

# A New Facile Synthesis of 4-Oxo-1,4-dihydrocinnolines

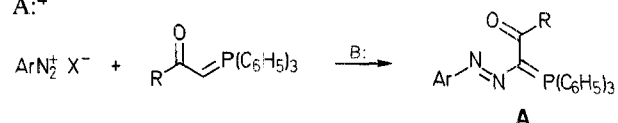
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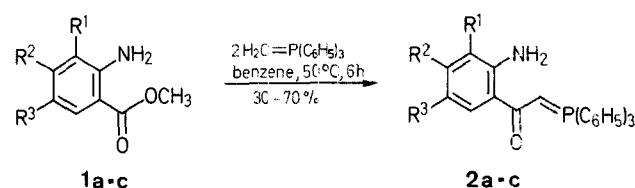
2-(2-Aminoaryl)-2-oxo-triphenylphosphoniummethylenes **2a–d**, readily available by standard procedures, are converted into 4-oxo-1,4-dihydrocinnolines **7a–d** by treatment with nitrous acid and subsequent basic hydrolysis of the intermediate phosphoranes **6a–d**.

Several methods for the synthesis of 4-oxo-1,4-dihydrocinnolines derivatives are reported in the literature. Those involving intramolecular cyclization of *ortho*-acyl and *ortho*-alkynyl substituted aryl diazonium ions are the usual preparative methods for these classes of compounds.<sup>2</sup> However the above syntheses depend on the availability of such starting materials as *o*-aminophenylpropionic acids or *o*-aminoacetophenones, few of which are readily available.

We became engaged in the synthesis of 4-oxo-1,4-dihydrocinnolines due to the pharmacological importance of some of them<sup>3</sup> and thus we needed a method to synthesize the 4-oxo-1,4-dihydrocinnoline ring system starting from easily available raw materials. In connection with our studies on arylazomethylene-triphenylphosphoranes we were aware that the coupling of diazonium ions with stabilized phosphoranes leads to the formation of a carbon-nitrogen bond and results in compounds A:<sup>4</sup>



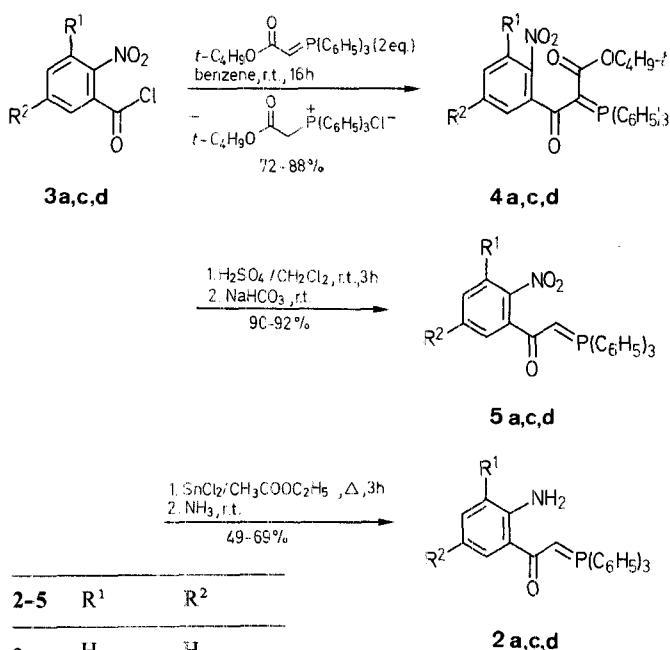
We report here that such a reaction, when applied intramolecularly to arenediazonium ions arising from appropriately built amines **2a–d**, conveniently affords the intermediates **6a–d**; which on hydrolysis give 4-oxo-1,4-dihydrocinnolines derivatives **7a–d** (Scheme C). Compounds **2a–d** were prepared in good yield by two different routes (Schemes A and B).



