

# Synthesis and Growth Inhibitory Properties of Glucosamine-Derived Glycerolipids

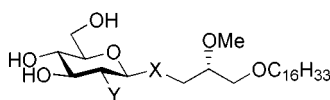
Guangli Yang,<sup>†</sup> Richard W. Franck,<sup>\*,†</sup> Robert Bittman,<sup>‡</sup> Pranati Samadder,<sup>§</sup> and Gilbert Arthur<sup>§</sup>

Department of Chemistry, Hunter College/CUNY, 695 Park Avenue,  
New York, New York 10021, Queens College/CUNY, Flushing, New York 11367, and  
Department of Biochemistry and Medical Genetics, University of Manitoba,  
Winnipeg, Manitoba, Canada R3E 0W3

rfranck@shiva.hunter.cuny.edu

Received October 26, 2000

## ABSTRACT



**1a** X = O, Y = NH<sub>2</sub> antiproliferative

**1b** X = CH<sub>2</sub>, Y = NH<sub>2</sub> an active C-analog

2-Amino C-glycerolipid **1b** was synthesized by using the Ramberg–Bäcklund rearrangement as the key step.  $\beta$ -C-Glycerolipid **1b** exhibits in vitro antiproliferative effects strikingly similar to those of O-glycoside analogue **1a**.

The study of C-glycoside analogues of bioactive O- and N-glycosides is a mature field.<sup>1</sup> Since the publication of the two books and several monograph chapters cited in ref 1, hundreds of articles have appeared. A great deal of work has focused on the structural and conformational properties of C-glycosides as probes of the anomeric and exo-anomeric effects. C-Glycosides are essentially inert to degradation by glycosidases because the anomeric carbon has been transformed from a hydrolytically labile O- or N-acetal linkage to an ether linkage. The underlying assumption for the use of C-glycoside analogues in glycobiology is that the conformational differences between the O- (or N-) linked natural material and the C-linked analogue will be minimal. The corollary to the minimal difference hypothesis is that the recognition and binding of the C-analogue will be similar to that of the natural material.

Until now, in contrast to the large number of C-glycosides that have been synthesized, few direct comparisons of O vs C biological activity have been made.<sup>2</sup> The most thorough comparison has been done for the C-lactose/O-lactose pair reported in significant papers in 1995, 1996, and 1998 by

(2) We are focusing on comparisons between O-glycosides and other identical materials with the simple replacement of glycoside O by CH<sub>2</sub>. We use the word “exact” to characterize the analogue. It is not possible to review the many bioactive C-glycoside materials that do not have an exact O-analogue for comparison. To cite a few notable cases: (a) Schmidt, R. R.; Dietrich, H. *Angew. Chem., Int. Ed. Engl.* **1991**, *30*, 1328–1329. (b) Michael, K.; Wittman, V.; König, W.; Sandow, J.; Kessler, H. *Int. J. Peptide Protein Res.* **1996**, *48*, 59–70. (c) Nagy, J. O.; Wang, P.; Gilbert, J. H.; Schefer, M. E.; Hill, T. G.; Callstrom, M. R.; Bednarski, M. D. *J. Med. Chem.* **1992**, *35*, 4501–4502. (d) Vyplel, H.; Scholz, D.; Macher, I.; Schindlmaier, K.; Schutze, E. *J. Med. Chem.* **1991**, *34*, 2759–2767. Refs 1a and 1b should be consulted for in-depth surveys of bioactive C-glycosides.

(3) Espinosa, J.-F.; Canada, F. J.; Asensio, J. L.; Martin-Pastor, M.; Dietrich, H.; Martin-Lomas, M.; Schmidt, R. R.; Jimenez-Barbero, J. *J. Am. Chem. Soc.* **1996**, *118*, 10862–10871. (b) Espinosa, J.-F.; Montero, E.; Vian, A.; Garcia, J. L.; Dietrich, H.; Schmidt, R. R.; Martin-Lomas, M.; Imberty, A.; Canada, F. J.; Jimenez-Barbero, J. *J. Am. Chem. Soc.* **1998**, *120*, 1309–1318.

(4) Wei, A.; Haudrechy, A.; Audin, C.; Hyuk-Sang, J.; Haudrechy-Bretel, N.; Kishi, Y. *J. Org. Chem.* **1995**, *60*, 2160–2169.

(5) Ravishankar, R.; Surolia, A.; Vijayan, M.; Lim, S.; Kishi, Y. *J. Am. Chem. Soc.* **1998**, *120*, 11297–11303.

