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Diastereoselective Cyclisation of N-Alkenylideneamines into 3,4-Dihydro-2*H*pyrrol-1-ium Halides

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Dedicated to Prof. Dr. Manfred Christl on the occasion of his 65th birthday

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A number of new chiral $2-(\alpha$ -bromoalkyl)pyrrolinium salts and $2-(\alpha$ -selenoalkyl)pyrrolidines were synthesized by the halocyclisation and selenocyclisation, respectively, of *N*-(alkenylidene)alkylamines and subsequent reduction. These cyclisations were implemented in a diastereomeric fashion for the first time. Substrate control (starting imines possessing chirality in the *N*-alkyl or the *N*-alkenyl substituent) and re-

Introduction

The asymmetric functionalisation of olefins is an important goal in synthetic organic chemistry. Thus, highly stereoselective epoxidations, dihydroxylations and hydroborations were developed. The addition of electrophiles to C=C bonds can also be combined with an intramolecular nucleophilic attack by ring formation, when alkenes with additional nucleophilic sites (O, S or N) are used as substrates. This electrophile-induced cyclisation is a versatile strategy for the construction of heterocycles. The halolactonisation, where unsaturated carboxylic acids are cyclised by a reaction with halogens to form lactones, is a wellknown example of this type of reaction. Other common starting materials are unsaturated carboxylic acid derivatives, imines (Scheme 1), alcohols and amines. Halogens, Nhalogenated succinimides, mercury(II) acetate, tert-butyl hydroperoxide^[1] (Sharpless conditions) and organoselenenyl halides^[2–5] commonly serve as electrophiles.^[6]

The general mechanism of the electrophile-induced cyclisation of unsaturated imines is shown in Scheme 1. Chemo-, regio- and diastereoselectivity are important aspects of such ring formations. Thus, the reaction of the electrophile can give just an addition to the double bond (e.g. to adduct 5) rather than undergoing a cyclisation to products 6 or 7. On the other hand, in accordance with the Baldwin rules^[7] the cyclisation can follow either the *exo-trig* or

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agent control (chiral organoselenenyl bromides) were

applied. Asymmetric induction by stereocentres of the alken-

ylidene or double asymmetric induction by chiral selenenyl

bromides on unsaturated imines with a chiral *N*-alkyl group

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resulted in diatereoselectivities up to 84:16.

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Scheme 1. General mechanism of the electrophile-induced cyclisation of unsaturated imines.

the *endo-trig* mode, leading to ring **3** or **4**, respectively. Finally, *cis-trans* isomers can be formed when nucleophiles are added to the primary cyclisation products affording cyclic amine **6** or **7**. A number of studies were devoted to the influence of different electrophiles and reaction conditions on the mode of the cyclisation of alcohols,^[2–4,6] thio-



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ethers,^[8–12] protected amines^[6,13–22] and hydrazines,^[23] carbonyl derivatives such as unsaturated oximes,^[24–31] hydrazones,^[30,32] imines,^[33–39] acetals,^[40–42] hemiaminals^[43,44] and aminals^[45] and carboxyl derivatives, such as amides,^[46,47] thioamides^[48,49] and carboxylic acids.^[2–4,6] A comprehensive overview of the whole spectrum of this type of reaction is given in the reviews by Cardillo et al.,^[6] Wessjohann et al.,^[2] Wirth^[3] and Petragnani et al.,^[4]

Remarkably, there have been no systematic, stereochemical studies on the ring-closing reaction of chiral N-(alkenylidene)alkylamines (unsaturated imines) **1**, even though the resulting product **3** or **4** is a suitable candidate for further functionalisation by reaction with nucleophiles at the iminium moiety, giving rise to the formation of interesting heterocycle (pyrrolidine or piperidine) **6** or **7**.

De Kimpe et al.^[33,34,36] investigated the reaction of achiral N-(alkenylidene)alkylamines with different electrophiles, focusing on its regio- and stereochemistry and the further derivatization of the resulting iminium salts 3 with nucleophiles, such as LiAlH₄,^[34] alkoxides^[36,50] and cyanides.^[51] The formation of five-membered, cyclic, iminium salts (pyrrolinium salts) 3 was observed in most cases. However, these pyrrolinium salts 3 can ring expand to six-membered ring products 7 in a subsequent reaction of nucleophiles with the C=N bond. An example of the cyclisation of racemic N-(alkenylidene)alkylamines^[33] was published, resulting in a diastereomeric ratio (dr) of 1:1. The cyclisation of unsaturated optically active oximes^[24,28,30,31] leads to cyclic nitrones with dr values from 50:50 to 86:14. A 5:1 diastereomeric mixture was obtained by the iodocyclisation of a racemic oxime.^[29] Optically active N-anilino-2-pyrrolidones were formed by the iodocyclisation of pentose-derived phenylhydrazones in a *dr* of 70:30.^[30]

So far, optically active products have not been obtained by the electrophile-induced cyclisation of *N*-(alkenylidene)alkylamines. Herein, we report three possible approaches to the stereoselective halo- or selenocyclisation of unsaturated imines. Following a substrate control, imines were used with chiral centres in the amine part of the unsaturated imine or in the alkene moiety. Reagent control was attempted by chiral selenyl bromides.

Results

Cyclisation of Chiral *N*-(Alkenylidene)alkylamines with Achiral Reagents

N-(Alkenylidene)alkylamines of the general type 1 are easily accessible by the reaction of an appropriate amine and an aldehyde. 2,2-Dimethylpent-4-enal **8** was chosen as a suitable aldehyde because unwanted enamine tautomerism is excluded, which probably would affect the stereoinduction in the cyclisation of the corresponding imines. Compound **8** was converted into the N-(alkenylidene)alkylamines **9** (Scheme 2, Table 1) with various chiral amines as well as achiral or racemic amines at room temperature in dichloromethane in the presence of magnesium sulfate (see the Supporting Information).



Scheme 2. Cyclisation of optically active N-(alkenylidene)alkylamines (chiral \mathbb{R}^1) with achiral electrophiles.

The electrophile-induced cyclisation of chiral and achiral imines **9** by bromine was carried out in dichloromethane at 0 °C (Table 1, entries 1, 3–7). In all cases, 2-bromomethyl-3,4-dihydro-2*H*-pyrrol-1-ium salts **10** were obtained in a 5-*exo-trig* mode. These compounds were isolated in high yields and fully characterized by NMR spectroscopy, showing typical ¹H and ¹³C chemical shifts for the characteristic functional groups in the products: 8.7–10.5 ppm/186–191 ppm (*CH*=N⁺), 3.3–4.4 ppm/32–34 ppm (*CH*₂Br) and 5.3–6.0 ppm/67–74 ppm (BrCH₂–*CH*–N⁺). Furthermore, the X-ray analysis of the tribromide of the 5*S* stereoisomer of **10e** (Figure 1) confirmed the proposed structure. The corresponding six-membered 6-*endo-trig* product was not observed.



Figure 1. X-ray crystal analysis of the tribromide of the 5S stereoisomer of **10e**.

The determination of dr values by ¹H NMR spectroscopy of the crude products revealed an unsatisfactory stereoselectivity (dr = 50:50-42:58) under these conditions. In order to overcome this problem, the reaction temperature was lowered in the cyclisation of compound **9f** from 0 °C to -60 °C or -78 °C (Table 1, entry 7). A modest increase in diastereoselectivity from 45:55 (0 °C) to 35:65 (-75 °C) was observed. Attempts to improve the diastereomeric ratio by the addition of Lewis acids to the reaction mixture were not successful. The addition of ZnBr₂ or LiBr did not cause a significant improvement, whereas strong Lewis acids such as BF₃·Et₂O or Et₂AlCl prevented the cyclisation (Table 1, entry 6).

Using less reactive iodine in dichloromethane at 0 °C for the cyclisation of the imine **9f** and the achiral *tert*-butylsubstituted **9k** led to mixtures of constitutionally isomeric pyrrolinium salts **11** and tetrahydropyridinium salts **13** as a result of competition between 5-*exo-trig* and 6-*endo-trig* cyclisation. The latter might occur as a result of thermodynamic control, achieved by iodine, which is less reactive than bromine. Imine **9f** gave rise to a constitutional isomer ratio of 86:14 (dr = 63:37 and 69:31, respectively) in favour of the five-membered ring **11a** (Table 1, entry 9), whereas in the case of the *N-tert*-butyl substrate **9k**, the pyridinium iodide **13b** was formed as the major isomer of a 19:81 mixture (Table 1, entry 15). Obviously, the bulkiness of the *N*-

Entry	N-R ¹	Imine 9	х	Hal	Iminium salts 10–13	Yield, $dr^{[a]}$ (0 °C)	Pyrrolidines 14	Yield, <i>dr</i> [a] (-78 °C)
1 2	N	9a	Br PhSe	Br Br	10a 12a	96 %[b]	14a	59 %[b]
3	N COOCH ₃	9b	Br	Br	10b	quant., 44:56		
4	N COOCH ₃	9c	Br	Br	10c	quant., 46:54		
5	N Si N	9d	Br	Br	10d	96 %, 50:50		
6	N COCH3	9e	Br	Br	10e	93 %, 42:58[c]		
7 8 9	N	9f	Br PhSe I	Br Br I	10f 12b 11a +13a	quant., 45:55[d] quant.[f]	14b	62 %, 47:53[e]
10	N	9g [a]	PhSe	Br	12c		14c	81 %, 48:52
11	N N N	9h	PhSe	Br	12d		14d	45 %, 41:59
12	$ \begin{array}{c} & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & $	9i	PhSe	Br	12e		14e	75 %, 33:67[e]
13	COOCH ₃	9j	PhSe	Br	12f		14f	59 %, 35:65[e,g]
14			Se	Br	12g		14g	61 %, 39:61[d]
15	tBu	9k	Ĩ	I	11b + 13b	67 %[h]		

Table 1. Synthesis of iminium salts 10-13 and pyrrolidines 14.