1-2	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>
a	H	H	H
b	H	–OCH <sub>2</sub> O–	H
c	CH <sub>3</sub>	H	H

Scheme A

As outlined in Scheme A, 2-amino-benzoic acid methyl esters **1a–c** were reacted with the “*in situ*” generated methylenetriphenylphosphorane to give **2a–c** (Table 1). The acylation of methylenetriphenylphosphorane using acylchlorides is a well known procedure<sup>5</sup> while the reaction using esters seems to be less common.<sup>6</sup> Such a method, which appears to have synthetic potential, is particularly useful when amino groups are present, as described here. Alternatively (Scheme B) the reaction of *o*-nitroarylloyl chlorides **3a, c, d** with *t*-butoxycarbonylmethylenetriphenylphosphorane afforded compounds **4a, c, d** (Table 2); acidic hydrolysis and decarboxylation of the *t*-butoxycarbonyl group gave compounds **5a, c, d** (Table 2). Reduction of the nitro

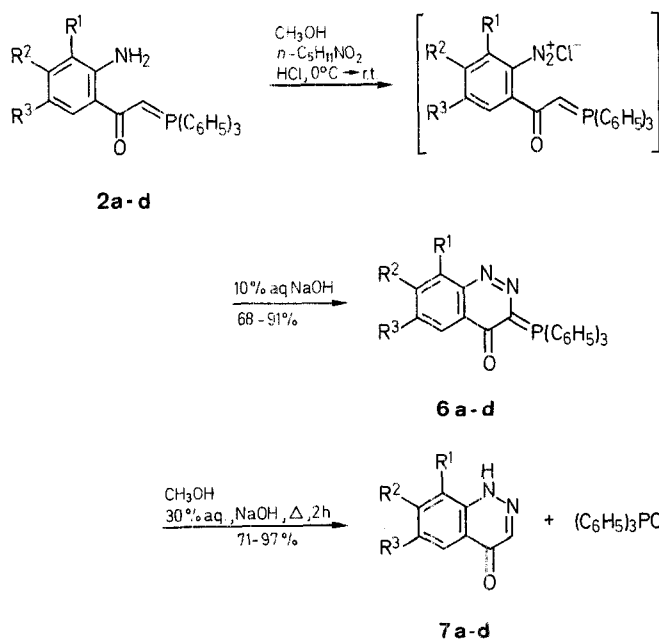


2-5	R <sup>1</sup>	R <sup>2</sup>
a	H	H
c	CH <sub>3</sub>	H
d	H	Cl

Scheme B

substituent with SnCl<sub>2</sub> led to compounds **2a, c, d** (Table 1). Overall yields of reactions reported in Schemes A and B are comparable, however *t*-butoxycarbonylmethylenetriphenylphosphorane is easier to handle and store than methylenetriphenylphosphorane.

2-(2-Aminoaryl)-2-oxo-triphenylphosphoniummethylenes **2a–d** thus prepared were very easily transformed into compounds **6a–d** (Table 3) by a spontaneous intramolecular coupling of the corresponding intermediates, the diazonium ions (Scheme C).



6-7	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>
a	H	H	H
b	H	–OCH <sub>2</sub> O–	H
c	CH <sub>3</sub>	H	H
d	H	H	Cl