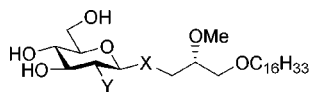
<sup>†</sup> Hunter College/CUNY.

<sup>‡</sup> Queens College/CUNY.

<sup>§</sup> University of Manitoba.

(1) Postema, M. H. D. *C-Glycoside Synthesis*; CRC Press: Boca Raton, FL, 1995. (b) Levy, D.; Tang, C. *The Synthesis of C-Glycosides*; Pergamon: Oxford, U.K.; 1995. (c) Beau, J.-M.; Gallagher, T. *Top. Curr. Chem.* **1997**, *187*, 1–54. (d) Nicotra, F. *Top. Curr. Chem.* **1997**, *187*, 55–83. (e) Du, Y.; Linhardt, R. J.; Vlahov, I. R. *Tetrahedron* **1998**, *54*, 9913–9959.

the Kishi and Schmidt and Jimenez-Barbero groups, which focused on NOE data and modeling results. There is partial but not complete agreement with regard to the similarities and differences in the conformation of ground-state and of binding conformations.<sup>3–5</sup> Nonetheless, the  $K_i$  values for *O*- and *C*-lactose for the competitive inhibition of  $\beta$ -galactosidase-catalyzed cleavage of *p*-nitrophenyl galactose are 1 and 3  $\mu$ M, respectively, which suggest a close similarity if not perfect identity of the two materials in their binding to the enzyme. Three other close comparisons include binding of a blood group trisaccharide to a leguminous lectin,<sup>6</sup> oligo- $\beta$ -1,6-galactosides to three monoclonal immunoglobulins,<sup>7</sup> and a trimannose analogue containing one *C*-linkage to concanavalin A.<sup>8</sup> In the first two of these comparisons, the affinities of the *O*-glycosides and their exact *C*-analogues were essentially identical. In the last report, the binding decreased by 66-fold (from 3  $\mu$ M for the *O*-trimannose to 198  $\mu$ M for the mono-*C*-analog). An early comparison is that between the antitumor activity of daunomycin and its nonexact *C*-analogue, with ED<sub>50</sub> values vs L1210 cells of 0.013 and 4  $\mu$ M, respectively.<sup>9</sup> Recently we had reported a comparison between an antiproliferative 2-deoxyglucosyl glycerolipid and its exact *C*-analogue in which the *C*-glycoside showed a severalfold weaker activity.<sup>10</sup> We now wish to describe an example in which the *O*- and *C*-glycerolipids of glucosamine display very similar micromolar antiproliferative activity against nine tumor cell lines.



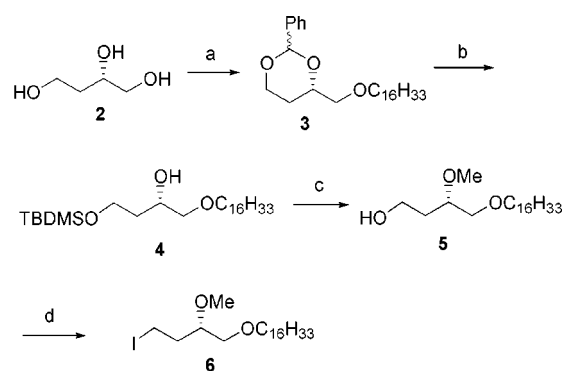
**1a** X = O, Y = NH<sub>2</sub>; **1b** X = CH<sub>2</sub>, Y = NH<sub>2</sub>; **1c** X = CH<sub>2</sub>, Y = H

Our plan was to compare glucosamine derivatives **1a** and **1b** since we had shown earlier that *O*-glycoside **1a** had micromolar antiproliferative activity in assays against several tumor cell lines.<sup>11</sup> This lead compound had been prepared via a zinc chloride catalyzed version of the Koenigs–Knorr reaction of 1-chlorotetra-*O*-acetylglucosamine (the intermediate in the conversion of **7** to **8**) and the appropriate modified glycerol. For the preparation of **1b**, we chose to test the Ramberg–Bäcklund (RB)<sup>12</sup> method for the synthesis of *C*-glycosides, under development by both our group and

Taylor's.<sup>13</sup> After the rather guarded outlook for the synthesis of *C*-glycosides of 2-amino sugars that was expressed in 1996,<sup>14</sup> several useful approaches have been reported.<sup>15</sup> However, we believed that our method offers both simplicity and certain  $\beta$ -anomeric selectivity.

