

## A STEREOCONTROLLED APPROACH TO 3-DEOXY-D-MANNO-2-OCTULOSONIC ACID CONTAINING DISACCHARIDES <sup>1</sup>

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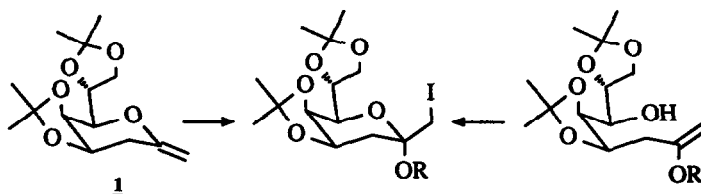
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**Summary:** The development of a synthetic approach to KDO containing disaccharides based on stereocontrolled iodocyclization of an enol ether is detailed.

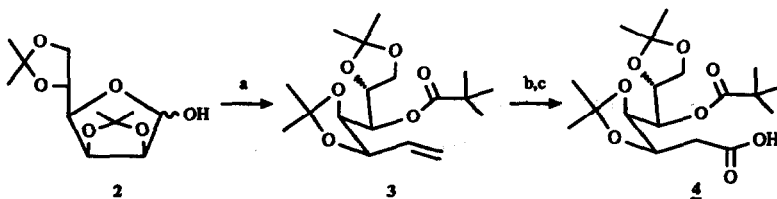
3-Deoxy-D-manno-2-octulosonic acid (KDO) occurs as a characteristic component of enterobacterial lipopolysaccharides (LPS) and has also been found in several acidic exopolysaccharides (K antigens), both located at the cell surface of Gram-negative bacteria <sup>2</sup>. The incorporation of KDO appears to be a vital step in growth of the bacteria. Therefore the synthetic chemistry involving KDO has become of increasing interest.

The synthesis of KDO-containing oligosaccharides has largely been approached so far by conventional glycosylation procedures <sup>3</sup> involving either methyl 3-deoxy-4,5,7,8-tetra-*O*-acetyl- $\alpha$ -D-manno-octulopyranosonate chloride, bromide, fluoride or the corresponding glycal. Anomerically selective *O*-alkylation has also been described <sup>4</sup>. Other solutions to this synthetic problem involve the use of a furan ring as a surrogate for the carboxylic acid residue of KDO <sup>5</sup> and the intramolecular oximercuration-demercuration of an appropriate acyclic precursor <sup>6</sup>. We would like to present a new solution, where the key step is a stereospecific intramolecular iodocyclization of a complex enol ether <sup>7</sup>.

We first considered the "exo-glycal" **1** <sup>8</sup> as a possible candidate for iodonium mediated glycosylation of "sugar alcohols". This approach was disappointing <sup>9</sup>, so that the development of the intramolecular version was a logical issue and indeed proved a successful stratagem:

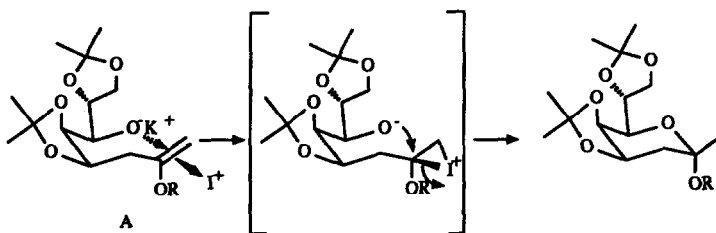


The carboxylic acid **4**, a general precursor of the KDO unit, was first prepared in three steps from known <sup>10</sup> 2,3,5,6-di-*O*-isopropylidene-D-mannose **2**. Wittig reaction followed by "in situ" pivaloylation (a,  $\text{Ph}_3\text{P}=\text{CH}_2$ , THF, HMPA then pivaloyl chloride) gave the derivative **3** (71%) <sup>11</sup>,  $[\alpha]_D -22^\circ$ . Hydroboration (b, 9-BBN then NaOH,  $\text{H}_2\text{O}_2$  <sup>12</sup>), followed by oxidation (c,  $\text{RuCl}_3$ ,  $\text{NaIO}_4$ ,  $\text{CH}_3\text{CN}-\text{CCl}_4-\text{H}_2\text{O}$  <sup>13</sup>) gave the carboxylic acid **4** (91%).



The carboxylic acid **4** was immediately esterified (a, ROH, DCC, DMAP, 1h, room temperature) with two "sugar alcohols", methyl 2,3,4-tri-*O*-benzyl- $\alpha$ -D-glucopyranoside <sup>14</sup> and 1,2,5,6-di-*O*-isopropylidene- $\alpha$ -D-glucofuranose <sup>10</sup> (Scheme 1); **5a** and **5b** were obtained (86% and 88%). Methylenation of the ester group (b, Tebbe's reagent, pyridine, 30min,  $-78^{\circ}\text{C}$  <sup>15</sup>) gave derivatives **6a** and **6b** (81% and 87%). Depivaloylation (c,  $\text{LiAlH}_4$ , THF, 1h, room temperature) gave the complex enol ethers **7a** and **7b** (93% and 90%). The key iodocyclization (d,  $\text{tBuOK}$ ,  $\text{I}_2$ , 1h,  $-78^{\circ}\text{C}$  <sup>7</sup>) gave the  $\alpha$ -glycosides **8a** and **8b** as only detectable isomers in 92% and 90% yield.

