

STUDIES ON THE TOTAL SYNTHESIS OF FREDERICAMYCIN A: DEVELOPMENT OF AN  
INTERMOLECULAR ALKYNE-CHROMIUM CARBENE COMPLEX CYCLIZATION APPROACH TO THE  
ABCDE RING SYSTEM

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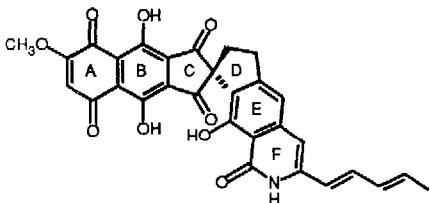
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**Abstract:** *The development of a synthetic approach to the fredericamycin A ABCDE ring system based on a regiospecific intermolecular alkyne-chromium carbene complex cyclization is detailed.*

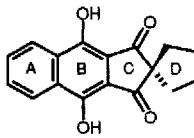
Fredericamycin A (**1**, NSC-305263), a quinone antitumor antibiotic<sup>2</sup> isolated from *Streptomyces griseus*<sup>3</sup> bearing a unique spiro[4.4]nonene central to its structure, has been shown to possess potent *in vitro* cytotoxic activity and confirmed *in vivo* antitumor activity that is derived from its inhibition of RNA and protein synthesis through nondiscriminant oxidative damage to DNA and/or discriminant inhibition of DNA processing enzymes.<sup>2,5</sup> Consequently, since the unambiguous establishment of its structure through a single crystal X-ray structure determination<sup>4</sup> after extensive spectroscopic studies<sup>5</sup> failed to resolve tautomeric structures, fredericamycin A continues to be the subject of biological<sup>2</sup> and extensive synthetic efforts<sup>6</sup> including one recently completed total synthesis.<sup>7</sup> Herein we detail preliminary studies on the development of a general approach to the construction of the fredericamycin A ABCDE ring system applicable to the total synthesis of fredericamycin A and structurally related agents based on the implementation of a regiospecific intermolecular alkyne-chromium carbene complex cyclization.<sup>8</sup>

Key to the development of this convergent assemblage of the fredericamycin A skeleton rests on the facility with which a simple aldol closure might provide for introduction of the spiro[4.4]nonene CD ring system; Scheme I, **5** → **4**,<sup>9</sup> and the feasibility for implementation of a regiospecific inter- or intramolecular alkyne-chromium carbene complex cyclization for introduction of the fully substituted B ring hydroquinone; Scheme I, **7** → **5/6**. Herein we detail preliminary studies resulting in the preparation of **2-3** that establish this as a viable approach to fredericamycin A.

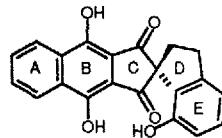
In contrast to initial expectations in which the electronic nature of the alkyne was anticipated to provide a useful and predominate element for control of the regioselectivity of an intermolecular alkyne-chromium carbene complex cyclization, a study of electronic and steric features of the alkyne that control