Previously, we had synthesized 2-deoxy *C*-glycoside **1c** by introducing a methyl ether into its thioglycoside precursor via *O*-methylation of the side chain hydroxyl immediately prior to the RB rearrangement. The corresponding *O*-methylation step is not clean in the 2-acetaminoglucose series because *N*-methylation also takes place. Therefore, the sequence was modified by installing the *O*-methyl group before the thioglycoside was prepared. The synthesis of the lipid (*S*)-4-*O*-hexadecyl-3-*O*-methyl-1-iodobutane (**6**) was easily accomplished starting from (*S*)-(-)-1,2,4-butanetriol **2** (Scheme 1). This procedure is based on selective protection

**Scheme 1<sup>a</sup>**



<sup>a</sup> Reagents and conditions: (a) ref 10; (b) (1) 80% AcOH, reflux, 81%, (2) TBDMSCl, CH<sub>2</sub>Cl<sub>2</sub>, imidazole, 87%; (c) (1) NaH, MeI, THF, 92%, (2) Bu<sub>4</sub>NF, THF, 83%; (d) Ph<sub>3</sub>P, I<sub>2</sub>, imidazole, toluene, reflux, 70%.

of **2** followed by *O*-alkylation.<sup>10</sup> Deprotection of **3** using 80% acetic acid at reflux, followed by selective silylation of the primary alcohol afforded silyl ether **4**. *O*-Methylation followed by deprotection of silyl group gave primary alcohol **5**. 4-*O*-Hexadecyl-3-*O*-methyl-1-iodobutane (**6**) was prepared from **5** and I<sub>2</sub>/Ph<sub>3</sub>P at reflux in toluene.

*N*-Acetyl-3,4,6-tri-*O*-acetyl-1-glucosamine thioacetate (**8**) was synthesized from commercial *N*-acetyl-D-glucosamine

(6) Wei, A.; Boy, K. M.; Kishi, Y. *J. Am. Chem. Soc.* **1995**, *117*, 9432–9436.

(7) Wang, J.; Kovac, P.; Sinay, P.; Glaudemans, C. P. J. *Carbohydr. Res.* **1998**, *308*, 191–193.

(8) Tsuruta, O.; Yuasa, H.; Kurono, S.; Hashimoto, H. *Bioorg. Med. Chem. Lett.* **1999**, *9*, 807–810.

(9) Acton, E. M.; Ryan, K. J.; Tracy, M.; Arora, S. K. *Tetrahedron Lett.* **1986**, *27*, 4245–4248. (b) Welch, S. C.; Levine, J. A.; Arimilli, M. N. *Synth. Commun.* **1993**, *23*, 131–134.

(10) Yang, G.; Franck, R. W.; Byun, H.-S.; Bittman, R.; Samadder, P.; Arthur, G. *Org. Lett.* **1999**, *1*, 2149–2151.

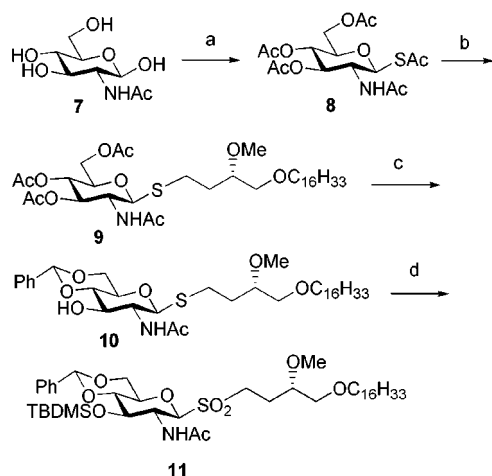
(11) Erukulla, R. K.; Zhou, X.; Samadder, P.; Arthur, G.; Bittman, R. *J. Med. Chem.* **1996**, *39*, 1545–1548.

(12) Paquette, L. A. *Org. React.* **1977**, *25*, 1–71. (b) Guziec, F. S., Jr.; San Filippo, L. J. *Tetrahedron* **1988**, *44*, 6241–6285. (c) Oae, S.; Uchida, Y. (Chapter 12); Braverman, S. (Chapter 13) In *The Chemistry of Sulfones and Sulfoxides*; Patai, S., Ed.; Wiley: New York, 1988. (d) Clough, J. M. In *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I., Eds.; Pergamon: Oxford, 1991; Vol. 3, Chapter 3.8.

