CAROTENOID METABOLISM AND THE BIOSYNTHESIS OF ABSCISIC ACID

ANDREW D. PARRY and ROGER HORGAN

Department of Biological Sciences, The University College of Wales, Aberystwyth, Dyfed, SY23 3DA, U.K.

(Received in revised form 7 August 1990)

Key Word Index—Lycopersicon esculentum; Solanaceae; Phaseolus vulgaris; Leguminosae; carotenoid metabolism; deuterium labelling; apo-carotenoids; neoxanthin; violaxanthin; abscisic acid.

Abstract—The conversion of all-trans-violaxanthin to 9'-cis-neoxanthin was shown to occur in fluridone-treated etiolated Lycopersicon and Phaseolus seedlings, following exposure to light. The results of deuterium oxide labelling experiments supported this precursor/product relationship, and provided further evidence for the origin of abscisic acid. Several apo-carotenoids, putative by-products of abscisic acid biosynthesis, were synthesised by chemical oxidation but were not detected in plant extracts. In vitro, lipoxygenase cleaved neoxanthin and violaxanthin down to small ($\leq C_{13}$) fragments. It may be that in vivo any apo-carotenoids formed by the specific cleavage of 9'-cis-neoxanthin, during abscisic acid biosynthesis, are rapidly metabolized by lipoxygenase or similar enzymes.

INTRODUCTION

There is now considerable evidence to support an indirect 'apo-carotenoid' pathway for abscisic acid (ABA; 7) biosynthesis [1-3, and refs therein] with the most likely precleavage precursors being 9'-cis-neoxanthin (4) and 9-cisviolaxanthin (3; [2, 3]). Cleavage of these xanthophylls across the 11,12 (11',12') double bond would produce xanthoxin (Xan; 5), which is readily converted to ABA by plant tissues [4, 5]. The regulation of ABA biosynthesis occurs prior to the production of Xan, and could involve control of violaxanthin and neoxanthin interconversion and/or the cleavage reaction [5, 6].

At present little is known about xanthophyll biosynthesis in higher plants, especially that of the *cis* isomers and allenic xanthophylls such as neoxanthin [7, 8]. A pathway for the biosynthesis of neoxanthin in plants has been proposed from zeaxanthin, via an epoxide such as violaxanthin [9, 10] but this lacks experimental support. A cell-free system from the alga *Amphidinium carterae* converted ¹⁴C-zeaxanthin to all-*trans*-neoxanthin (2) but no intermediates were isolated [11]. In part this paper describes attempts to ascertain the relationship between the all-*trans*- and 9-*cis*-isomers of violaxanthin and neoxanthin in leaves of higher plants.

A vast number of possible carotenoid breakdown products or apo-carotenoids, have been identified [12] but knowledge of the biochemistry of carotenoid catabolism is extremely limited. The majority of these apocarotenoids are volatile (C_9-C_{13}) compounds, many of which are important aroma constituents, but others range in size up to C_{30} , for example apo-8'-violaxanthal which has been extracted from orange peel [13]. The existence of such a variety of apo-carotenoids, both in the type of end-group and size, may indicate that in certain tissues non-specific cleavage of carotenoids is occurring.

The photooxidation of carotenoids in vitro has produced a range of C_9-C_{15} apo-carotenoids, including Xan and trans-xanthoxin (t-Xan), most of which have been identified in plant extracts [12, 14]. The bleaching of carotenoid solutions by lipoxygenase (LOX) has been described frequently but the products of such degradations have not usually been identified [15-17], although Xan and t-Xan were isolated from incubations of LOX and violaxanthin [17]. Pigment bleaching by cytochrome c and peroxidases also takes place in vitro [15, 18]. The relevance of such observations to the non-specific or specific degradation of carotenoids in vivo is uncertain. Several potential by-products of neoxanthin and violaxanthin cleavage have been synthesized and their occurrence in plant tissues and LOX/xanthophyll incubations investigated, with the aim of revealing more about the cleavage of these xanthophylls in vivo.

