Synthesis of Tetrahydropyridines from Morita–Baylis–Hillman Acetates of α,β-Unsaturated Aldehydes Via an Intramolecular 1,6-Conjugate Addition

Su Yeon Kim, Ko Hoon Kim, Hye Ran Moon, and Jae Nyoung Kim*

Department of Chemistry and Institute of Basic Science, Chonnam National University, Gwangju 500-757, Korea. *E-mail: kimjn@chonnam.ac.kr Received August 13, 2015, Accepted September 7, 2015, Published online December 17, 2015

Keywords: Tetrahydropyridines, Intramolecular 1,6-conjugate addition, Morita–Baylis–Hillman adducts, Double bond isomerization

Tetrahydropyridine-3-carboxylic acid derivatives have shown many interesting biological activities.^{1,2} The syntheses of tetrahydropyridine-3-carboxylic acids have been carried out by phosphine-catalyzed ring-forming reactions between allenoates and imines,^{2a–d} ring-closing metathesis (RCM) reaction of homoallylic amine derivatives,^{2e} chemical transformations of Morita–Baylis–Hillman (MBH) adducts,^{2f–i} and intramolecular 1,6-conjugate addition of 2,4-dienylamines.^{2j}

MBH adducts have been used for the synthesis of many interesting cyclic compounds.^{3–5} Recently the MBH adducts of α , β -unsaturated aldehydes have been studied extensively by us⁴ and other groups.⁵ During our recent studies using the MBH adducts of cinnamaldehydes,⁴ we reasoned out that introduction of a primary amine such as benzylamine at the primary position of MBH acetate **1a** could provide methyl tetrahydropyridine-3-carboxylate **2a**, as shown in Scheme 1. The S_N2' reaction between **1a** and benzylamine would afford 1:1 adduct **I**,⁶ and the following intramolecular 1,6-conjugate addition^{2j,7} would provide 1,2,5,6-tetrahydropyridine-3-carboxylate **2a**.

At the outset of our experiment, the reaction of 1a and benzylamine (4.0 equiv) was examined in CH₃CN.⁶ The corresponding 1:1 adduct I (*E*-isomer) was formed as a major product at room temperature in short time (1 h); however, the separation of I-*E* in pure form was somewhat difficult due to the formation of many side products including unreactive 1:1 adduct I-Z and 1:2 adduct IV,⁸ as also shown in Scheme 1. Thus, we decided to carry out the synthesis of 2a without separation of I-E. The cyclization of crude I to 2a did not proceed at room temperature even after a long time (20 h). After some trials, we found that 2a could be formed in a reasonable yield (52%) under the influence of K₂CO₃ in refluxing CH₃CN for 20 h. The cyclization was less effective in the absence of K₂CO₃ even under refluxing CH₃CN condition.⁹ The role of K₂CO₃ is unclear at this stage; however, similar observation for the requirement of base in conjugate addition reactions was reported.¹⁰

Encouraged by the successful synthesis of 2a, the reactions with some representative MBH acetates of α , β -unsaturated aldehydes **1a–e** were examined, and the results are summarized in Table 1. The reactions of **1a** with *p*-methoxybenzylamine, 2-phenethylamine, 3-phenyl-1-propylamine, and *n*-octylamine afforded **2b–2e** in moderate yields (48–58%). The reaction of MBH acetate of crotonaldehyde **1b** and benzylamine provided **2f** in good yield (68%). The reaction of **1b** and 1-phenethylamine afforded **2g** in a similar yield (63%) as an inseparable diastereomeric mixture (3:1).

The reaction of 4-methyl derivative **1c** and benzylamine afforded double bond-isomerized product **2h** in good yield (79%) instead of generally expected product **2h**'. As shown in Scheme 2, a double bond isomerization of initially formed product **2h** did not occur presumably due to stabilizing hyperconjugation effect of the methyl group. Similar observation



Scheme 1. Synthesis of tetrahydropyridine 2a.

Table 1. Synthesis of tetrahydropyridines.^a



^a Conditions: (i) MBH acetate (0.5 mmol), amine (2.0 mmol), CH₃CN, room temperature, 1 h; (ii) K₂CO₃ (1.0 mmol), reflux, 20 h.

^b PMB, *p*-methoxybenzyl.