[a] Determined by ¹H NMR spectroscopy (\pm 5%). [b] Racemate. [c] Lewis acids were added, and the corresponding values obtained are given in parentheses: ZnBr₂ (dr = 42:58), LiBr (dr = 42:58), BF₃·Et₂O (no reaction), Et₂AlCl (no reaction). [d] dr = 40:60 at -60 °C, dr = 35:65 at -78 °C, dr = 36:64 at -90 °C. [e] Separation of diastereomers by column chromatography. [f] A mixture of constitutional isomers was obtained: pyrrolidinium iodide **11a**, 86%, dr = 63:37; pyridinium iodide **13a**, 14%, dr = 69:31. [g] dr = 40:60 at -50 °C in CHCl₃. [h] A mixture of constitutional isomers was obtained: pyrrolidinium iodide **13b** = 19:81.

alkyl substituent governs the regioselectivity in these cases. The products of these iodocyclisations could not be separated but were unambiguously assigned by ¹³C NMR spectroscopy (e.g. CH₂I (9.7 ppm) and CH-I (12.7 ppm) for 11b and 13b, respectively).

Cyclisations of optically active, racemic (9g) and achiral (9a, 9k) imines 9 with arylselenyl bromide as the electrophile were carried out at -78 °C in dichloromethane (Table 1). In all cases, the 5-membered-ring iminium salts 12 (X = SePh) formed by the cyclisation were subsequently reduced by sodium borohydride. The resulting pyrrolidines 14 were stable enough for isolation and purification by column chromatography, but the yields were generally lower than those obtained in halocyclisations. This might be an indication that the reduction step from 12 (X = RSe) to pyrrolidines 14 is a low-yielding reaction. NMR spectroscopic studies [2.8-3.2 ppm /33-36 ppm (CH₂-Se), 2.9-3.2 ppm /59-63 ppm (SeCH₂-CH), 1.4-1.9 ppm/46-47 ppm (CH-CH₂-C_q) and 2.0-2.8 ppm/60-68 ppm (d, N-CH₂- C_{q}] and X-ray analysis of the 5S stereoisomer (S)-14f of the pyrrolidine 14 (Figure 2) verified the assigned structure. We determined the *dr* values of these products as ranging from 48:52 to 33:67, which were slightly higher than those obtained in the bromocyclisation. It turned out that N-(alkenylidene)alkylamines with an additional heteroatom in the N-alkyl substituent gave a somewhat higher induction in selenocyclisations than pure hydrocarbon substituents (Table 1).



Figure 2. X-ray crystal analysis of the 5S stereoisomer of the pyrrolidine 14f (major isomer).

Table 2. Syntheses of 14 and 17 and comparison of diastereomeric ratios.

Sterically hindered organoselenenyl halides were successfully used by Lipshutz et al.^[52] to improve the diastereoselectivity of the electrophile-induced cyclisation of alcohols. Interestingly, the application of 2,4,6-(triisopropylphenyl)selenenyl bromide in the cyclisation of the imine 9 caused the opposite effect; in other words, the diastereoselectivity was slightly lower than that observed with phenylselenyl bromide (Table 1, entries 13 and 14).

We further investigated the effect of introducing substituents into the alkylidene moiety of chiral imines 9 or of chain homologation. Thus, N(alkenylidene)alkylamines 15 were cyclised with phenylselenyl bromide, and in the case of 15c, also with bromine (Scheme 3). Again, the selenocyclisation products 16 (X = SePh) were reduced to the corresponding pyrrolidines 17 by sodium borohydride. The cyclisation of the N(hexenylidene)amine 15a (Table 2, entry 2), which represents a homolog of the imine 9j, afforded a six-membered ring product by the exo-trig mode. The dr determined for the corresponding piperidine 17a was lower than that determined in the case of 9j cyclisation. Thus, this homologation had a disadvantageous effect on the diastereoselectivity (Table 2, entries 1 and 2).



Scheme 3. Cyclisation of optically active N-(alkenylidene)alkylamines 15.

It can be expected that an R¹ methyl substituent shields the C=C bond to a certain extent, and thus, could give rise to an increase in stereoselectivity in electrophile-induced cyclisation. However, the opposite effect was observed (Table 2, compare entry 1 with entry 3 and entry 1 with entry 5). The application of bromine instead of phenylselenyl bromide gave the same unsatisfactory diastereoselectivity (Table 2, entries 1 and 4). Finally, we investigated the effect of a phenyl group at position 1 of the alkylidene moiety of imines 15 (Table 2, entry 6). Selenocyclisation followed by reduction gave rise to a mixture of four diastereomeric products 17d (dr = 13:21:38:28). The cyclisation

Entry	Imine	Iminium salt	Saturated heterocycle	\mathbb{R}^1	R ²	R ³	R ⁴	п	Х	dr	Yield [%]
1	9j	12f	14f	Н	Н	COOCH ₃	Ph	1	PhSe	35:65	58
2	15a	16a	17a	Н	Н	COOCH ₃	Ph	2	PhSe	46:54	64
3	15b	16b	17b	Me	Н	COOCH ₃	Ph	1	PhSe	46:54	36
4	15c	16c	_	Me	Н	COOCH ₃	Ph	1	Br	46:54	quant.
5	9f	12b	14b	Н	Н	Ph	Me	1	PhSe	47:53	62
6	15d	16d	17d	Н	Ph	Ph	Me	1	PhSe	13:21:38:28	57

to **16d** as well as the subsequent reduction to **17d**, where a second stereogenic centre was formed, occurred with unsatisfactory diastereoselectivity. In summary, homologation as well as the introduction of substituents into the alkylidene moiety of chiral imines did not lead to higher stereoselectivities.

We further approached diastereoselective seleno- or halocyclisation by introducing chirality into the alkenylidene moiety of unsaturated imines. Imines 18 were obtained from a ribose-derived aldehyde (see electronic Supporting Information).^[53–55] In the N(alkenvlidene)alkylamine 18a, chirality is only found in the alkylidene part, while the other imines 18 are chiral in both the alkylidene and the amine moiety (Scheme 4, Table 3). In all the imines 18, a chiral centre is near the C=C bond, and thus, a better facial selectivity of the primary electrophilic attack can be expected, which is probably crucial for the overall diastereoselective formation of products 19. Indeed, an increase in diastereoselectivity to dr = 20.80 was observed in the selenocyclisation of 18a with phenylselenyl bromide. But similar to the other selenocyclisations (vide supra), the product 20a was formed in low yield. Working at low temperature in CHCl₃ had no advantageous effect (Table 3, entry 1, footnote a). Remarkably, the application of the bulkier 2,4,6-(triisopropylphenyl)selenyl bromide gave lower stereoselectivity (38:62) as it did for 9j (Table 3, entry 3 and Table 2, entry 14). The same stereochemical result was obtained with bromine (Table 3, entry 2), and bromopyrrolinium salt 19b was obtained in high yield.



Scheme 4. Cyclisation of optically active *N*-(alkenylidene)alkylamines **18** with chiral alkylidene groups with achiral electrophiles.

Epimers **18b** and **18c** appeared as a matched/mismatched pair in the selenocyclisation. We observed dr values of 16:84 and 40:60, respectively. Obviously, the chirality in the alkylidene moiety of the unsaturated imine exerts the predominating asymmetric induction. This fact also matches with the very poor stereoselectivities in the cyclisation of imines **9** lacking chirality in the alkylidene part (see Table 1, entry



[a] 38%, dr = 75:25 at -50 °C in CHCl₃. [b] *cis/trans* assignments for the isomers were not possible by NMR spectroscopy.

8). The selenocyclisation of the phenylglycine derivative **18d** gave a marked diastereoselectivity of dr = 19:81, but the yields were even lower. NOE investigations of the products **20** and an X-ray crystal analysis of the pyrrolidine **20d** (see Figure 3) revealed that in all cases the *all-cis*-isomer (**R**)-**20** was the major isomer.



Figure 3. X-ray crystal analysis of the 5S stereoisomer of 20d.

Cyclisation of *N*-(Alkenylidene)alkylamines with Chiral Electrophiles

Inspired by the fact that chiral organoselenenyl halides can stereoselectively react with alkenes on one hand^[56,57] and that this methodology has not yet been applied to the inter- or intramolecular additions of imines to alkenes on the other hand, we tried to apply such chiral selenyl bromides to the selenocyclisation of unsaturated imines. The chiral selenyl bromides $22^{[58-67]}$ and $23^{[68-77]}$ generated in situ from the corresponding diselenides and bromine, were applied to the achiral imine **9a** at -78 °C (Scheme 5).