**1** fredericamycin A

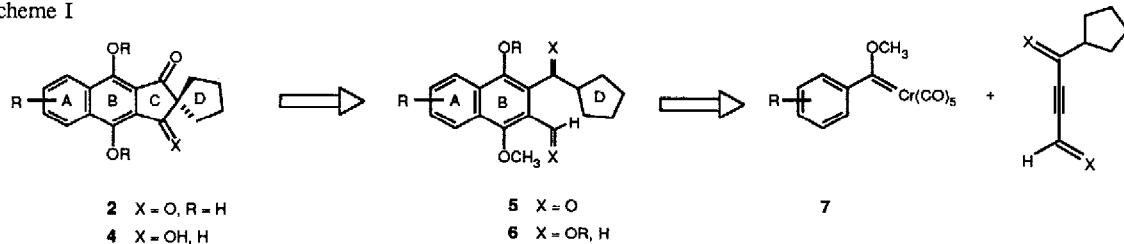


**2**



**3**

Scheme I



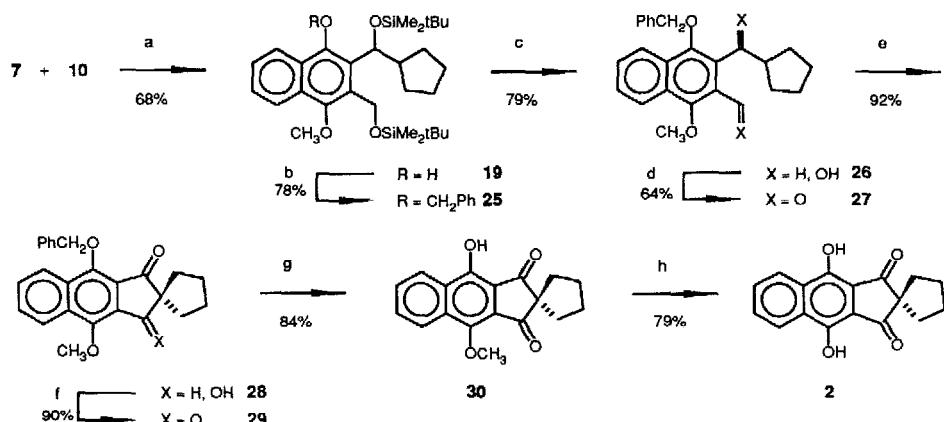
the cyclization mode<sup>8</sup> and regioselectivity<sup>8,10-11</sup> of the reactions of Fischer chromium carbene complexes with acetylenes revealed that the benzannulation chemical conversions were optimal with neutral alkynes (neutral alkynes > electron-deficient alkynes). In addition, modest steric differences in the substitution pattern at the alkyne adjacent carbons proved sufficient to permit complete regiocontrol in the intermolecular cyclization reactions.<sup>8,10-11</sup> Representative results of this study are detailed in Scheme II. As detailed in the recent efforts of Wulff,<sup>10</sup> the benzannulation reactions proved optimally conducted in heptane versus ether solvents (Et<sub>2</sub>O, THF) at concentrations of 0.3 M in the presence of 1.0-1.5 equivalents of alkyne. In addition, under the standard reaction conditions the product derived from the reaction of alkyne **10** with the chromium carbene complex **7** proved to be **20** (entry 9) presumably derived through in situ elimination of *t*-butyldimethylsilanol from the primary reaction product **19** with generation of an unstable orthoquinomethane that suffers a subsequent 1,5-hydrogen shift to provide **20**. Under the Yamashita reaction conditions which generally acylate phenols (entry 6)<sup>11</sup> the elimination was suppressed and, contrary to expectations, was attributed experimentally to the inclusion of acetic anhydride in the reaction mixture under conditions that do *not* acylate the phenol (entries 6-8) *and* that accelerate the rate of reaction. Thus, **19** or **20** may be obtained cleanly from the reaction of **10** with **7** depending on the reaction conditions selected.

Scheme II



Entry	X	R <sup>1</sup>	eq.	Conditions	Yield	R	Yield	R
1	<b>8a</b>	OSiMe <sub>2</sub> Bu, H	H	1.4 Ac <sub>2</sub> O:Et <sub>2</sub> N (1.5 eq:1.5 eq), 80°C, heptane, 1 h, 0.3 M	47%	Ac	<b>13</b>	-
2	<b>8b</b>	OSiPh <sub>2</sub> Bu, H	H	1.5 Ac <sub>2</sub> O:Et <sub>2</sub> N (1.5 eq:1.5 eq), 80°C, heptane, 3 h, 0.1 M	32%	Ac	<b>14</b>	-
3	<b>9a</b>	OSiMe <sub>2</sub> Bu, H	CO <sub>2</sub> CH <sub>3</sub>	1.5 65°C, THF, 24 h, 0.03 M	22%	H	<b>15</b> <b>16</b>	-
4	<b>9a</b>			1.5 Ac <sub>2</sub> O:Et <sub>2</sub> N (1.5 eq:1.5 eq), 80°C, heptane, 19 h, 0.1 M	10%	H	<b>15</b> <b>16</b>	8%
5	<b>9b</b>	OSiPh <sub>2</sub> Bu, H	CO <sub>2</sub> CH <sub>3</sub>	1.5 Ac <sub>2</sub> O:Et <sub>2</sub> N (1.5 eq:1.5 eq), 80°C, heptane, 19 h, 0.1 M	7%	H	<b>17</b> <b>18</b>	6% Ac
6	<b>10</b>	OSiMe <sub>2</sub> Bu, H	ClH <sub>2</sub> OSiMe <sub>2</sub> Bu	1.1 Ac <sub>2</sub> O:Et <sub>2</sub> N (1.0 eq:1.0 eq) 80°C, heptane, 4 h, 0.3 M	68%	H	<b>19</b> <b>20</b>	- H
7	<b>10</b>		0.8 Bu <sub>3</sub> N (1.0 eq), 80°C, heptane, 16 h, 0.3 M		30%	H	<b>19</b> <b>20</b>	17% H
8	<b>10</b>		1.0 Ac <sub>2</sub> O (1.0 eq), 80°C heptane, 3 h, 0.3 M		66%	H	<b>19</b> <b>20</b>	- H
9	<b>10</b>		1.0 80°C, heptane, 17 h, 0.3 M		-		<b>19</b> <b>20</b>	74% H
10	<b>11</b>	O	CH <sub>2</sub> OSiMe <sub>2</sub> Bu	1.5 65°C, THF, 9 h, 0.03 M	0%		<b>21</b>	
11	<b>12</b>	O	CO <sub>2</sub> CH <sub>3</sub>	1.5 65°C, THF, 9 h, 0.2 M	0%		<b>22</b>	