(13) Belica, P. A.; Franck, R. W. *Tetrahedron Lett.* **1998**, *39*, 8225–8228. (b) Ref 10. (c) Griffin, F. K.; Murphy, P. V.; Patterson, D. E.; Taylor, R. J. K. *Tetrahedron Lett.* **1998**, *39*, 8179–8182. (d) Alcaraz, M.-L.; Griffin, F. K.; Patterson, D. E.; Taylor, R. J. K. *Tetrahedron Lett.* **1998**, *39*, 8183–8186. (e) Taylor, R. J. K.; Griffin, F. K.; Paterson, D. E. *Angew. Chem., Int. Ed.* **1999**, *38*, 2939–2942. (f) Campbell, A. D.; Paterson, D. E.; Raynham, T. M.; Taylor, R. J. K. *J. Chem. Soc., Chem. Commun.* **1999**, 1599–1600. (g) Falconer, R. A.; Toth, I. *20th International Carbohydrate Symposium*, Hamburg, Aug. 28–Sept. 2, 2000, Poster B-348.

(14) Roe, B. A.; Bojajamra, C. G.; Griggs, J. L.; Bertozzi, C. R. *J. Org. Chem.* **1996**, *61*, 6442–6445. “Unfortunately, *C*-glycosyl derivatives of 2-amino sugars are among the most difficult to prepare as a result of the incompatibility of neighboring nitrogen-based functional groups (i.e., amides, carbamates, and azides) with common *C*-glycosylation strategies.”

(15) Free-radical methods: (a) ref 14. (b) Gaurat, O.; Xie, J.; Valery, J.-M. *Tetrahedron Lett.* **2000**, *41*, 1187–1189. (c) Junker, H.-D.; Phung, N.; Fessner, W.-D. *Tetrahedron Lett.* **1999**, *40*, 7063–7066. (d) Cui, J.; Horton, D. *Carbohydr. Res.* **1998**, *309*, 319–330. Organosamarium methods: (e) Andersen, L.; Munch, L.; Beau, J.-M.; Skrydstrup, T. *Synlett*

Scheme 2<sup>a</sup>

<sup>a</sup> Reagents and conditions: (a) (1) AcCl, overnight, (2) KSAc, acetone, 70%, two steps; (b)  $\text{NH}_2\text{NH}_2 \cdot \text{HOAc}$ , DMF, **6**,  $\text{Et}_3\text{N}$ , 85%; (c) (1) guanidine,  $\text{EtOH}/\text{CH}_2\text{Cl}_2$ , (2)  $\text{PhCH}(\text{OMe})_2$ , *p*-TsOH, DMF, 72%, two steps; (d) (1) TBDMSCl, imidazole, DMF, 93%, (2) MMPP, 95%.

**7** in two steps<sup>16</sup> (Scheme 2). After the *S*-acetate was selectively cleaved ( $\text{NH}_2\text{NH}_2 \cdot \text{HOAc}$ , DMF), alkylation with iodide **6** in  $\text{Et}_3\text{N}$  gave thioglycoside **9** in good yield.<sup>17</sup> Selective deprotection of the *O*-acetyl groups using guanidine,<sup>18</sup> followed by benzylidene acetal protection of the 4,6-diol, afforded thioglycoside **10**. Treatment of **10** with TBDMSCl followed by oxidation using MMPP provided sulfone **11**.<sup>19</sup> The RB rearrangement of sulfone **11** using 25% KOH on alumina in  $\text{CBrF}_2/\text{CBrF}_2$  at reflux gave alkene **12** (*Z* isomer only, which was confirmed by a NOE experiment) in 78% yield (Scheme 3).<sup>20</sup>

RB product 2-deoxy-2-*N*-acetyl glycal **12** is much more stable than the corresponding RB product in the 2-deoxy-glucose series. Exo glycal **12** can be stored at 0 °C for more than 1 month without decomposition. Simultaneous benzylidene deprotection and reduction of alkene **12** ( $\text{H}_2$ , 10% Pd/C) afforded  $\beta$ -*C*-glycoside **13** in 85% yield.<sup>21</sup> Of the

**1998**, 1393–1395. (f) Urban, D.; Skrydstrup, T.; Beau, J.-M. *J. Org. Chem.* **1998**, 63, 2507–2516. Organolithium methods: (g) Burkhart, F.; Kessler, H. *Tetrahedron Lett.* **1998**, 39, 255–256. (h) Schafer, A.; Thiem, J. *J. Org. Chem.* **2000**, 65, 24–29. Wittig methods: (i) Xie, J.; Molina, A.; Czernecki, S. *J. Carbohydr. Chem.* **1999**, 18, 481–498.