Scheme 5. Cyclisation of *N*-(alkenylidene)alkylamines **9** with chiral selenenyl bromides.

The camphor-derived selenyl bromide **22** gave a slightly higher diastereoselectivity than did **23**, but it was still unsatisfactory (dr = 41:59 and 48:52, Table 4, entries 1 and 2, respectively). In a double asymmetric induction^[78] approach, selenyl bromide **22** was reacted with the chiral amines **9j** and (*S*)-**9j**. Compound (*S*)-**9j** appeared as a matched case (38:62 and 24:76, Table 4, entries 3 and 4, respectively). Each of the diastereomeric products **21c** exists in two rotameric forms, as proven by temperature-dependent dynamic ¹H NMR investigations.

Table 4. Synthesis of pyrrolidines 21^[a].

Entry	\mathbb{R}^1	Imine	R ² –SeBr	Pyrroli- dines	Yield [%]	dr
1	Н	9a	22	21a	33	41:59
2	Н	9a	23 ^[b]	21b	72	48:52
3	CO ₂ Me	9j	22	21c	46	38:62 ^[c,d]
4	CO ₂ Me	(<i>S</i>)-9j	22	21d	40	24:76

[a] Reactions were performed at 0 °C. [b] $AgPF_6$ was added. [c] Separation of diastereomers by column chromatography. [d] Diastereomers were isolated as a mixture of rotamers.

Conclusions

In summary, a series of new chiral 2-(α -bromoalkyl)pyrrolinium salts **10**, **16c** and **19b** and 2-(α -selenoalkyl)pyrrolidines **14**, **17**, **20** and **21** were obtained by the halo- or selenocyclisations, respectively, of unsaturated imines (*N*-alkylidene-*N*-alkylamines) and subsequent reduction. Several effects on the stereoselectivity, such as chirality in the amine or the alkylidene substituent, the size of substituents in the alkylidene moiety, the type of electrophile (bromine, iodine, arylselenyl bromides, or chiral organoselenenyl bromides), temperature and the presence of Lewis acids were investigated. As a result, marked diastereoselectivities (up to dr = 84:16) were only obtained with a pentose-derived chiral alkylidene substituent and in the case of double asymmetric induction when the imine (*S*)-**9**, with a chiral *N*-alkyl group (phenylalanine derivative), was cyclised with a camphor-derived selenyl bromide. These are the first cases reported for stereoselective electrophile-induced cyclisations of unsaturated *N*-alkylidene-*N*-alkylamines.

Experimental Section

General Remarks: ¹H NMR and ¹³C NMR spectra were recorded at 300 MHz and 75.5 MHz, respectively, with a Bruker AC 300 spectrometer in CDCl₃ with TMS as an internal standard. Diastereomeric ratios were determined by ¹H NMR spectroscopy of the crude products. Cyclisation products were obtained as diastereomeric mixtures, and their spectra were recorded from the mixtures, if not otherwise mentioned. Silica gel (0.04–0.063 mm, Merck) was used for preparative column chromatography. Starting materials were purchased from commercial suppliers. All *N*(alkenylidene)alkylamines and 3,4-dihydro-2*H*-pyrrolium bromides decompose easily but can be stored in a freezer (–20 °C) for a few days.

General Procedure for the Synthesis of 3,4-Dihydro-2*H*-pyrrolium Bromides 10: The appropriate N(alkenylidene)alkylamine 9 (0.50 mmol) was dissolved in dry CH₂Cl₂ (10 mL) at 0 °C under argon in a flame-dried flask. A solution of bromine in dry CH₂Cl₂ (0.04 M, 0.50 mmol) was added over 30 min, and the temperature was maintained for 2 h at 0 °C. The mixture was warmed slowly to room temperature. The solution was concentrated under vacuum below room temperature, and the remaining product was stored under argon at -20 °C.

rac-1-Benzyl-2-bromomethyl-4,4-dimethyl-3,4-dihydro-2*H*-pyrrolium Bromide (10a): Starting material 9a (300 mg, 1.49 mmol). Yield: 518 mg (96%), colourless solid, m.p. 60 °C. ¹H NMR (300 MHz, CDCl₃): δ = 1.04 (s, 3 H, CH₃), 1.14 (s, 3 H, CH₃), 2.04 [dd, *J* = 7.5 Hz, *J* = 13.6 Hz, 1 H, C*H*H–C(CH₃)₂], 2.42 [dd, *J* = 8.7 Hz, *J* = 13.6 Hz, 1 H, C*H*H–C(CH₃)₂], 3.74 (dd, *J* = 1.9 Hz, *J* = 12.4 Hz, 1 H, C*H*H–Br), 4.12 (dd, *J* = 4.3 Hz, *J* = 12.2 Hz, 1 H, C*H*H–Br), 5.32 (d, *J* = 6.4 Hz 2 H, C*H*H–Ph), 5.44–5.55 (m, 1 H, C*H*H–Br), 7.37–7.39 (m, 3 H, C*H*_{ar}), 7.56–7.59 (m, 2 H, C*H*_{ar}), 8.99 (s, 1 H, C*H*=N⁺) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 24.8 (CH₃), 25.7 (CH₃), 32.2 (CH₂–Br), 39.1 [CH₂–C(CH₃)₂], 47.5 [C(CH₃)₂], 55.5 (CH₂–Ph), 70.5 (CH–N), 129.2 (*C*_{ar}), 129.8 (2 CH_{ar}), 130.2 (3 CH_{ar}), 186.2 (*C*H=N⁺) ppm. C₁₄H₁₉Br₂N (361.115): calcd. C 46.56, H 5.30, Br 44.25, N 3.88; found C 46.60, H 5.32, Br 43.78, N 3.87.

(2*S*)-2-Bromomethyl-1-[(1*S*,2*S*)-1-methoxycarbonyl-2-methylbutyl]-4,4-dimethyl-3,4-dihydro-2*H*-pyrrolium Bromide and (2*R*)-2-Bromomethyl-1-[(1*S*,2*S*)-1-methoxycarbonyl-2-methylbutyl]-4,4-dimethyl-3,4-dihydro-2*H*-pyrrolium Bromide (10b): Starting material 9b (60 mg, 0.25 mmol). Diastereomers were not separated. Yield: 100 mg (0.25 mmol). Diastereomers were not separated. Yield: 100 mg (0.25 mmol), quant.), dr = 44:56, yellow oil. ¹H NMR (300 MHz, CDCl₃): $\delta = 0.78-0.95$ (m, 4 H, C*H*-CH₃, CH₂-C*H*₃), 0.98 (s, 3 H, CH₃), 1.01 (s, 3 H, C*H*₃), 1.51, 1.54 (d, J = 6.1 Hz, J = 7.9 Hz, 3 H, CH-CH₃), 1.30–1.55 (m, 2 H, CH₂-CH₃), 1.94–2.56 [m, 2 H, CH₂-C(CH₃)₂], 3.82–4.41 (m, 2 H, CH₂-Br), 3.79, 3.84 (s, 3 H, COOCH₃), 4.88, 4.95 (d, J = 9.1 Hz, J = 5.6 Hz 1 H, CH-COO), 5.59–5.68, 5.89–5.98 (m, 1 H, CH-N), 9.38, 9.64 (s, 1 H, CH=N⁺) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 11.3$, 11.4 $\begin{array}{l} ({\rm CH_2-CH_3}),\,14.8,\,15.2\,\,({\rm CH-CH_3}),\,24.7,\,24.9\,\,({\rm C_q-CH_3}),\,25.8,\,26.8\\ ({\rm C_q-CH_3}),\,25.2,\,27.1\,\,({\rm CH_2-CH_3}),\,33.0,\,34.2\,\,({\rm CH_2-Br}),\,36.6,\,40.2\\ ({\rm CH-CH_3}),\,37.8,\,38.4\,\,[{\rm CH_2-C(CH_3)_2}],\,48.9,\,49.0\,\,[{\rm C(CH_3)_2}],\,53.9,\\ 54.2\,\,({\rm COOCH_3}),\,66.7,\,67.4\,\,({\rm CH-N}),\,73.3,\,74.6\,\,({\rm CH-COO}),\,166.8,\\ 167.9\,\,({\rm COO}),\,189.2,\,190.3\,\,({\rm CH=N^+})\,\,{\rm ppm.}\,\,{\rm HRMS}\,\,({\rm EI}):\,{\rm calcd.}\,\,{\rm for}\\ {\rm C_{14}H_{25}BrNO_2}\,\,318.1069/320.1048;\,\,{\rm found}\,\,318.1069/320.1049.\\ \end{array}$

