

Ring-opening of Indoles: An Unconventional Route for the Transformation of Indoles to 1H-Pyrazoles using Lewis Acid

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Ring-opening of Indoles: An Unconventional Route for the Transformation of Indoles to 1*H*-Pyrazoles using Lewis Acid

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KEYWORDS: Cyclization • indole • Lewis acid • 1*H*-pyrazole • ring-opening.

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ABSTRACT: An unusual transformation of indoles to pyrazoles via an aromatic ring-opening strategy has been developed. The salient feature of this strategy involves the C2-N1 bond opening and concomitant cyclization reaction of the C2=C3 bond of the indole moiety with the tosylhydrazone which proceeds under transition-metal and ligand free conditions. This ring-opening functionalization of indoles provides a wide scope of differently substituted pyrazoles.

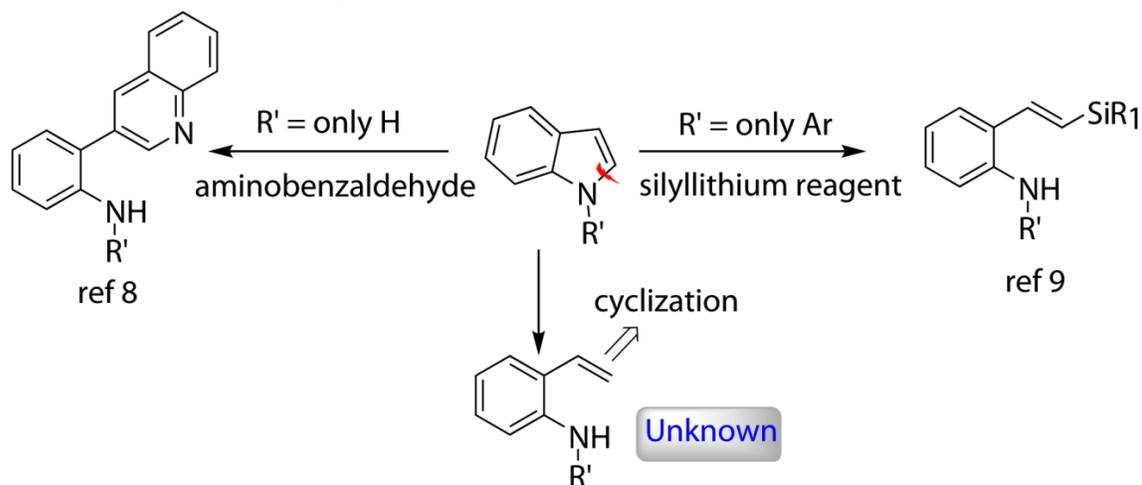
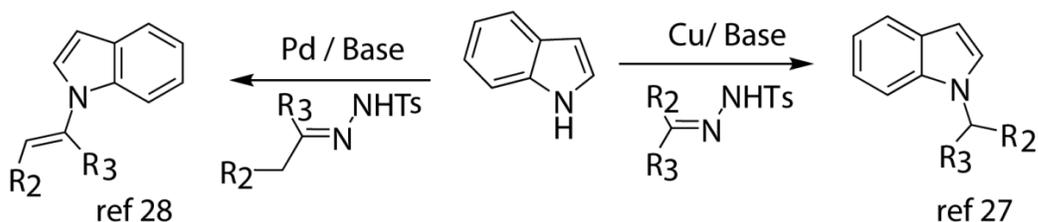
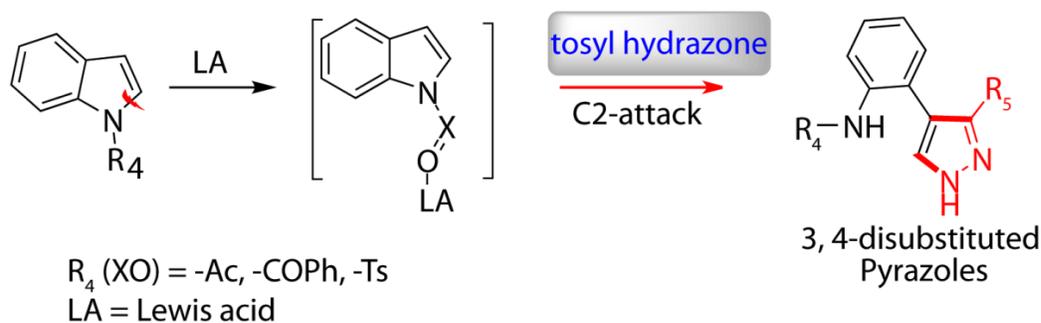
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The indole scaffold represents one of the most important structural subunits present in various pharmaceutically active and naturally occurring products. Conventional ways of indole functionalization follow nucleophilic and electrophilic reactions in which the indole scaffold

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3 retains its aromaticity.¹⁻³ Transition-metal catalyzed C-H functionalization of both the benzene
4 and pyrrole rings of indole has also been widely studied and considered as one of the fascinating
5 strategies for arene functionalization in modern chemistry.⁴⁻⁷ Ring-opening functionalization of
6 indole is uncommon and more challenging. The aromaticity of indole restricts such ring-opening
7 functionalization strategy. Recently, unconventional opening of C2-N1 bond of indoles was
8 described (Scheme 1a).⁸⁻¹¹ On the other hand, the usage of C2=C3 bond as electrophile for the
9 in-situ cyclization reaction followed by ring-opening of indole is unknown.

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12 In our previous report, we described the regioselective synthesis of 1*H*-pyrazoles through the
13 coupling reaction between the ambiphilic tosylhydrazones and alkenes or alkynes.¹² Other
14 references also describe that the aryl hydrazones are an important structural unit for the
15 transformation of alkene or alkyne to pyrazoles.¹³⁻²⁴ We and other research groups demonstrated
16 that tosylhydrazone have tremendous application in synthetic organic chemistry, especially in
17 cyclization reactions. However, its reactivity towards indole has not been greatly explored.^{12, 25-26}
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19 Recent studies described that the reaction of tosylhydrazone with indole favors N-alkylation or
20 N-vinylation of indole, rather than cyclization reactions with C2=C3 bond (Scheme 1b).²⁷⁻²⁹ We
21 hypothesized that C2-N1 bond opening could be one of the driving forces for C2=C3 bond
22 activation.

