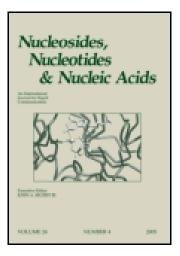
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## Nucleosides and Nucleotides

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lncn19

# A Serendipitous Synthesis of 8-Dimsyl-2'deoxyguanosine

Satyam Nampalli<sup>a</sup>, Inna Livshin<sup>a</sup> & Shiv Kumar<sup>a</sup> <sup>a</sup> Amersham Pharmacia Biotech, 26111 Miles Road, Cleveland, OHIO, 44128, USA Published online: 04 Oct 2006.

To cite this article: Satyam Nampalli , Inna Livshin & Shiv Kumar (1999) A Serendipitous Synthesis of 8-Dimsyl-2'-deoxyguanosine, Nucleosides and Nucleotides, 18:4-5, 697-699, DOI: 10.1080/15257779908041545

To link to this article: <u>http://dx.doi.org/10.1080/15257779908041545</u>

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### A SERENDIPITOUS SYNTHESIS OF 8-DIMSYL-2'-DEOXYGUANOSINE

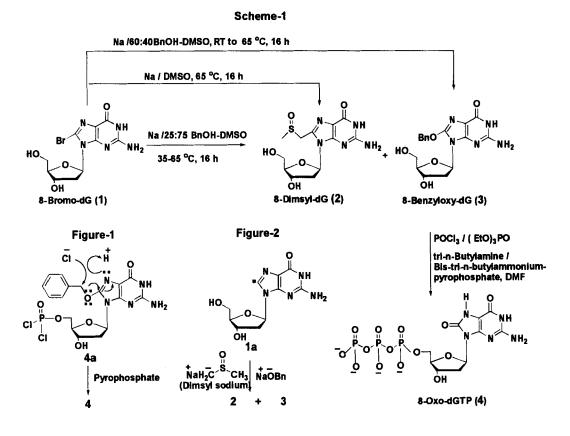
Satyam Nampalli, Inna Livshin and Shiv Kumar\* Amersham Pharmacia Biotech, 26111 Miles Road, Cleveland, OHIO, USA 44128.

**ABSTRACT:** A serendipitous synthesis of 8-dimsyl-dG (2) has been achieved along with the known 8-benzyloxy-dG (3) in a nucleophilic substitution reaction of 8-bromo-dG (1) with *in situ* generated dimsyl and benzyloxy sodium. Compound 3 was directly converted into the mutagenic oxidative DNA damage product, 8-oxo-dGTP (4).

As part of the research program directed towards the development of a random mutagenesis<sup>1</sup> kit, we needed to synthesize oxidative<sup>2</sup> DNA damage product, 8-Oxo-dGTP (4) (Scheme-1) in commercially viable quantities. 8-Bromo-2'-deoxyguanosine<sup>3</sup> (1) has been widely used to synthesize 8-substituted-2'-deoxyguanosine derivatives for different purposes<sup>4</sup>. We have decided to make use of suitably protected 8-benzyloxy-2'-deoxyguanosine (3), derivable from compound 1 for phosphorylation at the 5'-OH group in the face of problems encountered in achieving direct oxidation of dGTP<sup>5</sup> to yield 8-oxo-dGTP (4).

In an attempt to prepare 3 from 1 following the literature procedure<sup>6</sup>, sodium metal dissolution in a 2.5:7.5 BnOH:DMSO mixture appeared to be taking a long time at room temperature. Heating the reaction mixture at 35 °C for an hour ensured complete dissolution of the sodium metal. Addition of 8-bromo-dG (1) to the warm sodium dissolved solution and further heating at 65 °C for 16 h resulted in an unexpected 8-dimsyl-2'-deoxyguanosine (2)<sup>7</sup> as the major (45%) and 8-benzyloxy-2'-deoxyguanosine (3) as the desired, minor (25%) compounds.

