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Phosphorus, Sulfur, and Silicon and the Related Elements

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GENERAL APPROACH TO THE SYNTHESIS OF SYMMETRICAL 1,3,2-DIAZAPHOSPHOCANES

Alexandra I. Zavalishina ^a , Elena I. Orzhekovskaya ^a , Natalja M. Selezneva ^a , Larisa K. Vasyanina ^a , Vitaly K. Belsky ^b & Eduard E. Nifantyev ^a

^a Fuculty of Chemistry# knin Pedagogical State University, Nesvizhskii per. 3, Moscow, II 9021, RUSSIA

^b Karpov Research Institute of Physical Chemistry , ul. Vorontsovo pole 10, Moscow, 103064, RUSSIA Published online: 04 Oct 2006.

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GENERAL APPROACH TO THE SYNTHESIS OF SYMMETRICAL 1,3,2-DIAZAPHOSPHOCANES

ALEXANDRA I. ZAVALISHINA^a, ELENA I. ORZHEKOVSKAYA^a, NATALJA M. SELEZNEVA^a, LARISA K. VASYANINA^a, VITALY K. BELSKY^b and EDUARD E. NIFANTYEV^{a*}

^aFaculty of Chemistry, Lenin Pedagogical State University, Nesvizhskii per. 3, Moscow, 119021 RUSSIA and ^bKarpov Research Institute of Physical Chemistry, ul. Vorontsovo pole 10, Moscow, 103064 RUSSIA

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Synthesis of eight-membered phosphorus-nitrogen heterocycles has been performed by phosphocyclization of the corresponding diamines using different derivatives of phosphorous acid. The structures of diazaphosphocanes were supported by elemental analysis; ¹H, ¹³C, and ³¹P NMR spectroscopy; and x-ray diffraction analysis.

Keywords: Diazaphosphocan

1,3,2-Diazaphosphocyclanes containing 4–7-membered phosphorus-nitrogen heterocycles have been studied in detail^{1–3}. These compounds are shown to have peculiar chemical features and to be promising for development of fine organic synthesis.

The reaction of N,N-dimethyl-p-toluidine with phosphorus oxychloride and phosphorus thiochloride resulting in formation of eight-membered phosphorus-nitrogen heterocycles was studied previously by Shaw *et al.*^{4–} ⁶. No further publications concerned with 1,3,2-diazaphosphocanes are available.

This is to give notice that 1,3,2-diazaphosphocanes belonging to different chemical types can be obtained by the direct cyclophosphorylation of 2,2'-methylene-bis-*p*-toluidine (1) and its derivatives with phosphorus

^{*} Correspondence Author.

acid chlorides. The high lability of 1,3,2-diazaphospho(III)canes should be noted, in distinction from the 5-7 membered analogues known to date, wich significantly complicated the operations. So, the reaction of N,N'-diisopropyl-2,2'-methylene-bis-p-toluidine (2) with phosphorus trichloride in the presence of triethylamine gives 2-chloro-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'-dimethyl-1,3,2-diazaphosphocan. This is very unstable compound. It was stabilized using dimethylaminolysis in combination with sulfurization. Resulting 2-dimethylamido-2-thio-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'-dimethyl-1,3,2-diazaphosphocan (3) was isolated with a yield of 45%

The structure of this compound was supported by ${}^{1}H$ and ${}^{31}P$ NMR spectroscopy, and its spatial organization was confirmed by x-ray diffraction analysis (see figure).



Diamine 2 was also subjected to cyclophosphorylation with dichlorides of ethyl-, phenyl-, and diethylamidophosphorous acids. The corresponding 2-substituted 1,3,2-diazaphosphocanes were obtained and isolated as thiophosphates (4-6).

The phosphocyclization of diamine 1 with alkyldichlorophosphites follows an unusual pathway. Mixtures of two tautomeric forms are the products:



This equilibrium is evidenced by the appearance of two signals at 113 and 6.23 (J_{PH} 614 Hz) in the ³¹P NMR spectrum. It is important that the only product (**7**, **8**) forms at the treatment of the system obtained by sulfur in both cases.

The obtained eight-membered cyclic systems containing a trivalent phosphorus atom are very labile. For example, the interaction between the primary diamine 1 and phenyldichlorophosphite resulted in formation of the corresponding 2-phenoxy derivative, for which the above prototropy was not observed. However, the compound obtained was readily hydrolyzed with formation of an incomplete diazaphosphocan amide (9), when chromatographied on silica gel.

Some of the compounds studied are susceptible to dimerization, as was shown during their study using mass spectrometry.