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24 It is well documented that the carbonyl and sulfonyl groups form stable adducts with Lewis
25 acids.³⁰ Hence, we judiciously introduced the carbonyl or sulfonyl containing directing groups
26 like acyl, benzoyl and tosyl at the N1-position of indoles assuming that the complexation
27 between Lewis acids and these directing groups would induce the C2-N1 bond opening (Scheme
28 1c).^{10, 29, 31-40} Therefore, the decrease in electron density of the C2=C3 bond would lead to a
29 cyclization reaction with tosylhydrazones leading to the formation of pyrazoles. They are known
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Scheme 1. C2-N1 bond opening and the reactivity of indole towards hydrazone**a) Unconventional Ring Opening of Indole****b) Reactivity of tosylhydrazone with Indole****c) Our work**

to display a wide spectrum of biological activities including, anti-microbial, anti-fungal, anti-tubercular, anti-inflammatory, anti-convulsant, anti-cancer, anti-viral, neuroprotective.⁴¹⁻⁴² The

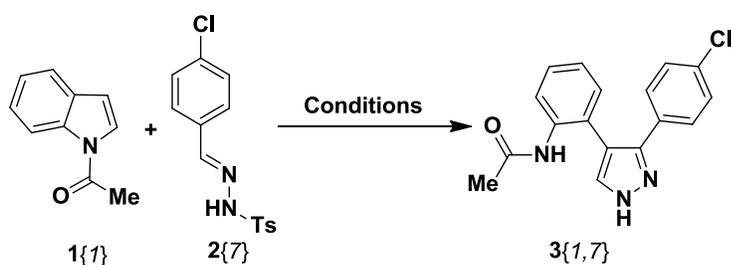
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3 pyrazole is a versatile synthetic intermediate and potent medicinal scaffold.⁴³⁻⁴⁴ The synthesis of
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5 pyrazoles still continues to attract considerable attention because of its applications in
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7 pharmaceutical and agrochemical industries. We reveal herein transition-metal and ligand free
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9 ring-opening functionalization and regioselective transformation of indole to 1*H*-pyrazoles. The
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11 product, 1*H*-pyrazole was obtained with moderate to excellent yield. The inhibitory activities of
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13 these synthesized compounds were performed against purified human immunosuppressive
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15 enzyme indoleamine 2,3-dioxygenase 1 (IDO1).
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20 Our initial investigations commenced with the aim of enhancing the electrophilicity of C2=C3
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22 bond of the indole moiety, so that the C2-N1 bond opening would become more facile leading to
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24 the annulation reaction with tosylhydrazone. For this purpose, we decided to study the reaction
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26 under mild acidic conditions using 1-acyl-1*H*-indole (**1**{*I*}) and tosylhydrazone (**2**{*7*}) as model
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28 substrates. To our delight, tosylhydrazone (**2**{*7*}) in the presence of BF₃.OEt₂ (catalytic amount)
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30 provided the regioisomeric 1*H*-pyrazole (**3**{*I,7*}) under ambient temperature (Table 1, entry1). As
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32 expected, the model reaction in the absence of BF₃.OEt₂ failed to provide the desired product
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34 even at higher temperature (Table 1, entry 2). This model reaction was also performed at higher
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36 temperatures and with different equivalents of BF₃.OEt₂. Reaction at 50 °C with 0.3 equivalents
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38 of BF₃.OEt₂ provided the target product in higher yield (Table 1, entry 4). Other Lewis and
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40 Bronsted acids such as AlCl₃, FeCl₃, Zn(OTf)₂, I₂, AcOH, TsOH, TfOH and TFA were found to
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42 be ineffective (Table 1). Solvent screening was also carried out, but dichloroethane (DCE)
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44 remained the best solvent for successful formation of pyrazole from indole (Table 1). This ring-
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46 opening reaction was unsuccessful under the basic reaction conditions.
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52 These optimized reaction conditions for the unusual indole ring-opening and cyclization
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54 reaction encouraged us to explore the scope and limitations of the synthesis of pyrazoles from
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the corresponding indoles (Figure 1). We hypothesized that the transformation of an indole into a pyrazole strongly depends on the C2-N1 bond opening proficiency of the indoles. The electronic effects of the substituents could play an important role for the formation of desired pyrazoles. Reaction between 1-acyl-1*H*-indole and electron-neutral phenyl tosylhydrazone provided the targeted pyrazole with moderate yield (55%, Table 2, **3**{1,1}).

Table 1. Optimization of the reaction conditions for the synthesis of 1*H*-pyrazole (**3**{1,7})^a

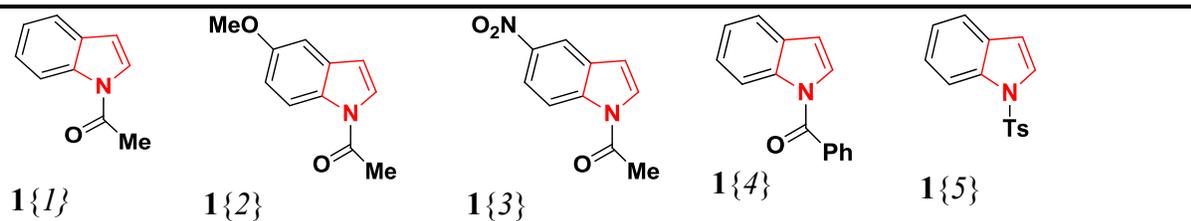


Entry	Acid / Base (equiv.)	Solvent	Time (h)	Temperature (°C)	Yield ^b (%)
1 ^c	BF ₃ .OEt ₂ (0.3)	DCE	14	RT	55
2 ^d	-	DCE	48	RT → 50	-
3 ^c	BF ₃ .OEt ₂ (0.1)	DCE	8	50	40
4 ^c	BF ₃ .OEt ₂ (0.3)	DCE	8	50	92
5 ^c	BF ₃ .OEt ₂ (0.5)	DCE	8	50	72
6 ^c	BF ₃ .OEt ₂ (0.3)	DCE	8	80	88
7 ^d	AlCl ₃ (0.3)	DCE	14	50	-
8 ^d	FeCl ₃ (0.3)	DCE	14	50	trace
9 ^d	Zn(OTf) ₂ (0.3)	DCE	14	50	-
10 ^d	Iodine (0.3)	DCE	14	50	-
11 ^d	AcOH (0.3)	DCE	14	50	-
12 ^d	TsOH (0.3)	DCE	14	50	-

