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## Synthesis of Novel 4,6-Diazaspiro[2.3]hex-1-en-5-ones by the Reaction of Diphenylcyclopropenone Oxime with Isocyanates <sup>#</sup>

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Diphenylcyclopropenone oxime hydrochloride (<u>3</u>) was prepared in an 83% yield from diphenylcyclopropenone and hydroxylamine hydrochloride in methanol. The salt <u>3</u> reacted with alkyl and aryl isocyanates in the presence of triethylamine to yield 1:2 addition products diazaspiro[2.3] hexenones in good yields.

Physical and chemical properties of cyclopropenones, cyclopropenium ions, and triafulvenes have been highly interested and currently investigated<sup>1)</sup> as microcyclic aromatics. In the continuation of our studies on the chemistry of cyclopropenium salts posessing heteroatom substituents we have found an easy route for the preparation of diphenylcyclopropenone oxime  $(\underline{4})$ ,<sup>2)</sup> the chemical nature of which has not been explored. Below we describe the preparation and reaction of  $\underline{4}$  with isocyanates to yield novel 4,6-diazaspiro[2.3]hexenes.

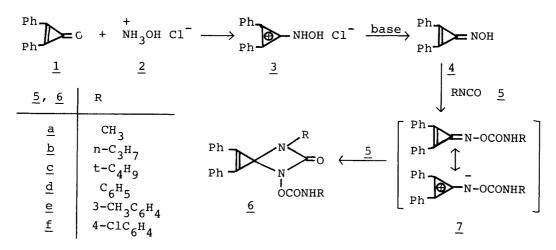
A solution of diphenylcyclopropenone (<u>1</u>)[20 mmol] and hydroxylamine hydrochloride (<u>2</u>)[60 mmol] in methanol(25 cm<sup>3</sup>) was allowed standing at room temperature for a day. The precipitated salt 1-hydroxyamino-2,3-diphenylcyclopropenium chloride (<u>3</u>) was obtained in an 83% yield by filtration. The salt <u>3</u> gave satisfactry spectroscopic data<sup>3</sup>) and yielded yellow needles of free <u>4</u><sup>4</sup>) on treating with aqueous sodium hydrogencarbonate or triethylamine in benzene.

An equimolar reaction of <u>3</u> with methyl isocyanate (<u>5a</u>) in the presence of triehtylamine gave 4-methyl-6-methylcarbamoyloxy-1,2-diphenyl-4,6-diazaspiro[2.3]-hex-1-en-5-one (<u>6a</u>) in a 40% yield. The use of twice excess moles of <u>5a</u> at room temperature for a day gave <u>6a</u> in a 70% yield. The structure of <u>6a</u> was assigned from its <sup>1</sup>H- and <sup>13</sup>C-NMR, and mass spectroscopic studies.<sup>5</sup>) Similar treatment of <u>3</u> with two moles of isocyanates such as propyl, t-butyl, phenyl, 3-methylphenyl, and 4-chlorophenyl isocyanates (<u>5b-f</u>) yielded in 63, 69, 71, 67, and 48% yield respectively. No 1:1 addition product was isolated in our hands. To our knowledge no spiro derivatives like these have been reported, though it has been shown that some C=N derivatives react with isocyanates to afford the 1:1 cycloaddition products, 1,3-diazetidines.<sup>6</sup>)

It is well known that aldoximes and ketoximes react with isocyanates to afford carbamates.<sup>6b)</sup> Although the isolation of (carbamoyloxy)iminocyclopropene

<sup>#</sup>This paper is dedicated to the late Professor Ryozo Goto, Kyoto University.

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 $(\underline{7})$ , a 1:1 addition product, failed, the intermediacy of  $\underline{7}$  was clear from the final product <u>6</u>. The reaction of highly polarized  $\underline{7}$  with <u>5</u> would be faster than that of <u>4</u> with <u>5</u>.

References

- Recent examples:R. A. Moss, S. Shen, K. Krogh-Jespessen, J. A. Potenza, H. J. Schugar, and R. C. Munjal, J. Am. Chem. Soc., <u>108</u>, 134(1986). F. J. Kaiser, G. Offermann, and G. Seitz, Chem. Ber., <u>119</u>, 2141(1986). T. Sugimoto, M. Shibata, S. Yoneda, Z. Yoshida, Y. Kai, K. Miki, N. Kasai, and T. Kobayashi, J. Am. Chem. Soc., <u>108</u>, 7032(1986). K. Takahashi, K. Ohnishi, and K. Takase, Chem. Lett., <u>1985</u>, 1447.
- 2) R. Breslow, T. Eicher, A. Krebs, R. A. Peterson, and J. Posner[J. Am. Chem. Soc., <u>87</u>, 1320(1965)] have reported the reaction of <u>1</u> with <u>2</u> in aqueous ethanol to yield deoxybenzoin oxime and diphenylisoxazolone. Y. Kitahara and M. Funamizu[Bull. Chem. Soc. Jpn., <u>37</u>, 1897(1964)] have obtained <u>4</u> as colorless needles from <u>1</u> and <u>2</u> in methanol by treating with aqueous NaHCO<sub>3</sub>. T. Eicher and G. Frenzel[Z. Naturforsch., B, <u>20</u>, 274(1965)] have reported <u>3</u> (BF<sub>4</sub> salt) from the ethoxy derivative of <u>1</u> and <u>2</u>, however the chemical nature of <u>3</u> and <u>4</u> have never been explored.
- 3) 3:mp 198-202 °C; IR(KBr) 2940, 2740, and 1920 cm<sup>-1</sup>; <sup>13</sup>C-NMR(CDCl<sub>3</sub>+CF<sub>3</sub>CO<sub>2</sub>H) S = 119.1(s), 119.2(s), 129.9(d), 130.3(d), 133.6(d), 133.8(d), 136.2(d), and 139.5(s).
- 4) 4:mp 133-136 °C; IR(KBr) 3150, 1880, and 1850 cm<sup>-1</sup>; MS(m/z) 221(M<sup>+</sup>).
- 5) <u>6a</u>:mp 135-137 <sup>O</sup>C; IR(KBr) 3450, 1770, and 1710 cm<sup>-1</sup>; <sup>1</sup>H-NMR(CDCl<sub>3</sub>)  $\boldsymbol{\$}$  = 2.59(s, 3H, CH<sub>3</sub>N), 2.85(d, J=6 Hz, 3H, NH<u>CH<sub>3</sub></u>), 6.13(q, J=6 Hz, 1H, NaOD/D<sub>2</sub>O exchange, NH), and 7.2-7.9(m, 10H, 2Ph); <sup>13</sup>C-NMR(CDCl<sub>3</sub>)  $\boldsymbol{\$}$  = 26.5(q), 26.6(q), 65.7(s), 118.4(s), 125.9(s), 128.9(d), 129.9(d), 130.4(d), 155.4(s), and 160.0(s); MS(m/z) 335(M<sup>+</sup>).
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