^c Inseparable diastereomeric mixture (3:1).

^d Each stereoisomer was separated in pure forms (47%/32%).

^e Each stereoisomer was separated in pure forms (34%/31%).



Scheme 2. Selective formation of 2h.

was also reported by Ramage and co-workers.^{2j} In addition, the yield of **2h** was higher than other entries presumably due to favorable *s*-*cis* conformation of the 1:1 adduct **V**, as shown in Scheme 2. It is interesting to note that the synthesis of **2h** could also be carried out even at room temperature without the assistance of K₂CO₃ in a similar yield (76%) after 20 h. Each stereoisomer (*cis/trans*) was separated in pure forms (47%/32%); however, the stereochemistry was not confirmed.¹¹ The reaction of **1c** and *p*-methoxybenzylamine afforded **2i** in good yield (65%).

The reaction of ethyl ester **1d** and benzylamine gave **2j** in moderate yield (51%). However, the reaction of acetyl derivative **1e** and benzylamine afforded **2k** in low yield (31%) due to unwanted formation of the corresponding 1:2 adduct in a larger amount. The 1:2 adduct was formed in appreciable quantity during the synthesis of 1:1 adduct, even though benzylamine was used in large excess (4.0 equiv).¹² The reaction of **1a** and aniline failed to produce **2l**. Although the corresponding 1:1 adduct was formed, a subsequent intramolecular



Scheme 3. Double bond isomerization of 2a.

1,6-conjugate addition to **2l** was ineffective under the reaction conditions.

The synthesis of *N*-unsubstituted derivative with ammonia source, such as NH₄OH or NH₄OAc, was unsuccessful due to the formation of 1:2 and 1:3 adducts.^{8c,13} Thus, we examined *N*-debenzylation of **2a** under the typical hydrogenolysis conditions, as shown in Scheme 3. Interestingly, a double bond isomerization to 1,4,5,6-tetrahydropyridine derivative **3** was observed unexpectedly.^{14,15}

In summary, various tetrahydropyridine-3-carboxylates were synthesized from MBH acetates of α , β -unsaturated aldehydes via $S_N 2'$ type introduction of primary alkylamines and the following intramolecular 1,6-conjugate addition reaction.

Experimental

Typical Procedure for the Synthesis of 2a. To a stirred solution of benzylamine (214 mg, 2.0 mmol) in CH₃CN (1.0 mL) was added a solution of **1a** (130 mg, 0.5 mmol, 0.5 mL of CH₃CN), and the reaction mixture was stirred for 1 h. K_2CO_3 (138 mg, 1.0 mmol) was added, and the reaction mixture was heated to reflux for 20 h. After the usual aqueous extractive workup and column chromatographic purification process (hexanes/EtOAc, 20:1), **2a** was obtained as a pale

yellow solid, 80 mg (52%). Other compounds were synthesized similarly, and the selected spectroscopic data of **2a**, **2b**, **2h** (major), **2h** (minor), and **3** are as follows.

Methyl 1-benzyl-6-phenyl-1,2,5,6-tetrahydropyridine-3-carboxylate (2a): 52%; pale yellow solid, mp 88–90 °C; IR (KBr) 1715, 1436, 1258 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 2.56–2.64 (m, 2H), 3.00–3.12 (m, 1H), 3.11 (d, J = 13.5 Hz, 1H), 3.47–3.57 (m, 1H), 3.62 (t, J = 6.6 Hz, 1H), 3.71 (s, 3H), 3.77 (d, J = 13.5 Hz, 1H), 7.07–7.12 (m, 1H), 7.19–7.43 (m, 10H); ¹³C NMR (CDCl₃, 75 MHz) δ 34.63, 50.04, 51.50, 58.56, 62.08, 126.85, 127.43, 127.78, 128.22, 128.56, 128.63, 128.74, 137.27, 138.92, 142.27, 166.21; ESIMS *m*/*z* 308 [M+H]⁺. Anal. Calcd for C₂₀H₂₁NO₂: C, 78.15; H, 6.89; N, 4.56. Found: C, 78.34; H, 6.97; N, 4.31.