RESULTS AND DISCUSSION

Origin of neoxanthin

Etiolated leaves contain the same major carotenoids as green leaves, namely lutein, violaxanthin, neoxanthin, antheraxanthin and β -carotene, but in different relative proportions [10]. Transfer of etiolated seedlings to the light results in a general stimulation of carotenoid synthesis, although the rates of accumulation for individual carotenoids vary widely [19, 20]. This accumulation of carotenoids is controlled both by phytochrome and the availability of chlorophyll and intact thylakoids [19]. After 72 hr in the light, levels of β -carotene and neoxanthin in leaves of five-day-old etiolated tomato seedlings increased 27- and 13-fold respectively (data not shown), whilst those of violaxanthin, antheraxanthin and lutein rose by only 1.7-, 6- and 4-fold respectively. Under

Abbreviations—DTT, dithiothreitol; LOX, lipoxygenase; not, notabilis; PA, phaseic acid; TEA, triethylamine; t-Xan, 2-transxanthoxin; Xan, xanthoxin.

these conditions it is difficult to determine the relationship between individual carotenoids.

Fluridone, which inhibits carotenogenesis at the stage of phytoene dehydrogenase [21], was used to prevent the synthesis of new carotenoids following the transfer of etiolated seedlings to light. The increases in levels of phytoene ($\lambda_{max}^{Ei0H} 287$ nm) and the constant low levels of β carotene confirmed that the inhibitor was successfully preventing carotenoid synthesis (data not shown). Data on carotenoid content are presented as percentages of total xanthophylls because on a fresh weight or leaflet basis overall carotenoid levels remained constant or fell, especially in *Phaseolus*, due to light-induced leaf expansion (data not shown).

After a time lag (6-9 hr in Lycopersicon; 12-25 hr in Phaseolus), during which chlorophyll synthesis began and some phytoene accumulated, a decrease occurred in the percentage of all-trans-violaxanthin concommitant with an increase in the percentage of 9'-cis-neoxanthin (Figs 1-3). The combined percentage of these two xanthophylls remained virtually constant (42.5±1.3, Lycopersicon wild-type; 37.7 ± 0.8 , L. notabilis; 42.6 ± 1.4 , Phaseolus) while the ratio of all-trans-violaxanthin: 9'-cis-neoxanthin changed from ca 5:1 to ca 1:1. There was a fall in the percentage of violaxanthin and neoxanthin in Phaseolus after 60 hr but this was probably due to preferential photodestruction of these compounds over lutein as leaf expansion occurred [see 22]. Longer term experiments have shown that the ratio of 9'-cis-neoxanthin: all-transviolaxanthin does not change from ca 1:1 to 3:2, possibly as a result of structural constraints within the pigmentprotein complexes.

These results suggest that during the transition of etioplasts into chloroplasts all-trans-violaxanthin acts as



Time (hours after exposure to light)







Time (hours after exposure to light)

Fig. 2. Stacked bar graph showing changes in xanthophyll composition in leaves of fluridone-treated seven-day-old etiolated *L. esculentum notabilis* seedlings, following transfer from the dark to the light. Key as in Fig. 1.



% of total ranthophylls

Time (hours after exposure to light)



a precursor of 9'-cis-neoxanthin, and that this conversion is stimulated by light, either directly or as a consequence of chloroplast formation (Fig. 4). This is consistent with previous hypotheses regarding neoxanthin formation and the results of carotenoid interconversions in an algal cellfree system [9-11].



Fig. 4. Probable biosynthetic pathway for ABA from all-trans-violaxanthin.

There were no significant changes in the percentage of either all-*trans*-neoxanthin or 9-*cis*-violaxanthin during the conversion of all-*trans*-violaxanthin to 9'-*cis*neoxanthin, and so it is not possible to predict which, if either, is an intermediate in this conversion.

Notabilis (not) is a wilty tomato mutant whose ABA deficiency may result from a lesion in the C_{40} part of the biosynthetic pathway [5, 6]. The combined percentage of all-trans-violaxanthin and 9'-cis-neoxanthin were consistently lower in leaves of not than those of wild-type Lycopersicon or Phaseolus, (Fig. 2) but the conversion of all-trans-violaxanthin to 9'-cis-neoxanthin proceeded as normal. The significance of this is unknown but examinations of the sub-plastidal localization of the xanthophylls in wild-type and not plastids would be interesting.