(2S)-2-Bromomethyl-1-[(1S)-1-methoxymethyl-2-methylpropyl]-4,4dimethyl-3,4-dihydro-2*H*-pyrrolium Bromide and (2*R*)-2-Bromomethyl-1-[(1S)-1-methoxymethyl-2-methylpropyl]-4,4-dimethyl-3,4-dihydro-2H-pyrrolium Bromide (10e): Starting material 9e (379 mg, 1.79 mmol). Diastereomers were not separated. Yield: 616 mg (1.66 mmol, 93%), dr = 42:58, yellow oil. ¹H NMR (300 MHz, CDCl₃): δ = 1.09 (d, J = 6.4 Hz, 3 H, CH–CH₃), 1.17 (d, J = 6.8 Hz, 3 H, CH–CH₃), 1.52 (s, 3 H, CH₃), 1.61 (s, 3 H, CH₃), 2.16–2.62 [m, 4 H, CH–CH₃, CH–CH₂O, CH₂–C(CH₃)₂], 3.43, 3.45 (s, 3 H, OCH₃), 3.70-4.14 (m, 4 H, CH₂-Br, CH₂O), 5.30-5.41 (m, 1 H, CH-N), 8.72, 8.86 (s, 1 H, CH=N⁺) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 18.8, 19.3 (CH–CH₃), 19.5, 20.4 (CH-CH₃), 24.8, 25.0 (C_q-CH₃), 26.0, 26.6 (C_q-CH₃), 27.7, 31.0 (CH-CH₃), 32.2, 33.0 (CH₂-Br), 38.4, 38.5 [CH₂-C(CH₃)₂], 48.3, 48.5 [C(CH₃)₂], 59.5, 59.6 (OCH₃), 67.8, 68.5 (CH-CH₂O), 68.6, 68.7 (CH₂–O), 71.9, 74.0 (CH–N), 187.3, 187.8 (CH=N⁺) ppm. HRMS (EI): calcd. for C₁₃H₂₅BrNO 290.1120; found 290.1118. Tribromide (S)-10e was obtained from a CHCl₃ solution with excess Br₂ as an orange solid with m.p. 114 °C (CH₂Cl₂/Et₂O).

X-ray Crystal Analysis of the Corresponding Tribromide (S)-10e:^[79] Empirical formula: C13H25Br4NO, formula weight: 530.98 g/mol, temperature: 180(2) K, wavelength: 0.71073 Å, crystal system: orthorhombic, space group: P212121, unit cell dimensions: a =9.341(3) Å, b = 13.381(4) Å, c = 13.381(4) Å, V = 1928.1(12) Å³, Z = 4, density (calculated): 1.829 Mg/m³, absorption coefficient: 8.343 mm^{-1} , F(000) = 1032, crystal size: $0.80 \times 0.32 \times 0.28 \text{ mm}$, theta range for data collection: 2.01 to 25.90°, limiting indices: -8 $\leq h \leq 11, 0 \leq k \leq 16, 0 \leq l \leq 18$; reflections collected: 4078, unique: 3586 [*R*(int.) = 0.0515], completeness to θ = 25.90 is 99.9%, absorption correction: empirical (psi-scan), max./min. transmission: 0.2035/0.0573, refinement method: full-matrix leastsquares on F², data/restraints/parameters: 3586-0-189, goodnessof-fit on F^2 : 1.112, final R indices $[I > 2\sigma(I)]$: $R_1 = 0.0631$, $wR_2 =$ 0.1328, R indices (all data): $R_1 = 0.1124$, $wR_2 = 0.1650$, absolute structure parameter: 0.03(4), extinction coefficient: 0.0010(4), largest diff. peak and hole: 0.889 and -1.144 e/Å³.

2-Iodomethyl-4,4-dimethyl-1-[(R)-1-phenylethyl]-3,4-dihydro-2H-pyrrolium Iodide (11a) and 3-Iodo-5,5-dimethyl-1-[(R)-1-phenylethyl]-2,3,4,5-tetrahydropyridinium Iodide (13a): 9f (69 mg, 0.32 mmol) was dissolved in dry CH₂Cl₂ (10 mL) at 0 °C under argon in a flame-dried flask. A solution of iodine (81 mg, 0.32 mmol) in dry CH₂Cl₂ (10 mL) was added over 30 min, and the temperature was maintained for 2 h at 0 °C. The mixture was warmed slowly to room temperature and was concentrated under vacuum below room temperature to give 0.15 g (0.32 mmol, quant. yield) of a yellow oil. Isomers were not separated. 11a/13a = 86:14; dr (11a) = 37:63; dr (13a) = 31:69). HRMS (EI): calcd. for C₁₅H₂₁IN 342.0713; found 342.0713. Major isomer 11a: ¹H NMR (300 MHz, $CDCl_3$): $\delta = 1.46, 1.49$ (s, 3 H, CH_3), 1.58, 1.65 (s, 3 H, CH_3), 1.82– 1.96 (m, 1 H, CH_2 – C_q), 2.03, 2.00 (d, J = 10.0 Hz, 3 H, CH– CH_3), 2.40, 2.57 (dd, J = 8.0 Hz, J = 13.5 Hz, 1 H, CH*H*-C_q), 3.54–3.62 (m, 1 H, CHH-I), 3.74-3.80 (m, 1 H, CHH-I), 5.07-5.16, 5.33-5.42 (m, 1 H, CH-CH₃), 4.13-4.25, 5.40-5.49 (m, 1 H, CH), 7.32-7.48 (m, 5 H, CH_{ar}), 9.92, 10.10 (s, 1 H, CH=N⁺) ppm. ¹³C NMR $(75.5 \text{ MHz}, \text{ CDCl}_3): \delta = 5.9, 7.8 (CH_2-I), 20.9, 21.6 (CH-CH_3),$ 24.6, 25.3 (CH₃), 26.1, 26.4 (CH₃), 41.6, 42.0 (CH₂), 47.6 $\begin{bmatrix} C(CH_3)_2 \end{bmatrix}, 62.5, 63.0 (CH-CH_3), 69.1, 72.0 (CH-N), 126.6, 127.7 \\ (2 CH_{ar}), 128.1, 128.8 (CH_{ar}), 130.1, 130.2 (2 CH_{ar}), 134.2, 134.4 \\ (CH-C_{ar}), 185.1, 185.2 (CH=N^+) ppm. Minor isomer$ **13a**: ¹H $NMR (300 MHz, CDCl_3): <math>\delta = 0.99, 1.10$ (s, 3 H, CH₃), 1.53, 1.58 (s, 3 H, CH₃), 2.03–2.00 (m, 4 H, CH-CH₃, CHH-C_q), 2.07–2.30 (m, 1 H, CHH-C_q), 4.61–4.70, 4.92–5.03 (m, 3 H, CH-CH₃, CH₂–N), 4.65–4.74 (m, 1 H, CH-I), 7.32–7.48 (m, 5 H, CH_{ar}), 9.22, 9.56 (s, 1 H, CH=N⁺) ppm.

1-tert-Butyl-2-iodomethyl-4,4-dimethyl-3,4-dihydro-2H-pyrrolium Iodide (11b) and 1-tert-Butyl-3-iodo-5,5-dimethyl-2,3,4,5-tetrahydropyridinium Iodide (13b): 9k (100 mg, 0.598 mmol) was dissolved in dry CH₂Cl₂ (10 mL) at 0 °C under argon in a flame-dried flask. A solution of iodine (152 mg, 0.598 mmol) in dry CH₂Cl₂ (10 mL) was added over 30 min, and the temperature was maintained at 0 °C for 2 h. The mixture was warmed slowly to room temperature and was concentrated under vacuum. The crude product was recrystallized from CH2Cl2/Et2O to give a white solid (170 mg, 0.404 mmol, 67% yield). Isomers (19:81) were not separated, m.p. 214 °C (CH₂Cl₂/Et₂O). HRMS (ESI⁺): calcd. for C₁₁H₂₁IN 294.0713; found 294.0710. C₁₁H₂₁I₂N (421.100): calcd. C 31.37, H 5.03, N 3.33; found C 31.52, H 5.30, N 3.40. Major isomer 13b: ¹H NMR (300 MHz, [D₆]DMSO): δ = 1.04 (s, 3 H, CH₃), 1.10 (s, 3 H, CH₃), 1.25 [s, 9 H, C(CH₃)₃], 1.93 (m, 1 H, CHH-C_a), 2.07 (m, 1 H, CHH-C_a), 3.75 (m, 1 H, CHH-N), 4.11 (m, 1 H, CHH-N), 4.23-4.36 (m, 1 H, CH-I), 8.54 (s, 1 H, CH=N⁺) ppm. ¹³C NMR (75.5 MHz, [D₆]DMSO): δ = 12.7 (CH– I), 25.0 (CH₃), 25.5 (CH₃), 26.9 [3 C, C(CH₃)₃], 46.8 [C(CH₃)₂], 42.0 (CH₂), 53.6 (CH₂–N), 69.2 [C(CH₃)₃], 178.9 (CH=N⁺) ppm. Minor isomer 11b: ¹H NMR (300 MHz, $[D_6]DMSO$): $\delta = 1.17$ (s, 3 H, CH₃), 1.25 (s, 3 H, CH₃), 1.32 [s, 9 H, C(CH₃)₃], 1.84–1.93 (m, 1 H, CHH-C_q), 2.32–2.43 (m, 1 H, CHH-C_q), 3.59–3.68 (m, 1 H, CHH-I), 3.91-4.02 (m, 1 H, CHH-I), 4.98-5.09 (m, 1 H, CH), 9.25 (s, 1 H, CH=N⁺) ppm. ¹³C NMR (75.5 MHz, [D₆]DMSO): δ = 9.7 (CH₂-I), 19.0 (CH₃), 25.6 (CH₃), 28.8 [3 C, C(CH₃)₃], 46.7 [C(CH₃)₂], 41.8 (CH₂), 65.2 [C(CH₃)₃], 72.6 (CH–N), 187.9 $(CH=N^+)$ ppm.

