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Organocatalytic asymmetric multi-component [C+NC+CC] synthesis of highly functionalized pyrrolidine derivatives

Ismail Ibrahem, Ramon Rios, Jan Vesely and Armando Córdova*

Department of Organic Chemistry, Arrhenius Laboratory, Stockholm University, SE-106 91 Stockholm, Sweden

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Abstract—The highly chemo- and enantioselective organocatalytic [C+NC+CC] coupling process between aldehydes, dialkyl 2aminomalonates and α , β -unsaturated aldehydes is presented. The reaction gives access to highly functionalized pyrrolidine derivatives in good yields with >10:1 dr and 90–98% ee. © 2007 Elsevier Ltd. All rights reserved.

The pyrrolidine ring is an important structural motif found in several bioactive molecules. For example, it is present in the neuroprotective agent kaitocephalin,¹ the influenza drug A-192558,² and the antitumor antibiotic bioxalomycin β 1.³ Pyrrolidines also serve as useful molecular scaffolds for the exploration and exploitation of pharmacophore space via diversity-oriented synthesis (DOS),^{4–6} which has led to the findings of new drug leads for the treatment of cancer⁷ and hepatitis C viral infections.⁸ Thus, intense research is ongoing for the development of new stereocontrolled routes to chiral pyrrolidines. For example, asymmetric methods that rely on 1,3-dipolar cycloaddition reactions have been developed for their preparation.^{9,10} In this area, organo-

electron-deficient alkenes (normal electron demand reaction) have been successfully employed for the synthesis of functional pyrrolidines.^{4,11} In the rapidly growing research field of organocatalysis, MacMillan first reported a catalytic enantioselective synthesis of isoxazolidines based on the chiral imidazolidinone catalyzed 1,3-dipolar cycloaddition between preformed nitrones and enals.^{12–18} The iminium activation mechanism, which is central in this type of catalysis, has been successfully applied in several asymmetric reactions. In this context, we recently reported a catalytic, highly enantioselective isoxazolidine synthesis based on an organocatalytic asymmetric multi-component 1,3-dipolar cycloaddition (Eq. 1).¹⁹

metallic complex-catalyzed enantioselective cycloaddition transformations between azomethine ylides and Recently, Garner et al. reported a stereoselective synthesis of functional pyrrolidines based on the union of an aldehyde ('C'), an amino acid derivative ('NC'), and an electron-deficient alkene ('CC') using a chiral auxiliary in what they termed a three-component

^{*} Corresponding author. Tel.: +46 8 162479; fax: +46 8 154908; e-mail addresses: acordova@organ.su.se; acordova1a@netscape.net

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Scheme 1. Organocatalytic asymmetric aldehyde \rightarrow imine \rightarrow dipole \rightarrow cycloadduct reaction cascade.

[C+NC+CC] coupling reaction.²⁰ Notably, Schreiber reported one example of a one-pot, three-component synthesis of a highly substituted pyrrolidine derivative catalyzed by a chiral silver complex.⁴ Inspired by these findings and the importance of asymmetric multi-component reactions,^{4,19–21} we envisioned that a small chiral amine would be able to catalyze an asymmetric threecomponent cascade imine → azomethine ylide → 1,3dipolar cycloaddition [C+NC+CC] sequence according to Scheme 1. To the best of our knowledge, no report of an organocatalytic highly enantioselective pyrrolidine synthesis based on an asymmetric multi-component 1,3dipolar cycloaddition has been disclosed to date.²²

Herein, we present a highly chemo- and enantioselective one-pot, three-component organocatalytic route to the synthesis of valuable pyrrolidine derivatives in good yields with 3:1->10:1 dr and 90-98% ee.