Dithiothreitol (DTT, 1 mM) has been reported to inhibit the de-epoxidation of violaxanthin to zeaxanthin, during the operation of the 'xanthophyll cycle', by an as yet unknown mechanism [23]. DTT was fed to etiolated *Phaseolus* seedlings via the transpiration stream 24 hr before the seedlings were transferred to light, but no effect on the conversion of violaxanthin to neoxanthin was apparent (data not shown).

We have previously shown that deuterium from D_2O was incorporated into carotenoids in etiolated *Phaseolus* seedlings [2]. Water stress-induced ABA was labelled with deuterium to a similar extent as neoxanthin and violaxanthin. We considered this as confirmatory evid-

ence for the apo-carotenoid nature of ABA. To further investigate the relationship between violaxanthin, neoxanthin and ABA extracts were made of Phaseolus seedlings grown on 50% D₂O, in the dark for five days, and then either kept in the dark on 50% D_2O or water, or transferred to the light and fed with water or 50% D₂O for a further three days. Following transfer from D₂O to water, both in the dark and in the light, a reduction in the amount of labelling occurred (Table 1). In the dark this reduction was between 25 and 28% for violaxanthin and lutein and 16% for neoxanthin. In the light the extent of deuteration of leutein and violaxanthin fell by 44-50% compared to a fall of 32% for neoxanthin, even though the levels of neoxanthin had increased by over six-fold compared to increases of only 1.5-1.7-fold for the other xanthophylls.

This indicates that violaxanthin and lutein were being synthesized from newly formed precursors while neoxanthin was derived from an existing, labelled compound. These data further support the hypothesis that neoxanthin is synthesized from violaxanthin, at least in 'greening' leaves.

The extent of deuteration of ABA and phaseic acid (PA) produced when plants grown under the above conditions were stressed was also determined. As can be seen from Table 1 the extent of labelling of ABA and neoxanthin was comparable in all treatments. We therefore conclude that the xanthophyll pool available to the

Table 1. Extent of deuteration of selected xanthophylls, ABA and PA extracted from leaves of *P. vulgaris*, grown on 50% D_2O in the dark for 5 days and then either kept in the dark on 50% D_2O or water, or transferred to the light and fed with 50% D_2O or water

Treatment, after five days in the dark grown on 50% D ₂ O	Violaxanthin		Neoxanthin		Lutein		104	ПА
	% ² H	Conc.*	% ² H	Conc.	% ² H	Conc.	ава % ² Н	га % ² Н
Dark, 3 days, 50% D2O	18	147	19	30	16	199	15	16
Dark, 3 days, H ₂ O	13	111	16	25	12	134	12	12
Light, 3 days, 50% D ₂ O	14	197	16	118	14	284	17	n.d.
Light, 3 days, H ₂ O	9	256	13	188	9	299	11	n.d.

 $* = nmol g^{-1}$ fr. wt.

n.d. = not determined.

cleavage enzymes is in equilibrium with the bulk of these carotenoids produced by light-induced *de novo* synthesis. Thus while it seems very likely that the neoxanthin utilized for ABA biosynthesis is spatially separate from the bulk of the neoxanthin, which is presumed to exist in the thylakoid pigment-protein complexes, this neoxanthin cannot be detected as a separate pool by the above method.

Neoxanthin and violaxanthin cleavage

In order to investigate the nature of the *in vivo* cleavage of these xanthophylls, potential breakdown products were synthesized by zinc permanganate oxidation. All*trans*-violaxanthin and 9'-cis-neoxanthin were purified in bulk from green *Phaseolus* leaves. Following oxidation fragments were purified by RP-HPLC (system 1) and identified on the basis of UV/VIS absorbance spectra, reactions with dilute acid and mass spectra (Table 2). All*trans*-violaxanthin gave rise to *t*-Xan and the all-*trans* C_{25} and C_{27} epoxy apo-aldehydes (11, 12), while 9'-cisneoxanthin produced Xan, the C_{15} allenic apo-aldehyde (8), the 9'-cis C_{25} and C_{27} epoxy apo-aldehydes (9, 10), consistent with previous studies (Fig. 5; [24-26]). Specific cleavage of 9-cis-violaxanthin or 9'-cisneoxanthin across the 11,12 (11',12') double bond to produce Xan would also produce either the C_{25} epoxy apo-aldehyde (11) or the C_{25} allenic aldehyde (9). Random cleavage would lead to other fragments being formed. A RP-HPLC system was developed (system 2) in which the apo-aldehydes were well resolved, and in which no major components of the plant extracts examined (absorbing ca 300-500 nm) co-chromatographed with the C_{25} and C_{27} apo-aldehydes.