(16) Horton, D.; Wolfrom, M. L. *J. Org. Chem.* **1962**, 27, 1794–1799.

(17) Park, W. K. C.; Meunier, S. J.; Zanini, D.; Roy, R. *Carbohydr. Lett.* **1995**, 1, 179–184.

(18) Kunesch, N.; Meit, C.; Poisson, J. *Tetrahedron Lett.* **1987**, 28, 3569–3572.

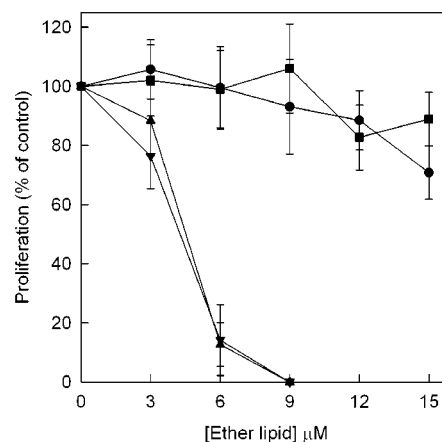
(19) This protection scheme was implemented (i) to replace KOH-sensitive acetates and (ii) to avoid alkylations with benzyl bromide, which would react with the amide function.

(20) We found that the yield of the RB reaction using freshly prepared  $\text{KOH}/\text{Al}_2\text{O}_3$  is much higher than using the material that has been stored for 1 month in the desiccator.

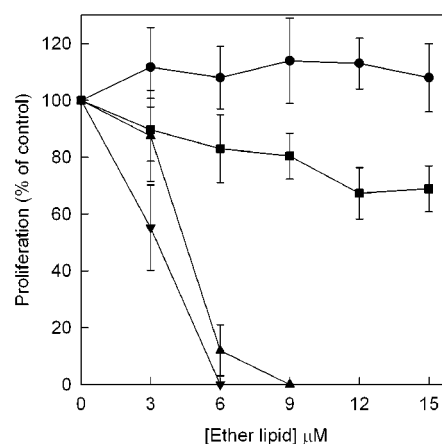
(21) Both Taylor's group and ours have observed clean  $\beta$ -*C*-glycoside formation upon hydrogenation deoxyglycals, almost certainly due to a chairlike transition state when the hydrogen is transferred to the  $\alpha$ -face. Conversely,  $\beta$ -face approach of hydrogen requires a twist-boat like TS during H-transfer.

(22) King, S. A.; Pipik, B.; Thompson, A. S.; Decamp, A.; Verhoeven, T. R. *Tetrahedron Lett.* **1995**, 36, 4563–4566.

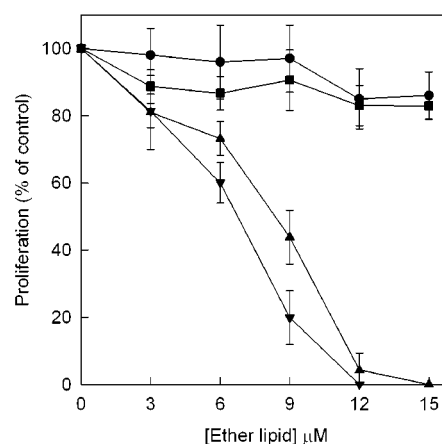
(a) SK-N-MC



(b) HS578T

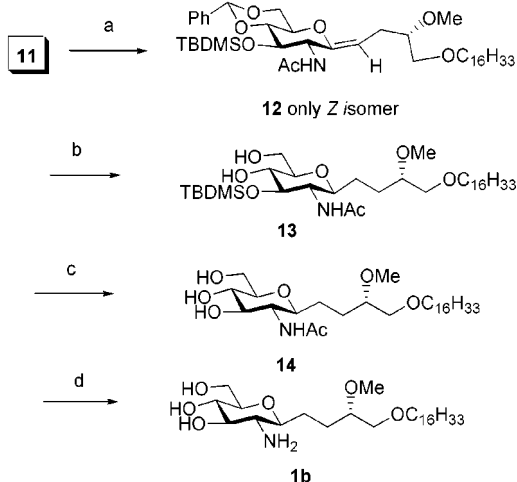


(c) DU145



**Figure 1.** Effects of antitumor ether lipids **1a** (▼), **1b** (▲), **1c** (●), and **14** (■) on the proliferation of (a) SK-N-MC, (b) HS578T, and (c) DU145 cells. Cells were treated with each compound for 48 h.

