Efficient Synthesis of 6-Prenylcoumarins; Total Syntheses of Suberosin, Toddaculin, *O*-Methylapigravin (*O*-Methylbrosiperin) and *O*-Methylbalsamiferone

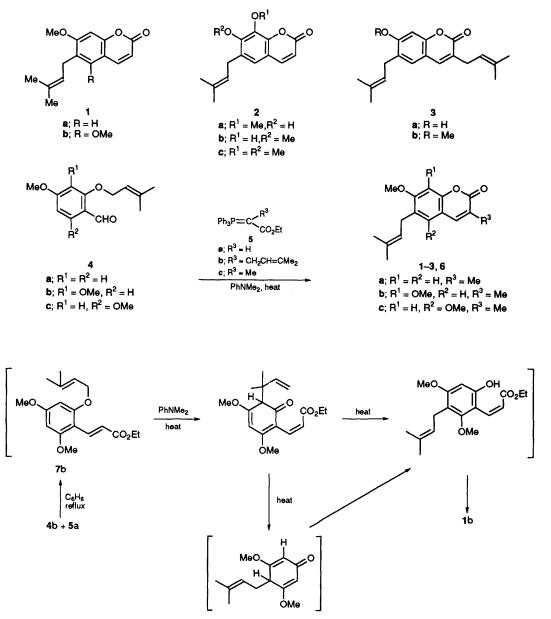
Raghao S. Mali,* Paramjeet Kaur Sandhu and Anita Manekar-Tilve

Garware Research Centre, Department of Chemistry, University of Poona, Ganeshkhind, Pune-411 007, India

Synthesis of naturally occurring 6-prenylcoumarins **1a**, **b**, **2c**, and **3b** and their derivatives **6a–c** is described, starting from 2-prenyloxybenzaldehydes **4a–c**, using a tandem Claisen rearrangement and Wittig reaction.

Several 6-prenylcoumarins such as suberosin 1a, toddaculin 1b, apigravin 2a, brosiperin 2b and balsamiferone 3a have been isolated from natural sources.¹ A large number of 6-allyl- and 6-prenyl-coumarins have been used as intermediates for the synthesis of biologically active compounds,¹ naturally occurring 6-substituted coumarins and linear furo-coumarins.¹⁻³ In view of this, various approaches have been developed^{1,4,5} for 6-allyl- and 6-prenyl-coumarins.

Claisen rearrangement of allyloxy benzene provides *ortho*allylphenol; most of the reported methods utilize 7-allyloxycoumarins as starting materials to obtain allylcoumarins.¹ Since 7-allyloxycoumarins on Claisen rearrangement provide exclusively 8-allylcoumarins,¹ the C-8 position is blocked to obtain 6-allylcoumarins.^{6.7} In a recent approach 7-alkoxycoumarins have been initially converted to methyl 2-allyloxy-4-alkoxycinnamates and then to 6-allylcoumarins such as suberosin **1a** and related compounds.⁸ In an alternative approach the propynylic ether of umbeliferone has been used for the synthesis of demethylsuberosin,⁹ which was subsequently converted to 3,6-diprenyl-7-hydroxycoumarin (balsamiferone **3a**).¹⁰ A route utilizing 3-prenyl-7-hydroxycoumarin¹¹ has also been reported for balsamiferone **3a**.



Scheme 1

Literature methods^{12–14} for toddaculin **1b** either involve multistep sequences and/or provide **1b** in very low yields. Most of these approaches^{13,14} utilize 5,7-dihydroxycoumarin as the starting material. As 7-(1,1-dimethylallyloxy)coumarins provide 8-allylcoumarins on Claisen rearrangement, it was necessary to synthesize 5-(1,1-dimethylallyloxy)-7-methoxycoumarin to obtain toddaculin¹³ **1b**. The major obstacle in this case was the selective allylation of C-5-hydroxyl group of 5,7-dihydroxycoumarin.