Extracts of a variety of tissues (etiolated and green leaves of *Lycopersicon* and *Phaseolus*, non-stressed and stressed, both for varying times and to varying degrees; green leaves of the wilty ABA-deficient mutants of *Lycopersicon*, not, flacca and sitiens; and roots of *Lycopersicon*) were analysed using this HPLC system but no traces of any of the C₂₅ or C₂₇ apo-aldehydes were detected. In one case the levels of all-*trans*-violaxanthin and 9'-cisneoxanthin in etiolated *Phaseolus* leaves fell by almost 20 μ g g⁻¹ fr. wt following stress, but no C₂₅ or C₂₇ apo-aldehydes were found, despite a limit of detection estimated to be <10 ng g⁻¹.

The C_{25} and C_{27} epoxy apo-aldehydes of interest here (11, 12) have been isolated from various fruits [13, 27] but the corresponding apo-alcohols, which may be formed by

Table 2. Some properties of all-trans-violaxanthin, 9'-cis-neoxanthin and the apo-carotenoids formed from them by zinc permanganate oxidation

Compound	R, (min; RP- HPLC system 2)	λ _{max} (nm; in eluate)*	Shift + acid (nm)	λ_{\max} compared to published data			
				Present	Published	Ref./solvent	
C ₁₅ Allenic apo-ald.	6.0	292	n.d.	291	290	24/EtOH	
Xan/t-Xan	11.0	285	n.d.	282	283	37/MeOH	
C23 Allenic apo-ald.	18.5	419	0	430, 405 (385)	431, 408, 389	24/C ₆ H ₁₂	
C ₂₇ Allenic apo-ald.	20.0	443	0	456, 430 (408)		$-/C_6H_{12}$	
C25 t ⁺ -Epoxy apo-ald.	21.9	420	22-26	(439) 414	(440) 414	26/C6H6	
C ₂₅ c-Epoxy apo-ald.	22.2	414	19-22	(435) 411	(434) 410	26/C H	
C ₂₇ t-Epoxy apo-ald.	23.2	442	22	(469) 440	(470) 442	26/C6H6	
C ₂₇ c-Epoxy apo-ald.	23.6	439	13	(462) 438	(468) 439	26/C6H6	
9'-cis-Neoxanthin	27.0	465, 437, 413	14-16	466, 437, 414	467, 438, 415	35/EtOH	
All-t-violaxanthin	29.5	472, 442, 418	39-43	470, 441, 420	470, 441, 422	36/EtOH	

*Spectra recorded in eluate (HPLC system 2) similar (± 1 -2 nm) to those recorded in ethanol.

 $\dagger t = all - trans, c = 9 - cis.$

n.d. = not determined.



Fig. 5. Apo-aldehydes formed from zinc permanganate oxidation of violaxanthin and neoxanthin.

reduction of the apo-aldehydes, are more commonly found [27-30]. Oxidation of apo-aldehydes to the carboxylic acids could also occur, as several apo-acids have been isolated from plants [8, 31, 32]. Comparisons were made between extracts of non-stressed and stressed etiolated bean leaves, by HPLC (systems 2-4), but no compounds were found to accumulate which could have been C_{25} or C_{27} apo-alcohols or apo-acids. Retention times of the corresponding apo-carotenals and apo-carotenols on **RP-HPLC** are similar [8]. Methylated extracts were also examined, in the hope that any methyl-apo-acids present would have polarities similar to the apo-aldehydes, but no evidence was obtained to suggest that any apo-acids were present.

A further possibility was that any apo-carotenoid acids formed were being esterified. HPLC (systems 1–4) analysis of normal or hydrolysed extracts (pectinase for 18 hr or pH $11/60^{\circ}$ for 30 min) did not suggest that this was occurring.