Scheme C

**Table 1.** 2-(2-Aminoaryl)-2-oxo-triphenylphosphoniummethyldes **2a-d** Prepared

Product	Method	Yield (%)	m.p. (°C) <sup>a</sup> (solvent)	Molecular Formula <sup>b</sup>	<sup>1</sup> H-NMR (CDCl <sub>3</sub> /TMS) <sup>c</sup> δ (ppm)
<b>2a</b>	A	70	161–162 (C <sub>2</sub> H <sub>5</sub> OH)	C <sub>26</sub> H <sub>22</sub> NOP (395.4)	4.15 (d, 1H, –CH=P, <i>J</i> <sub>HP</sub> = 24 Hz); 5.6 (br s, 2H, –NH <sub>2</sub> ); 6.3–6.6 (m, 2H, H-3, H-5); 6.8–7.1 (m, 1H, H-4); 7.2–7.8 (m, 15H <sub>arom</sub> + H-6)
	B	69			
<b>2b</b>	A	60	143–145 (C <sub>2</sub> H <sub>5</sub> OH)	C <sub>27</sub> H <sub>22</sub> NO <sub>3</sub> P (437.4)	4.1 (br d, 1H, –CH=P, <i>J</i> <sub>HP</sub> = 24 Hz); 5.7 (br s, 2H, –NH <sub>2</sub> ); 5.8 (s, 2H, –O–CH <sub>2</sub> –O–); 6.1 (s, 1H, H-3); 7.1–7.8 (m, 15H <sub>arom</sub> + H-6)
<b>2c</b>	A	165	182–183 (C <sub>2</sub> H <sub>5</sub> OH)	C <sub>27</sub> H <sub>24</sub> NOP (407.4)	2.1 (s, 3H, CH <sub>3</sub> -3); 4.3 (br d, 1H, –CH=P, <i>J</i> <sub>HP</sub> = 24 Hz); 5.8 (br s, 2H, –NH <sub>2</sub> ); 6.5 (dd, 1H, H-5, <i>J</i> <sub>5,6</sub> = 9 Hz, <i>J</i> <sub>5,4</sub> = 8 Hz); 7.05 (dd, 1H, H-4, <i>J</i> <sub>4,5</sub> = 8 Hz, <i>J</i> <sub>4,6</sub> = 1 Hz); 7.3–8.0 (m, 15H <sub>arom</sub> + H-6)
	B	49			
<b>2d</b>	B	67	134–135 ( <i>i</i> -C <sub>3</sub> H <sub>7</sub> OH)	C <sub>26</sub> H <sub>21</sub> ClNOP (427.8)	4.2 (br d, 1H, –CH=P, <i>J</i> <sub>HP</sub> = 24 Hz); 5.6 (br s, 2H, –NH <sub>2</sub> ); 6.45 (d, 1H, H-3, <i>J</i> <sub>3,4</sub> = 9 Hz); 7.0 (dd, 1H, H-4, <i>J</i> <sub>4,3</sub> = 9 Hz, <i>J</i> <sub>4,6</sub> = 2.5 Hz); 7.1–8.0 (m, 15H <sub>arom</sub> + H-6)

<sup>a</sup> Melting points are uncorrected.<sup>b</sup> Satisfactory microanalyses obtained: C ± 0.4, H ± 0.2, N ± 0.3.<sup>c</sup> Recorded on a Varian EM 390 spectrometer.**Table 2.** 2-(2-Nitroaryl)-2-oxo-triphenylphosphoniumalkylides **4a, c, d** and **5a, c, d** Prepared

Prod-uct	Yield (%)	m.p. (°C) <sup>a</sup> (solvent)	Molecular Formula <sup>b</sup>	<sup>1</sup> H-NMR (CDCl <sub>3</sub> /TMS) <sup>c</sup> δ (ppm)
<b>4a</b>	88	195–196 (CH <sub>3</sub> OH)	C <sub>31</sub> H <sub>28</sub> NO <sub>5</sub> P (525.5)	0.9 [s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ]; 7.1–8.1 (m, 19H <sub>arom</sub> );
<b>4c</b>	72	179–180 (CH <sub>3</sub> OH)	C <sub>32</sub> H <sub>30</sub> NO <sub>5</sub> P (539.5)	0.9 [s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ]; 2.35 (s, 3H, CH <sub>3</sub> ); 7.0–7.9 (m, 18H <sub>arom</sub> );
<b>4d</b>	82	213–214 (CH <sub>3</sub> OH)	C <sub>31</sub> H <sub>27</sub> ClNO <sub>5</sub> P (559.9)	0.9 [s, 9H, C(CH <sub>3</sub> ) <sub>3</sub> ]; 7.1–8.0 (m, 18H <sub>arom</sub> );
<b>5a</b>	92	161–162 (benzene)	C <sub>26</sub> H <sub>20</sub> NO <sub>3</sub> P (425.4)	7.1–7.8 (m, 19H <sub>arom</sub> , 1H, CH=P)
<b>5c</b>	91.5	178–180 (benzene)	C <sub>27</sub> H <sub>22</sub> NO <sub>3</sub> P (439.4)	2.4 (s, 3H, CH <sub>3</sub> ); 7.0–7.8 (m, 18H <sub>arom</sub> , 1H, CH=P)
<b>5d</b>	90	201–202 (CH <sub>3</sub> OH)	C <sub>26</sub> H <sub>19</sub> ClNO <sub>3</sub> P (459.8)	7.1–7.8 (m, 18H <sub>arom</sub> + 1H, CH=P)

<sup>a</sup> Melting points are uncorrected.<sup>b</sup> Satisfactory microanalyses obtained: C ± 0.5, H ± 0.1, N ± 0.2.<sup>c</sup> Recorded on a Varian EM 390 spectrometer.**Table 3.** 4-Oxo-3-triphenylphosphoranylidene-3,4-dihydrocinnolines **6a-d** Prepared