General Procedure for the Selenocyclisation and Reduction to Pyrrolidines 14: The appropriate N(alkenylidene)alkylamine 9 (0.50 mmol) was dissolved in dry CH₂Cl₂ (10 mL) at -78 °C under argon in a flame-dried flask. A solution of phenylselenyl bromide (0.60 mmol, 1.2 equiv.) in dry CH₂Cl₂ was added over 30 min. After 2 h at -78 °C, MeOH (10 mL) and NaBH₄ (0.74 mmol) were added. The mixture was warmed slowly to room temperature. The organic layer was separated and washed with water (10 mL). After separating the layers, the aqueous layer was washed with CH₂Cl₂ (2 × 5 mL). The combined organic layers were dried with MgSO₄ and concentrated under vacuum. The remaining yellow oil was purified by column chromatography (40 g silica gel, 2.3 cm column diameter, cyclohexane/EtOAc mixture).

rac-1-Benzyl-4,4-dimethyl-2-[(phenylselenyl)methyl]pyrrolidine (14a): Starting material 9a (244 mg, 1.21 mmol). Yield: 258 mg (0.719 mmol, 59%), colourless oil, $R_f = 0.14$ (hexane/EtOAc, 20:1). ¹H NMR (300 MHz, CDCl₃): $\delta = 0.98$ (s, 3 H, C_q-CH₃), 1.10 (s, 3 H, C_q-CH₃), 1.65 (dd, J = 7.9 Hz, J = 12.8 Hz, 1 H, CH-CH'H-C_q), 1.83 (dd, J = 8.1 Hz, J = 12.6 Hz, 1 H, CH-CH'H-C_q), 2.01 (d, J = 9.0 Hz, 1 H, N-CH'H-C_q), 2.65 (d, J = 9.0 Hz, 1 H, N-CH'H-C_q), 2.88 (m, 1 H, CH-CH₂Se), 3.11 (dd, J = 3.1 Hz, J = 11.4 Hz, 1 H, CH'H-Se), 3.15 (dd, J = 4.1 Hz, J = 11.5 Hz, 1 H, CH'H-Se), 3.17 (d, J = 13.2 Hz, 1 H, Ph-CH'H), 4.04 (d, J = 13.2 Hz, 1 H, Ph-CH'H), 7.19-7.48 (m, 10 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 28.9$ (CH₃), 30.1 (CH₃), 33.2 (CH₂Se), 35.9 [C(CH₃)₂], 46.5 (CH-CH₂-C_q), 58.3 (CH₂-Ph), 63.6

(CH), 68.2 (N– CH_2-C_q), 126.5 (CH_{ar}), 126.9 (CH_{ar}), 128.2 (2 CH_{ar}), 128.7 (2 CH_{ar}), 129.1 (2 CH_a), 131.4 (C_arSe), 132.3 (2 CH_a), 139.8 (CH– C_{ar}) ppm. C₂₀H₂₅NSe (358.379): calcd. C 67.03, H 7.03, N 3.91; found C 67.26, H 7.07, N 3.82.

(2R)-4,4-Dimethyl-1-[(1R)-1-phenylethyl]-2-[(phenylselenyl)methyl]pyrrolidine and (2S)-4,4-Dimethyl-1-[(1R)-1-phenylethyl]-2-[(phenylselenyl)methyl|pyrrolidine (14b): Starting material 9f (82 mg, 0.38 mmol). Diastereomers were separated, dr = 47:53. Minor isomer: yield 42 mg (0.11 mmol, 29%), colourless oil, $R_{\rm f} = 0.57$ (hexane/EtOAc, 9:1). $[a]_D^{25} = -45.9$ (c = 1.4, CHCl₃). ¹H NMR (300 MHz, CDCl₃): δ = 0.92 (s, 3 H, C_q-CH₃), 1.06 (s, 3 H, C_q-CH₃), 1.21 (d, J = 6.7 Hz, 3 H, CH–CH₃), 1.54 (dd, J = 6.3 Hz, J = 12.6 Hz, 1 H, CH–CH'H–C_q), 1.74 (dd, J = 8.3 Hz, J = 12.6 Hz, 1 H, CH–CH'H–C_q), 2.24 (d, J = 8.9 Hz, 1 H, N–CH'H), 2.44 (d, J = 8.9 Hz, 1 H, N–CH'H), 2.79–2.82 (m, 2 H, CH₂Se), 3.06–3.10 (m, 1 H, CH–CH₂Se), 3.78 (q, J = 6.6 Hz, 1 H, CH–CH₃), 7.10– 7.34 (m, 10 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, $CDCl_3$): δ = 15.1 (CH-CH₃), 28.3 (C_q-CH₃), 29.0 (C_q-CH₃), 33.3 (CH₂Se), 35.4 [C(CH₃)₂], 45.8 (CH–CH₂), 57.8 (CH), 60.7 (CH), 62.1 (N–CH₂), 126.0 (CH_{ar}), 126.3 (CH_{ar}), 127.2 (2 CH_{ar}), 127.7 (2 CH_{ar}), 128.5 $(2 CH_{ar})$, 130.6 (C_{ar} Se), 131.8 (2 CH_{ar}), 144.9 (CH- C_{ar}) ppm. HRMS (EI): calcd. for C₂₁H₂₇NSe 373.13087; found 373.13084. Major isomer: yield of 47 mg (0.13 mmol, 33%), colourless oil, $R_{\rm f}$ = 0.46 (hexane/EtOAc, 9:1). $[a]_{D}^{25}$ = 142.7 (c = 1.1, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 0.74$ (s, 3 H, C_q-CH₃), 1.03 (s, 3 H, C_q - CH_3), 1.30 (d, J = 6.9 Hz, 3 H, CH- CH_3), 1.52 (dd, J = 6.5 Hz, $J = 12.6 \text{ Hz}, 1 \text{ H}, \text{ CH-CH'}H-C_q), 1.64 \text{ (dd}, J = 8.0 \text{ Hz}, J =$ 12.6 Hz, 1 H, CH–CH'H–C_g), 2.15 (d, J = 8.9 Hz, 1 H, N–CH'H), 2.59 (d, J = 8.9 Hz, 1 H, N-CH'H), 2.90-2.94 (m, 1 H, CH-CH₂Se), 3.00–3.13 (m, 2 H, CH₂Se), 3.75 (q, J = 6.8 Hz, 1 H, CH-CH₃), 7.06-7.41 (m, 10 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 21.6 (CH–CH₃), 28.3 (C_q–CH₃), 29.1 (C_q–CH₃), 34.3 [C(CH₃)₂], 35.9 (CH₂Se), 46.2 (CH-CH₂), 59.3 (CH), 59.4 (CH), 62.9 (N-CH₂), 126.6 (CH_{ar}), 126.8 (CH_{ar}), 127.9 (2 CH_{ar}), 128.0 (2 CH_{ar}), 129.0 (2 CH_{ar}), 131.0 (C_{ar}Se), 132.7 (2 CH_{ar}), 142.1 (CH-Car) ppm. C₂₁H₂₇NSe (372.406): calcd. C 67.73, H 7.31, N 3.76; found C 67.69, H 7.07, N 3.74.

(R)-Methyl 2-[(2R)-4,4-Dimethyl-2-(phenylselenylmethyl)pyrrolidin-1-yl]-2-phenylacetate [(R)-14f] and (R)-Methyl 2-[(2S)-4,4-Dimethyl-2-(phenylselenylmethyl)pyrrolidin-1-yl]-2-phenylacetate [(S)-14f]: Starting material 9j (98 mg, 0.38 mmol). Diastereomers were separated. dr = 35:65. (R)-14f: yield of 57 mg (0.14 mmol, 37%), colourless crystals, $R_f = 0.26$ (hexane/EtOAc, 9:1), m.p. 46–48 °C (CHCl₃). $[a]_D^{20} = -7.9$ (c = 1.00, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 0.72$ (s, 3 H, C_q-CH₃), 1.03 (s, 3 H, C_q-CH₃), 1.48 (dd, J = 6.4 Hz, J = 12.8 Hz, 1 H, CH–CH'H–C_q), 1.72 (dd, J =8.1 Hz, J = 12.6 Hz, 1 H, CH–CH'H–C_q), 2.20 (d, J = 9.0 Hz, 1 H, N-CH'H), 2.81 (d, J = 9.4 Hz, 1 H, N-CH'H), 2.85 (d, J = 9.8 Hz, 1 H, SeCH'H), 2.98 (dd, J = 2.6 Hz, J = 11.7 Hz 1 H, SeCH'H), 3.02-3.09 (m, 1 H, CH-CH₂Se), 3.56 (s, 3 H, COOCH₃), 4.43 (s, 1 H, CH–COOCH₃), 7.08–7.37 (m, 10 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 28.1 (C_q-CH₃), 28.9 (C_q-CH₃), 32.9 (CH₂Se), 36.0 [C(CH₃)₂], 46.0 (CH-CH₂), 51.8 (COOCH₃), 60.5 (CH), 63.6 (N-CH₂), 67.4 (CH), 126.7 (CH_{ar}), 128.0 (CH_{ar}), 128.3 (2 CH_{ar}), 129.0 (2 CH_{ar}), 129.2 (2 CH_{ar}), 130.5 (C_{ar}Se), 132.8 (2 CH_{ar}), 134.9 (C_{ar}), 172.5 (COO) ppm. HRMS (ESI⁺): calcd. for C₂₂H₂₇NO₂Se 417.1207; found 417.1207. (S)-14f: yield of 35 mg (0.084 mmol, 22%), colourless oil, $R_{\rm f} = 0.35$ (hexane/EtOAc, 9:1). $[a]_{D}^{20} = 1.2 \ (c = 1.00, \text{ CHCl}_3)$. ¹H NMR (300 MHz, CDCl₃): $\delta =$ 0.95 (s, 3 H, C_q – CH_3), 1.00 (s, 3 H, C_q – CH_3), 1.60 (dd, J = 7.3 Hz, $J = 12.6 \text{ Hz}, 1 \text{ H}, \text{ CH-CH'}H-C_q), 1.79 \text{ (dd}, J = 7.9 \text{ Hz}, J = 12.6 \text{ Hz}, 1 \text{ H}, \text{ CH-CH'}H-C_q)$ 12.8 Hz, 1 H, CH–CH'H–C_q), 2.46 (d, J = 9.0 Hz, 1 H, N–CH'H), 2.68 (d, J = 9.0 Hz, 1 H, N–CH'H), 2.91 (dd, J = 3.2 Hz, J =