In tomato leaves a C_{10} 'by-product' of ABA biosynthesis has been identified, which may be derived from the central portion of a C_{40} xanthophyll [33]. If this is the case then the 'other' C_{15} end of 9'-cis-neoxanthin or 9-cisviolaxanthin might also accumulate. Previous work has shown that following stress t-Xan, the 'other' end of 9-cisviolaxanthin, does accumulate to a limited extent in mature tomato leaves but not at all in etiolated bean leaves [2, 5]. The presence of the C_{15} allenic apoaldehyde (8) that would be produced from neoxanthin was investigated by HPLC (systems 2, 3 and 5) and GC-MS but none was found in any extract examined in this present study. Of all the possible C_{13} apo-carotenoids derivable by the simple oxidation of tobacco carotenoids only two have not been isolated, one of these is the C_{13} allenic ketone related to neoxanthin [12, 34]. It has been suggested that this may result from its rapid metabolism to more stable compounds [34].

Lipoxygenase (LOX), in the presence of certain fatty acids such as linoleic, acid, can cause carotenoid cleavage in vitro [15-17], and was used here to bleach solutions of neoxanthin and violaxanthin. Conditions were optimized using zeaxanthin. The bleaching reaction could be stopped completely by the addition of the anti-oxidant tertbutylated hydroxyquinoline (TBHQ; [16]). Incubations interrupted with TBHQ were found to contain low levels of C_{25} and C_{27} apo-aldehydes (<5% of the peak area of remaining C₄₀), both epoxy and allenic apo-aldehydes from neoxanthin and epoxy apo-aldehydes from violaxanthin. However the presence of these compounds was transient and after 15-20 min no traces were found. Xan/t-Xan and the C15 allenic apo-aldehyde were produced as well but in low yields (Xan/t-Xan, 1.9-4.1% of initial violaxanthin, 1.7-3.0% of neoxanthin; $C_{1.5}$ allene, only ca 25% of the levels of Xan/t-Xan based on A_{280} nm), even though over 95% of the C₄₀ xanthophylls were cleaved during the incubations. LOX/fatty acid free radical cleavage of xanthophylls appeared to show no substrate specificity and resulted in complete fragmentation, producing molecules of $\leq C_{13}$.

The stoichiometry observed between losses of neoxanthin and violaxanthin and increases in ABA and its metabolites in stressed etiolated bean leaves suggests that on a molar basis a 1:1 relationship exists [2, 3]. This implies that a specific cleavage, across the 11,12 (11',12') double bond, of the 9-cis xanthophylls is occurring to liberate Xan. It may be that a specific enzyme, possibly a dioxygenase, performs this first cleavage after which LOX and/or related enzymes rapidly degrade any C_{25} apoaldehydes released.

EXPERIMENTAL

Plant material. Seeds of Lycopersicon esculentum Mill. cv Ailsa Craig wild-type and notabilis were obtained from Dr I. B. Taylor (University of Nottingham School of Agriculture, Sutton Bonington, Leics., U.K.). Etiolated L. esculentum and Phaseolus vulgaris seedlings were grown on vermiculite at 22° in the dark for up to 7 days. Plants to be treated with fluridone were grown on H₂O for 5–6 days and then sprayed with fluridone (13 mg1⁻¹) for 2 days before being transferred to weak light (17 µmol m⁻² sec⁻¹), after which they were sprayed twice daily with 1 mg1⁻¹ fluridone. Leaves were harvested at intervals, frozen and stored in liquid N₂. Water stressed and light-grown material was obtained as previously [2].

Determination and identification of carotenoids and ABA. As described in ref. [2].

Deuterium oxide labelling experiments. These were performed as described previously [2], using Phaseolus vulgaris cv Canadian Wonder seedlings. After 5 days on 50% D_2O in the dark some seedlings were placed in weak light (17 μ mol m⁻² sec⁻¹), and some transferred to H₂O. The latter had their 'growth pouches'/root systems thoroughly flushed with H₂O. The extent of deuteration of the carotenoids, ABA and PA was calculated using a BASIC computer programme bassed on the method of Wendt and McClosky [38], as in ref. [2].