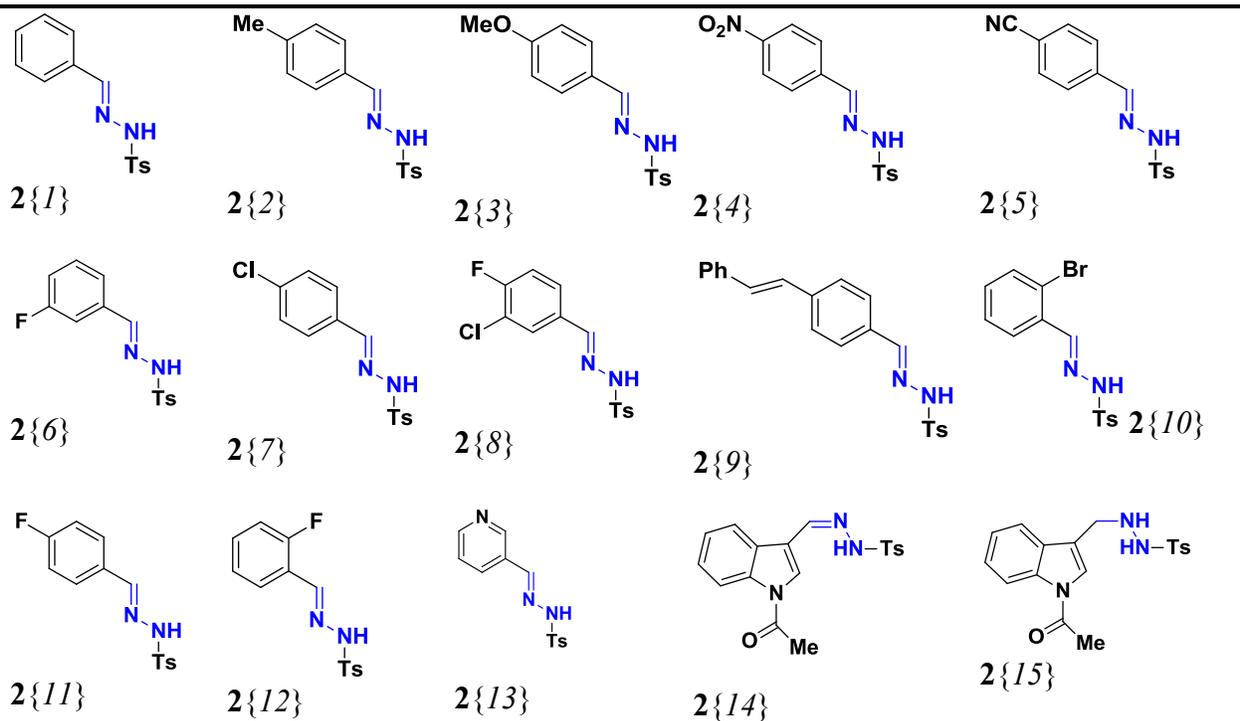
13 ^d	TfOH (0.3)	DCE	14	50	-
14 ^d	TFA (0.3)	DCE	14	50	-
15 ^c	BF ₃ .OEt ₂ (0.3)	CHCl ₃	14	50	55
16 ^c	BF ₃ .OEt ₂ (0.3)	CH ₂ Cl ₂	14	50	60
17 ^d	BF ₃ .OEt ₂ (0.3)	DMF	14	80	-
18 ^d	BF ₃ .OEt ₂ (0.3)	DMSO	14	80	-
19 ^d	BF ₃ .OEt ₂ (0.3)	toluene	14	70	-
20 ^d	BF ₃ .OEt ₂ (0.3)	CH ₃ CN	14	70	-
21 ^d	BF ₃ .OEt ₂ (0.3)	CH ₃ OH	14	50	-
22 ^c	Et ₃ N (0.3)	DCE	14	50	N.D
23 ^c	Cs ₂ CO ₃ (0.3)	DCE	14	50	N.D

^aAll the reactions were performed using 0.1 mmol (1 equiv) of **1**{*I*} and 0.11 mmol (1.1 equiv) of **2**{*7*} under acidic conditions. ^bIsolated yield of product (1*H*-pyrazole). ^c5-60% starting material **1**{*I*} and **2**{*7*} was recovered. ^d100% Starting materials **1**{*I*} and **2**{*7*} were recovered.

Alternation in the electronic environments on both indole and hydrazone moieties affected the reaction yield under the optimized reaction conditions. The aryl-tosylhydrazones bearing electron-rich substituents such as –Me and –OMe produced the corresponding pyrazoles in lower yields (**3**{*I,2*}, **3**{*I,3*}). While, the presence of an electron-deficient aryl-tosylhydrazone allowed the formation of corresponding pyrazoles (**3**{*I,4-8*}, **3**{*2,7*}, **3**{*3,7*}) in higher yields. It is important to mention that the synthesis of pyrazoles containing the nitro-, cyano- and halogen groups are highly significant as these functional groups could be deployed in transition-metal catalyzed cross-coupling reactions. The use of tosylhydrazones with such reactive groups was well tolerated in the presence of BF₃.OEt₂ thus providing the opportunity for further

1*H*-indoles 1:

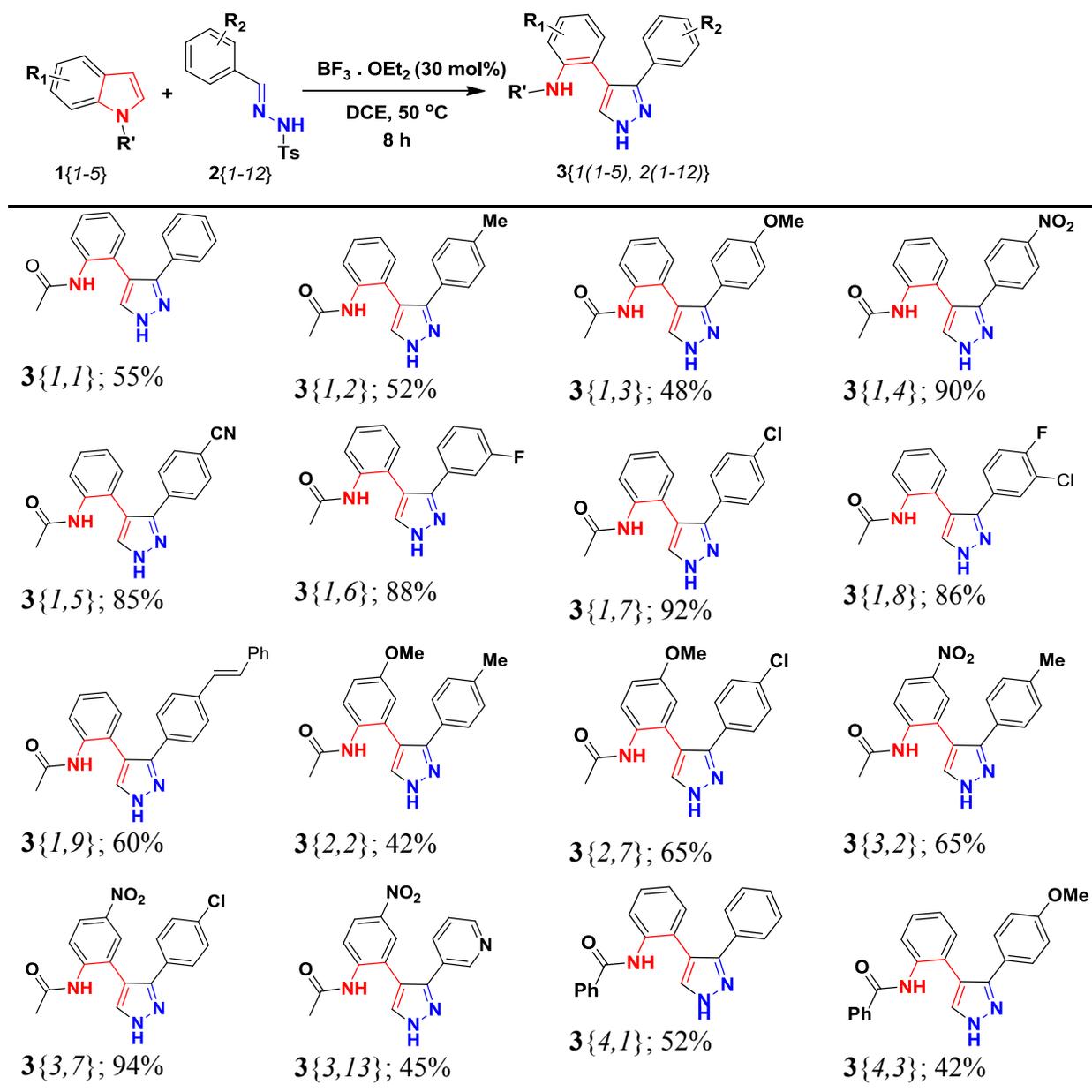
Tosylhydrazones 2:

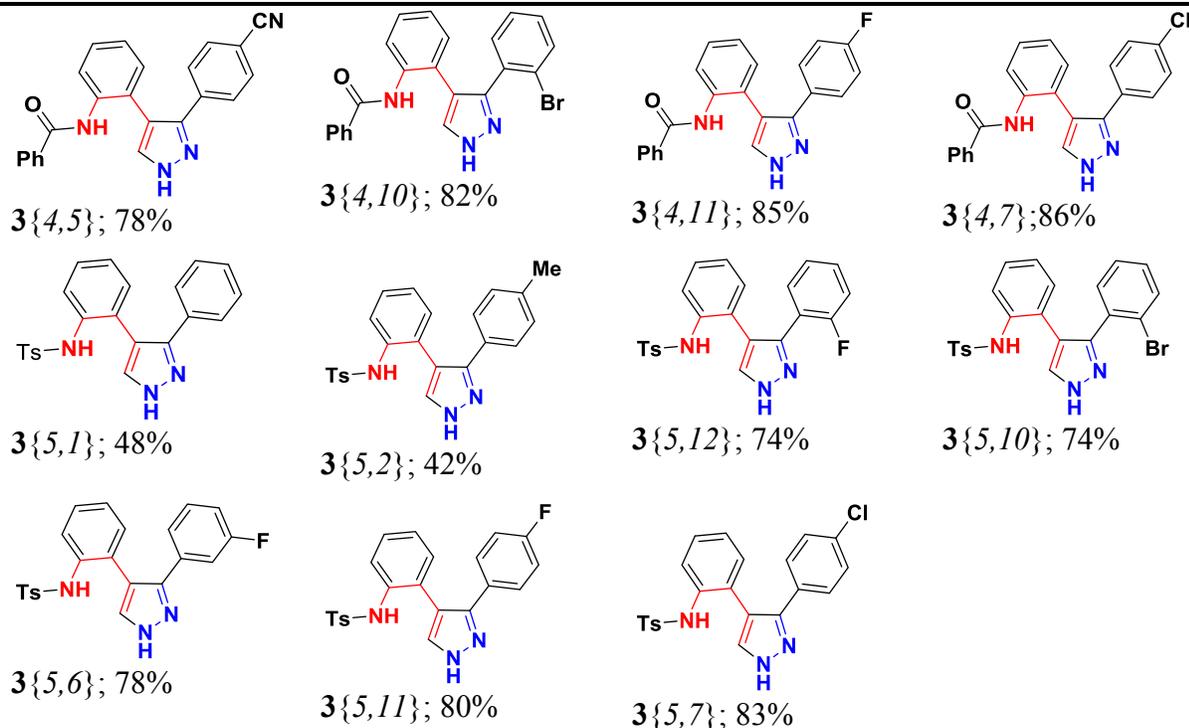
**Figure 1.** Substrates used in the synthesis of pyrazole derivatives

functionalization of these pyrazoles. Interestingly, tosylhydrazone with alkene as substituent shows high selectivity for the synthesis of regioselective pyrazole 3{1,9} with only a moderate yield. Reactions with substituted indoles also displayed quite interesting results. The presence of electron-withdrawing group ($-\text{NO}_2$) in the benzene ring of the indole produced the corresponding pyrazole in higher yield. However, the desired pyrazole was obtained in lower yields when an

electron-donating group (-OMe) was present in the benzene ring of the indole. The reaction of methyl substituted (at C2 or C3 position) indoles was unsuccessful. A combination of both

Table 2. Substrate scope for the synthesis of 1*H*-pyrazoles^a.





^aAll the reactions were performed using 0.2 mmol of **1**{1-5} (1 equiv) and 0.22 mmol of **2**{1-12} (1.1 equiv) in the presence of 0.06 mmol of $\text{BF}_3 \cdot \text{OEt}_2$ (0.3 equiv) in 2 mL of DCE at 50 °C.