In an initial catalyst screen for the reaction between benzaldehyde 1a (0.375 mmol), diethyl 2-aminomalonate 2a (0.375 mmol), and 2-heptenal 3a (0.25 mmol), we found that proline 6 and primary amino acids such as valine 7 catalyzed the chemoselective formation of functionalized pyrrolidine 4a in high conversion and moderate enantioselectivity (Table 1). Simple chiral pyrrolidines such as 8–12 also catalyzed the asymmetric formation of 4a. In addition, MacMillan's amino acid derivative 13 catalyzed the formation of 4a in low conversion and with excellent diastereoselectivity (19:1 endo/exo) but low enantioselectivity under our reaction conditions (entry 8). To our delight, the protected diarylprolinol 11^{23} catalyzed the formation of 4a with high efficiency and excellent enantioselectivity in CHCl₃ (entry 6).²⁴ The chiral pyrrolidine **4a** was the only product observed by proton NMR analyses of the crude reaction mixture, however, the chiral pyrrolidine 4a is rather unstable and decomposed upon isolation and storage. The highest diastereo- and enantioselectivity for the reaction was achieved when CHCl₃ or toluene was used as the solvent. For example, the highest dr (endo/exo) was accomplished in toluene when the aliphatic enal 3a was used as the substrate (entry 9). Moreover, the addition of a stoichiometric amount of triethylamine (TEA) improved the diastereoselectivity of the reaction.

Thus, we decided to investigate the scope of the catalytic asymmetric one-pot, three-component reaction using toluene and $CHCl_3$ as the solvent, TEA as an additive and chiral amine **11** as the catalyst (Table 2).



| | Ph H | + CO ₂ EtO ₂ C + NH ₂ 2a | Et + Bu | Catalyst H (20 mol%) H <u>(20 mol%)</u> Ph` Solvent, rt Et ₃ N (1 equiv) | CO_2Et N CO_2Et H H 4a | |
|-------|----------|--|----------------------|---|---|----------------------|
| | | Соон Н ₂ м | соон С | OH H OH H | | |
| | | 6 | 7 8 | 9 ₀ 1 | 0 ~ | |
| | | | h —Ph CN TMS H | Ar Ar OTMS | 1 | |
| | | 11 | 12 : A | Ar=di(3,5-CF ₃)C ₆ H ₃ 13 | | |
| Entry | Catalyst | Solvent | Time (h) | Conv. ^a (%) | dr ^b | ee ^c (%) |
| 1 | 6 | CHCl ₃ | 20 | 100 | 8:1 | 50 |
| 2 | 7 | CHCl ₃ | 66 | 66 | 2:1 | 17 |
| 3 | 8 | CHCl ₃ | 20 | 76 | 5:1 | 33 |
| 4 | 9 | CHCl ₃ | 20 | <10 | n.d. | n.d. |
| 5 | 10 | CHCl ₃ | 20 | 90 | 2:1 | 33 |
| 6 | 11 | CHCl ₃ | 44 | 100 | 3:1 | 97 |
| 7 | 12 | CHCl ₃ | 20 | 52 | 3:2 | 39 |
| 8 | 13 | CHCl ₃ | 20 | 19 | 19:1 | 13 |
| 9 | 11 | Toluene | 20 | $100(51)^{d}$ | $10:1 (10:1)^d$ | 95 (95) ^d |
| 10 | 11 | DMF | 20 | 52 | 3:2 | 29 |
| 11 | 11 | CH ₃ CN | 20 | 70 | 3:2 | 26 |

^a Conversion of **3a** as determined by NMR analyses of the crude reaction mixture.

^b endo/exo-Ratio determined by NMR analyses of the crude reaction mixture.

^c Determined by chiral-phase HPLC analyses of **4a**.

^d The results for the isolated product after silica-gel column chromatography.