Large-scale extraction of carotenoids. Two-week-old lightgrown P. vulgaris leaves (106 g) were homogenized in 500 ml of redistilled Me₂CO with 20 g of Celite-535, filtered, reduced to aq. (ca 50 ml) and partitioned against C_6H_{12} (30 ml). The sample, in C_6H_{12} , was subjected to flash chromatography (17 × 5 cm i.d.; silica gel; eluted with C_6H_{12} -iso-PrOH-TEA, 800:200:1), 100 ml frs collected and UV/VIS spectra recorded. Violaxanthin (frs 5-6) and neoxanthin (frs 7-11) were collected separately and reduced to dryness. Further purification was by RP/HPLC (system 1), after which the xanthophylls (all-*trans*-violaxanthin, 0.7 mg; 9'-cis-neoxanthin, 1.6 mg) were stored in EtOH at -20° .

Zinc permanganate oxidation. This was based on methods used by [24-26] but performed on a smaller scale. The technique was optimized using commercially prepared zeaxanthin. Between 250 and 500 μ g of xanthophyll was dissolved in Me₂CO (2:1; v/w) in a mini-vial, covered with foil, and aq. Zn(MnO₄)₂ (2 mg ml⁻¹) added dropwise [0.25 mg Zn(MnO₄)₂/mg xanthophyll] over 1 hr with vigorous stirring. After 1 hr the mixt. was reduced to aq., made up to 5 ml with H₂O and partitioned twice against EtOAc (10 ml). The EtOAc was washed with H₂O prior to being reduced to dryness. The samples were stored in EtOAc at -20°. Analysis was by RP-HPLC (systems 1 and 2, samples injected in 20-30 μ l EtOAc), linked to a photodiode-array detector, and GC-MS.

C₁₅ allenic apo-aldehyde (8) (3,5-dihydroxy-6,7-didehydro-5,6-dihydro-apo- β -caroten-11-al): EIMS (GC-MS), m/z (rel. int.): 250 [M]⁺ (7), 232 (100), 214 (33), 199 (61), 149 (57), 133 (64), 105 (42), 91 (39), 77 (36), 69 (25), 43 (57).

 $cis-C_{25}$ epoxy apo-aldehyde (9'-cis-5',6'-epoxy-3'-hydroxy-5',6'-dihydro-12-apo- β -caroten-12-al): probe MS, m/z (rel. int.): 382 [M]⁺ (100), 364 (9), 302 (32), 287 (19), 234 (10), 221 (26), 173 (26), 159 (26), 145 (32), 119 (35), 105 (37), 91 (49).

trans- C_{25} epoxy apo-aldehyde (11) (5,6-epoxy-3-hydroxy-5,6dihydro-12'-apo- β -caroten-12'-al): probe MS, m/z (rel. int.): 382 [M] ⁺ (100), 364 (30), 302 (35), 287 (21), 234 (14), 221 (32), 173 (37), 159 (39), 145 (46), 119 (49), 105 (54), 91 (88).

 $cis-C_{27}$ epoxy apo-aldehyde (9'-cis-5',6'-epoxy-3'-hydroxy-5'6'-dihydro-10-apo- β -caroten-10-al): probe MS, m/z (rel. int.): 408 [M]⁺ (88), 390 (5), 328 (22), 299 (10), 287 (10), 234 (21), 221 (41), 181 (38), 145 (43), 119 (55), 105 (60), 91 (100).

trans- C_{27} epoxy apo-aldehyde (12) (5,6-epoxy-3-hydroxy-5,6dihydro-10'-apo- β -caroten-10'-al): probe MS, m/z (rel. int.): 408 [M]⁺ (31), 390 (9), 328 (7), 299 (4), 287 (5), 234 (12), 221 (22), 181 (22), 145 (26), 119 (36), 105 (38), 91 (100).

C₂₅ allenic apo-aldehyde (9) (3,5-dihydroxy-6,7-didehydro-5,6-dihydro-12'-apo- β -caroten-12'-al): probe MS, m/z (rel. int.): 382 [M]⁺ (21), 364 (40), 346 (76), 331 (19), 275 (12), 263 (15), 247 (24), 233 (24), 221 (27), 207 (29), 197 (45), 167 (100), 119 (43), 105 (45), 91 (59).

 C_{27} allenic apo-aldehyde (10) (3,5-dihydroxy-6,7-didehydro-5,6-dihydro-10'-apo- β -caroten-10'-al): probe MS, m/z (rel. int.): 408 [M]⁺ (12), 390 (24), 372 (31), 357 (8), 311 (12), 267 (20), 247 (16), 230 (35), 215 (43), 197 (67), 167 (55), 119 (53), 105 (61), 91 (100).