Prod-uct	Yield (%)	m.p. (°C) <sup>a</sup> (solvent)	Molecular Formula <sup>b</sup>	<sup>1</sup> H-NMR (CDCl <sub>3</sub> /TMS) <sup>c</sup> δ (ppm)
<b>6a</b>	91	172–173 (CH <sub>3</sub> CN)	C <sub>26</sub> H <sub>19</sub> N <sub>2</sub> OP (405.8)	7.2–8.2 (m, 19H <sub>arom</sub> )
<b>6b</b>	78	240–241 ( <i>i</i> -C <sub>3</sub> H <sub>7</sub> OH)	C <sub>27</sub> H <sub>19</sub> N <sub>2</sub> O <sub>3</sub> P (450.4)	6.0 (s, 2H, –OCH <sub>2</sub> O–); 7.2–7.7 (m, 17H <sub>arom</sub> );
<b>6c</b>	68	213–214 (CH <sub>3</sub> CN)	C <sub>27</sub> H <sub>21</sub> N <sub>2</sub> OP (420.4)	2.9 (s, 3H, CH <sub>3</sub> ); 7.2–8.1 (m, 18H <sub>arom</sub> );
<b>6d</b>	80	209–210 (dioxane)	C <sub>26</sub> H <sub>18</sub> ClN <sub>2</sub> OP (440.9)	7.2–8.3 (m, 18H <sub>arom</sub> )

<sup>a</sup> Melting points are uncorrected.<sup>b</sup> Satisfactory microanalyses obtained: C ± 0.35, H ± 0.2, N ± 0.3.<sup>c</sup> Recorded on a Varian EM 390 spectrometer.**Table 4.** 4-Oxo-1,4-dihydrocinnolines **7a-d** Prepared

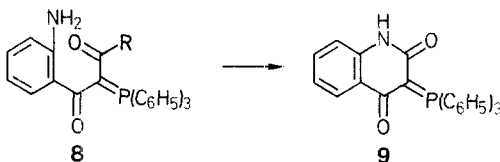
Prod-uct	Yield (%)	m.p. (°C) <sup>a</sup> (solvent)	Molecular Formula or Lit. m.p. (°C)	<sup>1</sup> H-NMR (Solvent/TMS) <sup>b</sup> δ (ppm)
<b>7a</b>	97	225–227 (CH <sub>3</sub> OH)	225 <sup>4</sup> 233.5–234 <sup>4</sup>	(DMSO- <i>d</i> <sub>6</sub> ): 7.35–8.15 (m, 5H <sub>arom</sub> ); 13.5 (br s, 1H, OH)
<b>7b</b>	93	327–328 (dec) (AcOH)	316–318 <sup>4</sup>	(DMSO- <i>d</i> <sub>6</sub> ): 6.3 (s, 2H, –O–CH <sub>2</sub> –O–); 7.4 (s, 1H, H-8); 7.6 (s, 1H, H-5); 8.8 (s, 1H, H-3)
<b>7c</b>	78	219–221 (CH <sub>3</sub> OH)	C <sub>9</sub> H <sub>8</sub> N <sub>2</sub> O <sup>c</sup> (160.2)	(CF <sub>3</sub> COOH): 3.0 (s, 3H, CH <sub>3</sub> ); 7.9–8.6 (m, 3H <sub>arom</sub> ); 9.15 (s, 1H, H-3)
<b>7d</b>	71	296–297 (CH <sub>3</sub> OH)	294–295 <sup>4</sup>	(DMSO- <i>d</i> <sub>6</sub> ): 3.2 (br s, 1H, OH); 7.3–7.9 (m, 4H <sub>arom</sub> )

<sup>a</sup> Melting points are uncorrected.<sup>b</sup> Recorded on a Varian EM 390 spectrometer.<sup>c</sup> calc. C 67.81 H 5.06 N 17.53

found 67.47 5.07 17.48

Basic hydrolysis of phosphoranes **6a-d** gave the 4-oxo-1,4-dihydrocinnolines derivatives **7a-d** and triphenylphosphine oxide (Table 4).