The organocatalytic enantioselective three-component reactions were highly chemo- and enantioselective. The corresponding chiral pyrrolidine derivatives 4 were obtained in 50-63% yields with 3:1->10:1 dr and 90-98% ee. For example, the reaction between benzaldehyde 1a, 2-aminomalonate 2a, and cinnamic aldehyde 3b gave the corresponding product 4d in 63% yield with 10:1 dr and 95% ee (entry 4). Several different aromatic and aliphatic enals 3 could be used as acceptors for the reaction. Moreover, various benzaldehyde derivatives were used as the aldehyde moiety. The reactions were also readily performed in parallel. Thus, the one-pot, three-component organocatalytic cascade [C+NC+CC] coupling process should be useful as a platform in diversity-oriented synthesis (DOS) of pyrrolidine derivatives with three contiguous chiral centers. The chiral pyrrolidine catalyzed asymmetric cascade [C+NC+CC] reaction could also be extended to a one-pot in situ oxidation to obtain the more stable acid functionalized chiral pyrrol-idine derivative **14** (Scheme 2).²⁵ For example, the onepot organocatalytic asymmetric tandem 1,3-reaction dipolar/oxidation sequence between aldehyde 1a, aminomalonate 2a, and enal 3b gave the corresponding acid 14d in 52% yield with 10:1 dr and 95% ee. Next, pure acid 14d was converted quantitatively to the corresponding chiral proline derivative 15d (1:1 mixture of two diastereoisomers) under acidic conditions.²⁶ Quantitative methylation of diacid **15d** gave proline derivatives $16d^{27}$ and 16d' as single diastereomers after isolation. Thus, the chiral pyrrolidine catalyzed one-pot reaction can be used as a novel route for the preparation of useful chiral proline derivatives with four contiguous stereocenters.

Comparison with the literature revealed the relative configuration of **16d** and established the stereochemistry of pyrrolidine derivatives **4**.²⁷ This is in accordance with previous chiral pyrrolidine **11** catalyzed 1,3-dipolar cycloadditions.^{12d,19} Thus, efficient shielding of the *Si*-face of the chiral iminium intermediate by the bulky aryl groups of **11** leads to stereoselective *Re*-facial *endo*-addition to the activated olefin via the plausible transition state depicted in Scheme 1.

In summary, we have described a simple highly chemoand enantioselective organocatalytic one-pot, threecomponent, cascade [C+NC+CC] reaction sequence between aldehydes, dialkyl aminomalonate, and α , β -unsaturated aldehydes. The reaction represents a versatile asymmetric entry to a variety of valuable highly functionalized pyrrolidine derivatives in good yields with 3:1–>10:1 dr and 90–98% ee. Mechanistic studies, synthetic applications of this transformation in DOS as well as the development of other enantioselective multi-component reactions are ongoing in our laboratory.

Table 2. Scope of the organocatalytic three-component reaction



| | | | | 0 () | | | | |
|-------|---------------|-----------------------------------|------------|----------|------------------------|-----------------|---------------------|--|
| Entry | R | \mathbb{R}^1 | Product | Time (h) | Yield ^a (%) | dr ^b | ee ^c (%) | |
| 1 | <i>n</i> -Bu | Ph | 4a | 20 | 51 ^d | 10:1 | 95 | |
| 2 | <i>n</i> -Bu | $4-BrC_6H_4$ | 4b | 20 | 58 ^d | 8:1 | 92 | |
| 3 | <i>n</i> -Bu | $4 - NO_2C_6H_4$ | 4 c | 20 | 59 ^d | 5:1 | 93 | |
| 4 | Ph | Ph | 4d | 20 | 63 ^e | 10:1 | 95 | |
| 5 | <i>n</i> -Bu | $4-MeOC_6H_4$ | 4e | 20 | 61 ^e | 3:1 | 98 | |
| 6 | Ph | Ph | 4f | 20 | 58 ^d | >10:1 | 90 | |
| 7 | $4-BrC_6H_4$ | Ph | 4g | 20 | 60 ^e | 10:1 | 92 | |
| 8 | $4-ClC_6H_4$ | Ph | 4h | 20 | 52 ^e | 8:1 | 97 | |
| 9 | 2-Naphth | Ph | 4i | 20 | 60 ^e | 8:1 | 96 | |
| 10 | <i>n</i> -Pr | 4-BrC ₆ H ₄ | 4j | 44 | 57 ^e | 5:1 | 98 | |
| 11 | $4-ClC_6H_4$ | $4-ClC_6H_4$ | 4k | 44 | 50 ^e | 8:1 | 97 | |
| 12 | Ph | $4-CNC_6H_4$ | 41 | 44 | 52 ^e | 6:1 | 92 | |
| 13 | <i>n</i> -Pr | $4-ClC_6H_4$ | 4 m | 44 | 58 ^e | 6:1 | 96 | |
| 14 | <i>n</i> -Hex | 4-CNC ₆ H ₄ | 4n | 44 | 55 ^e | 5:1 | 95 | |
| 15 | <i>n</i> -Et | 4-ClC ₆ H ₄ | 4 o | 20 | 55 ^e | 5:1 | 92 | |
| 16 | <i>n</i> -Pr | 4-MeC ₆ H ₄ | 4p | 20 | 62 ^e | 4:1 | 98 | |

^a Isolated yield of the pure product **4** after silica-gel chromatography.