Lipoxygenase incubations. The procedure used was that of ref. [15] but on a small scale. Aq. linoleic acid soln: linoleic acid in EtOH (7.5% w/v; 1 ml) mixed with Tween 80 in EtOH (10% v/v; 0.3 ml), aq. EDTA (0.5%; 5 ml) added, the pH adjusted to 9 (0.1 M NaOH) and the vol. made up to 10 ml with H₂O. Aq. carotenoid solns: 250 µg of carotenoid plus 9 µl Tween 80 dissolved in 250 μ l CHCl₃, reduced to dryness under vacuum, dissolved in EDTA (0.25%; 1 ml). Incubation: 1 ml of linoleic acid soln and 1 ml of carotenoid soln made up to 12.5 ml with H_2O , aliquots removed (2-10 ml) to plastic cuvettes and LOX (lipoxidase; EC 1.13.11.12; Sigma type V; 737 000 units mg⁻¹) added (300 units ml⁻¹ of incubation). Bleaching monitored continuously at 460 nm (zeaxanthin), 445 nm (neoxanthin) or 450 nm (violaxanthin). TBHQ added in EtOH (5 μ l ml⁻¹ of incubation) to final conc. of 480 µm. After 5-20 min the incubation mixt. was loaded on to a C18 Sep-Pak cartridge (pre-washed

with 5 ml MeOH and 10 ml H_2O), washed with 10 ml of H_2O and the reaction products eluted with EtOAc (2-3 ml). Analysis of the products was by RP-HPLC (systems 2-4) and HPLC (system 5).

Chromatography. HPLC systems: 1. ODS-Spherisorb (250 \times 10 mm i.d.) eluted with a linear gradient of 10-50% B in A over 25 min and maintained at 100% B for a further 15 min, at 2 ml min⁻¹, where A is 85% MeOH (+0.1% TEA; v/v) and B is CH2Cl2-MeOH (3:2). Eluate monitored between 300 and 500 nm. 2. ODS-Spherisorb (250 × 4.5 mm i.d.) eluted with 30-100% A₁ in B₁ over 15 min followed by 0-100% C₁ in A₁ over 30 min at 1 ml min⁻¹, where A_1 is MeCN-H₂O (9:1; +0.1% TEA; v/v), B_1 is H_2O (+0.1% TEA; v/v) and C_1 is EtOAc. Eluate monitored between 220 and 500 nm. 3. As 2 except eluted with 0-100% A_1 in B_1 over 45 min. 4. ODS-Spherisorb (250 × 4.5 mm i.d.) eluted with a linear gradient of 20-100% MeOH in 0.1 MHOAc over 40 min at 1 mlmin⁻¹ eluate monitored between 220-500 nm. 5. Spherisorb (250 \times 4.5 mm i.d.) eluted isocratically with C₆H₁₂-iso-PrOH (22:3) at 2 ml min⁻¹, eluate monitored between 220 and 500 nm.

GC-MS. GC: Phase Sep. OV-1 bonded-phase column (25 m $\times 0.32$ mm i.d.; 0.4 μ m film thickness) with a temp. prog. of 35–180° (ballistic)-250° (at 8° min⁻¹), He at 20 ml min⁻¹, linked to MS via a single stage jet separator at 230°. MS: resolution > 600, ionising voltage 70 eV, source 190°, probe temp. 190–220°.

Acknowledgements—We thank Mr J. K. Heald for his expert operation of the GC-MS, Dr D. C. Walton for many useful discussions and Dr B. H. Davies for his advice. ADP was supported by the AFRC, and the HPLC-photodiode-array detector and GC-MS were purchased with funds provided by the AFRC.