several methods attempted for cleavage of the silyl group,  $\text{Bu}_4\text{NF}$ , formic acid, acidic ionic exchange resin (Dowex

Scheme 3<sup>a</sup>

<sup>a</sup> Reagents and conditions: (a) CBrF<sub>2</sub>CBBrF<sub>2</sub>, *t*-BuOH, 25% KOH/Al<sub>2</sub>O<sub>3</sub>, reflux, 70%; (b) H<sub>2</sub>, 10% Pd/C, EtOAc, 80%; (c) BF<sub>3</sub>·Et<sub>2</sub>O, CH<sub>3</sub>CN, 0 °C, 93%; (d) 2 N KOH/EtOH, 120 °C, 75%.

50W), and BF<sub>3</sub>·Et<sub>2</sub>O, only BF<sub>3</sub>·Et<sub>2</sub>O in CH<sub>3</sub>CN<sup>22</sup> gave a clean reaction. The *N*-acetyl group was cleaved by using 2 N KOH in EtOH at 120 °C to afford final product **1b**.

In summary, our synthesis proved to be quite facile. Worthy of note is the very high stereoselectivity observed in each synthetic step, particularly in the RB sequence to afford exo glycal **12** and its reduction to afford *C*-glycoside **13**. It is of interest to note that the most rigorous conditions in the sequence involved the deacylation of **14** to afford **1b**. In addition to stereoselectivity, the combination of simplicity and convergence makes our approach an attractive and novel method for *C*-glycolipid synthesis.

Table 1 summarizes the comparative test results for **1a** and **1b**; also included are the data for **1c**, the 2-deoxy

(23) Samadder, P.; Byun, H.-S.; Bittman, R.; Arthur, G. *Anticancer Res.* **1998**, *18*, 465–470.

(24) Hu, Y.; Meullet, E. J.; Qiao, L.; Berggren, M. M.; Powis, G.; Kozikowski, A. P. *Tetrahedron Lett.* **2000**, *41*, 7415–7418.

**Table 1.** Growth Inhibitory Properties of **1a**, **1b**, and **1c**: IC<sub>50</sub> Values for Inhibition of Cell Proliferation<sup>a</sup>

cell line	IC <sub>50</sub> (μM)		
	<b>1a</b>	<b>1b</b>	<b>1c</b>
MCF-7, breast	8.0	8.1	25.6
MDA-MB-468, breast	7.0	9.0	34.4
MDA-MB-231, breast	7.1	9.1	40.0
HS578T, breast	3.1	5.1	21.0
BT549, breast	6.5	8.9	28.5
A498, kidney	6.9	8.5	ND
SK-N-SH, neuronal	3.8	4.1	ND
SK-N-MC, neuronal	4.1	4.1	ND
DU145, prostate	6.5	7.9	ND

<sup>a</sup> The IC<sub>50</sub> values for **1a**, **1b**, and **1c** were determined as described in ref 23. Briefly, exponentially growing cells were incubated with the drugs (0–15 μM), and the increase in cell numbers after 48 h was determined and expressed as a percentage of the controls, which had no drug. ND means not determined but > 15 μM.

analogue of **1b**, which was described previously.<sup>10</sup> Figure 1 shows the data for the assay against SK-N-MC, HS578T, and DU145 cells. In all nine examples, the IC<sub>50</sub> values<sup>23</sup> (drug concentrations required to inhibit growth by 50%) indicate that *C*-glycoside analogue **1b** shows antiproliferative activity remarkably parallel to that of the parent *O*-glycoside **1a**. It is interesting that a similar level of activity of ether glycerophospholipids bearing a deoxyinositol headgroup was reported recently by Kozikowski et al.<sup>24</sup> We are developing additional *C*-glycolipids in the aminosugar family, which will be described in the future.

**Acknowledgment.** Support for the research at Hunter has come from NIH Grant GM 51216, PSC-CUNY grants, and RCMI Grant NIH RR 03037, which supports the instrumentation infrastructure of the Chemistry Department at Hunter.

**Supporting Information Available:** Experimental procedures and spectroscopic data for compounds **3–14**, and **1b**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL006783A