

S0960-894X(96)00155-2

SYNTHESIS AND ANTIBACTERIAL ACTIVITY OF NEW TROPONE-SUBSTITUTED PHENYLOXAZOLIDINONE ANTIBACTERIAL AGENTS. 2. MODIFICATION OF THE PHENYL RING - THE POTENTIATING EFFECT OF FLUORINE SUBSTITUTION ON *IN VIVO* ACTIVITY.

Michael R. Barbachyn,*^{,†} Dana S. Toops,[†] Kevin C. Grega,[†] Susan K. Hendges,[†] Charles W. Ford,[‡] Gary E. Zurenko,[‡] Judith C. Hamel,[‡] Ronda D. Schaadt,[‡] Douglas Stapert,[‡] Betty H. Yagi,[‡] Jerry M. Buysse,[§] William F. Demyan,[§] James O. Kilburn,^I and Suzanne E. Glickman^I

[†]Medicinal Chemistry Research, [‡]Cancer and Infectious Diseases Research, [§]Molecular Biology Research, Pharmacia & Upjohn, Inc., Kalamazoo, MI 49001 and ^ICenters for Disease Control, Atlanta, GA 30333

Abstract: Various electron-withdrawing groups were incorporated into the *meta* position of tropone-substituted 3-phenyl-2-oxazolidinones and their influence on antibacterial activity examined. Consideration of *in vitro* and *in vivo* test results indicated that one or two fluorine atoms flanking the *para* tropone appendage is the optimum arrangement for these compounds. Synthetic routes to enantiomerically enriched analogues are reported. Copyright © 1996 Elsevier Science Ltd

The oxazolidinones are an important new class of orally active, synthetic antibacterial agents.¹ In the preceding paper of this series we described the genesis of antibacterially active tropone-substituted phenyloxazolidinones, for example 1.² Preliminary investigations into the effect of the tropone regioisomer on activity revealed that the 5- and 6-substituted analogues, 1b and 1c, respectively, were essentially equivalent in activity. Both 1b and 1c were found to be dramatically more active than the corresponding 4-substituted isomer 1a.



We were intrigued by the early oxazolidinone structure-activity relationships reported by workers at DuPont.³ In particular, their finding that conjugated electron-withdrawing groups (e.g., acetyl) in the *para*

position of phenyloxazolidinones afforded the most active analogues suggested to us that further structural modifications leading to reduced electron density in the phenyl ring might provide compounds with increased potency. With this in mind, we contemplated the introduction of inductive electron-withdrawing substituents (e.g., F, Cl, and CF_3) into the *meta* position(s) of the phenyl ring.⁴ Targets of generic structure 2 emerged after a decision to utilize the more readily accessible 5-bromo-2-methoxytropone as starting material.⁵ Racemic analogues were judged adequate to explore preliminary structure–activity relationships in this series. The more active analogues were then prepared in enantiomerically enriched form.



(R^1 = alkoxy, amino; R^2 = H, F; R^3 = F, Cl, CF₃)

Chemistry

The synthetic procedures described in the preceding paper² for appending a methoxytropone moiety to an intermediate phenyloxazolidinone derivative were found to be applicable to the preparation of (\pm) -2. Therefore, preparation of the requisive *para* bromo- or iodophenyloxazolidinone precursors became a principal focus. As shown in Scheme 1, Cbz derivatives of substituted anilines (3) can be readily elaborated to the racemic 5-(iodomethyl)oxazolidinone intermediates 4 through an allylation/iodocyclocarbamation reaction sequence.⁶⁻⁸ Treatment of 4 with sodium azide then affords a 5-azidomethyl derivative which, depending on the nature of the *para* X group, is reduced to the corresponding amine by one of two routes. When X = Br or I, reduction of the azide is achieved by formation and then hydrolysis of an intermediate iminophosphorane.⁹ In cases where X = H, the desired 5-(aminomethyl)oxazolidinones were prepared either by the former route or by catalytic hydrogenation. The resultant amines were generally not isolated, but rather acetylated to afford intermediates 5 (X = H, Br, I). Examples of 5 with X = H were then iodinated to generate the corresponding *para* iodo derivatives. Further elaboration of 5 (X = Br, I), via a synthetic protocol involving a Stille coupling as its key step,² led to smooth generation of the targeted tropone-substituted phenyloxazolidinones 2.

