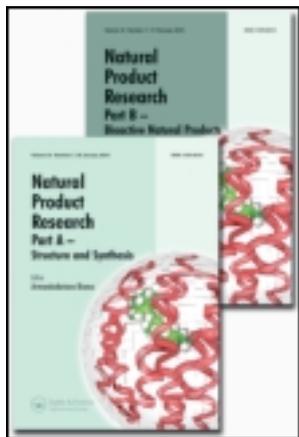


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A new triterpenoid saponin from *Glinus oppositifolius* with α -glucosidase inhibitory activity

Deepak Kumar^a, Vrunda Shah^a, Rina Ghosh^b and Bikas C. Pal^{a*}

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Evaluation of α -glucosidase inhibitory activity led to the isolation of six triterpene saponins from the aerial parts of *Glinus oppositifolius* including one new and five known constituents. The structure of the new saponin, glinoside C (**1**), was established as 16-*O*-(β -D-glucopyranosyl)-3 β ,12 β ,16 β ,21 α ,22-pentahydroxy hopane by extensive use of 1-D, 2-D NMR and mass spectral techniques. The other constituents identified were 3-*O*-(β -D-xylopyranosyl)-spergulagenin A (**2**), spergulacin (**3**), spergulin A (**4**), spergulacin A (**5**) and spergulin B (**6**). Compound **1** exhibited the greatest inhibition of the enzyme with IC₅₀ of 127 \pm 30 μ M. Kinetics study for the compound **1** demonstrated mixed type of inhibition (K_i = 157.9 μ M).

Keywords: *Glinus oppositifolius*; Molluginaceae; glinoside-C; α -glucosidase; hopanoid triterpene saponin

1. Introduction

α -Glucosidase is an intestinal enzyme that catalyses the final step in the process of carbohydrate hydrolysis. Inhibition of this enzyme plays an important role in the management of post-prandial hyperglycaemia, as it delays the digestion of carbohydrates. There are several reports describing the α -glucosidase inhibitory potential of various constituents (Feng, Yang, & Wang, 2011; Li et al., 2008; Wafo et al., 2011) including triterpene saponins (Li et al., 2009; Luo, Ma, & Kong, 2008; Mbaze et al., 2007) isolated from plants. With the aim to uncover new leads with α -glucosidase inhibitory potential, the less explored plant *Glinus oppositifolius* (L.) Aug. DC. (Syn: *Mollugo spergula* L., Family: Molluginaceae) has been investigated in this study. The plant, commonly known as Gima, is an edible plant traditionally used for the treatment of skin diseases (Anonymus, 1962) and is a good source of triterpenoid saponins (Barua et al., 1980; Sahu, Koike, Banerjee, Achari, & Nikaido, 2001; Traore et al., 2000). This article describes the isolation of six saponins, out of which one is a new one, and another one is being reported for the first time from this plant. All the constituents were evaluated for their α -glucosidase inhibitory activity, and the new constituent was observed to exert the greatest activity.

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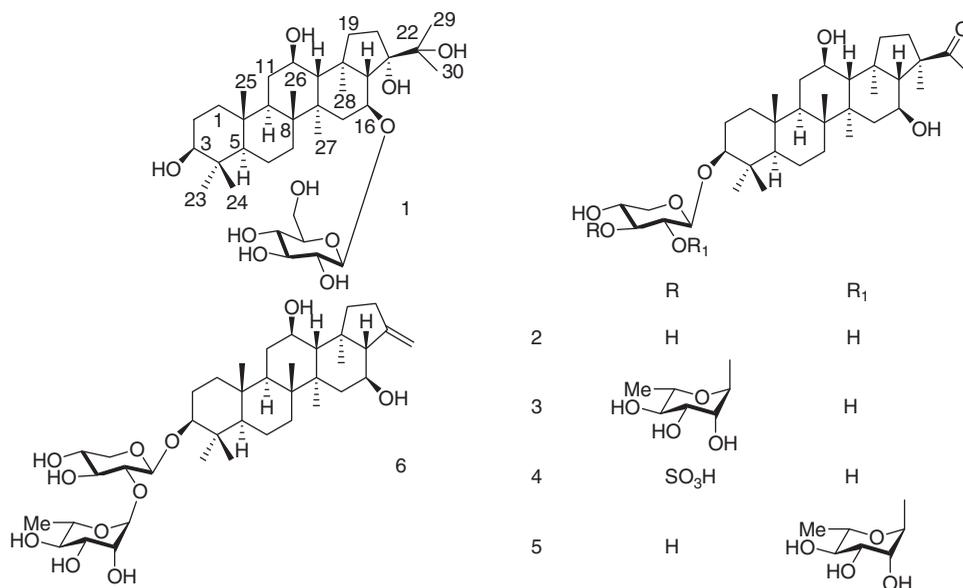


Figure 1. Structure of compounds 1–6.