^b endo/exo-Ratio determined by NMR analyses of the crude reaction mixture.

^c Determined by chiral-phase HPLC analyses.

^d Reaction run in toluene.

^e Reaction run in CHCl₃.



Scheme 2. Reagents and conditions: (a) (i) (S)-11 (20 mol %), Et₃N, CHCl₃, rt, 20 h, 85%; (ii) NaClO₂, isobutene, KHPO₄, *t*-BuOH–H₂O 2:1, 52%. (b) HCl (6 N), AcOH, reflux, 16 h, >99%. (c) TMSCHN₂, benzene–MeOH, rt, 1.5 h, >99%.

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References and notes

 (a) Kobayashi, H.; Shin-ya, K.; Furihata, K.; Hayakawa, Y.; Seto, H. *Tetrahedron Lett.* **2001**, *42*, 4021; (b) Ma, D.; Yang, J. J. Am. Chem. Soc. **2001**, *123*, 9706; (c) Kawasaki, M.; Shinada, T.; Hamada, M.; Ohfune, Y. Org. Lett. 2005, 7, 4165.

- (a) Hanessian, S.; Bayrakdarian, M.; Luo, X. J. Am. Chem. Soc. 2002, 124, 4716; (b) DeGoey, D. A.; Chen, H.-J.; Flosi, W. J.; Grampovnik, D. J.; Yeung, C. M.; Klein, L. L.; Kempf, D. J. J. Org. Chem. 2002, 67, 5445.
- (a) Zaccardi, J.; Alluri, M.; Ashcroft, J.; Bernan, V.; Korshalla, J. D.; Morton, G. O.; Siegel, M.; Tsao, R.; Williams, D. R.; Maeiese, W.; Ellestad, G. A. J. Org. Chem. 1994, 59, 4045; (b) Scott, J. D.; Williams, R. M. Chem. Rev. 2002, 102, 1669.
- Lo, M. M.-C.; Neimann, C. S.; Nagayama, S.; Perlstein, E. O.; Schreiber, S. L. J. Am. Chem. Soc. 2004, 126, 16077.
- (a) Hanessian, S.; Bayrakdarian, M. Bioorg. Med. Chem. Lett. 2000, 10, 427; (b) Hanessian, S.; Bayrakdarian, M. Tetrahedron Lett. 2002, 43, 9441.
- MacClean, D.; Schullek, J. R.; Murphy, M. M.; Ni, Z.-J.; Gordon, E. M.; Gallop, M. A. Proc. Natl. Acad. Sci. U.S.A. 1997, 94, 2805.
- Ding, K.; Lu, Y.; Nikolovska-Coleska, Z.; Qiu, S.; Ding, Y.; Gao, W.; Stuckey, J.; Krajewski, K.; Roller, P. P.; Tomita, Y.; Parrish, D. A.; Deschamps, J. R.; Wang, S. J. Am. Chem. Soc. 2005, 127, 10130.
- Burton, G.; Ku, T. W.; Carr, T. J.; Kiesow, T.; Sarisky, R. T.; Lin-Goerke, J.; Baker, A.; Earnshaw, D. L.; Hofmann, G. A.; Keenan, R. M.; Dhanak, D. *Bioorg. Med. Chem. Lett.* 2005, 15, 1553.