Since only the oxazolidinone (S)-enantiomer exhibits antibacterial activity,¹ the preparation of optically active 2 was an important goal. Two distinct synthetic approaches, exemplified by the monofluorophenyl series depicted in Scheme 2, were utilized to prepare enantiomerically enriched analogues. In the first of these, 3-fluorophenyl isocyanate (6) was reacted with (R)-glycidyl butyrate, followed by cleavage of the butyryl group





with sodium methoxide, to give the key (R)-5-(hydroxymethyl)oxazolidinone 7.^{1,10} A sequence of reactions involving mesylation or tosylation of 7, azide displacement, reduction, and acetylation, then affords the (S)-5-(acetamidomethyl)oxazolidinone 8. Intermediate 8 was then iodinated and elaborated via a Stille cross-coupling reaction² to generate optically active examples of structure 2. The recent revelation that lithiated carbamate derivatives of anilines have general utility in preparing enantiomerically enriched 3-aryl-5-(hydroxymethyl)oxazolidinones¹¹ was also found to be applicable to the synthesis of intermediates 7.

Biological Evaluation and Discussion

A preliminary evaluation of the antibacterial activity of selected tropone-substituted phenyloxazolidinones bearing one or two substituents in the *meta* position(s) of the phenyl ring was conducted. The *in vitro* activity was assessed by determination of minimum inhibitory concentration (MIC) values, utilizing standard agar dilution methods, and *via* a cell-free prokaryotic transcription coupled translation (T/T) assay, employing S-30 *Escherichia coli* extracts. In the latter test system, expression of plasmid encoded *lacZ* yields the functional reporter enzyme (β -galactosidase). In vivo efficacy was evaluated by effective dose₅₀ (ED₅₀) determinations, using standard lethal-systemic infection models in mice.

As shown in Table 1, the appended R^2 and R^3 groups on the phenyl ring of racemic analogues 9-13 have a profound effect on the antibacterial activity of these compounds. Using 11 as a reference point, it can be seen that the addition of a chlorine (10) results in reduced *in vitro* and *in vivo* antibacterial activity. The activity loss





is even more pronounced for the corresponding trifluoromethyl analogue 9. In contrast, the addition of one fluorine in the *meta* position affords a compound (12) with *in vitro* activity identical to that of 11, but with significantly improved *in vivo* efficacy. Interestingly, the addition of a second fluorine substituent (13) imparts an additional two-fold improvement in the *in vivo* activity, relative to its monofluoro congener 12. This trend is repeated in the potent *in vivo* activities (see Table 2) observed for the antibacterially active (S)-enantiomers¹ (*cf.* ED₅₀s of 14 and 15). Compounds 14–17 demonstrate *in vitro* activity greater than that of the comparator, vancomycin, against aerobic gram-positive bacteria, including methicillin-resistant strains. *In vitro* activity against anaerobic bacteria (e.g., *Bacteroides fragilis*) and *Mycobacterium tuberculosis* is also indicated. Overall, the optically active compounds exhibit approximately twice the activity of their racemic counterparts (*cf.* MICs/ED₅₀s of 12 to 14 and 13 to 15). Analogues 15 and 17 are particularly noteworthy in that their level of *in vivo* efficacy is comparable to or greater than, respectively, the current clinical benchmark vancomycin.

Based on the above results, the often significant potentiating effect of fluorine substitution is evident. The optimum arrangement in the tropone-substituted oxazolidinone subclass appears to be one or two fluorine substituents flanking the appended para tropone moiety. Further evaluation of these agents is warranted.

Table 1. Activity of Selected Racemic Tropone-Substituted Phenyloxazolidinones

MeO O H³ N O NHAC

				MIC (µ	ED ₅₀ ,	T/T		
Compound	R ²	R ³	S.a1	S.e.	E.f.	S.p.	$(mg/kg)^2$	(%) ³
9	Н	CF ₃	32	4	16	4	>20 (1.3)	58
10	Н	Cl	4	1	2	1	>20 (2.2)	71
11	н	Н	2	0.5	1	0.25	12.2 (1.8)	89
12	Н	F	1	0.5	1	0.25	7.7 (1.7)	
13	F	F	2	0.5	1	0.5	3.7 (2.2)	85
vancomycin			1	2	4	0.5	1.3-2.2	

¹Minimum inhibitory concentration: lowest concentration of drug (μ g/mL) that inhibits visible growth of the organism. ²Effective dose₅₀: amount of drug required (mg/kg body weight/dose) to cure 50% of infected mice subjected to a lethal systemic infection; oxazolidinones administered orally; vancomycin ED₅₀ (sc administration) in parentheses. ³Transcription/translation assay; results reported as % inhibition, compound concentration of 2 μ g/mL. S.a.-1, Staphylococcus aureus UC[®]9213 (methicillin-susceptible); S.e., Staphylococcus epidermidis UC[®]12084 (methicillin-resistant); E.f., Enterococcus faecalis UC[®]9217; S.p., Streptococcus pneumoniae UC[®]9912.