2. Results and discussion

Compound **1** (Figure 1) was obtained as an amorphous solid. The molecular formula was assigned as $C_{36}H_{62}O_{10}$ on the basis of its HR-FAB-MS ion at m/z 677.4242 [$M + Na$]⁺ (Calcd for $C_{36}H_{62}O_{10}Na$, 677.4241). Infrared (IR) absorption bands (3389, 1067 cm^{-1}) were suggestive of a glycoside (Mimaki, Doi, Kuroda, & Yokosuka, 2007). The compound displayed 36 carbon signals in its ^{13}C -NMR spectrum, of which 6 could be assigned to a hexose unit and 30 attributable to the aglycone part. The multiplicity of the individual resonances, determined using DEPT, revealed the presence of 8 methyl, 9 methylene, 12 methine and 7 aliphatic quaternary resonances. In support, the 1H -NMR spectrum showed the presence of eight tertiary methyl signals at δ_H 0.71, 0.84, 1.00, 1.09, 1.21, 1.32, 1.52 and 1.56 (each 3H, s). Acid hydrolysis of the compound **1** with 5% HCl-dioxan gave glucose as sugar component, which was identified by TLC, and the absolute configuration was determined by the measurement of optical rotation (Hamed et al., 1996). The 1H - and ^{13}C -NMR spectra revealed that the aglycone was a triterpenoid with a hopane skeleton (Sahu et al., 2001; Traore et al., 2000) and contained five hydroxyl groups (three secondary ones at C-3, C-12, C-16 and two tertiary ones at C-21 and C-22). The presence of eight-OH groups was indicated by HMBC correlated 1H - and ^{13}C -NMR signals at δ_H/δ_C 7.33/76.0, 7.31/66.9, 7.28/75.8, 7.19/79.0, 6.29/72.7, 5.88/63.8, 5.80/78.4 and 4.96/86.1. The 1H -NMR spectrum also showed a signal at δ_H 5.13 (d, $J=7.8$ Hz), and the corresponding carbon peak appeared at δ_C 102.0, assigned to the anomeric CH. The exact linkage of the sugar was determined by the combined use of 1H - 1H correlation spectroscopy (COSY), NOESY, HSQC and HMBC (Figure 2), where there were cross peaks between H-1' (δ_H 5.13) and C-16 (δ_C 75.4). Chemical shift of the anomeric proton and its coupling constant suggested that the sugar linkage at C-16 was a β -D-glucopyranoside. The chemical shift of C-17 appears somewhat higher but may be explained by the contribution of C-21-OH substitution as reported for

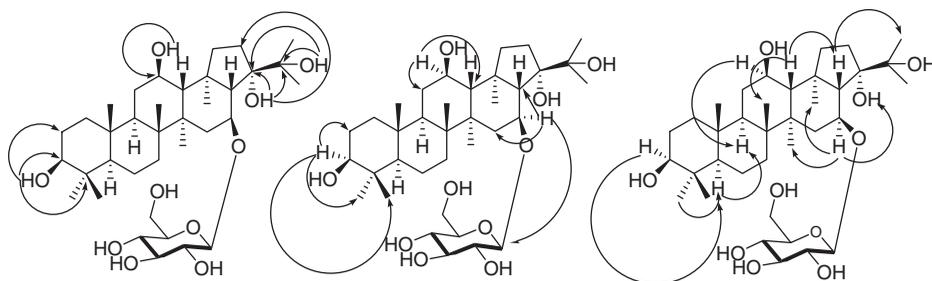


Figure 2. Key HMBC hydroxyl proton–carbon correlations, HMBC proton–carbon correlation and ^1H – ^1H NOESY correlations of **1**.

Table 1. α -Glucosidase inhibitory activity of *G. oppositifolius* extract, fractions and isolated saponins.