- For reviews on 1,3-dipolar cycloadditions see: (a) Pellissier, H. Tetrahedron 2007, 63, 3235; (b) Karlsson, S.; Högberg, H.-E. Org. Prep. Proced. Int. 2001, 33, 103; (c) Torssell, K. B. G. Nitrile Oxides, Nitrones and Nitronates in Organic Synthesis; VCH: Weinheim, 1998; (d) Gothelf, K. V.; Jørgensen, K. A. Chem. Rev. 1998, 98, 863; (e) Gothelf, K. V. Synthesis 2002, 211; (f) Gothelf, K. V.; Jørgensen, K. A. Chem. Commun. 2000, 1449.
- (a) Pandey, G.; Banerjee, P.; Gadre, S. R. Chem. Rev. 2006, 106, 4484; (b) Najea, C.; Sansano, J. M. Angew. Chem., Int. Ed. 2005, 44, 6272; (c) Husinec, S.; Savic, V. Tetrahedron: Asymmetry 2005, 16, 2047.
- For selected examples see: (a) Saito, S.; Tsubogo, T.; Kobayashi, S. J. Am. Chem. Soc. 2007, 129, 5364; (b) Gothelf, A. S.; Gothelf, K. V.; Hazell, R. G.; Jørgensen, K. A. Angew. Chem., Int. Ed. 2002, 41, 4236; (c) Zeng, W.; Chen, G.-Y.; Zhou, Y.-G.; Li, Y.-X. J. Am. Chem. Soc. 2007, 129, 750; (d) Yan, X.-X.; Peng, Q.; Zhang, Y.; Zhang, K.; Hong, W.; Hou, X.-L.; Wu, Y.-D. Angew. Chem., Int. Ed. 2006, 45, 1979; (e) Dogan, O.; Koyuncu, H.; Garner, P.; Bulut, A.; Youngs, W. J.; Panzner, M. Org. Lett. 2006, 8, 4687; (f) Cabrera, S.; Arrayas, R. G.; Carretero, J. C. J. Am. Chem. Soc. 2005, 127, 16394; (g) Alemparte, C.; Blay, G.; Jørgensen, K. A. Org. Lett. 2005, 7, 4569; (h) Bonini, B. F.; Boschi, M.; Comes-Franchini, M.; Fochi, M.; Fini, F.; Mazzanti, A.; Ricci, A. Synthesis 2005, 543; (i) Stohler, R.; Wahl, F.; Pfaltz, A. Synthesis 2005, 1431.
- (a) Jen, W. S.; Wiener, J. J. M.; MacMillan, D. W. C. J. Am. Chem. Soc. 2000, 122, 9874; For other studies see: (b) Karlsson, S.; Högberg, H.-E. Eur. J. Org. Chem. 2003, 15, 2782; (c) Karlsson, S.; Högberg, H. E. Tetrahedron: Asymmetry 2002, 13, 923; (d) Chow, S. S.; Nevalainen, M.; Evans, C. A.; Johannes, C. W. Tetrahedron Lett. 2007, 48, 277; (e) Chen, W.; Yuan, X.-H.; Li, R.; Du, W.; Wu, Y.; Ding, L.-S.; Chen, Y.-C. Adv. Synth. Catal. 2006, 348, 1818.
- For reviews see: (a) Dalko, P. I.; Moisan, L. Angew. Chem., Int. Ed. 2001, 40, 3726; (b) List, B. Chem. Commun. 2006, 819; (c) Dalko, P. I.; Moisan, L. Angew. Chem., Int. Ed. 2004, 43, 5138; (d) Marigo, M.; Jørgensen,