All the known methods, for the synthesis of 6-allyl and 3,6-diallylcoumarins make use of preformed coumarins.8-10 We report herein a novel and general route for naturally occurring 6-prenylcoumarins 1a, b, 2c and 3b and their derivatives 6a-c from 2-prenyloxybenzaldehydes 4a-c. The aldehydes 4a-c were prepared by prenylation¹³ (prenyl bromide, K₂CO₃, tetrabutylammonium iodide, acetone, reflux) of the corresponding 2-hydroxybenzaldehydes. Thus, the reaction of 4a with phosphorane 5a in N,N-dimethylaniline at 200°C for 6 h under nitrogen atmosphere, directly gave suberosin 1a, mp 87 °C (lit.15 87-88 °C) in 47% yield. A similar reaction of 4c and 4b with phosphorane 5a for 8 and 12 h provided toddaculin 1b, mp 93 °C (lit.¹³ 93-94 °C) and O-methylapigravin (O-methylbrosiperin, 2c), mp 93 °C (lit.¹⁶ 93-95 °C) in 50 and 55% yields, respectively. It was anticipated that the reaction of 4a-c with 5a would initially give (E)-esters 7a-c, which would isomerise thermally to the (Z)-isomer and then cyclize after Claisen rearrangement to give 1a, b and 2c (Scheme 1). Thus, when 4b was reacted with **5a** in refluxing benzene for 6 h the (E)-ester **7b**, mp 92 °C, was obtained, which on heating in refluxing N,N-dimethylaniline at 200 °C for 6 h provided toddaculin 1b. The α and β olefinic protons in 7b appeared in ¹H NMR (CDCl₃) as doublets (J 16 Hz) at δ 7.00 and 8.42, respectively, which confirmed its geometry. O-Methylbalsamiferone 3b was obtained in 49% yield, by a similar reaction of 4a with phosphorane 5b.

To demonstrate the generality of this approach the aldehydes 4a-c were reacted with phosphorane 5c to obtain coumarins 6a-c in 48, 58 and 45% yields, respectively. The present approach, which does not require preformed coumarin, demonstrates the synthetic utility of this tandem Claisen rearrangement and Wittig reaction for the synthesis of 6-prenyl- and 3,6-diprenyl-coumarins. IR and ¹H NMR spectral data of coumarins 1a, b and 2c are identical with literature data.^{5,13,15} The new coumarins 3b and 6a-c also exhibited satisfactory analytical and spectral data.[†]

J. CHEM. SOC., CHEM. COMMUN., 1994

The authors thank the UGC, New Delhi for the financial support. One of them (A. M. T.) thanks the CSIR, New Delhi for the award of a Senior Research Fellowship.

Received, 9th August 1993; Com. 3/04794F

Footnote

[†] Selected spectral data for compound **6c**: IR (Nujol, v_{max} /cm⁻¹): 1720 (C=O); ¹H NMR (CDCl₃) δ : 1.8 and 1.9 (s, 3 H each, 2 × Me), 2.2 (s, 3 H, C₃-Me), 3.4 (d, 2 H, ArCH₂), 3.8 and 3.9 (s, 3 H each, 2 × OMe), 5.2 (t, 1 H, CH), 6.8 (s, 1 H, C₈-H), 7.8 (s, 1 H, C₄-H).

References

- 1 R. D. H. Murray, J. Mendez and S. A. Brown, *The Natural Coumarins, Occurence, Chemistry and Biochemistry*, Wiley-Interscience New York, 1982.
- 2 R. D. H. Murray, M. M. Ballantyne and K. P. Mathai, *Tetrahedron*, 1971, 27, 1247.
- 3 T. R. Seshadri and M. S. Sood, Indian J. Chem., 1963, 1, 291.
- 4 S. K. Koul, S. C. Taneja and K. L. Dhar, Indian J. Chem., Sect. B., 1987, 26, 574.
- 5 M. B. Raizada, S. K. Garg and S. R. Gupta, *Indian J. Chem., Sect.* B, 1981, 20, 918.
- 6 N. H. Pardanani, Y. A. Shaikh and K. N. Trivedi, J. Indian Chem. Soc., 1975, 52, 45.
- 7 G. M. Massanet, E. Pendo, F. Rodriguez-Luis and J. Salva, *Heterocycles*, 1987, 26, 1541.
- 8 N. Cairns, L. M. Harwood and D. P. Astles, J. Chem. Soc. Chem. Commun., 1986, 1264.
- 9 N. Cairns, L. M. Harwood and D. P. Astles, J. Chem. Soc. Chem. Commun., 1986, 750.
- 10 N. Cairns, L. M. Harwood and D. P. Astles, J. Chem. Soc., Chem. Commun., 1987, 400.
- 11 D. Swaroop, P. B. Sharma and R. S. Kapil, *Indian J. Chem.*, Sect. B, 1983, 22, 408.
- 12 P. N. Sharma, A. Shoeb, R. S. Kapil and S. P. Popli, Indian J. Chem., Sect. B., 1980, 19, 938.
- 13 R. D. H. Murray, M. M. Ballantyne, T. C. Hogg and P. H. McCabe, *Tetrahedron*, 1975, **31**, 2960.
- 14 S. Mahey, T. R. Seshadri and S. K. Mukerjee, Indian J. Chem., 1974, 12, 29.
- 15 P. W. Austin and T. R. Seshadri, Indian J. Chem., 1968, 6, 412.
- 16 S. K. Garg, S. R. Gupta and N. D. Sharma, *Phytochemistry*, 1979, 18, 1580.