Sample	Inhibitory concentration (IC_{50}) ^a
Methanolic extract	$2.548 \pm 0.106 \text{ mgmL}^{-1}$
Ethyl acetate fraction	$2.138 \pm 0.117 \text{ mgmL}^{-1}$
n-Butanol fraction	$0.858 \pm 0.074 \text{ mgmL}^{-1}$
Aqueous fraction	$> 3 \text{ mgmL}^{-1}$
Glioside C (1)	$127 \pm 30 \mu\text{M}$
3- <i>O</i> -(β -D-xylopyranosyl)-spergulagenin A (2)	$1654 \pm 170 \mu\text{M}$
Spergulacin (3)	$628 \pm 80 \mu\text{M}$
Spergulin A (4)	$143 \pm 20 \mu\text{M}$
Spergulacin A (5)	$694 \pm 60 \mu\text{M}$
Spergulin B (6)	$1783 \pm 290 \mu\text{M}$
Acarbose	$15 \pm 3 \mu\text{M}$

Note: ^aValues are expressed as mean \pm SEM ($n = 3$).

glisoside A (Traore et al., 2000). The stereochemistry of **1** was determined by NOESY experiment. From the above data, we concluded that compound **1** was a glucoside with a new aglycone, 3β , 12β , 16β , 21α , 22 -pentahydroxy hopane. The known compounds (Figure 1) were identified as 3-*O*-(β -D-xylopyranosyl)-spergulagenin A (**2**), spergulacin (**3**), spergulin A (**4**), spergulacin A (**5**) and spergulin B (**6**) by direct comparison of their spectral data with those in the literature (Sahu et al., 2001). Compound **2** is reported for the first time from this plant.

The *n*-BuOH fraction showed maximum inhibition of the enzyme *in vitro* among the three fractions. The isolated saponins were tested for their potential to inhibit α -glucosidase, *in vitro*, and the result showed varying degree of inhibition. The α -glucosidase exhibited Michaelis–Menten kinetics with *p*-nitrophenyl α -D-glucopyranoside at 37°C (K_m : 1.513 mM ; V_{max} : $0.910 \text{ mM min}^{-1}$). Compound **1** affected the greatest inhibition of enzyme (Table 1) and demonstrated a mixed type of enzyme inhibition ($K_i = 157.9 \mu\text{M}$; Figure 3). The inhibition demonstrated by these constituents was less active compared to acarbose, a standard drug, hence these constituents may be classified as moderate inhibitors of the enzyme. On the basis of these results, the edible plant *G. oppositifolius* may be used for the management of diabetic complications as food additive.

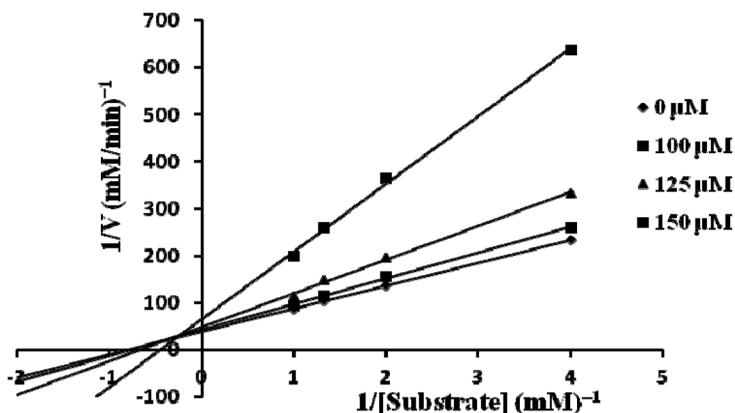


Figure 3. Kinetic analysis of α -glucosidase inhibition by glinoside C. Lineweaver–Burke plot of hydrolysis of *p*-nitrophenyl α -D-glucopyranoside (0.25, 0.5, 0.75 and 1 mM) by α -glucosidase (0.5 U mL⁻¹) in the presence or absence of glinoside C (1).

3. Experimental

3.1. General experimental procedures

Optical rotation was recorded in MeOH using a Jasco P-1020 polarimeter. IR spectra were recorded with a Jasco-FT-IR-model 410 spectrophotometer as KBr disks. ¹H-, ¹³C- and 2D NMR were recorded on a Bruker Ultrashield NMR (600 MHz) in pyridine-d₅ with TMS as internal standard. HR-FAB-MS was recorded on a MStation JMS 700 (JEOL Ltd, Japan). Silica gel (Merck, 100–200 mesh) was used for column chromatography; silica gel (60 F254) was used for TLC and spots were visualised by spraying with Liebermann–Burchard reagent followed by heating.