11.5 Hz, 1 H, SeCH'H), 2.98 (dd, J = 7.9 Hz, J = 11.7 Hz 1 H, SeCH'H), 3.21 (dq, J = 3.1 Hz, J = 7.7 Hz 1 H, CH–CH₂Se), 3.61 (s, 3 H, COOCH₃), 4.65 (s, 1 H, CH–COOCH₃), 7.13–7.35 (m, 10 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 28.5$ (C_q–CH₃), 29.2 (C_q–CH₃), 33.0 (CH₂Se), 35.8 [C(CH₃)₂], 46.3 (CH–CH₂), 51.3 (COOCH₃), 60.3 (CH), 62.6 (N–CH₂), 66.1 (CH), 126.5 (CH_{ar}), 127.8 (CH_{ar}), 128.3 (2 CH_{ar}), 128.5 (2 CH_{ar}), 128.9 (2 CH_{ar}), 130.8 (C_{ar}Se), 132.2 (2 CH_{ar}), 137.1 (C_{ar}), 172.0 (COO) ppm. C₂₂H₂₇NO₂Se (416.415): calcd. C 63.45, H 6.54, N 3.36; found C 63.40, H 6.39, N 3.34.

X-ray Crystal Analysis of (S)-14f:^[79] Empirical formula: C₂₂H₂₇NO₂Se, formula weight: 416.41 g/mol, temperature: 180(2) K, wavelength: 0.71073 Å, crystal system: monoclinic, space group: P21, unit cell dimensions: a = 6.006(2) Å, b = 14.311(3) Å, $\beta = 95.82(4)^{\circ}$, c = 11.812(3) Å, V = 1009.9(5) Å³, Z = 2, density (calculated): 1.369 Mg/m³, absorption coefficient: 1.874 mm⁻¹, F(000): 432, crystal size: $0.60 \times 0.48 \times 0.32$ mm, theta range for data collection: 2.24 to 25.24°, limiting indices: $-7 \le h \le 7, -17 \le$ $k \le 17, -14 \le l \le 14$; reflections collected: 6613, unique: 3487 [R(int.) = 0.1335], completeness to θ = 25.24 is 96.8%, absorption correction: empirical (psi-scan), max./min. transmission: 0.5854/ 0.3993, refinement method: full-matrix least-squares on F^2 , data/ restraints/parameters: 3487/1/239, goodness-of-fit on F²: 0.960, final R indices $[I > 2\sigma(I)]$: $R_1 = 0.0867$, $wR_2 = 0.2052$, R indices (all data): $R_1 = 0.1140$, $wR_2 = 0.2235$, absolute structure parameter: -0.06(3), extinction coefficient: 0.026(6), largest diff. peak and hole: 1.560 and -1.293 e/Å³.

(R)-Methyl 2-{(2R)-4,4-Dimethyl-2-[(2,4,6-triisopropylphenylselenyl)methyl]pyrrolidin-1-yl}-2-phenylacetate and (R)-Methyl 2-{(2S)-4,4-Dimethyl-2-[(2,4,6-triisopropylphenylselenyl)methyl]pyrrolidin-1yl}-2-phenylacetate (14g): Methyl (R)-2-(2,2-dimethylpent-4-enylideneamino)-2-phenylacetate (9j) (807 mg, 3.11 mmol) was dissolved in dry CH₂Cl₂ (40 mL) at -78 °C under argon in a flamedried flask. A solution of (2,4,6triisopropylphenyl)selenyl bromide (5.10 mmol, 1.6 equiv.), generated in situ from bis(2,4,6-triisopropylphenyl) diselenide (1.44 g, 2.55 mmol) and bromine (407 mg, 2.55 mmol) at -78 °C in dry CH₂Cl₂ (30 mL), was added over 30 min. After 2 h at -78 °C, MeOH (10 mL) and NaBH₄ (241 mg, 6.37 mmol) were added. The mixture was warmed slowly to room temperature. The organic layer was separated and washed with water (10 mL). The aqueous layer was extracted with CH₂Cl₂ $(2 \times 5 \text{ mL})$. The combined organic layers were dried with MgSO₄ and concentrated under vacuum. The remaining oil was purified by column chromatography (400 mL silica gel, 4.5 cm column diameter, cyclohexane/EtOAc, 9:1). Diastereomers were not separated. dr = 61:39. Major isomer: yield of 0.63 g (37%), colourless oil, $R_{\rm f}$ = 0.70 (hexane/EtOAc, 9:1). ¹H NMR (300 MHz, CDCl₃): δ = 0.96 (s, 3 H, C_q – CH_3), 1.00 (s, 3 H, C_q – CH_3), 1.14 [d, J = 6.5 Hz, 6 H, CH(CH₃)₂], 1.15 [d, J = 5.3 Hz, 6 H, CH(CH₃)₂], 1.18 [d, J = 6.8 Hz, 6 H, CH(CH₃)₂], 1.62 (dd, J = 7.2 Hz, J = 12.4 Hz, 1 H, CH–CH'H–C_q), 1.81 (dd, J = 7.7 Hz, J = 12.6 Hz, 1 H, CH– $CH'H-C_q$), 2.40 (d, J = 8.7 Hz, 1 H, N-CH'H), 2.57 (dd, J =2.6 Hz, J = 10.9 Hz, 1 H, SeCH'H), 2.70 (d, J = 8.7 Hz, 1 H, N-CH'H), 2.73–2.82 [m, 2 H, SeCH'H, CH(CH₃)₂], 3.17 (dq, J =2.7 Hz, J = 7.6 Hz, 1 H, CH–CH₂Se), 3.54 (s, 3 H, COOCH₃), 3.83 $[q, J = 6.9 \text{ Hz}, 2 \text{ H}, CH(CH_3)_2], 4.62 (s, 1 \text{ H}, CH-COOCH_3), 6.92$ (s, 2 H, CH_{ar}), 7.17–7.28 (m, 5 H, CH_{ar}) ppm. ¹³C NMR $(75.5 \text{ MHz}, \text{CDCl}_3): \delta = 24.1 [2 \text{ C}, \text{CH}(CH_3)_2], 27.0 [4 \text{ C}, 2$ CH(CH₃)₂], 28.8 (C_q-CH₃), 29.5 (C_q-CH₃), 34.3 [CH(CH₃)₂], 34.4 [2 C, CH(CH₃)₂], 35.0 (CH₂Se), 35.9 [C(CH₃)₂], 46.6 (CH-CH₂), 51.3 (COOCH₃), 60.9 (CH), 62.1 (N-CH₂), 65.5 (CH), 121.7 (2 CH_{ar}), 127.8 (CH_{ar}), 128.4 (2 CH_{ar}), 128.5 (2 CH_{ar}), 129.3 (C_{ar}Se), 137.3 (C_{ar}CH), 149.4 [C_{ar}CH(CH₃)₂], 153.2 [2 C, C_{ar}CH(CH₃)₂],