REFERENCES

- Zeevaart, J. A. D. and Creelman, R. A. (1988) Ann. Rev. Plant Physiol. Mol. Biol. 39, 439.
- Parry, A. D., Babiano, M. J. and Horgan, R. (1990) Planta 182, 118.
- 3. Li, Y. and Walton, D. C. (1990) Plant Physiol. 92, 551.
- Taylor, H. F. and Burden, R. S. (1974) in *Biochemistry and* Chemistry of Plant Growth Regulators (Schreiber, K., ed.), p. 187. Academy of Sciences, G.D.R., Halle (Saale).
- Parry, A. D., Neill, S. J. and Horgan, R. (1988) Planta 173, 397.
- Parry, A. D. (1989) Ph.D. Thesis. University College of Wales, Aberystwyth.
- 7. Britton, G. (1982) Physiol. Veg. 20, 735,
- Britton, G. (1988) in *Plant Pigments* (Goodwin, T. W., ed.), p. 133. Academic Press, London.
- 9. Weedon, B. C. L. (1970) Rev. Pure Appl. Chem. 20, 51.
- Goodwin, T. W. (1980) The Biochemistry of the Carotenoids Vol. 1. Chapman & Hall, London.
- 11. Swift, I. E., Milborrow, B. V. and Jeffrey, S. W. (1982) Phytochemistry 21, 2859.
- 12. Enzell, C. (1985) Pure Appl. Chem. 57, 693.
- 13. Molnar, P. and Szabolcs, J. (1980) Phytochemistry 19, 633.
- 14. Taylor, H. F. and Burden, R. S. (1970) Phytochemistry 9, 2217.
- Ben-Aziz, A., Grossman, S., Ascarelli, I. and Budowski, P. (1971) Phytochemistry 10, 1445.
- Klein, B. P., Grossman, S., King, D., Cohen, B. S. and Pinsky, A. (1984) Biochem. Biophys. Acta 793, 72.
- 17. Firn, R. D. and Friend, J. (1972) Planta 103, 263.
- 18. Matile, P. and Martinoia, E. (1982) Plant Cell Rep. 1, 244.

- Malhotra, K., Oelze-Karow, H. and Mohr, H. (1982) Planta 154, 361.
- Britton, G. (1986) in Regulation of Chloroplast Differentiation (Akoyunoglu, G. and Senger, H., eds), p. 125. Alan R. Liss, New York.
- Ridley, S. M. (1982) in Carotenoid Chemistry and Biochemistry (Britton, G. and Goodwin, T. W., eds), p. 353. Pergamon Press, Oxford.
- 22. Barry, P., Young, A. J. and Britton, G. (1990) J. Exp. Botany 41, 123.
- 23. Winter, K. and Koniger, M. (1989) Planta 180, 24.
- Bonnett, R., Mallams, A. K., Spark, A. A., Tee, J. L., Weedon, B. C. L. and McCormick, A. (1969) J. Chem. Soc. (C), 429.
- Molnar, P. and Szabolcs, J. (1979) Acta Chim. Acad. Sci. Hung. 99, 155.
- 26. Szabolcs, J. (1976) Pure Appl. Chem. 47, 147.
- 27. Gross, J. and Eckhardt, G. (1981) Phytochemistry 20, 2267.
- Molnar, P., Szabolcs, J. and Radics, L. (1987) Phytochemistry 26, 1493.
- 29. Gross, J., Gabai, M., Lifshitz, A. and Sklarz, B. (1974) Phytochemistry 13, 1917.

- 30. Gross, J. and Eckhardt, G. (1978) Phytochemistry 17, 1803.
- 31. Diallo, B. and Vanhaelen, M. (1987) Phytochemistry 26, 1491.
- Jandiko, I. J. O. and Pattenden, G. (1989) Phytochemistry 28, 3159.
- Taylor, I. B. (1987) in Developmental Mutants in Higher Plants (Thomas, H. and Grierson, D., eds), p. 197. C.U.P., Cambridge.
- 34. Enzell, C. R., Wahlberg, I. and Aasen, A. J. (1977) in Progress in the Chemistry of Organic Natural Products (Herz, W., Grisebach, H. and Kirby, G. W., eds), p. 1. Springer, Wien.
- Cholnoky, L., Gyorgyfy, K., Ronai, A., Szabolcs, J., Toth, Gy., Galasko, G., Mallams, A. K., Waight, E. S. and Weedon, B. C. L. (1969) J. Chem. Soc. (C), 1256.
- Davies, B. H. (1976) in The Chemistry and Biochemistry of Plant Pigments Vol. 2 (Goodwin, T. W., ed.), p. 38. Academic Press, London.
- Burden, R. S. and Taylor, H. F. (1970) *Tetrahedron Letters* 47, 4071.
- 38. Wendt, G. and McClosky, J. A. (1970) Biochemistry 9, 4854.