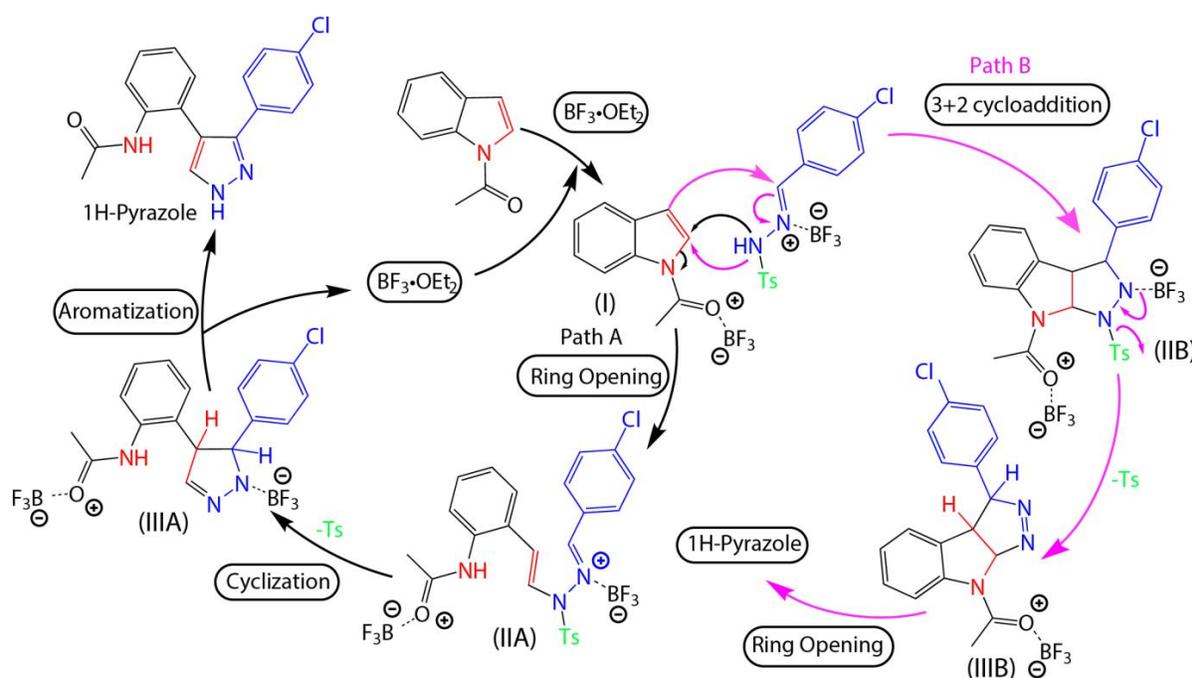
electron deficient indole and tosylhydrazone provided the targeted pyrazoles with maximum yield ($3\{3,7\}$). The reaction with tosylhydrazones bearing pyridine ring produced the target pyrazole with only moderate yield. Unfortunately, the reaction of C2/C3-methyl substituted N-acyl indole failed to provide the targeted pyrazole.

This uncommon reactivity of 1-acyl-indole with aryl-tosylhydrazone prompted us to extend our heterocycles compound library using various N1-substituted indoles. The N1-benzoyl and N1-tosyl indoles successfully participated in this ring-opening and concomitant cyclization reaction. The reaction between N1-benzoyl and N1-tosyl indoles with both electron-rich and electron-deficient aryl-tosylhydrazone showed similar reactivity trends (Table 2) as that with the acyl group, but the yields of the desired pyrazoles were reduced to some extent. The XRD analysis of the compound $3\{4,5\}$ confirmed the structure of the desired pyrazole (Figure S1 and

Table S1). Unfortunately, N1-methyl and N1-benzyl containing indole or unsubstituted (N1-H) indole failed to provide the corresponding pyrazole under these optimized reaction conditions (most of the starting materials were recovered). The reaction between 1-acyl-1*H*-indole and the acetophenone tosylhydrazone under the similar experimental conditions failed to provide the target product.

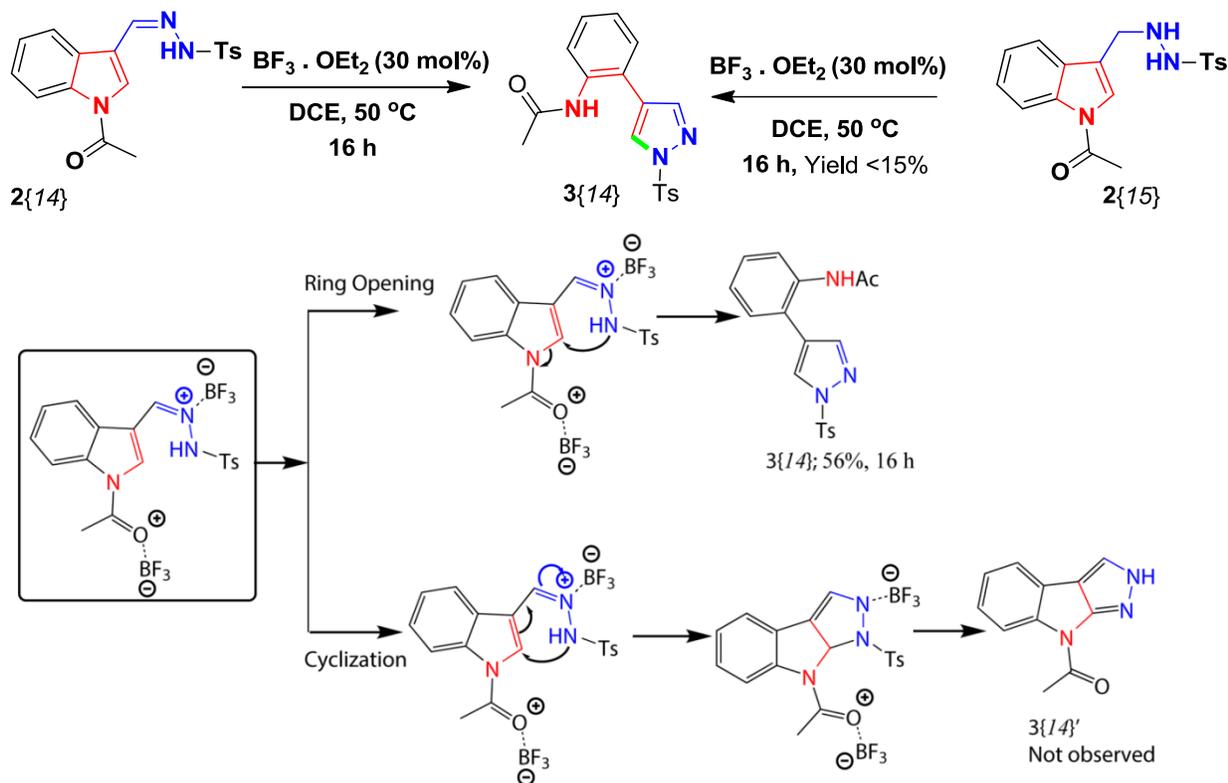
Based on the reported literatures and our results, we propose the following atypical mechanistic pathways for the synthesis of 1*H*-pyrazoles from indoles (Scheme 2). The complexation of Lewis acid ($\text{BF}_3 \cdot \text{OEt}_2$) with the oxygen of N-acyl / N-benzoyl / N-tosyl indoles sequestered the nitrogen lone-pair and allowing the activation of C2=C3 bond.^{10, 45} The

Scheme 2. Plausible coupling reaction of activated indoles with tosylhydrazones.