172.1 (COO) ppm. Minor isomer: yield of 0.40 g (24%). $R_{\rm f} = 0.60$ (hexane/EtOAc, 9:1). ¹H NMR (300 MHz, CDCl₃): $\delta = 0.67$ (s, 3) H, $C_{q}-CH_{3}$), 1.04 (s, 3 H, $C_{q}-CH_{3}$), 1.14 [d, J = 7.2 Hz, 6 H, $CH(CH_3)_2$], 1.16 [d, J = 6.8 Hz, 6 H, $CH(CH_3)_2$], 1.20 [d, J =6.8 Hz, 6 H, $CH(CH_3)_2$], 1.47 (dd, J = 6.8 Hz, J = 12.4 Hz, 1 H, $CH-CH'H-C_q$, 1.74 (dd, J = 7.9 Hz, J = 12.8 Hz, 1 H, CH- $CH'H-C_{g}$, 2.17 (d, J = 9.0 Hz, 1 H, N-CH'H), 2.63 (d, J =6.4 Hz, 2 H, SeCH₂), 2.80–2.98 [m, 3 H, N–CH'H, CH(CH₃)₂, CH- CH_2Se], 3.48 (s, 3 H, $COOCH_3$), 3.83 [q, J = 7.0 Hz, 2 H, CH(CH₃)₂], 4.32 (s, 1 H, CH–COOCH₃), 6.89–6.92 (m, 2 H, CH_{ar}), 6.96 (s, 2 H, CH_{ar}), 7.14–7.18 (m, 3 H, CH_{ar}) ppm. ¹³C NMR $(75.5 \text{ MHz}, \text{CDCl}_3)$: $\delta = 24.9 [2 \text{ C}, \text{CH}(CH_3)_2], 27.0 [4 \text{ C}, 2$ CH(CH₃)₂], 28.2 (C_q-CH₃), 29.1 (C_q-CH₃), 34.3 [CH(CH₃)₂], 34.4 [2 C, CH(CH₃)₂], 35.3 (CH₂Se), 36.2 [C(CH₃)₂], 46.3 (CH-CH₂), 51.8 (COOCH₃), 60.8 (CH), 63.1 (N-CH₂), 66.9 (CH), 121.8 (2 CH_{ar}), 127.6 (C_{ar}Se), 128.0 (CH_{ar}), 128.3 (2 CH_{ar}), 129.3 (2 CH_{ar}), 134.4 (CarCH), 149.6 [CarCH(CH₃)₂], 153.2 [2 C, CarCH(CH₃)₂], 172.6 (COO) ppm. HRMS (ESI⁺): calcd. for C₃₁H₄₆NO₂Se 544.2688; found 544.2683.

(2R)-2-Bromomethyl-1-[(R)-(methoxycarbonyl)(phenyl)methyl]-2,4,4-trimethyl-3,4-dihydro-2H-pyrrolium Bromide and (2S)-2-Bromomethyl-1-[(R)-(methoxycarbonyl)(phenyl)methyl]-2,4,4-trimethyl-3,4-dihydro-2H-pyrrolium Bromide (16c): Starting material 15c (240 mg, 0.88 mmol). Yield: 381 mg (0.88 mmol, quant.). Isomers were not separated, dr = 46:54, yellow foam. HRMS (ESI⁺ in MeCN): calcd. for C₁₇H₂₃BrO₂N 352.0907; found 352.0907. Major isomer: ¹H NMR (300 MHz, CDCl₃): δ = 1.48 (s, 3 H, CH₃), 1.66 (s, 3 H, CH_3), 1.94 (s, 3 H, N-C_a-CH₃), 2.23 (d, J = 7.0 Hz, 1 H, C_q -CH'*H*- $C_q x$), 2.53 (d, *J* = 13.9 Hz, 1 H, C_q -CH'*H*- C_q), 3.82 (d, J = 12.0 Hz, 1 H, Br–CH'H), 3.86 (s, 3 H, CH₃), 4.67 (d, J =12.8 Hz, 1 H, Br-CH'H), 6.63 (s, 1 H, CH-COOCH₃), 7.37-7.78 (m, 5 H, CH_{ar}), 9.41 (s, 1 H, CH=N) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 24.7 (N-C_q-CH_3)$, 26.6 (CH₃), 27.8 (CH₃), 37.5 (CH₂Br), 48.1 [C(CH₃)₂], 45.9 (CH₂), 54.9 (COOCH₃), 63.3 (CH), 83.8 (C_q), 128.4 (2 CH_{ar}), 130.5 (2 CH_{ar}), 131.5 (CH_{ar}), 131.0 (C_{ar}), 166.7 (COO), 189.7 (CH=N) ppm. Minor isomer: ¹H NMR (300 MHz, CDCl₃): δ = 1.49 (s, 3 H, CH₃), 1.67 (s, 3 H, CH₃), 1.96 (s, 3 H, N–C_q–CH₃), 2.26 (d, J = 13.9 Hz, 1 H, C_q–CH'H–C_q), 2.50 (d, J = 13.8 Hz, 1 H, C_q -CH'H- C_q), 3.84 (d, J = 12.7 Hz, 1 H, Br-CH'H), 3.90 (s, 3 H, CH_3), 4.93 (d, J = 12.8 Hz, 1 H, Br-CH'H), 6.79 (s, 1 H, CH-COOCH₃), 7.37-7.78 (m, 5 H, CH_{ar}), 8.87 (s, 1 H, CH=N) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 25.2 (N-C_q-CH₃), 25.3 (CH₃), 27.0 (CH₃), 38.3 (CH₂Br), 48.2 [C(CH₃)₂], 45.8 (CH₂), 55.2 (COOCH₃), 64.3 (CH), 83.1 (C_q), 129.9 (2 CH_{ar}), 130.2 (2 CH_{ar}), 131.2 (CH_{ar}), 130.0 (C_{ar}), 167.6 (COO), 189.7 (CH=N) ppm.

(R)-Methyl 2-[(2R)-5,5-Dimethyl-2-(phenylselenylmethyl)piperidin-1-yl]-2-phenylacetate and (R)-Methyl 2-[(2S)-5,5-Dimethyl-2-(phenylselenylmethyl)piperidin-1-yl]-2-phenylacetate (17a): Starting material 15a (0.15 g, 0.55 mmol). Yield: 0.15 g (0.35 mmol, 64%). Isomers were not separated, dr = 54:46, colourless oil, $R_f = 0.34$ and 0.29 (hexane/EtOAc, 8:2). C23H29NO2Se (430.442): calcd. C 64.18, H 6.79, N 3.25; found C 64.04, H 6.91, N 3.25. Major isomer: ¹H NMR (300 MHz, CDCl₃): $\delta = 0.91$ (s, 3 H, CH₃), 1.00 (s, 3 H, CH₃), 1.10–1.27 (m, 2 H, CH₂), 1.76–1.82 (m, 2 H, CH₂), 1.99 (d, J = 11.6 Hz, 1 H, CH'H), 2.56 (d, J = 11.1 Hz, 1 H, CH'H), 2.85-2.90 (m, 1 H, CH-CH₂Se), 2.91-2.96 (m, 1 H, CH'H), 3.16-3.24 (m, 1 H, CH'H), 3.65 (s, 3 H, CH₃), 4.49 (s, 1 H, CH-COOCH₃), 7.15–7.52 (m, 10 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ $= 25.2 (CH_2Se), 25.5 (CH_3), 28.1 (CH-CH_2), 28.4 (CH_3), 30.7$ [C(CH₃)₂], 33.3 (C_q-CH₂), 51.4 (COOCH₃), 54.5 (SeCH₂-CH), 56.7 (N-CH₂), 69.8 (CH-COO), 126.8 (CH_{ar}), 127.8 (CH_{ar}), 128.6 $(2 CH_{ar}), 129.0 (2 CH_{ar}), 129.2 (2 CH_{ar}), 131.1 (C_{ar}Se), 132.6 (2 CH_{ar}), 132.6 (2 CH_{ar}), 131.1 (C_{ar}Se), 132.6 (2 CH_{ar}), 131.1 (C_{ar}Se), 132.6 (2 CH_{ar}), 132.6 (2 CH_{a$

CH_{ar}), 137.1 (C_{ar}), 171.8 (COO) ppm. Minor isomer: ¹H NMR (300 MHz, CDCl₃): $\delta = 0.78$ (s, 3 H, CH₃), 0.84 (s, 3 H, CH₃), 1.23–1.39 (m, 2 H, CH₂), 1.63–1.72 (m, 2 H, CH₂), 2.15–2.20 (m, 1 H, CH'H), 2.25 (d, J = 12.1 Hz, 1 H, CH'H), 2.98–3.02 (m, 1 H, CH-CH₂Se), 3.20 (t, J = 12.1 Hz, 1 H, CH'Hx), 3.46 (dd, J = 6.7 Hz, J = 11.7 Hz, 1 H, CH'H), 3.69 (s, 3 H, CH₃), 5.06 (s, 1 H, CH-COOCH₃), 7.15–7.52 (m, 10 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 24.8$ (CH₃), 27.0 (CH₂Se), 28.6 (CH₃), 30.8 [C(CH₃)₂], 31.7 (CH-CH₂), 36.4 (C_q-CH₂), 52.0 (COOCH₃), 56.9 (N-CH₂), 57.3 (SeCH₂-CH), 66.1 (CH-COO), 126.8 (CH_{ar}), 128.2 (CH_{ar}), 128.3 (2 CH_{ar}), 128.6 (2 CH_{ar}), 129.1 (2 CH_{ar}), 130.6 (C_{ar}Se), 132.6 (2 CH_{ar}), 135.8 (C_{ar}), 173.2 (COO) ppm.

