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New Reaction Mode of the Horner-Wadsworth-Emmons Reaction for the Preparation of α -Fluoro- α , β -unsaturated Esters

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Abstract: Excellent *E*-selectivity was observed in the Horner-Wadsworth-Emmons (HWE) reactions of ethyl 2-fluoro-2-diethylphosphonoacetate **1** with alkyl aryl ketones **2a-f** using Sn(OSO₂CF₃)₂ and *N*-ethylpiperidine. Mg(II)-promoted HWE reactions of **1** with aldehydes **2h,i** afforded α -fluoro- α , β -unsaturated esters **3h,i** in a *Z*-selective manner depending on the reaction temperatures.

 α -Fluoro- α , β -unsaturated esters play an important role in the preparation of biologically active fluorine compounds, and various synthetic methods have been developed.¹ Most of the construction methods for the α -fluoro- α , β -unsaturated esters have shown Zselectivity.² However, Horner-Wadsworth-Emmons (HWE) reactions of aldehydes with ethyl 2-fluoro-2-diethylphosphonoacetate 1 can preferentially furnish E- α -fluoro- α , β -unsaturated esters.³⁻⁵ There have been a few reports on the HWE reactions with ketones in the presence of some base but the stereoselectivity is not clear or poor.⁶ Recently, we have developed a new reaction mode of the HWE reaction using Sn(OSO₂CF₃)₂ and N-ethylpiperidine to obtain excellent Z-selectivity in the reactions of methyl bis(trifluoroethyl)phosphonoacetate with l alkyl aryl ketones.⁷ We now wish to report highly E-selective HWE reactions of ethyl 2-fluoro-2-diethylphosphonoacetate 1 with alkyl aryl ketones 2a-f using $Sn(OSO_2CF_3)_2$ and N-ethylpiperidine as shown in Scheme 1. A tendency to afford Z-alkenes in the Mg(II)-mediated HWE reactions of fluorophosphonate 1 with aldehydes is also described. All reaction conditions and results are summarized in Tables 1-3.





The HWE reactions of alkyl aryl ketones **2a-f** with **1** in the presence of sodium hydride in THF 0 °C gave the corresponding α -fluoro- α , β -unsaturated esters **3a-f** with modest *E*-selectivity (Table 1, entries 1-6). On the other hand, treatment of **2a-f** with **1** in the presence of Sn(OSO₂CF₃)₂ and *N*-ethylpiperidine in CH₂Cl₂ at 0 °C afforded alkenes **3a-f** in a highly *E*-selective manner, respectively (Table 2,

| Table 1. NaH / THF mediated Horner-Wadsworth-Emmons reactions |
|---|
| of 1 with ketones 2a-g and aldehydes 2h,i ^a |

| Entry | Ketone or Aldehyde | <i>t /</i> h | Yield (%) ^b | Alkene (<i>E/Z</i>) ^c |
|-------|------------------------|--------------|---------------------------|---------------------------------------|
| 1 | 2a | 1 | 78 | 3a (85 : 15) |
| 2 | 2b | 2 | 93 | 3b (87 : 13) |
| 3 | 2c | 18 | 49 | 3c (78:22) |
| 4 | 2d ^{<i>d</i>} | 43 | 40 | 3d (72:28) |
| 5 | 2e | 1 | 97 | 3e (86 : 14) |
| 6 | 2f | 1 | 84 | 3f (90:10) |
| 7 | 2g | 1 | 86 | 3g (49:51) ^e |
| 8 | 2h | 1 | 82 | 3h (83 : 17) |
| 9 | 2 i | 1 | 82 | 3i (83 : 17) |

^a Conditions: THF, 0 °C, 1 / NaH / 2 (1.7 : 1.5 :1)

^b Isolated yield

^{c 1}H NMR (400 MHz, CDCl₃) analysis

^d Reflux

^e HPLC (TSK-GEL Silica 60, hexane - propan-2-ol) analysis

| aldenydes 2n,I" | | | | | | | |
|-----------------|-----------------------|---------------------------------|--------------|---------------------------|---------------------------------------|--|--|
| Entry | Ketone or Aldehyde | Solvent | <i>t /</i> h | Yield (%) ^b | Alkene (<i>E/Z</i>) ^c | | |
| 1 | 2a | CH ₂ Cl ₂ | 23 | 91 | 3a (97 : 3) | | |
| 2 | 2b | CH ₂ Cl ₂ | 21 | 92 | 3b (99:1) | | |
| 3 | 2b | THF | 18 | 97 | 3b (93 : 7) | | |
| 4 | 2c | CH ₂ Cl ₂ | 20 | 95 | 3c (97:3) | | |
| 5 | 2d | CH ₂ Cl ₂ | 20 | 97 | 3d (98 : 2) | | |
| 6 | 2e | CH ₂ Cl ₂ | 19 | 94 | 3e (98 : 2) | | |
| 7 | 2f | CH ₂ Cl ₂ | 22 | 73 | 3f (99 : 1) | | |
| 8 | 2g | CH ₂ Cl ₂ | 22 | 86 | 3g (54 : 46) ^a | | |
| 9 | 2h | THF | 14 | 92 | 3h (84 : 16) | | |
| 10 | 2i | THF | 16 | 86 | 3i (95 : 5) | | |