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3 formation of activated N-acyl indole (**I**) in the presence of $\text{BF}_3 \cdot \text{OEt}_2$ was supported by the ^{13}C
4 NMR and 2D-NMR (HSQC) experiments (Figure S2 and S3). The unsuccessful ring-opening
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6 reaction of alkyl or aryl substituted indoles also support this hypothesis. The nucleophilic attack
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8 of the NH of Lewis acid-tosylhydrazone complex at the C2 of activated indole would lead to the
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10 formation of intermediate **IIA** followed by the cleavage of C2-N1 bond (Path A).⁴⁶ Isolation of
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12 pyrazole **3**{14} from compound **2**{14} supported both the nucleophilic attack at the C2-centre
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14 and C2-N1 bond cleavage (Scheme 3). However, we failed to isolate any intermediates,
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16 indicating stronger reactivity of the reactants/intermediates under the optimized experimental
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18 reaction conditions. The intramolecular cyclization of intermediate **IIA** would lead to the
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20 formation of intermediate **IIIA** which could produce the desired pyrazoles through
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22 aromatization.
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28 The other probable pathway (Path B) describe the 3+2 cycloaddition reaction between the
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30 tosylhydrazone, (complexed with $\text{BF}_3 \cdot \text{OEt}_2$) and activated indole (intermediate I). Then the
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32 aromatization through cleavage of indole ring of intermediate **IIIB** would lead to the desired
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34 pyrazole. However, the intramolecular cyclization of compound **2**{14} failed to produce the
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36 desired pyrazole **3**{14}' indicating the preference for Path A over Path B mechanistic pathway
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39 The cyclization reaction strongly depends on the electrophilic nature of imine bond of
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41 hydrazone. The withdrawal of electron density from the C=N bond of the hydrazones should be
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43 facile for the electron deficient aryl group. Hence, the cyclization reactions for hydrazones with
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45 electron withdrawing groups are more effective in comparison with the electron rich hydrazones.
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48 We also performed the intramolecular cyclization of compound **2**{15}, which is the reduced
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50 product of compound **2**{14}. However, the compound **2**{15} also produced the pyrazole **3**{14},
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52 which could be due to the oxidized aromatization.
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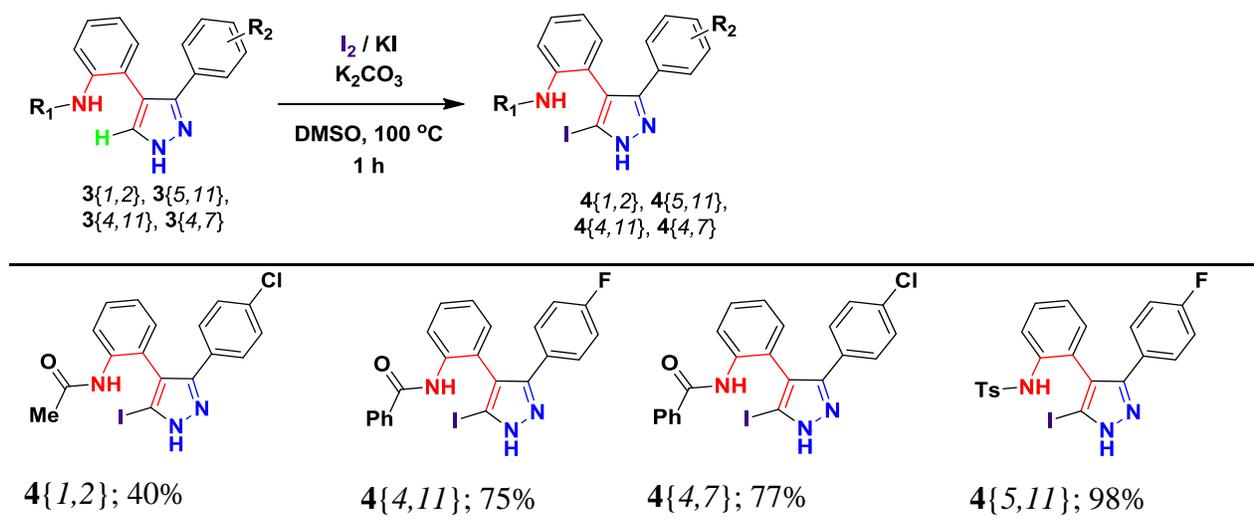
Scheme 3. Plausible mode of reaction of compound **2{14}** for the synthesis of 1*H*-pyrazole

The mechanistic studies suggested that the ring opening reaction is directed by the complex formation ability of the N-acyl (or tosyl or benzoyl) indole with the Lewis acid. The formation of complex of the Lewis acid ($\text{BF}_3 \cdot \text{OEt}_2$) with the $-\text{C}=\text{O}$ (or sulphonyl) group of the protected indole results in the cleavage of C2-N1 bond of the indole ring. The absence of such carbonyl or sulphonyl group in N1-methyl, N1-benzyl or unsubstituted (N1-H) indole failed to undergo through the ring opening and concomitant cyclization reaction with tosylhydrazine. Interestingly, these observations also demonstrate the importance of carbonyl or sulphonyl group at N1 position of the indole. We hypothesize that the formation of intermediate **III A** requires the

presence of H-atom at the imine carbon for the final aromatization reaction. For this reason, the hydrazone of acetophenone failed to produce the target product.

To demonstrate further the synthetic utility of these pyrazoles, we performed iodination reactions. The newly generated amide or sulphonamide group can act as powerful directing groups for the functionalization of the C5-position of pyrazoles (Table 3). To our delight, the selective C-H iodination of the pyrazole over the two other aromatic rings was performed with excellent yield (**4**{1,2}, **4**{4,11}, **4**{4,7} and **4**{5,11}; 40 – 98%). The carbon- iodine bond has tremendous synthetic applications in modern chemistry.⁴⁷⁻⁴⁸ The presence of sulphonamide and benzamide with the aryl ring of pyrazoles shows better result in comparison with the acetamide group for this iodination reaction, which indicates the importance of these directing groups. The directing group of compound **3**{4,11} was deprotected under basic condition using ethanolic solution at high temperature (Scheme 4). The free amine group of compound **4**{4,11} can be utilized for further functionalization.⁴⁹

Table 3. Synthetic applications of this novel class of pyrazoles.