K. A. Chem. Commun. 2006, 2001; (e) Enders, D.; Grondal, C.; Hüttl, M. R. M. Angew. Chem., Int. Ed. 2007, 46, 1570.

- 14. For selected examples of C-nucleophiles see: (a) Ahrendt, K. A.; Borths, C. J.; MacMillan, D. W. C. J. Am. Chem. Soc. 2000, 122, 4243; (b) Northrup, A. B.; MacMillan, D. W. C. J. Am. Chem. Soc. 2002, 124, 2458; (c) Kunz, R. K.; MacMillan, D. W. C. J. Am. Chem. Soc. 2005, 127, 3240; (d) Paras, N. A.; MacMillan, D. W. C. J. Am. Chem. Soc. 2001, 123, 4370; (e) Halland, N.; Aburell, P. S.; Jørgensen, K. A. Angew. Chem., Int. Ed. 2004, 43, 1272; (f) Brandau, S.; Landa, A.; Franzen, J.; Marigo, M.; Jørgensen, K. A. Angew. Chem., Int. Ed. 2006, 45, 4305; (g) Halland, N.; Hansen, T.; Jørgensen, K. A. Angew. Chem., Int. Ed. 2003, 42, 4955; (h) Halland, N.; Aburel, P. S.; Jørgensen, K. A. Angew. Chem., Int. Ed. 2003, 42, 661; (i) Carlone, A.; Cabrera, S.; Marigo, M.; Jørgensen, K. A. Angew. Chem., Int. Ed. 2007, 46, 1101, and references therein; (j) Gotoh, H.; Masui, R.; Ogino, M.; Shoji, H.; Hayashi, Y. Angew. Chem., Int. Ed. 2006, 45, 6853; (k) Hanessian, S.; Pham, V. Org. Lett. 2000, 2, 2975; (1) Enders, D.; Hüttl, M. R. M.; Grondal, C.; Raabe, G. Nature 2006, 441, 861.
- For selected examples of *H*-nucleophiles see: (a) Yang, J.
 W.; Hechevarria Fonseca, M. T.; Vignola, N.; List, B.
 Angew. Chem., Int. Ed. 2005, 44, 108; (b) Guellet, S. G.; Tuttle, J. B.; MacMillan, D. W. C. J. Am. Chem. Soc.
 2005, 127, 32; (c) Mayer, S.; List, B. Angew. Chem., Int. Ed. 2006, 45, 4193; (d) Wilson, R. M.; Jen, W. S.; MacMillan, D. W. C. J. Am. Chem. Soc. 2005, 127, 11616; (e) Zhao, G.-L.; Córdova, A. Tetrahedron Lett. 2006, 47, 7417.
- For selected examples of S-nucleophiles see: (a) Marigo, M.; Schulte, T.; Franzen, J.; Jørgensen, K. A. J. Am. Chem. Soc. 2005, 127, 15710; (b) Wang, W.; Li, H.; Wang, J.; Zu, L. J. Am. Chem. Soc. 2006, 128, 10354; (c) Rios, R.; Sunden, H.; Ibrahem, I.; Zhao, G.-L.; Eriksson, L.; Córdova, A. Tetrahedron Lett. 2006, 47, 8547; (d) Brandau, S.; Maerten, E.; Jørgensen, K. A. J. Am. Chem. Soc. 2006, 128, 14986.
- For selected examples of N-nucleophiles see: (a) Chen, Y. K.; Yoshida, M.; MacMillan, D. W. C. J. Am. Chem. Soc. 2006, 128, 9328; (b) Ibrahem, I.; Rios, R.; Vesely, J.; Zhao, G.-L.; Córdova, A. Chem. Commun. 2006, 849; (c) Vesely, J.; Ibrahem, I.; Rios, R.; Zhao, G.-L.; Xu, Y.; Córdova, A. Tetrahedron Lett. 2007, 48, 2193; (d) Dinér, P.; Nielsen, M.; Marigo, M.; Jørgensen, K. A. Angew. Chem. 2007, 46, 1983; (e) Vesely, J.; Ibrahem, I.; Zhao, G.-L.; Rios, R.; Córdova, A. Angew. Chem., Int. Ed. 2007, 46, 778; (f) Uria, U.; Vicario, J. L.; Badia, D.; Carillo, L. Chem. Commun. 2007, 46, 1983.
- For selected examples of O-nucleophiles see: (a) Bertelsen, S.; Dinér, P.; Johansen, R. L.; Jørgensen, K. A. J. Am. Chem. Soc. 2007, 129, 1536; (b) Govender, T.; Hojabri, L.; Moghaddam, F. M.; Arvidsson, P. I. Tetrahedron: Asymmetry 2006, 17, 1763; (c) Sundén, H.; Ibrahem, I.; Zhao, G.-L.; Eriksson, L.; Córdova, A. Chem. Eur. J. 2007, 13, 574.
- Rios, R.; Ibrahem, I.; Vesely, J.; Zhao, G.-L.; Córdova, A. *Tetrahedron Lett.* 2007, 48, 5701.
- (a) Garner, P.; Kaniskan, H. Ü.; Hu, J.; Youngs, W. J.; Panzner, M. Org. Lett. 2006, 8, 3647; For other examples of this approach see: (b) Garner, P.; Kaniskan, H. Ü. J. Org. Chem. 2005, 70, 10868, and references therein.
- 21. For an excellent review on the importance of developing asymmetric multi-component reactions see: Ramon, D. J.; Yus, M. Angew. Chem., Int. Ed. 2005, 44, 1602.
- 22. After completion of this manuscript an excellent paper by Vicario et al. (Vicario, J. L.; Reboredo, S.; Badia, D.; Carrillo L. Angew. Chem., Int. Ed. 2007, early view)