3.2. Chemicals and reagents

α -Glucosidase (Maltase, ex microorganism, EC 3.2.1.20) and *p*-nitrophenyl α -D-glucopyranoside were purchased from SRL Mumbai, India. Acarbose was purchased from Sigma. Na₂HPO₄, NaH₂PO₄, DMSO, Na₂CO₃, other chemicals and solvents were of highest purity grade and purchased from Merck, India.

3.3. Plant material

The aerial part of the plant was collected in Kolkata (India) in August 2010 and identified by the Botanical Gardens, Howrah, India. A voucher specimen (NIP-K/BCP/004) has been deposited at National Institute of Pharmaceutical Education and Research, Kolkata.

3.4. Extraction and isolation

The air-dried plant (1 kg) was subjected to extraction using MeOH (3.5 L × 5) at room temperature for 48 h. The extract was filtered, and the solvent was evaporated under reduced pressure using a rotary evaporator at 45°C. The resultant extract was lyophilised to afford the crude MeOH extract (138 g; yield 13.8% w/w). A part (100 g) of the extract was then suspended in Milli-Q H₂O and partitioned successively with EtOAc and *n*-BuOH (Kumar, Mallick, Vedasiromoni, & Pal, 2010). Each fraction was evaporated under

vacuum to yield the EtOAc fraction (25 g), *n*-BuOH fraction (39 g) and aqueous fraction (35 g). All the fractions were stored at 4°C till further use.

The *n*-BuOH fraction (20 g) was chromatographed on silica gel (200 g) using a step gradient, petroleum ether–CHCl₃ 1:1 (500 mL, fr-1), 1:9 (500 mL, fr-2), CHCl₃ (500 mL, fr-3), CHCl₃–MeOH 9:1 (500 mL, fr-4), 8:2 (500 mL, fr-5) and 7:3 (500 mL, fr-6), to get a total of six fractions. Repeated silica gel column chromatography of fr-4 led to the isolation of compounds **2** (40 mg) and **3** (200 mg). Fr-5 was further chromatographed over a silica gel column and eluted with increasing polarity of CHCl₃–MeOH, 1:0 to 7:3 (v/v), to give a total 30 fractions. Among these, fractions 15–17 were mixed together because HPLC analysis (Discovery[®] RP amide C16, 5 μm, 25 cm × 4.6 mm, eluted with MeOH–H₂O 60:40, at 1 mLmin⁻¹, ELS-detector) suggested them to have similar pattern with four major peaks (*t*_R = 9.6, 16.2, 19.9, 29.8 min). The mixed fraction was purified using Prep-HPLC (Discovery[®] RP amide C16, 5 μm, 25 cm × 10 mm, MeOH–H₂O 60:40 at 2.5 mLmin⁻¹, RI-detector). All the four peaks were collected and evaporated under reduced pressure using rotary evaporator at 45°C to afford compounds **4** (48 mg), **1** (30 mg), **5** (67 mg) and **6** (40 mg).