Methyl 1-(4-methoxybenzyl)-6-phenyl-1,2,5,6-tetrahydropyridine-3-carboxylate (2b): 53%; pale yellow oil; IR (film) 1714, 1511, 1258 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 2.54–2.62 (m, 2H), 2.97–3.09 (m, 1H), 3.05 (d, J = 12.9Hz, 1H), 3.43–3.54 (m, 1H), 3.58 (t, J = 6.3 Hz, 1H), 3.69 (d, J = 12.9 Hz, 1H), 3.71 (s, 3H), 3.79 (s, 3H), 6.83 (d, J =8.4 Hz, 2H), 7.05–7.10 (m, 1H), 7.20 (d, J = 8.4 Hz, 2H), 7.24–7.31 (m, 1H), 7.32–7.42 (m, 4H); ¹³C NMR (CDCl₃, 75 MHz) δ 34.68, 49.85, 51.53, 55.21, 57.94, 62.04, 113.63, 127.42, 127.81, 128.63, 128.78, 129.76, 130.81, 137.29, 142.34, 158.58, 166.27; ESIMS *m*/z 338 [M+H]⁺. Anal. Calcd for C₂₁H₂₃NO₃: C, 74.75; H, 6.87; N, 4.15. Found: C, 74.49; H, 6.82; N, 4.03.

Methyl 1-benzyl-5-methyl-6-phenyl-1,2,3,6-tetrahydropyridine-3-carboxylate (2h, major): 47%; pale yellow oil; IR (film) 1739, 1453, 1170 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 1.45 (s, 3H), 2.54–2.64 (m, 1H), 3.09–3.20 (m, 2H), 3.27 (d, J = 13.5 Hz, 1H), 3.65 (d, J = 13.5 Hz, 1H), 3.66 (s, 3H), 3.84 (s, 1H), 5.71 (s, 1H), 7.17–7.41 (m, 10H); ¹³C NMR (CDCl₃, 75 MHz) δ 21.32, 41.09, 47.96, 51.68, 58.12, 69.06, 118.51, 126.84, 127.33, 127.97, 128.19, 128.60, 129.39, 137.61, 139.12, 141.00, 173.93; ESIMS m/z 322 [M+H]⁺. Anal. Calcd for C₂₁H₂₃NO₂: C, 78.47; H, 7.21; N, 4.36. Found: C, 78.76; H, 7.50; N, 4.23.

Methyl 1-benzyl-5-methyl-6-phenyl-1,2,3,6-tetrahydropyridine-3-carboxylate (2h, minor): 32%; pale yellow oil; IR (film) 1739, 1453, 1171 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 1.40 (s, 3H), 2.48 (t, *J* = 10.5 Hz, 1H), 3.14 (dd, *J* = 11.4 and 5.1 Hz, 1H), 3.19 (d, *J* = 13.5 Hz, 1H), 3.34–3.48 (m, 1H), 3.66 (s, 3H), 3.71 (d, *J* = 13.5 Hz, 1H), 3.81 (s, 1H), 5.71 (s, 1H), 7.14–7.44 (m, 10H); ¹³C NMR (CDCl₃, 75 MHz) δ 21.25, 42.12, 48.88, 51.79, 58.46, 69.61, 118.93, 126.82, 127.41, 128.10, 128.24, 128.58, 129.17, 137.43, 139.10, 141.70, 173.53; ESIMS *m/z* 322 [M+H]⁺.

Methyl 1-benzyl-6-phenyl-1,4,5,6-tetrahydropyridine-3-carboxylate (3): 54%; colorless oil; IR (film) 1681, 1620, 1296, 1152 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 1.88–2.06 (m, 3H), 2.32–2.50 (m, 1H), 3.74 (s, 3H), 4.20 (d, *J* = 15.3 Hz, 1H), 4.28 (t, *J* = 3.9 Hz, 1H), 4.43 (d, *J* = 15.3 Hz, 1H), 7.12–7.24 (m, 4H), 7.26–7.42 (m, 6H), 7.82 (s, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 16.21, 28.76, 50.64, 57.45, 57.66, 95.12, 126.37, 127.34, 127.41, 127.76, 128.58, 128.73, 137.02, 141.48, 146.52, 168.88; ESIMS *m/z* 308 [M+H]⁺. Anal. Calcd for C₂₀H₂₁NO₂: C, 78.15; H, 6.89; N, 4.56. Found: C, 78.44; H, 7.03; N, 4.61.