appeared on the web on 30th May describing an organocatalytic reaction between a preformed azomethine ylide derived from diethyl 2-aminomalonate and enals. Notably, catalyst **11** did not catalyze the reaction under their reaction conditions.

- (a) Marigo, M.; Wabnitz, T. C.; Fielenbach, D.; Jørgensen, K. A. Angew. Chem., Int. Ed. 2005, 44, 794; (b) Hayashi, Y.; Gotoh, H.; Hayashi, T.; Shoji, M. Angew. Chem., Int. Ed. 2005, 44, 4212.
- 24. Typical experimental procedure for the organocatalytic three-component synthesis of chiral pyrrolidines. To a stirred solution of aldehyde 1 (0.375 mmol, 1.5 equiv) in CHCl₃ or toluene, diethyl aminomalonate 2 (0.375 mmol, 1.5 equiv) was added. The reaction was stirred at room temperature for 1 h and then catalyst 11 (0.05 mmol, 20 mol %), TEA (0.25 mmol, 1.0 equiv), and α , β -unsaturated aldehyde 3 (0.25 mmol, 1.0 equiv) were added. The reaction was then stirred at room temperature for the time shown in Table 2. Next, the crude was purified by silica-gel column chromatography to afford the pyrrolidine derivative 4. Compound 4a: colorless oil. ¹H NMR (400 MHz, CDCl₃): $\delta = 9.03$ (d, J = 4.4 Hz, 1H), 7.35–7.20 (m, 5H), 5.13 (d, J = 9.2 Hz, 1H), 4.37-4.21 (m, 4H), 3.32-3.25 (m, J)1H), 3.07-3.00 (br s, 1H), 2.95-2.87 (m, 1H), 1.80-1.70 (m, 2H), 1.36–1.20 (m, 10 H), 0.92–0.84 (m, 3H); ¹³C NMR (100 MHz, CDCl₃): 202.2, 171.8, 169.6, 138.9, 128.6, (100 MH2, CDC13). 202.2, 1713, 100.6, 100.6, 100.7, 120.6, 120.7, 127.1, 75.1, 61.9, 61.7, 59.6, 44.0, 30.4, 30.2, 22.7, 14.2, 14.0, 13.8. $[\alpha]_{D}^{23} - 33.5$ (*c* 1.0, CHCl₃). HRMS (ESI): calcd for $[M+H]^+$ (C₂₁H₃₀NO₅) requires *m/z* 376.2118, found 376.2108. The enantiomeric excess was determined by HPLC with an AD column. (n-hexane-i- $PrOH = 98:2, \lambda = 250 \text{ nm}$), 1.0 mL/min; $t_R = major \text{ enan-}$ tiomer 15.7 min, minor enantiomer 10.7 min.
- 25. Typical experimental procedure for the one-pot organocatalytic three-component synthesis of acid functionalized pyrrolidine 14d: To a stirred solution of aldehyde 1a (0.375 mmol) in CHCl₃, diethyl aminomalonate 2a (0.375 mmol) was added. The reaction was stirred at room temperature for 1 h and then catalyst 11 (0.05 mmol, 20 mol%), TEA (0.25 mmol, 1.0 equiv), and cinnamic aldehyde 3b (0.25 mmol) were added. After stirring for 20 h at room temperature, the reaction temperature was decreased to 0 °C and then isobutene (0.1 mL), tertbutanol (0.4 mL), H₂O (0.2 mL), KH₂PO₄ (54.4 mg, 4 mmol), and NaClO₂ (36 mg, 4 mmol) were added sequentially. The reaction was allowed to reach room