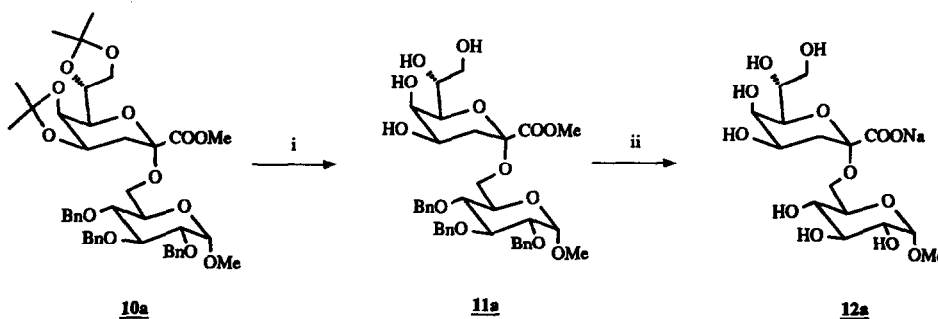
The remarkable stereoselectivity of the cyclization reaction may be rationalized in assuming (i) a chair product like  $^5\text{C}_2(\text{D})$  conformation for the transition state and (ii) a conformational preference of **7** as A:

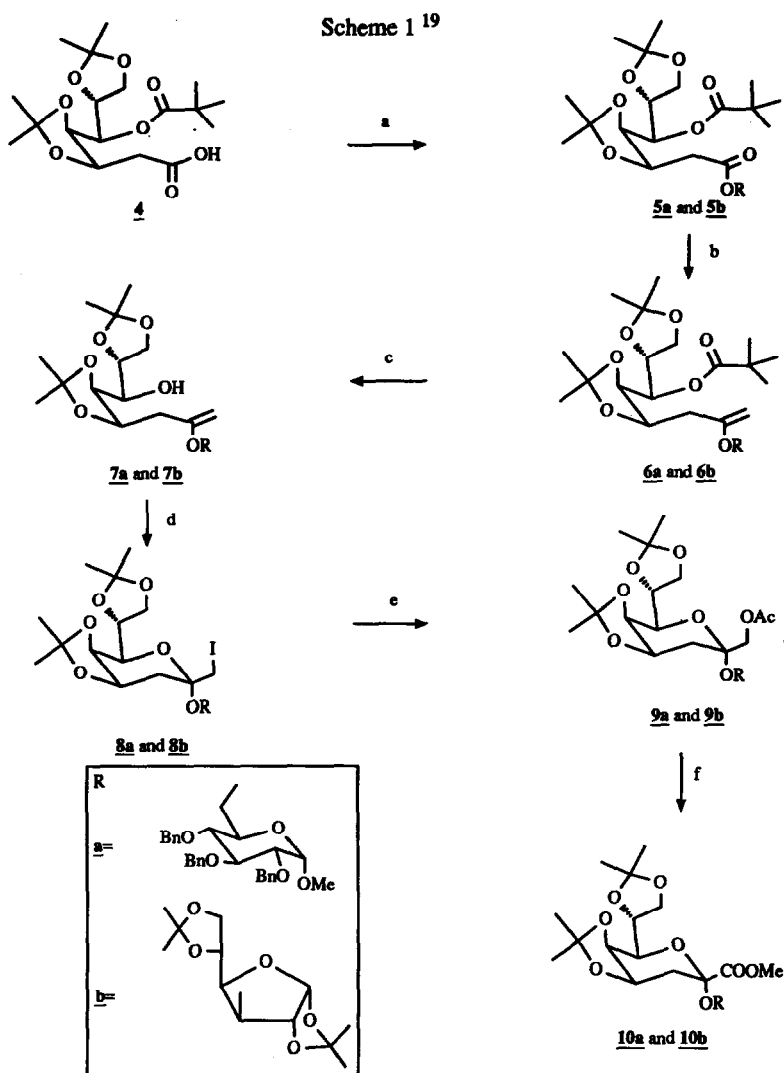


$\alpha$ -configuration of **8a** and **8b** was assigned from the characteristic chemical shift of H-4 <sup>16</sup>.