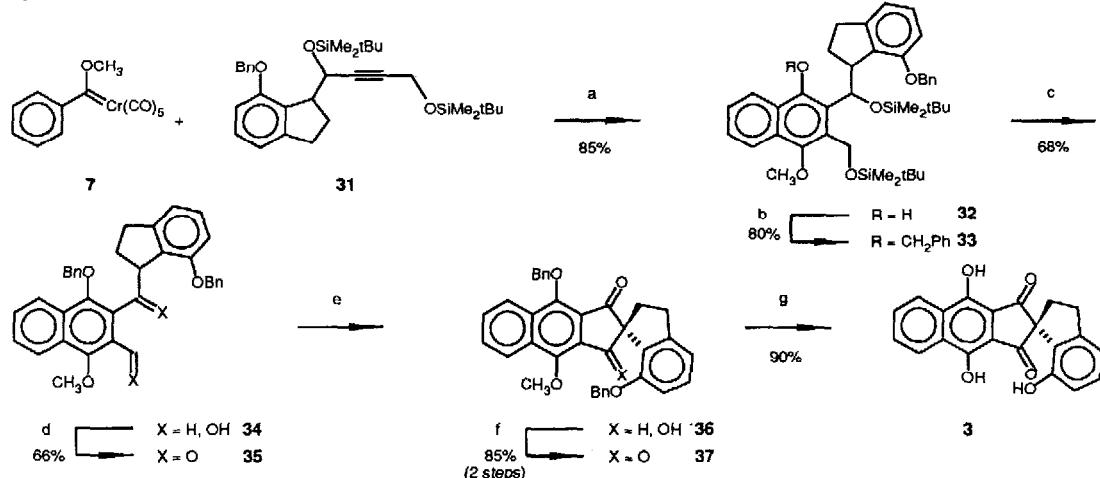
Scheme III



(a)  $80^{\circ}\text{C}$ , heptane,  $\text{Ac}_2\text{O}:\text{Et}_3\text{N}$  (1.5 eq:1.5 eq), 3 h. (b)  $\text{PhCH}_2\text{Br}$  (1.2 eq),  $\text{K}_2\text{CO}_3$  (10 eq),  $(\text{Bu})_4\text{NI}$  (0.1 eq),  $\text{DMF}$ ,  $25^{\circ}\text{C}$ , 15 h. (c)  $\text{HOAc}:\text{THF}:\text{H}_2\text{O}$  (3:1:1),  $40^{\circ}\text{C}$ , 48 h. (d)  $(\text{COCl})_2$  (2.2 eq),  $\text{DMSO}$  (4.8 eq),  $\text{Et}_3\text{N}$  (10 eq),  $\text{CH}_2\text{Cl}_2$ ,  $-60^{\circ}\text{C}$ , 1 h. (e)  $\text{NaOCH}_3$  (0.1 eq),  $\text{CH}_3\text{OH}$ ,  $40^{\circ}\text{C}$ , 6 h. (f)  $\text{PCC}$  (1.5 eq),  $\text{CH}_2\text{Cl}_2$ ,  $25^{\circ}\text{C}$ , 5 h. (g) 10%  $\text{Pd-C}$  (1 eq),  $\text{HCO}_2\text{NH}_2$  (5 eq),  $\text{CH}_3\text{OH}$ ,  $25^{\circ}\text{C}$ , 2 h. (h)  $\text{BH}_3$  (2.5 eq),  $\text{NaI}$  (5.0 eq),  $\text{DMF}$ ,  $150^{\circ}\text{C}$ , 3 h.