temperature. After vigorous stirring overnight, the crude was purified by column chromatography on silica-gel (pentane/EtOAc mixtures) to afford the desired acid **14d**. Colorless oil. ¹H NMR (400 MHz, CDCl₃): $\delta = 7.40-7.05$ (m, 10H), 5.21 (d, J = 9.2Hz, 1H), 4.65 (d, J = 11.2 Hz, 1H), 4.32–4.20 (m, 2H), 3.94–3.85 (m, 2H), 3.58–3.49 (m, 1H), 1.25 (t, J = 7.2 Hz, 3H), 0.77 (t, J = 7.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): 174.2, 171.7, 169.4, 140.4, 136.5, 128.5, 128.2, 128.1, 127.7, 127.5, 127.2, 76.5, 62.4, 61.8, 58.8, 49.8, 14.0, 13.3. $[\alpha]_{D}^{23} + 28.3$ (*c* 1.0, CHCl₃). HRMS (ESI): calcd for $[M+H]^+$ (C₂₃H₂₆NO₆) requires m/z 412.1755, found 412.1740.

- 26. Experimental procedure for the synthesis of highly functionalized proline derivatives: Compound 14d (0.073 mmol) was dissolved in AcOH (0.15 mL) and HCl (6 N, 0.6 mL). The reaction was stirred at reflux for 16 h. Next, the crude mixture was diluted with water and washed with EtOAc. The aqueous layer was dried under reduced pressure to afford the proline derivative 15d in a 1:1 diastereomeric mixture in quantitative yield. Compound 15d (1:1 mixture of diastereoisomers): colorless solid. Mp 214–216 °C. ¹H NMR (400 MHz, D₂O): δ = 7.60–7.20 (m, 10H), 5.60–5.50 (m, 1H), 5.48–5.40 (m, 1H), 5.10–5.02 (m, 1H), 4.05–3.95 (m, 1H), 4.65–4.55 (m, 1H), 4.30–4.20 (m, 1H), 4.05–3.95 (m, 1H), 3.90–3.80 (m, 1H). HRMS (ESI): calcd for [M+H]⁺ (C₁₈H₁₈NO₄) requires *m/z* 312.1230, found 312.1239.
- 27. To a 5 mL round bottom flask containing compound 15d (0.1 mmol) in benzene (1 mL) was added MeOH (0.2 mL) and the reaction mixture was stirred for 5 min. Next, TMSCHN₂ (0.150 mL) was added dropwise. The reaction mixture was stirred for 90 min followed by the removal of the solvent under reduced pressure. The crude mixture was purified by silica-gel column chromatography (pentane/ EtOAc mixtures) to afford the pure diastereomers. Compound 16d: colorless solid. Mp 64-65 °C. ¹H NMR (400 MHz, CDCl₃): $\delta = 7.32-7.22$ (m, 10H), 4.87 (d, J = 8.8 Hz, 1H), 4.09 (d, J = 8.8 Hz, 1H), 3.90 (t, J =8.4 Hz, 1H), 3.72 (s, 3H), 3.54 (t, J = 8.4 Hz, 1H), 3.16 (s, 3H): ¹³C NMR (100 MHz, CDCl₃): 173.1, 172.0, 140.2, 139.3, 129.0, 128.5, 128.0, 127.8, 127.4, 127.2, 67.8, 65.5, 59.0, 52.5, 52.4, 51.5. $[\alpha]_{D}^{23}$ +19.2 (*c* 1.0, CHCl₃). HRMS (ESI): calcd for [M+Na]⁺ (C₂₀H₂₁NO₄Na) requires *m*/*z* 362.1363, found 362.1372. The data matched those for previously reported 16d see: Tsuge, O.; Kanemasa, S.; Yoshioka, M. J. Org. Chem. 1988, 53, 1384.