Acknowledgments. This research was supported by Basic Science Research Program through the National Research Foundation of Korea (NRF) funded by the Ministry of Education, Science and Technology (2014R1A1A2053606). Spectroscopic data were obtained from the Korea Basic Science Institute, Gwangju branch.

Supporting Information. Additional supporting information is available in the online version of this article.

References

- For synthesis and biological activities of tetrahydropyridine-3carboxylic acid derivatives, see: (a) G. Zheng, A. M. Smith, X. Huang, K. L. Subramanian, K. B. Siripurapu, A. Deaciuc, C.-G. Zhan, L. P. Dwoskin, J. Med. Chem. 2013, 56, 1693; (b) E. K. Moltzen, H. Pedersen, K. P. Bogeso, E. Meier, K. Frederiksen, C. Sanchez, H. Love Lembol, J. Med. Chem. 1994, 37, 4085; (c) G. A. Showell, T. L. Gibbons, C. O. Kneen, A. M. MacLeod, K. Merchant, J. Saunders, S. B. Freedman, S. Patel, R. Baker, J. Med. Chem. 1991, 34, 1086; (d) Y. Matsubara, R. Yoneda, S. Harusawa, T. Kurihara, Chem. Pharm. Bull. 1988, 36, 1597; (e) C. Muller-Uri, E. A. Singer, W. Fleischhacker, J. Med. Chem. 1986, 29, 125; (f) P. Krogsgaard-Larsen, K. Thyssen, K. Schaumburg, Acta Chem. Scand. 1978, B32, 327.
- 2. For synthesis of tetrahydropyridine-3-carboxylic acid derivatives, see: (a) Z. Wang, S. Castellano, S. S. Kinderman, C. E. Argueta, A. B. Beshir, G. Fenteany, O. Kwon, Chem. Eur. J. 2011, 17, 649; (b) S. Castellano, H. D. G. Fiji, S. S. Kinderman, M. Watanabe, P. de Leon, F. Tamanoi, Kwon, J. Am. Chem. Soc. 2007, 129, 5843; О. (c) R. P. Wurz, G. C. Fu, J. Am. Chem. Soc. 2005, 127, 12234; (d) G.-L. Zhao, M. Shi, Org. Biomol. Chem. 2005, 3, 3686; (e) P. V. Ramachandran, T. E. Burghardt, L. Bland-Berry, J. Org. Chem. 2005, 70, 7911; (f) S. Gowrisankar, H. S. Lee, J. M. Kim, J. N. Kim, Tetrahedron Lett. 2008, 49, 1670; (g) K. H. Kim, S. H. Kim, H. J. Lee, J. N. Kim, Adv. Synth. Catal. 2013, 355, 1977; (h) S. Takizawa, N. Inoue, H. Sasai, Tetrahedron Lett. 2011, 52, 377; (i) X. Meng, Y. Huang, R. Chen, Chem. Eur. J. 2008, 14, 6852; (j) K. Clinch, C. J. Marquez, M. J. Parrott, R. Ramage, Tetrahedron 1989, 45, 239; (k) V. N'Goka, G. Schlewer, J.-M. Linget, J.-P. Chambon, C.-G. Wermuth, J. Med. Chem. 1991, 34, 2547; (1) S. Duttwyler, C. Lu, A. L. Rheingold, R. G. Bergman, J. A. Ellman, J. Am. Chem. Soc. 2012, 134, 4064; (m) S. Duttwyler, S. Chen, M. K. Takase, K. B. Wiberg, R. G. Bergman, J. A. Ellman, Science 2013, 339, 678; (n) T. Mesganaw, J. A. Ellman, Org. Process Res. Dev. 2014, 18, 1097; (o) H.-J. Wang, L.-P. Mo, Z.-H. Zhang, ACS Comb. Sci. 2011, 13, 181.
- For general reviews on MBH reaction, see:(a) D. Basavaiah, A. J. Rao, T. Satyanarayana, *Chem. Rev.* 2003, 103, 811;
 (b) D. Basavaiah, B. S. Reddy, S. S. Badsara, *Chem. Rev.* 2010, 110, 5447;
 (c) V. Singh, S. Batra, *Tetrahedron* 2008,

64, 4511; (d) J. N. Kim, K. Y. Lee, *Curr. Org. Chem.* 2002, 6, 627; (e) K. Y. Lee, S. Gowrisankar, J. N. Kim, *Bull. Korean Chem. Soc.* 2005, 26, 1481; (f) S. Gowrisankar, H. S. Lee, S. H. Kim, K. Y. Lee, J. N. Kim, *Tetrahedron* 2009, 65, 8769; (g) M. Shi, F.-J. Wang, M.-X. Zhao, Y. Wei, *The Chemistry of the Morita–Baylis–Hillman Reaction*, RSC Publishing, Cambridge, 2011.