The application of the regiospecific benzannulation reaction of **7** with **10** to the assemblage of the fredericamycin ABCD carbon framework is detailed in Scheme III. Protection of the free phenol of the benzannulation product **19** as its benzyl ether **25** was accomplished cleanly under mild basic conditions that proceeded without competitive elimination of *t*-butyldimethylsilanol followed by deprotection of the primary and secondary benzylic alcohols afforded **26**. Direct oxidation of diol **26** to keto aldehyde **27** was accomplished cleanly only under the conditions of Swern oxidation<sup>12</sup> and required carefully controlled reaction conditions that ensure activation of both alcohols through formation of the bisalkoxysulfonium salt prior to base-catalyzed elimination of dimethyl sulfide with formal oxidation of the primary and secondary benzylic alcohols. Keto aldehyde **27** cleanly closed to the spirocyclic keto alcohol **28** upon exposure to sodium methoxide thus providing the functionalized spiro[4.4]nonene and establishing the viability of this approach to the fredericamycin A ABCD ring system. Oxidation of the secondary alcohol afforded **29** without detection of a competitive retro aldol reaction and subsequent sequential deprotection of **29** provided **2**.

Scheme IV



(a)  $80^{\circ}\text{C}$ , heptane,  $\text{Ac}_2\text{O}$  (1 eq), 3 h. (b)  $\text{PhCH}_2\text{Br}$  (1.5 eq),  $\text{K}_2\text{CO}_3$  (10 eq),  $(\text{Bu})_4\text{NI}$  (0.1 eq),  $\text{DMF}$ ,  $25^{\circ}\text{C}$ , 48 h. (c)  $(\text{Bu})_4\text{NF}$  (5 eq),  $\text{THF}$ ,  $25^{\circ}\text{C}$ , 72 h. (d)  $(\text{COCl})_2$  (2.2 eq),  $\text{DMSO}$  (4.8 eq),  $\text{Et}_3\text{N}$  (10 eq),  $\text{CH}_2\text{Cl}_2$ ,  $-67^{\circ}\text{C}$ , 1 h. (e)  $\text{NaOCH}_3$  (0.1 eq),  $\text{CH}_3\text{OH}$ ,  $60^{\circ}\text{C}$ , 3 h. (f)  $\text{PCC}$  (3 eq),  $\text{CH}_2\text{Cl}_2$ ,  $25^{\circ}\text{C}$ , 14 h. (g)  $\text{BBr}_3$  (3.6 eq),  $\text{CH}_2\text{Cl}_2$ ,  $-78$  to  $25^{\circ}\text{C}$ , 19 h.

The extension of these observations to the assemblage of the fredericamycin A ABCDE ring system is detailed in Scheme IV. Regiospecific cyclization of **31**<sup>13</sup> with **7** under the conditions previously developed (0.3 M **7**, 1.0 equiv **31**, 80°C, heptane, 1.0 equiv Ac<sub>2</sub>O, 3 h) cleanly provided **32** (85%) as the predominant or exclusive benzannulation product ( $\geq 95\%$ ) without the detection of subsequent products derived from elimination of *t*-butyldimethylsilanol. Conversion of **32** to the keto aldehyde **35**, base-catalyzed aldol closure of **35** to the spirocyclic keto alcohol **36**, and subsequent oxidation provided **37**. Deprotection of **37** provided **3** constituting the partially functionalized fredericamycin A ABCDE ring system.

**Acknowledgments.** We gratefully acknowledge the financial support of the National Institutes of Health (CA42056, NIDDK CA40884) and the Alfred P. Sloan Foundation.

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(Received in USA 13 January 1989)