 (m, 2H), 1.94–2.03 (m, 2H), 2.35 (m, 2H), 2.52 (m, 1H), 2.57 (dd, 2H, J = 9.2, 6.8 Hz), 3.42 (s, 2H), 4.68 (m, 1H), 5.38 (d, 1H, J = 4.8 Hz), 7.17–7.30 (m, 5H).

¹³C NMR (100 MHz, CDCl₃): δ = 206.4, 166.6, 143.1, 139.4, 128.3 (2C), 128.2 (2C), 125.5, 122.8, 75.0, 56.7, 56.1, 54.0, 50.0, 48.7, 42.3, 39.7, 39.5, 38.4, 38.0, 36.9, 36.6, 36.2, 35.8, 35.1, 33.2, 33.1, 33.1, 31.9, 31.8, 31.1, 28.0, 27.7, 25.6, 24.4, 24.3, 24.1, 23.8, 22.8, 22.6, 21.0, 19.3, 18.7, 11.9, 11.6, 10.7.

MS (ESI): m/z = 687 (MH⁺).

HRMS (CI): m/z calcd for $C_{47}H_{75}O_3$ (MH⁺): 687.5716. Found: 687.5718.

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