- 4. For our recent synthesis and synthetic applications of MBH adduct of α , β -unsaturated aldehydes, see: (a) K. H. Kim, S. Lee, J. Lee, M. J. Go, J. N. Kim, Tetrahedron Lett. 2013, 54, 5739; (b) C. H. Lim, S. H. Kim, K. H. Kim, J. N. Kim, Tetrahedron Lett. 2013, 54, 2476; (c) K. H. Kim, S. Lee, J. Lee, J. N. Kim, Tetrahedron Lett. 2015, 56, 4349; (d) K. H. Kim, H. R. Moon, J. Lee, J. Kim, J. N. Kim, Adv. Synth. Catal. 2015, 357, 1532; (e) J. W. Lim, K. H. Kim, S. H. Kim, J. N. Kim, Tetrahedron 2014, 70, 6831; (f) J. W. Lim, K. H. Kim, S. H. Kim, J. N. Kim, Tetrahedron Lett. 2013, 54, 2595; (g) J. W. Lim, S. H. Kim, J. Yu, J. N. Kim, Bull. Korean Chem. Soc. 2013, 34, 3503; (h) C. H. Lim, S. H. Kim, H. J. Lee, H. J. Kim, J. N. Kim, Bull. Korean Chem. Soc. 2013, 34, 993; (i) J. W. Lim, K. H. Kim, S. H. Kim, J. N. Kim, Tetrahedron Lett. 2012, 53, 5449; (j) K. H. Kim, H. S. Lee, S. H. Kim, J. N. Kim, Tetrahedron Lett. 2011, 52, 5605; (k) K. H. Kim, S. H. Kim, S. Park, J. N. Kim, Tetrahedron 2011, 67, 3328.
- 5. For synthesis and synthetic applications of MBH adducts of α,β -unsaturated aldehydes by other groups, see: (a) N. M. Garrido, M. R. Sanchez, D. Diez, F. Sanz, J. G. Urones, Tetrahedron: Asymmetry 2011, 22, 872; (b) N. M. Garrido, M. Garcia, D. Diez, M. R. Sanchez, F. Sanz, J. G. Urones, Org. Lett. 2008, 10, 1687; (c) J. S. Yadav, B. V. S. Reddy, A. P. Singh, N. Majumder, Tetrahedron Lett. 2010, 51, 2291; (d) P. Srihari, A. P. Singh, A. K. Basak, J. S. Yadav, Tetrahedron Lett. 2007, 48, 5999; (e) S. Rajesh, B. Banerji, J. Iqbal, J. Org. Chem. 2002, 67, 7852; (f) X. Li, X. Xu, Y. Tang, Org. Biomol. Chem. 2013, 11, 1739.
- For selected S_N2' type introduction of amine to MBH adduct, see:(a) A. Mishra, N. Rastogi, S. Batra, *Tetrahedron* 2012, *68*, 2146; (b) S. Ghosh, R. Dey, K. Chattopadhyay, B. C. Ranu, *Tetrahedron Lett.* 2009, *50*, 4892; (c) H. S. Lee, E. S. Kim, S. H. Kim, J. N. Kim, *Tetrahedron Lett.* 2009, *50*, 2274.
- For selected examples of intramolecular 1,6-conjugate additions, see:(a) J. M. Sorbetti, K. N. Clary, D. A. Rankic, J. E. Wulff, M. Parvez, T. G. Back, J. Org. Chem. 2007, 72, 3326; (b) T. G. Back, D. A. Rankic, J. M. Sorbetti, J. E. Wulff, Org. Lett. 2005, 7, 2377; (c) J. E. Toth, P. L. Fuchs, J. Org. Chem. 1986, 51, 2594; (d) S. Nara, H. Toshima, A. Ichihara, Tetrahedron Lett. 1996, 37, 6745; (e) R. W. Bates, P. Song, Synthesis 2009, 655; (f) R. W. Bates, R. H. Snell, S. Winbush, SYNLETT 2008, 1042; (g) T. Gallagher, I. Derrick, P. M. Durkin,

C. A. Haseler, C. Hirschhauser, P. Magrone, *J. Org. Chem.* **2010**, 75, 3766; (h) O. E. Hutt, L. N. Mander, *J. Org. Chem.* **2007**, 72, 10130.