It is noteworthy that the 1-(2-aminoaryl)-1,3-dioxo-2-alkylidenetriphenylphosphoranes **8**, obtained by reduction of the corresponding nitro compound, was not stable and cyclized spontaneously to give the 2,4-dioxo-1,2,3,4-tetrahydro-3-(triphenylphosphoranylidene)-quinoline **9**.<sup>7</sup>



Anthranilic acid methyl ester **1a** is commercially available. Compounds **1b, c** were prepared as previously described.<sup>8,9</sup>

**1-*t*-Butoxycarbonyl-2-(2-nitroaryl)-2-oxo-triphenylphosphoniummethylenes 4a, c, d; General Procedure:**

A solution of the appropriate *o*-nitro-aryl chloride **3** (10 mmol) in dry benzene (10 ml) is slowly added to a solution of *t*-butoxycarbonylmethylenetriphenylphosphorane (7.5 g, 20 mmol) in dry benzene (30 ml). The resulting solution is kept at room temperature overnight. The precipitated *t*-butoxycarbonylmethylenetriphenylphosphorane hydrochloride is collected by filtration and the organic layer is washed with 5% sodium hydrogen carbonate solution (2 × 10 ml), dried with sodium sulfate and evaporated. The products are then purified by crystallization (Table 2).

**Hydrolysis and Decarboxylation of 4a, c, d: 2-(2-Nitroaryl)-2-oxo-triphenylphosphoniummethylenes 5a, c, d; General Procedure:**

To a solution of phosphoranes **4a, c, d** (5 mmol) in dichloromethane (10 ml), 85% sulfuric acid (2 ml) is added. After stirring for 3 h at room temperature, the solution is washed with 5% sodium hydrogencarbonate solution (2 × 75 ml), dried with sodium sulfate, and evaporated. From the residue, the phosphoranes **5a, c, d** are recovered by crystallization (Table 2).

**2-(2-Aminoaryl)-2-oxo-triphenylphosphoniummethylenes 2a-c; General Procedure:**

**Method A:** To a stirred slurry of methylenetriphenylphosphoranes [prepared from methyl triphenylphosphonium iodide (12.2 g, 30 mmol) and sodium amide (1.57 g, 40 mmol) in anhydrous benzene (120 ml) under nitrogen with stirring at room temperature overnight] a solution of anthranilic acid methyl esters (15 mmol) in anhydrous benzene (20 ml) is added. The resulting mixture is kept at 50°C for 6 h. Sodium iodide is then filtered off at room temperature and the solvent evaporated. From the crude residue the compounds **2a-c** are recovered as solid compounds by treatment with a little benzene at room temperature and purified by crystallization (Table 1).

**Method B:** Tin(II)chloride dihydrate (5.64 g, 25 mmol) is added to a solution of *o*-nitroarylmethylenephosphoranes **5a, c, d** (5 mmol) in ethyl acetate (40 ml). After refluxing for 3 h, the mixture is allowed to cool to room temperature, made alkaline with concentrated ammonia and extracted with benzene (2 × 50 ml). The organic layer is then dried with sodium sulfate and evaporated. From the crude residue the phosphoranes **2a, c, d** are recovered by crystallization (Table 1).

**4-Oxo-3-triphenylphosphoranylidene-3,4-dihydrocinnolines 6a-d; General Procedure:**

To a solution of phosphoranes **2a-d** (5 mmol) in methanol (10 ml) and 36% hydrochloric acid (0.5 ml), *n*-pentyl nitrite (1.12 g, 9.5 mmol) is added in 10 min at 0°C. After 15 min the solution is allowed to warm to room temperature, made alkaline (pH = 8–9) with 10% aqueous sodium hydroxide, diluted with water (40 ml) and extracted with chloroform (2 × 30 ml). The organic layer is dried with sodium sulfate and evaporated. The products are collected by filtration after treatment of the crude residue with ether and purified by crystallization (Table 3).

**4-Oxo-1,4-dihydrocinnolines 7a-d; General Procedure:**

A solution of 3-triphenylphosphoranylidene-4(1*H*)oxo-cinnolines **6a-d** (1 mmol) in methanol (10 ml) and 30% sodium hydroxide (1 ml) is refluxed for 2 h. Methanol is evaporated, the residue treated with water and the triphenylphosphine oxide is extracted with dichloromethane (2 × 20 ml). The aqueous layer is then treated with 10% hydrochloric acid (3.5 ml) and the 4(1*H*)oxo-cinnolines **7a-d** collected by filtration. The products are then purified by crystallization (Table 4).

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