 $\begin{array}{l} \text{P-N}(1) = 1.648(4), \ \text{P-N}(3) = 1.663(3), \ \text{P-N}(2) = 1.667(3), \ \text{N}(2)\text{-C}(1) = \\ 1.442(5), \ \text{N}(2)\text{-C}(16) = 1.501(5), \ \text{N}(3)\text{-C}(9) = 1.439(2), \ \text{N}(3)\text{-C}(19) = \\ 1.500(5), \ \text{C}(6)\text{-C}(7) = 1.507, \ \text{C}(7)\text{-C}(8) = 1.512(5), \ \text{C}(1)\text{-C}(6) = 1.394, \\ \text{C}(8)\text{-C}(9) = 1.400(5). \end{array}$

$$\begin{split} N(1)-P-N(3) &= 110.37(17), \quad N(1)-P-N(2) &= 102.95(17), \quad N(3)-P-N(2) &= \\ 103.92(16), \quad N(1)-P-S &= 111.15(15), \quad N(3)-P-S &= 110.97(12), \quad N(2)-P-S &= \\ 116.98(12), \quad C(22)-N(1)-P &= 125.6(4), \quad C(23)-N(1)-P &= 120.8(4), \quad C(1)-N(2)-C(16) &= 118.0(3), \quad C(1)-N(2)-P &= 122.1(2), \quad C(16)-N(2)-P &= 117.8(3), \\ C(9)-N(3)-C(19) &= 118.1(3), \quad C(9)-N(3)-P &= 121.9(2), \quad C(19)-N(3)-P &= \\ \end{split}$$



FIGURE The structure of 3 from the x-ray diffraction analysis data

= 119.6(3), C(2)-C(1)-N(2) = 119.3(3), C(6)-C(1)-N(2) = 121.8(3), C(1)-C(6)-C(7) = 122.6(4), C(6)-C(7)-C(8) = 114.2(3), C(8)-C(9)-N(3) = 122.0(3)

EXPERIMENTAL

³¹P NMR spectra were recorded on a Bruker WP-80 instrument; ¹H NMR spectra were recorded on a Bruker WM-250 spectrometer. Mass-spectrometric studies were performed by the electron impact method on a Kratos MS-890 instrument. X-ray diffraction analysis was carried out on a Syntex P-1. An X-ray analysis of **3** was performed on a Syntex P-1 diffractometer using Cu-K_α radiation (λ = 1.54178 A). Crystals are orthorhombic, space

group Pbca, $C_{23}H_{34}N_3PS$, Z 8, d_{calc} 1.179 g/cm³, a 14.908(3), b 16.547(3), c 18.988(3) A. The structure was solved by a direct method and was refined anisotropically basing on 2148 independent reflections with I 2 $\sigma(I)$ to a final R=0.029. All calculations have been done using SHELX97 package.

General procedure

Phosphorus trichloride or dichloride of alkyl- (aryl- or dialkylamido-) phosphorous acid was added slowly to a solution of 0.01 mol of diamine (1, 2) and 0.02 mol of triethylamine in dry benzene under stirring at 5–10°C. The reaction mixture was stirred at the room temperature; triethylamine hydrochloride was filtered off, and the raw cyclic reaction product was used. In order to obtain the corresponding thioderivatives (3–8), sulfur was added. The reaction mixture was stirred for 2–3 h; the solvent was removed, and the target product was isolated either by recrystallization from benzene (4) or using the column chromatography on silica gel 45/75 (3, 5–8).

2-Thio-2-dimethylamido-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'dimethyl-1,3,2-diazaphosphocan (3)

Yield = 45%, m.p. 141–142°C. $\delta_P(CHCl_3)$ 69.80. MS, m/e: 415. ¹H NMR (CDCl₃, δ): 1.24 d (6H, CH₃^a, ³J_{HH} 6.4 Hz), 1.3 d (6H, CH₃^B, ³J_{HH} 7.83 Hz), 2.33 d (6H, N(CH₃)₂, ³J_{HP}10.9Hz), 2.23 s (3H, CH₃¹¹), 2.27 s (3H, CH₃^{11'}), 3.09 m (1H, CH^{1Pr}, ³J_{HP}15.3Hz, ³J_{HH}7.83Hz, ³J_{HH}6.4Hz), 3.74 d (1H, CH₂, ²J_{HH} 14.74 Hz), 4.06 m (1H, CH^{i-Pr} ³J_{HP}15.3Hz, ³J_{HH}7.83Hz, ³J_{HH}7.83Hz, ³J_{HH}6.4Hz), 4.7 d (1H, CH₂, ²J_{HH} 14.74 Hz), 6.56–7.09 m (6H, Ar). Found, %: C 66.3; H 8.39; P 7.07 C₂₃H₃₄N₃PS. Calcd., %: C 66.5; H 8.19; P 7.14.