- For the formation of 1:2 adduct, see:(a) R. Wang, J. Li, C. Li, J. Liu, D. Li, X. Jia, H. Zhai, *Chin. J. Chem.* **2010**, *28*, 1253; (b) M. Akssira, F. E. Guemmout, P. Bauchat, A. Foucaud, *Can. J. Chem.* **1994**, *72*, 1357; (c) R. Pathak, V. Singh, S. N. Nag, S. Kanojiya, S. Batra, *Synthesis* **2006**, 813.
- 9. Compound **2a** was obtained in moderate yield (36%) without K₂CO₃ under refluxing conditions for 20 h.
- For the use of base in conjugate addition reactions of amine nucleophiles, see:(a) D. Rosenthal, G. Brandrup, K. H. Davis Jr., M. E. Wall, J. Org. Chem. 1965, 30, 3689;
 (b) C.-E. Yeom, M. J. Kim, B. M. Kim, Tetrahedron 2007, 63, 904;
 (c) V. B. Labade, S. S. Pawar, M. S. Shingare, Monatsh Chem. 2011, 142, 1055;
 (d) A. O. Maldaner, R. A. Pilli, SYN-LETT 2004, 1343;
 (e) H. Toya, T. Satoh, K. Okano, K. Takasu, M. Ihara, A. Takahashi, H. Tanaka, H. Tokuyama, Tetrahedron 2014, 70, 8129;
 (f) H. Toya, K. Okano, K. Takasu, M. Ihara, A. Takahashi, H. Tanaka, H. Tokuyama, Org. Lett. 2010, 12, 5196.
- NOE experiment was carried out with both stereoisomers of 2h; however, a significant NOE increment of the proton at C5-position (-CHCOOMe) was not observed for both isomers by irradiation of the proton at C2-position (-CHPh). Thus the cis/trans stereochemistry could not be confirmed by NOE experiment (see, Supporting Information).
- 12. The 1:2 adduct of **1e** and benzylamine was obtained in appreciable yield (37%) when we separate it before treating with K₂CO₃ in a separate experiment.
- For the formation of 1:3 adduct, see:(a) V. Singh, R. Pathak, S. Kanojiya, S. Batra, SYNLETT 2005, 2465; (b) P. Bauchat, A. Foucaud, Tetrahedron Lett. 1989, 30, 6337; (c) A. Foucaud, F. El Guemmout, Bull. Soc. Chim. Fr. 1989, 3, 403.
- For similar double bond isomerization with palladium catalyst, see:(a) K. T. Wanner, A. Kaertner, Arch. Pharm. 1987, 320, 1050; (b) K. T. Wanner, A. Kartner, Heterocycles 1987, 26, 917; (c) K. H. Kim, H. S. Lee, S. H. Kim, K. Y. Lee, J.-E. Lee, J. N. Kim, Bull. Korean Chem. Soc. 2009, 30, 1012.
- For similar 1,4,5,6-tetrahydropyridines, see:(a) S. Yu, W. Zhu, D. Ma, J. Org. Chem. 2005, 70, 7364; (b) H. M. C. Ferraz, F. L. C. Pereira, E. R. S. Goncalo, L. S. Santos, M. N. Eberlin, J. Org. Chem. 2005, 70, 110; (c) P. Jakobsen, J. M. Lundbeck, M. Kristiansen, J. Breinholt, H. Demuth, J. Pawlas, M. P. Torres Candela, B. Andersen, N. Westergaard, K. Lundgren, N. Asano, Bioorg. Med. Chem. 2001, 9, 733; (d) S. Long, F. R. Stefani, S. Biondi, G. Ghiselli, M. Panunzio, Bioorg. Med. Chem. 2013, 21, 5811.