Table 2. Activity of Optically Active Tropone-Substituted Mono- and Difluorophenyloxazolidinones



			MIC (µg/mL) ¹							ED ₅₀ ,
Compound	R ¹	R ²	S.a1	S.a2	S.e.	E.f.	S.p.	B.f.	M.tb.	(mg/kg) ²
14	MeO	Н	1	0.5	0.25	0.5	0.125	8	≤1 ³	4.1 (1.7)
15	MeO	F	1	0.5	0.5	1	0.25	8		2.3 (2.0)
16	MeNH	Н	0.5	0.25	0.125	0.25	0.125	2		
17	propargylamino	н	1	0.5	0.25	0.5	0.125	2		1.3 (2.1)
vancomycin			1	1	2	4	0.5	32		1.7-2.1

¹Minimum inhibitory concentration. ²Effective dose₅₀; oxazolidinones administered orally; vancomycin ED₅₀ in parentheses (sc administration). ³Isoniazid MIC = 0.2 μ g/mL. S.a.-1, Staphylococcus aureus UC[®]9213 (methicillin-susceptible); S.a.-2, Staphylococcus aureus UC[®]6685 (methicillin-resistant); S.e., Staphylococcus epidermidis UC[®]12084 (methicillin-resistant); E.f., Enterococcus faecalis UC[®]9217; S.p., Streptococcus pneumoniae UC[®]9912; B.f., Bacteroides fragilis UC[®]12199; M.tb., Mycobacterium tuberculosis H37Rv.

Acknowledgment

The authors thank Debra A. Ulanowicz and Steven J. Brickner for starting materials.

References and Notes

- 1. Gregory, W. A.; Brittelli, D. R.; Wang, C.-L. J.; Wuonola, M. A.; McRipley, R. J.; Eustice, D. C.; Eberly, V. S.; Bartholomew, P. T.; Slee, A. M.; Forbes, M. J. Med. Chem. 1989, 32, 1673.
- Barbachyn, M. R.; Toops, D. S.; Ulanowicz, D. A.; Grega, K. C.; Brickner, S. J.; Ford, C. W.; Zurenko, G. E.; Hamel, J. C.; Schaadt, R. D.; Stapert, D.; Yagi, B. H.; Buysse, J. M.; Demyan, W. F.; Kilburn, J. O.; Glickman, S. E. *Bioorg. Med. Chem. Lett.* 1996, 6, preceding paper in this issue.
- Gregory, W. A.; Brittelli, D. R.; Wang, C.-L. J.; Kezar, H. S., III; Carlson, R. K.; Park, C.-H.; Corless, P. F.; Miller, S. J.; Rajagopalan, P.; Wuonola, M. A.; McRipley, R. J.; Eberly, V. S.; Slee, A. M.; Forbes, M. J. Med. Chem. 1990, 33, 2569.
- 4. During the course of these investigations, oxazolidinones bearing multiple substituents on their phenyl ring, including fluorinated derivatives, were disclosed: Park, C.-H.; Brittelli, D. R.; Wang, C. L.-J.; Marsh, F. D.; Gregory, W. A.; Wuonola, M. A.; McRipley, R. J.; Eberly, V. S.; Slee, A. M.; Forbes, M. J. Med. Chem. 1992, 35, 1156. The analogues described are distinct from those shown in this paper. Interestingly, a blanket statement by Park and co-workers that 3,4,5-trisubstituted phenyloxazolidinones "are devoid of antibacterial activity" has been shown to be erroneous for the 3,4,5-trisubstituted phenyloxazolidinones (e.g., 13 and 15) described herein.
- 5. Banwell, M. G.; Lambert, J. N.; Reum, M. E.; Onrust, R. Org. Prep. Proc. Intl. 1988, 393.
- 6. Satisfactory spectra (IR, ¹H NMR, low and high resolution MS) were obtained for all new compounds.
- 7. Grega, K. C.; Barbachyn, M. R.; Brickner, S. J.; Mizsak, S. A. J. Org. Chem. 1995, 60, 5255.
- 8. A manuscript describing a comprehensive study of the iodocyclocarbamation reaction is under preparation: Brickner, S. J. et al.
- 9. Vaultier, M.; Knouzi, N.; Carrie, R. Tetrahedron Lett. 1983, 24, 763.
- 10. Herweh, J. E.; Kauffmann, W. J. Tetrahedron Lett. 1971, 809.
- 11. Brickner, S. J.; Manninen, P. R.; Ulanowicz, D. A.; Lovasz, K. D.; Rohrer, D. C. Abstracts of Papers, 206th National Meeting of the American Chemical Society, Chicago, IL, August, 1993; American Chemical Society: Washington, DC, 1993; ORGN 089.

(Received in USA 9 February 1996; accepted 19 March 1996)