Table 2. $Sn(OSO_2CF_3)_2 / N$ -ethylpiperidine / THF or CH_2Cl_2 mediatedHorner-Wadsworth-Emmons reactions of 1 with ketones 2a-g andaldehydes $2h.i^a$

 a Conditions: THF or CH_2Cl_2, 0 °C, 1 / Sn(OSO_2CF_3)_2 / N-ethylpiperidine / (1.4 : 1.68 : 1.54 : 1)

^b Isolated yield

^{c 1}H NMR (400 MHz, CDCl₃) analysis

^d HPLC (TSK-GEL Silica 60, hexane - propan-2-ol) analysis

Table 3. MgBr_2 / Et_3N / THF or i-PrMgBr / THF mediated Horner-Wadsworth-Emmons reactions of 1 with aldehydes $2h,i^a$

| Entry | Aldehyde | Conditions | T/°C | Yield (%) ^b | Alkene (<i>E/Z</i>) ^c |
|-------|------------|------------|--------|---------------------------|---------------------------------------|
| 1 | 2h | (A) | 0 | 89 | 3h (19:81) |
| 2 | 2h | (B) | reflux | 86 | 3h (25 : 75) |
| 3 | 2h | (B) | 40 | 95 | 3h (26:74) |
| 4 | 2h | (B) | 0 | 82 | 3h (23:77) |
| 5 | 2h | (B) | -40 | 43 | 3h (33:67) |
| 6 | 2h | (B) | -78 | 34 | 3h (51 : 49) |
| 7 | 2h | (B) | -100 | 13 | 3h (64 : 36) |
| 8 | 2i | (A) | 0 | 77 | 3i (53 : 47) |
| 9 | 2i | (B) | reflux | 74 | 3i (37:63) |
| 10 | 2i | (B) | 0 | 82 | 3i (70 : 30) |
| 11 | 2 i | (B) | -78 | 12 | 3i (95 : 5) |

^a Conditions: (A); THF, 1 h, 1 / MgBr₂ / Et₃N / 2 (1.4 : 1.68 : 1.54 : 1), (B); THF, 1 h, 1 /i-PrMgBr / 2 (1.7 : 1.5 : 1)

^b Isolated yield

^{c1}H NMR (400 MHz, CDCl₃) analysis

entries 1,2, and 4-7).⁸ The geometry of α -fluoro- α , β -unsaturated esters **3a-g** was assigned on the basis of ¹H-¹H NOE experiments (400 MHz, CDCl₃) of the corresponding primary alcohols derived by reduction of 3a-g with DIBAL-H in CH₂Cl₂ at 0 °C. When THF was employed as the solvent instead of CH2Cl2 in the same Sn(II)-promoted reaction of alkyl aryl ketone 2b with 1, E-selectivity of the alkenic product 3b decreased from 99 : 1 to 93 : 7. Interestingly, the NaH-promoted reactions of the ketones 2c,d bearing bulky i-Pr or t-Bu group in THF was hard to obtain the highly *E*-selective products **3c**,**d** (Table 1, entries 3 and 4), while the Sn(II)-promoted reactions of 2c,d in CH₂Cl₂ gave 3c,d in an excellent E-selective manner and in good yields (Table 2, entries 4 and 5) as well as those of the ketones 2a,b. Thus, the stereochemical outcome with high E-selectivity in the Sn(OSO₂CF₃)₂mediated HWE reactions of ketones 2a-f with fluorophosphonate 1 can be rationalized in terms of a six-membered transition state as shown in Fig. 1 (e.g. 2b).⁷ The stereoselectivity in the NaH-promoted HWE reactions of 2a-g with 1 seems to depend on the relative bulkiness between both substituents of the ketones in a plausible non-chelationcontrolled transition state. Under both reaction conditions as described above, the HWE reactions of ketone 2g with 1 gave a ca. 1 : 1 mixture of E- and Z-isomers of 3g (Table 1, entry 7; Table 2, entry 8), respectively.