2-Thio-2-ethoxy-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'-dimethyl-1,3,2-diazaphosphocan (4)

Yield = 45%, m.p. 164–166°C. δ_{P} (CHCl₃) 71.51. MS, m/e: 416. ¹H NMR (CDCl₃, δ): 0.25 (0.28) dd (6H, CH₃^{a,B}, ³J_{HH} 2.56 Hz,), 1.1 t (3H, CH₃CH₂O, ³J_{HH} 6.83 Hz), 1.35 d (3H, CH₃^a, ³J_{HH} 6.83 Hz), 1.44 d (3H, CH₃^B, ³J_{HH} 6.83 Hz), 2.28 s (6H, CH₃^{11,11'}), 3.77 m (1H, CH^{i-Pr}, ³J_{HP} 15.37 Hz, ³J_{HH}2.56Hz, ³J_{HH}6.83Hz), 3.90 m (1H, CH^{i-Pr}, ³J_{HP} 15.37 Hz, ${}^{3}J_{HH}$ 2.56Hz, ${}^{3}J_{HH}$ 6.83Hz), 3.95 m (2H CH₃CH₂O, ${}^{3}J_{HP}$ 8.53 Hz, ${}^{3}J_{HH}$ 6.83Hz), 4.04 d (1H, CH₂, ${}^{2}J_{HH}$ 14.51 Hz), 4.17 d (1H, CH₂, ${}^{2}J_{HH}$ 14.51 Hz), 6.94–7.11 m (6H, Ar). Found, %: C 65.90; H 8.26; P 6.26. C₂₃H₃₃N₂OPS. Calcd., %: C 66.3; H 7.9; P 7.4.

2-Thio-2-phenoxy-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'-dimethyl-1,3,2-diazaphosphocan (5)

Yield = 30%, m.p. 165–166°C. $\delta_{P}(C_{6}H_{6})$ 67.57. ¹H NMR ($C_{6}H_{6}$, δ): 0.36 (0.40) dd (6H, $CH_{3}^{a,B}$, ³J_{HH} 6.8 Hz), 1.39 d. (3H, CH_{3}^{a} , ³J_{HH} 6.4 Hz), 1.47 d. (3H, CH_{3}^{B} , ³J_{HH} 6.4 Hz), 2.06 s (3H, CH_{3}^{11}), 2.1 s (3H, $CH_{3}^{11'}$), 4.1 m (1H, CH^{iPr} ³J_{HP}15Hz, ³J_{HH}6.8Hz, ³J_{HH}6.4Hz), 4.20 d (1H, CH_{2}^{2} , ²J_{HH} 14.52 Hz), 4.28 d (1H, CH_{2} , ²J_{HH} 14.52 Hz), 4.64 m (1H, CH^{iPr} ³J_{HH}6.8Hz, ³J_{HH}6.4Hz), 6.78–7.34 m (11H, Ar).

2-Thio-2-diethylamido-1,3-diisopropyl-4,5;7,8-dibenzo-11,11'dimethyl-1,3,2-diazaphosphocan (6)

Yield = 42%, m.p. 170–171°C. $\delta_P(CHCl_3)$ 70.97. MS, m/e: 443. ¹H NMR (CDCl₃, δ): 1.23 d (12H, CH₃^{iPr}, ³J_{HH}7.46 Hz), 1.27 m (6H, CH₃CH₂-, ³J_{HH}6.83Hz), 2.28 s (3H, CH₃¹¹), 2.32 s (3H, CH₃^{11'}), 3.16 q (1H, CH^{iPr}, ³J_{HH}7.46Hz), 3.25 m (2H, -<u>CH₂-CH₃</u>, ³J_{HH}6.83Hz, ³J_{HP} 12.81 Hz), 3.48 m (2H, -<u>CH₂-CH₃</u>, ³J_{HH}6.83Hz, ³J_{PH} 13.23 Hz), 3.79 d (1H, CH₂, ²J_{HH} 14.72 Hz), 4.03 q (1H, CH^{iPr}, ³J_{HH}6.83Hz), 4.64 d (1H, CH₂, ²J_{HH} 14.72 Hz), 6.97–7.12 m (6H, Ar).

2-H-2-Oxo-4,5;7,8-dibenzo-11,11'-dimethyl1,3,2-diazaphosphocan (9)

Yield = 35%. m.p. 172–173°C. δ_P (CHCl₃) 10.31. MS, m/e: 272. ¹H NMR (CDCl₃, δ): 2.29 s (6H, CH₃^{11,11}), 3.93 d (1H, CH₂, ²J_{HH} 14.29 Hz), 4.03 d (1H, CH₂, ²J_{HH} 14.29 Hz), 4.97 s (2H, NH^{1,3}, ³J_{HH} 2.2 Hz), 7.68 d (1H, J_{PH} 614 Hz), 6.86–7.11m (6H, Ar).

Note: Letters a and e denote two magnetically unequivalent methyl groups in isopropyl radicals.

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