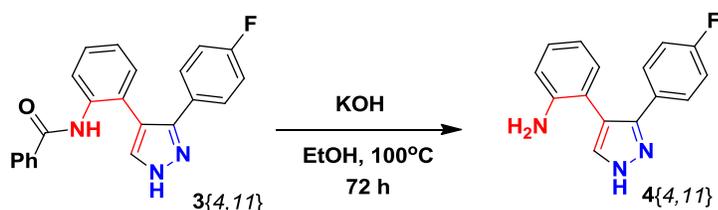


^aAll the reactions were performed using 1 equiv of **3**{1,2} or **3**{5,11} or **3**{4,11} or **3**{4,7}, 1 equiv. of iodine, 2 equiv. of KI and 2 equiv. of K_2CO_3 in the presence of 0.3 equiv. of $BF_3 \cdot OEt_2$

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3 in 1 mL of DMSO at 100 °C.
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6 Synthesis of substituted pyrazoles using such quick and easy to perform mild reaction
7 conditions is a part of our on-going heterocyclic drug discovery program for cancer
8 immunotherapy. Immunotherapeutic approach is being considered as one of the most promising
9 approaches in the battle against cancer.⁵⁰⁻⁵¹ In this regard, we explored the IDO1 inhibitory
10 activity of these compounds (HPLC purified). The IC₅₀ values were measured against purified
11 human IDO1 enzyme using standard spectrophotometric method (Table 4 and S2). Compounds,
12 **3**{2,2} (IC₅₀ = 40.33 μM) and **3**{5,1} (IC₅₀ = 25.0 μM) showed moderate IDO1 inhibitory
13 activity, which could be due to the presence of substituted aryl containing pyrazole and
14 sulfonamide moieties.⁵²⁻⁵³ Nonetheless, the inhibitory activity study suggests that this pyrazole
15 scaffold can be used as the lead compound and its modifications may lead to the development of
16 potent IDO1 inhibitors with favorable biochemical and biophysical properties.
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34 **Scheme 4.** Removal of the directing group
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46 In summary, we described an unusual ring-opening and cyclization reaction of a stable
47 aromatic heterocyclic compound. This transition-metal and ligand free synthesis of pyrazoles
48 from the corresponding indoles through the C2-N1 bond cleavage is unknown. Complexation of
49 the Lewis acid with indoles and tosylhydrazones could be the driving force for this reaction. This
50 method could be a useful alternative to the existing methods for regioselective synthesis of
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pyrazoles. The inhibitory activities of these synthesized pyrazoles were tested against immune suppressive enzyme, IDO1. The preliminary results suggest that the pyrazole moiety can be used as synthetically amenable lead for future development of IDO1 inhibitors.

Table 4. Inhibitory activity of the 1*H*-pyrazoles against purified human IDO1 enzyme.

Compound	IDO1	Compound	IDO1
	IC ₅₀ (μM) ^a		IC ₅₀ (μM) ^a
3 {1,4}	84.50 ± 3.6	3 {5,1}	25.0 ± 3.5
3 {2,2}	40.33 ± 0.3	3 {5,2}	115.5 ± 5.8
3 {4,3}	74.17 ± 2.3	3 {5,6}	96.8 ± 5.8

^aIC₅₀ values are the mean of three independent assays.

ASSOCIATED CONTENT

The Supporting Information is available free of charge on the ACS Publications website at DOI:

Spectroscopic data for all compounds; crystallographic information for compounds **3**{4,5} (PDF)

X-ray data for compounds **3**{4,5} (CIF).

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Author Contributions

The manuscript was written through contributions of all authors.

Notes

The authors declare no competing financial interest.

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ABBREVIATIONS

IDO1, indoleamine 2,3-dioxygenase; DCE, dichloroethane.

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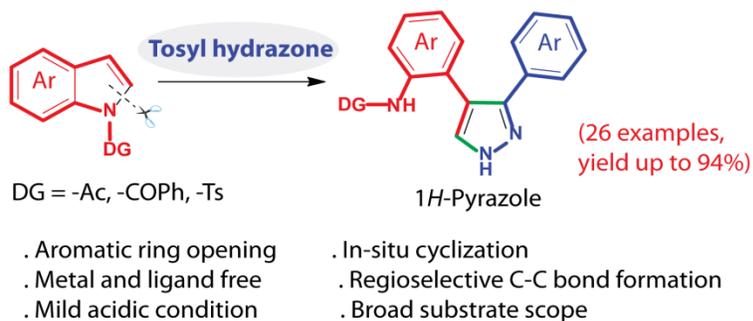
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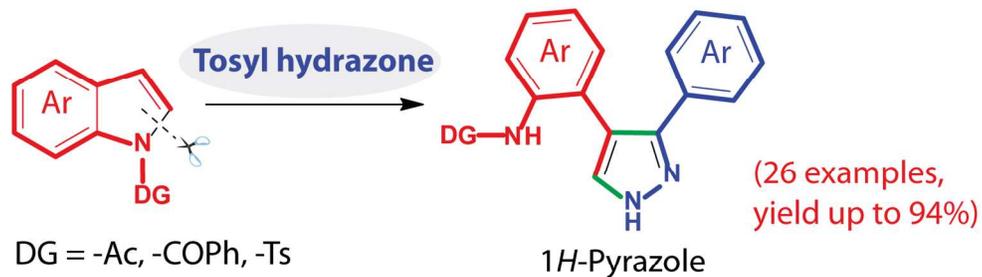
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TOC



Ring-opening: Lewis acid catalyzed ring-opening of the indoles afforded regioselective pyrazole in the presence of tosylhydrazones. *In-situ* cyclization after the cleavage of C2-N1 bond of the indoles proceeds in the absence of any transition-metal and ligand.



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- . Aromatic ring opening
 - . Metal and ligand free
 - . Mild acidic condition
 - . In-situ cyclization
 - . Regioselective C-C bond formation
 - . Broad substrate scope

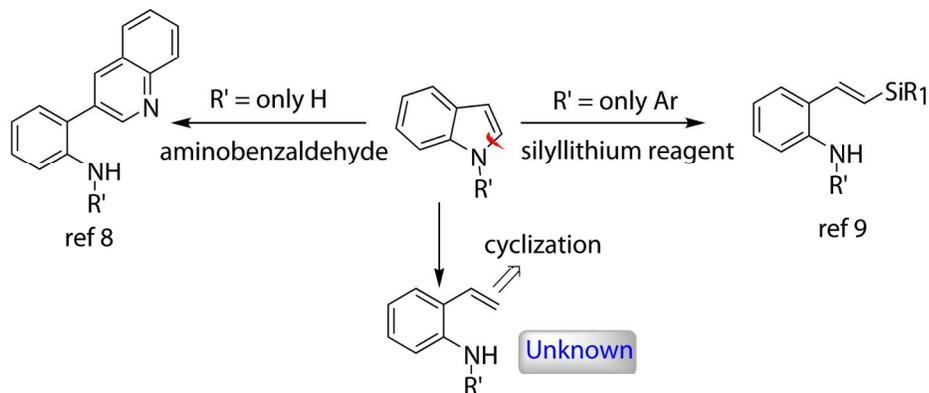
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Ring-opening: Lewis acid catalyzed ring-opening of the indoles afforded regioselective pyrazole in the presence of tosylhydrazones. In-situ cyclization after the cleavage of C2-N1 bond of the indoles proceeds in the absence of any transition-metal and ligand.