Subsequently, the similar Sn(II)- and NaH-promoted reactions of aldehydes **2h**,**i** with fluorophosphonate **1** afforded the corresponding α -fluoro- α , β -unsaturated esters **3h**,**i** with a good or excellent *E*-selectivity (Table 1, entries 8 and 9; Table 2, entries 9 and 10). The geometry of **3h**,**i** was determined by the coupling constants between fluorine and the adjacent olefinic proton in their ¹H NMR analysis (400 MHz, CDCl₃) as follows; *E*-**3h** ($J_{\text{H-F}} = 22.3 \text{ Hz}$, lit.⁵ 22 Hz), *Z*-**3h** ($J_{\text{H-F}} = 35.1 \text{ Hz}$), *E*-**3i** ($J_{\text{H-F}} = 21.4 \text{ Hz}$), and *Z*-**3i** ($J_{\text{H-F}} = 32.8 \text{ Hz}$). Previously, we reported a characteristic reaction mode with Mg(II) in the aldol type reactions,



Figure 1. Plausible six-membered transition state involving Sn(II) chelation

which was quite different from that with Sn(II).⁹ Thus, tentative HWE reactions of benzaldehyde **2h** with **1** at various temperatures employing MgBr₂-triethylamine or *i*-PrMgBr gave Z-selective **3h** (Table 3, entries 1-5), which was contrary to the *E*-selectivity in the cases of Sn(II)- and NaH-promoted reactions at 0 °C (Table 1, entry 8; Table 2 entry 9). The Mg(II)-promoted HWE reaction of **2i** with **1** gave **3i** in the modest *Z*-selective manner (Table 3, entry 9). Interestingly, the stereoselectivity of α -fluoro- α , β -unsaturated esters **3h**,**i** in the Mg(II)-promoted reactions of aldehydes **2h**,**i** with **1** was variable depending on the reaction temperatures. The *Z*-selective manner (Table 3, entries 2-5 and 9) in the Mg(II)-promoted reactions of **2h**,**i** changed to *E*-selective one (Table 3, entries 7, 10, and 11) with decreasing the reaction temperature.⁴ These results made it possible to prepare the *Z*-isomers of α -fluoro- α , β -unsaturated esters by exploiting the Mg(II)-promoted HWE reaction.

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- (8) A typical procedure is as follows. To a suspension of $Sn(OSO_2CF_3)_2$ (820 mg, 1.97 mmol) and 2-fluoro-2-

diethylphosphonoacetate **1** (330 µL, 1.64 mmol) in anhydrous CH₂Cl₂ (5 mL) was added *N*-ethylpiperidine (247 µL, 1.80 mmol) at 0 °C. The mixture was stirred at 0 °C for 1 h under argon, and alkyl aryl ketone **2b** (155 µL, 1.17 mmol) was added. After being stirred at 0 °C for 21 h under argon, the reaction mixture was poured into H₂O (15 mL) and then extracted with CHCl₃ (3 x 50 mL). To the CHCl₃ extract was added *n*-hexane (300 mL), and the mixture was submitted to filtration through a silica gel short column. The filtrate was evaporated *in vacuo* to afford a crude product **3b** (*E* : *Z* = 99 : 1), which was purified by chromatography on a silica gel column eluting with *n*-hexane/AcOEt (20 : 1) to obtain α -fluoro- α , β -unsaturated esters *E*-**3b** (37.6 mg, 91%) and *Z*-**3b** (2.4 mg, 1%) as a pale yellow oil, respectively. *E*-**3b**: ¹H NMR (400 MHz, CDCl₃) δ 0.99 (3 H, t, *J* = 7.3 Hz), 1.00 (3 H, t, *J* = 7.3 Hz), 2.55 (2 H, dq, ⁴*J*_{H,F} = 3.6 Hz, *J* = 7.3 Hz), 4.02 (2 H,

q, *J* = 7.3 Hz), 7.10-7.17 (2 H, m), 7.30-7.38 (3 H, m); IR (NaCl) 2980, 2937, 1729, 1656, 1444, 1265, 1151, 763, 700 cm⁻¹; HREI-MS calcd for $C_{13}H_{15}O_2F$ MW 222.1056, found *m/e* 222.1057 (M⁺); Anal. Calcd for $C_{13}H_{15}O_2F$: C, 70.25; H, 6.80. Found: C, 69.96; H, 6.92. *Z*-**3b**: ¹H NMR (400 MHz, CDCl₃) δ 1.03 (3 H, t, *J* = 7.3 Hz), 1.37 (3 H, t, *J* = 7.3 Hz), 2.90 (2 H, dq, ⁴*J*_{H,F} = 1.8 Hz, *J* = 7.3 Hz), 4.34 (2 H, q, *J* = 7.3 Hz), 7.28-7.43 (5 H, m); IR (NaCl) 2975, 2932, 1725, 1642, 1445, 1247, 1142, 766, 699 cm⁻¹; HREI-MS calcd for $C_{13}H_{15}O_2F$ MW 222.1056, found *m/e* 222.1069 (M⁺).

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