3.4.1. *Glinoside C (1)*

Amorphous solid; $[\alpha]_D^{28} = +17.2^\circ$ (*c* 0.02, MeOH); IR (KBr cm⁻¹) ν_{\max} : 3389, 2945, 1654, 1563, 1455, 1384, 1067, 644; HR–FAB–MS *m/z*: 677.4242 [M + Na]⁺, Calcd for C₃₆H₆₂O₁₀Na, 677.4241. ¹H–NMR (C₅D₅N; 600 MHz) aglycone moiety, δ : 1.49 (1H, m, H-1a), 0.79 (1H, m, H-1b), 1.78 (1H, m, H-2a), 1.24 (1H, m, H-2b), 3.43 (1H, m, H-3), 0.73 (1H, d, *J* = 12.0 Hz, H-5), 1.49 (1H, m, H-6a), 1.29 (1H, m, H-6b), 1.17 (1H, m, H-7a), 1.41 (1H, m, H-7b), 1.24 (1H, m, H-9), 2.30 (1H, m, H-11a), 1.45 (1H, m, H-11b), 4.42 (1H, m, H-12), 1.86 (1H, d, *J* = 7.8 Hz, H-13), 1.67 (1H, m, H-15a), 1.89 (1H, m, H-15b), 4.44 (1H, m, H-16), 2.32 (1H, d, *J* = 9.0 Hz, H-17), 1.91 (1H, m, H-19a), 2.69 (1H, m, H-19b), 1.82 (1H, m, H-20a), 1.91 (1H, m, H-20b), 1.21, 1.00, 0.71, 0.84, 1.09, 1.32, 1.52, 1.56 (each 3H, s, H₃–23, –24, –25, –26, –27, –28, –29, –30), sugar moiety, 5.13 (1H, d, *J* = 7.8 Hz, H-1'), 4.08 (1H, m, H-2'), 4.35 (1H, m, H-3'), 4.25 (1H, m, H-4'), 4.05 (1H, m, H-5'), 4.40 (1H, m, H-6'a), 4.62 (1H, m, H-6'b), hydroxyl proton, 5.80 (1H, d, *J* = 4.2 Hz, 3-OH), 7.31 (1H, s, 12-OH), 4.96 (1H, s, 21-OH), 7.28 (1H, s, 22-OH), 7.33 (1H, s, 2'-OH), 7.19 (1H, s, 3'-OH), 6.29 (1H, s, 4'-OH), 5.88 (1H, br s, 6'-OH). ¹³C–NMR (C₅D₅N; 600 MHz) aglycone moiety, δ : 39.1 (C-1), 28.6 (C-2), 78.4 (C-3), 39.8 (C-4), 56.0 (C-5), 19.1 (C-6), 33.8 (C-7), 42.0 (C-8), 48.9 (C-9), 37.6 (C-10), 27.9 (C-11), 66.9 (C-12), 54.6 (C-13), 45.5 (C-14), 46.0 (C-15), 75.4 (C-16), 74.1 (C-17), 47.9 (C-18), 42.8 (C-19), 38.1 (C-20), 86.1 (C-21), 75.8 (C-22), 29.0 (C-23), 16.6 (C-24), 16.2 (C-25), 17.2 (C-26), 20.0 (C-27), 17.2 (C-28), 26.9 (C-29), 27.6 (C-30), sugar moiety, 102.0 (C-1'), 76.0 (C-2'), 79.0 (C-3'), 72.7 (C-4'), 78.4 (C-5'), 63.8 (C-6').

3.5. *Acid hydrolysis of 1*

A part (15 mg) of compound **1** was heated with 2 (N) HCl in Dioxan–H₂O (1:1 (v/v), 10 mL) for 2 h on H₂O bath (80°C). Dioxan was distilled off under reduced pressure and the product after dilution with H₂O was extracted with EtOAc (15 mL × 2). Evaporation furnished a product (5 mg), which showed three spots in TLC perhaps due to artefact formation and was not proceed with further. The aqueous layer was neutralised with Ag₂CO₃, then filtered and evaporated. The sugar was identified as D-glucose with authentic sample by TLC in EtOAc–MeOH–H₂O–AcOH (13:4:4:3; *R*_f 0.35). Sugar was isolated by preparative TLC with same solvent system and the optical rotation of sugar was measured. $[\alpha]_D^{28} + 103.8$ (*c* 0.10, H₂O) (Hamed et al., 1996).

3.6. α -Glucosidase enzyme inhibition assay

The α -glucosidase enzyme inhibitory assay was performed in a 96 well plate as described earlier by Hou et al. (2009), with some modifications. For the analysis of the type of inhibition, kinetics of enzyme inhibition by glicoside C (**1**) was studied by double reciprocal plot at four different substrate concentrations –0.25, 0.5, 0.75 and 1 mM.

3.7. Statistical analysis

Statistical calculations were carried out with GraphPad Prism (GraphPad Software, Inc., USA). Results are expressed as the mean \pm SEM.

4. Conclusion

α -glucosidase inhibitory activity guided fractionation and purification led to isolation of six triterpene saponins from the aerial part of the *G. oppositifolius*, including one new saponin, glicoside C (**1**). This exhibited the greatest inhibition of the enzyme (IC_{50} $127 + 30 \mu M$) and its kinetics study showed mixed type of inhibition.

Supplementary material

Experimental details relating to this article are available online, alongside Table S1 and Figures S1–S22.

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