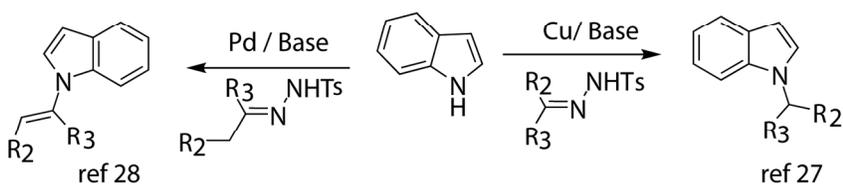
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146x68mm (300 x 300 DPI)

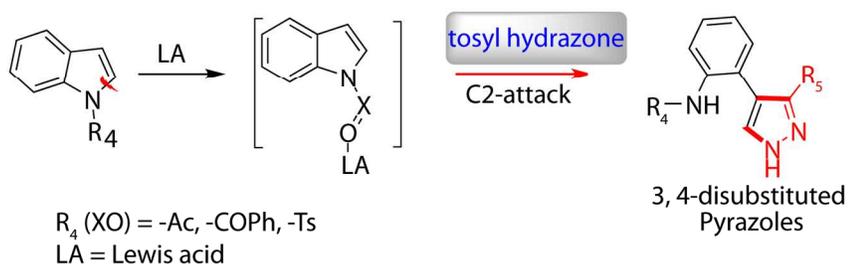
a) Unconventional Ring Opening of Indole



b) Reactivity of tosylhydrazone with Indole

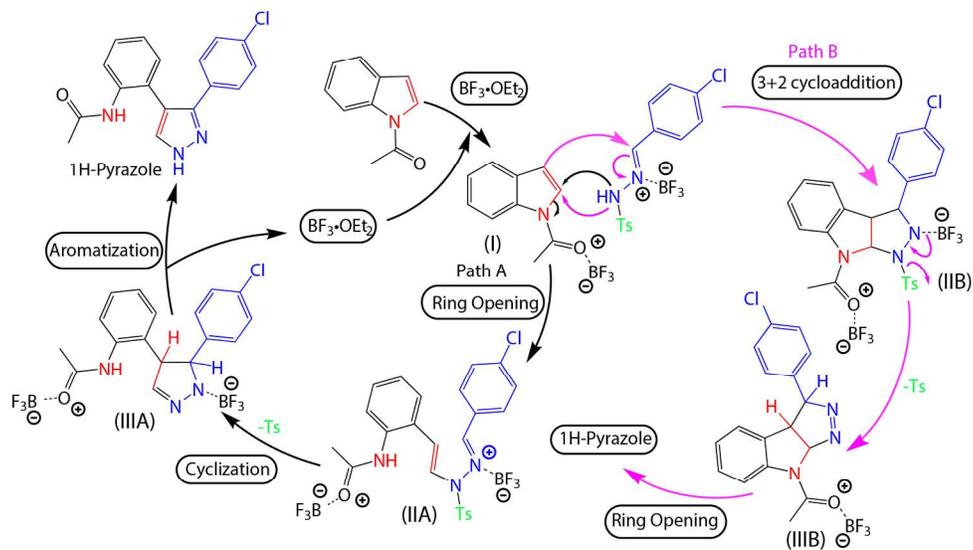


c) Our work



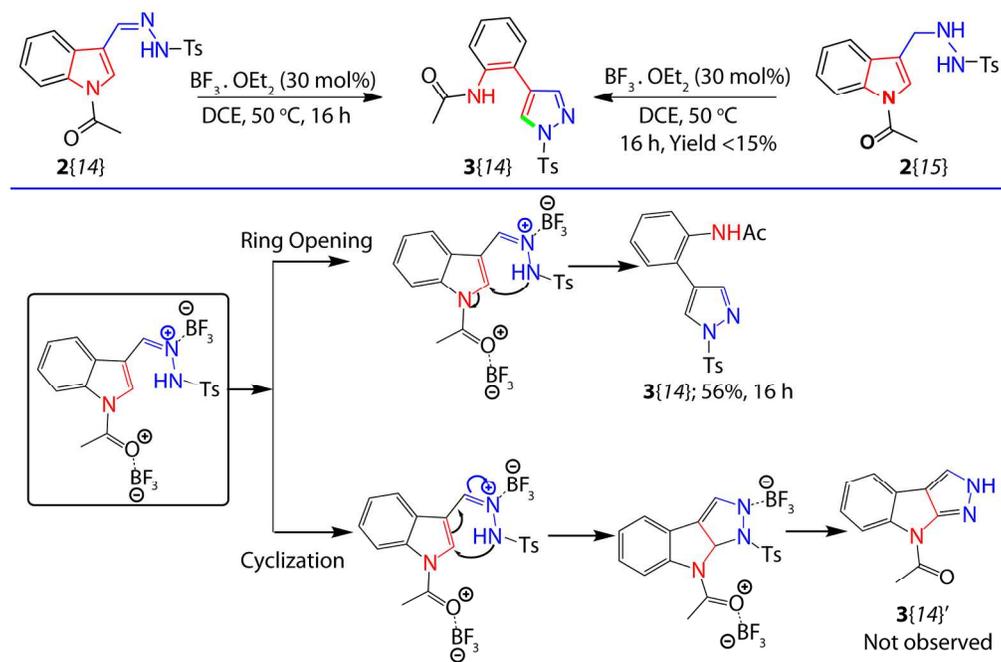
C2-N1 bond opening and the reactivity of indole towards hydrazone.

158x167mm (300 x 300 DPI)



Plausible coupling reaction of activated indoles with tosylhydrazones.

188x119mm (300 x 300 DPI)



29 Plausible mode of reaction of compound **2{14}** for the synthesis of 1H-pyrazole.

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31 183x125mm (300 x 300 DPI)