In contrast, addition of 8-bromo-dG (1) to the clearly dissolved solution of sodium metal in a 60:40 BnOH:DMSO mixture, at room temperature and heating at 65  $^{\circ}$ C, exclusively afforded the desired compound **3** in 80% yield. In the absence of BnOH, heating 8-bromo-dG (1) in sodium metal dissolved solution of DMSO at 65  $^{\circ}$ C for 16 h provided 8-dimsyl-dG (2) in 35% yield.SRN1 free radical<sup>8</sup> mechanism (Figure-1) has been



invoked to have been operated to help explain the formation of compounds 2 and 3 via iminyl radical (1a)being attacked by dimsyl sodium<sup>9</sup> and sodium benzyloxide.

Having obtained the desired 8-benzyloxy-dG (3) in an improved yield, it was phosphorylated to directly produce 8-oxo-dGTP (4) in 45% yield. Mechanistically (Figure-2), in situ generated HCl from the reaction of 3 with POCl<sub>3</sub> is believed to have catalyzed the cleavage of benzyl iminol ether in 5'-dichlorophosphate intermediate (4a), which upon treatment with pyrophosphate directly generated 8-oxo-dGTP (4).

In summary, a serendipitous formation of 8-dimsyl-dG (2) has been unraveled along with the desired 8-benzyloxy-dG (3), which was converted directly into 8-oxodGTP (4), needed for the development of a random mutagenesis kit. SRN1 mechanistic pathway for the formation of hitherto unknown 8-dimsyl-dG (2) as well as 8-benzyloxydG (3) and HCl catalyzed pathway for the cleavage of benzyl group in 3 during phosphorylation to 8-oxo-dGTP (4) have been invoked.

Currently, efforts are underway to convert 8-dimsyl-dG (2) to its triphosphate, 8dimsyl-dGTP and study its DNA polymerase substrate activity including anti-HIV.

#### REFERENCES

- 1. Zaccolo, M.; Williams, D. M.; Brown, D.M.; Gherardi, E. J. Mol. Biol. 1996, 255, 589-603.
- Lipscomb, L. N.; Peek, M.E; Morningstar, M. L.; Verghis, S. M.; Miller, E. M.; Rich, A.; Essigmann, J. M.; Williams, L. D. Proc. Natl. Acad. Sci. USA., 1995, 92, 719-723.
- Koizume, S.; Kamiya, H.; Inoue, H.; Ohtsuka, E. Nucleosides & Nucleotides., 1994, 13, 1517-1534.
- 4. Cho, B. P.; Kadlubar, F. F.; Culp, S. J.; Evans, F. K. Chem. Res. Toxicol., 1990, 3, 445-452.
- 5. Purmal, A. A.; Kow, Y. W.; Wallace, S. S. Nucleic Acids Research., 1994, 22, 3930-3935.
- 6 Lin, T.S.; Cheng, J-C.; Ishiguro, K.; Sartorelli, A. C. J. Med. Chem., 1985, 28, 1194-1198.
- 7. 'H-NMR (DMSO-d<sub>6</sub>):  $\delta$  (ppm) 10.65 (1H, bs, D<sub>2</sub>O exch. NH), 6.37 (2H, bs, D<sub>2</sub>O exch. NH<sub>2</sub>), 6.13 (1H, dd, J= 6.0 Hz, 9.0Hz, 1'-H), 5.24 (1H, d, J = 6.0 Hz, D<sub>2</sub>O exch. 3'-OH), 4.86 (1H, t, J = 6.0 Hz, D<sub>2</sub>O exch. 5'- OH), 4.35 (1H, m, 3'-H), 3.80 (1H, m, 4'-H), 3.56 (2H, m, 5'-H<sub>2</sub>), 3.30 (1H, m, CH<sub>a</sub>-SO-CH<sub>3</sub>), 2.97 (1H, m, 2'-H<sub>a</sub>), 2.82 (1H, m, CH<sub>a</sub>-SO-CH<sub>3</sub>), 2.58 (3H, s, -SO-CH<sub>3</sub>), 2.08 (1H, m, 2'-H<sub>b</sub>). UV: (Tris, pH 7.4)  $\lambda$ max 271 nm. C<sub>12</sub>H<sub>17</sub>N<sub>5</sub>O<sub>5</sub>S: Calculated S, 9.33; Found S, 9.78.
- See review- Norris in Patai & Rappoport "The chemistry of functional groups, supplement D", Pt.1, 1983, 681-701. Wiley, New York.
- 9. Liedholm, B.; J. Chem. Soc. Perkin I., 1992, 2235-2237.