COPPER CATALYZED REDUCTIVE METALLATION OF A PROPARGYLIC EPOXIDE TO AN ALLENYL LITHIUM REAGENT

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Summary: Phenethypyl cyclohexene oxide undergoes a reductive metaliation by Buli and catalytic amount of Cu salt. The resulting allenyl lithium reagent reacts, then, normally, with various electrophiles.

In the preceding letter 1 we have shown that Grignard reagents react with propargylic epoxides under copper I catalysis to afford substituted allenols. Such copper I catalyzed reactions are not usually performed with organolithium reagents 2 . We have already published a notable exception with allylic epoxides 3 . We report herein that in the case of phenethynyl cyclohexene oxide $\underline{\mathbf{1}}$ the reaction takes an entirely different course from what we expected :

After hydrolysis, the allenol $\underline{3}$ was obtained quantitatively with 91% diastereoselectivity. That an intermediate organolithium reagent $\underline{2}$ was involved, was shown by deuteration with D₂0. Reduction products, such as $\underline{3}$, have already been obtained, in a similar reaction but with stoichiometric amounts of copper 1 (R₂CuLi)⁴. Its formation was interpreted as a quench of an intermediate Cu^{III} organometallic⁵. In our case, where only 5% of copper salt is present, $\underline{2}$ has to be an organolithium reagent, whose formation may be accounted by the following catalytic cycle:

 $\underline{2}$ reacts normally and diastereoselectively, as an allenyl lithium reagent, with various electrophiles 6 (diastereomeric purity : \geq 90%)

Noteworthy is the fact that the reagent $\underline{\mathbf{2}}$ retains its stereochemistry during all the above transformations. The reaction cannot be performed without added Cu^{I} salts whatever the solvent. We are currently exploring the scope and limitations of this reaction.

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References and notes :

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- 6. All the products were fully characterized and their analytical and spectroscopic data will be reported soon in the full paper.

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