The  $\text{CH}_2\text{I}$ -appendage was modified as follows: nucleophilic displacement under drastic conditions (e,  $\text{CsOAc}$ , HMPA, 36h,  $140^{\circ}\text{C}$ ) gave the acetates **9a** and **9b** (68% and 84%), which were easily converted into **10a** and **10b** (f,  $\text{MeONa}$ ,  $\text{MeOH}$ , 2h, room temperature, then  $(\text{COCl})_2$ ,  $\text{DMSO}$ ,  $\text{Et}_3\text{N}$ ,  $-78^{\circ}\text{C}$  <sup>17</sup>, then  $\text{NaClO}_2$ ,  $\text{H}_2\text{O}_2$  30%,  $\text{CH}_3\text{CN}$ , 12h, room temperature <sup>18</sup>, then  $\text{CH}_2\text{N}_2$ ,  $0^{\circ}\text{C}$ , 67% and 62%).

Compound **10a** was converted (i,  $\text{CF}_3\text{COOH}$ ,  $\text{MeOH}$ ,  $\text{H}_2\text{O}$ , 48h, room temperature) into **11a** (90%), which, after catalytic hydrogenolysis (ii,  $\text{H}_2$ ,  $\text{Pd/C}$ ) and saponification ( $\text{NaOH}$ ), provided the known KDO-disaccharide **12a** <sup>6</sup>:





1. Part of this work was presented at the 5<sup>th</sup> European Symposium on Carbohydrates (Eurocarb V), Prague, August 1989, A-141 and at *Journées de Chimie Organique* (JCO 89), Palaiseau, September 1989, A-233.
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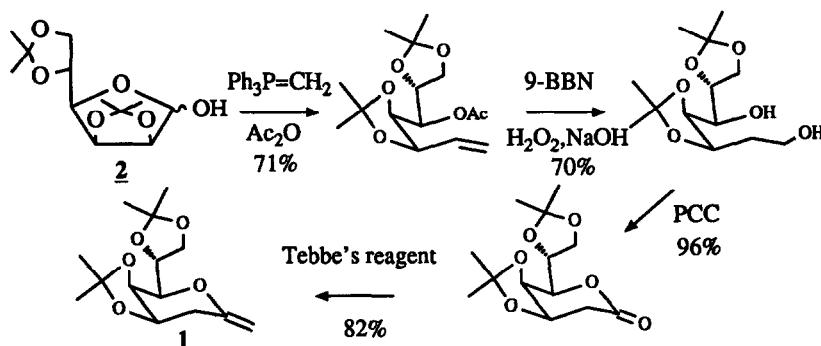
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5. S.J. Danishefsky, M.P. DeNinno and S. Chen, *J. Am. Chem. Soc.*, 110, 3929 (1988).

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7. While this work was in progress, a similar reaction has independently been reported: A.G.M. Barrett, B.C.B. Bezuidenhout, A.F. Gasiecki, A.R. Howell and M.A. Russell, *J. Am. Chem. Soc.*, 111, 1392 (1989).

8. The compound **1**,  $[\alpha]_D +2^\circ$ , has been prepared from 2,3-5,6-di-*O*-isopropylidene-D-mannose **2**<sup>10</sup> as follows:



9. We found in a model study that NIS mediated addition reaction on fully benzylated 1-methylene-D-glucose occurred only with reactive alcohols. See also: D. Noort, G.H. Veeneman, G.J.P.H. Boons, G.A. van der Marel, G.J. Mulder, J.H. van Boom, *Synlett*, 205 (1990).

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11. All new compounds had satisfactory microanalytical and spectral properties. Optical rotations were measured for solutions in chloroform at  $20^\circ\text{C}$ , unless otherwise stated.

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16. **8a**:  $\delta$  4.45 (ddd,  $J_{5,4}$  2.5,  $J_{3a,4}$  11.5,  $J_{3e,4}$  3.5 Hz); **8b**:  $\delta$  4.49 (ddd,  $J_{5,4}$  3.5,  $J_{3a,4}$  11.5,  $J_{3e,4}$  3 Hz).

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19. **5a**:  $[\alpha]_D +11^\circ$ ; **5b**:  $[\alpha]_D -14^\circ$ ; **6a**:  $[\alpha]_D +6^\circ$ ; **6b**:  $[\alpha]_D -23^\circ$ ; **7a**:  $[\alpha]_D +44^\circ$  (c 0.9, pyridine); **7b**:  $[\alpha]_D -13^\circ$ ; **8a**:  $[\alpha]_D +12^\circ$ ; **8b**:  $[\alpha]_D +5^\circ$ ; **9a**:  $[\alpha]_D +26^\circ$ ; **9b**:  $[\alpha]_D -2^\circ$ ; **10a**:  $[\alpha]_D +51^\circ$ , m.p.  $101^\circ\text{C}$  (from hexane); **10b**:  $[\alpha]_D +1^\circ$ ; **11a**:  $[\alpha]_D +38^\circ$ , m.p.  $100^\circ\text{C}$  (from hexane).