(R)-Methyl 2-Phenyl-2-[(2R)-2,4,4-trimethyl-2-(phenylselenylmethyl)pyrrolidin-1-yl]acetate and (R)-Methyl 2-Phenyl-2-[(2S)-2,4,4-trimethyl-2-(phenylselenylmethyl)pyrrolidin-1-yl]acetate (17b): Starting material 15b (376 mg, 1.38 mmol). Yield: 216 mg (0.502 mmol, 36%). Isomers were not separated, dr = 46:54, colourless oil, $R_{\rm f} =$ 0.36 (hexane/EtOAc, 20:1). C₂₃H₂₉NO₂Se (430.442): calcd. C 64.18, H 6.79, N 3.25; found C 64.03, H 6.54, N 3.09. Minor isomer: ¹H NMR (300 MHz, CDCl₃): $\delta = 0.98$ (s, 3 H, CH₃), 0.99 (s, 3 H, CH_3), 1.13 (s, 3 H, N-C_q-CH₃), 1.41 (d, J = 12.7 Hz, 1 H, C_q- $CH'H-C_q$, 1.90 (d, J = 13.1 Hz, 1 H, $C_q-CH'H-C_q$), 2.33 (d, J= 9.1 Hz, 1 H, N-CH'H), 2.95 (d, J = 11.4 Hz, 1 H, Se-CH'H),3.02 (d, J = 9.2 Hz, 1 H, N–CH'H), 3.27 (d, J = 11.3 Hz, 1 H, Se– CH'H), 3.65 (s, 3 H, CH₃), 4.68 (s, 1 H, CH–COOCH₃), 7.14–7.43 (m, 10 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, $CDCl_3$): $\delta = 23.4$ (N– C_q-CH₃), 29.2 (CH₃), 29.9 (CH₃), 35.3 [C(CH₃)₂], 40.4 (CH₂Se), 51.7 (COOCH₃), 53.0 (CH₂), 59.6 (CH₂), 61.7 (CH), 64.3 (C_q), 126.6 (CHar), 128.6 (CHar), 127.7 (2 CHar), 128.4 (2 CHar), 129.2 (2 CH_{ar}), 137.9 (C_{ar}Se), 132.2 (2 CH_{ar}), 138.1 (C_{ar}), 174.1 (COO) ppm. Major isomer: ¹H NMR (300 MHz, CDCl₃): $\delta = 0.90$ (s, 3) H, CH₃), 0.96 (s, 3 H, CH₃), 1.21 (s, 3 H, N–C_q–CH₃), 1.45 (d, J = 12.6 Hz, 1 H, C_q–CH'*H*–C_q), 2.02 (d, J = 13.0 Hz, 1 H, C_q– $CH'H- C_q$), 2.44 (d, J = 9.1 Hz, 1 H, N-CH'H), 2.94 (d, J =11.1 Hz, 1 H, Se–CH'H), 2.91 (d, J = 9.2 Hz, 1 H, N–CH'H), 3.15 (d, J = 11.1 Hz, 1 H, Se-CH'H), 3.63 (s, 3 H, CH₃), 4.64 (s, 1 H, CH-COOCH₃), 7.14-7.43 (m, 10 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 25.1 (N–C_q–CH₃), 28.8 (CH₃), 29.4 (CH₃), 34.7 [C(CH₃)₂], 39.8 (CH₂Se), 51.8 (COOCH₃), 53.4 (CH₂), 60.4 (CH₂), 62.3 (CH), 64.7 (C_q), 126.7 (CH_{ar}), 128.5 (CH_{ar}), 128.2 (2 CH_{ar}), 128.4 (2 CH_{ar}), 129.1 (2 CH_{ar}), 137.9 (C_{ar}Se), 132.3 (2 CH_{ar}), 138.4 (C_{ar}), 174.5 (COO) ppm.

3,3-Dimethyl-2-phenyl-1-[(*R***)-1-phenylethyl]-5-(phenylselenylmethyl)pyrrolidine (17d):** Starting material **15d** (0.25 g, 0.86 mmol). Yield: 0.22 g (0.49 mmol, 57%). Isomers were not separated, dr = 13:21:38:28. After being column chromatographed twice, three of these isomers could be isolated as a mixture, colourless oil, $R_f = 0.22$ (hexane/EtOAc, 20:1). ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 14.6$, 18.2, 24.2 (*C*H₃), 24.9, 25.7, 26.3 (*C*H₃), 28.5, 28.9, 31.0 (*C*H₃), 34.6, 34.9, 37.1 (*C*H₂Se), 40.3, 40.6, 40.9 [*C*(*C*H₃)₂], 44.7, 45.7, 46.6 (*C*H-*C*H₂), 54.8, 55.7, 56.7 (*C*H-*C*H₃), 58.4, 59.4, 60.5 (*C*H-*C*H₂Se), 74.6, 75.2, 78.0 (*C*H-Ph), 125.9–133.0 (15*C*H_{ar}), 129.6, 129.8, 130.7 (SeC_{ar}), 140.7, 142.1, 142.3 (*C*_{ar}), 143.8, 144.6, 146.7 (*C*_{ar}) ppm. HRMS (EI): calcd. for C₂₇H₃₁NSe 449.1616; found 449.1612.

(3aR,4R,6aS)-5-Benzyl-4-bromonium-2,2-dimethyl-4,6a-dihydro-3aH-[1,3]dioxolo[4,5c]pyrrol-5-ium Bromide and (3aR,4S,6aS)-5-Benzyl-4-bromonium-2,2-dimethyl-4,6a-dihydro-3aH-[1,3]dioxolo-[4,5-c]pyrrol-5-ium Bromide (19b): N-Benzyl[(4S,5R)-(2,2-dimethyl-5-vinyl[1,3]dioxolane-4-yl)methylene]amine (18a) (0.56 g, 2.3 mmol) was dissolved in dry CH₂Cl₂ (20 mL) at -78 °C under argon in a flame-dried flask. A 1.0 m solution of bromine (2.3 mL,

1.0 equiv.) in CH₂Cl₂, which was diluted with dry CH₂Cl₂ (20 mL) was added over 2 h. After 1 h at -78 °C, the mixture was concentrated under vacuum (T < 30 °C). Yield: 0.93 g (2.3 mmol, quant.), dr = 38:62 (HPLC/MS in MeOH of the crude product), brown foam. HRMS (ESI⁺): calcd. for C₁₅H₁₉BrNO₂ 324.0599/326.0579; found 324.0601/326.0581.

General Procedure for the Selenocyclisation and Reduction to Tetrahydro[1,3]dioxolo-[4,5-c]pyrroles 20: The appropriate N(alkenylidene)alkylamine 18 (2.1 mmol) was dissolved in dry CH₂Cl₂(20 mL) at -78 °C under argon in a flame-dried flask. A solutionof the appropriate arylselenyl bromide (1.3 equiv.) in dry CH₂Cl₂(15 mL) was added over 20 min. After 2.5 h at -78 °C, MeOH(10 mL) and NaBH₄ (4.2 mmol) were added. The mixture waswarmed slowly to room temperature. The organic layer was separated and washed with water (20 mL). After separating the layers,the aqueous layer was washed with CH₂Cl₂ (2×5 mL). The combined organic layers were dried with MgSO₄ and concentrated under vacuum. The remaining yellow oil was purified by columnchromatography (40 g silica gel, 2.3 cm column diameter, cyclohexane/EtOAc mixture).

(3aR,4R,6aS)-5-Benzyl-tetrahydro-2,2-dimethyl-4-(phenylselenylmethyl)-3aH-[1,3]dioxolo[4,5-c]pyrrole [(R)-20a] and (3aR,4S,6aS)-5-Benzyl-tetrahydro-2,2-dimethyl-4-(phenylselenylmethyl)-3aH-[1,3]dioxolo[4,5-c]pyrrole [(S)-20a]: Starting material 18a (0.52 g, 2.1 mmol). Diastereomers were not separated. Yield: 0.36 g (0.89 mmol, 42%), dr = 20.80, colourless oil, $R_f = 0.26$ and 0.31 (hexane/EtOAc, 8:2). C21H25NO2Se (402.389): calcd. C 62.68, H 6.26, N 3.48; found C 62.69, H 6.43, N 3.43. Major isomer (R)-**20a**: ¹H NMR (300 MHz, CDCl₃): δ = 1.33 (s, 3 H, CH₃), 1.55 (s, 3 H, CH₃), 2.06 (dd, J = 4.7 Hz, J = 11.1 Hz, 1 H, CHH), 2.44-2.48 (m, 1 H, CHN), 3.06 (d, J = 10.9 Hz, 1 H, CHH), 3.16 (d, J = 14.0 Hz, 1 H, CH*H*–Ph), 3.17 (d, J = 9.4 Hz, 2 H, CH₂Se), 3.97 (d, J = 13.6 Hz, 1 H, CHH–Ph), 4.55 (dd, J = 4.5 Hz, J = 6.4 Hz, 1 H, CH–CH–O), 4.74 (dd, J = 4.9 Hz, J = 6.4 Hz, 1 H, CH₂CH– O), 7.23–7.27 (m, 8 H, CH_{ar}), 7.52–7.55 (m, 2 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 25.3 (CH₂Se), 25.7 (CH₃), 26.5 (CH₃), 57.0 (CH₂Ph), 59.7 (CH₂N), 67.9 (CHN), 77.6 (CHCH₂), 81.1 (CHCH), 111.5 [C(CH₃)₂], 127.0 (CH_{ar}), 127.1 (CH_{ar}), 128.4 (2 CH_{ar}), 128.7 (2 CH_{ar}), 129.2 (2 CH_{ar}), 130.9 (C_{ar}Se), 132.8 (2 CH_{ar}), 138.2 (C_{ar}) ppm. Minor isomer (S)-20a: ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3)$: $\delta = 1.31$ (s, 3 H, CH₃), 1.52 (s, 3 H, CH₃), 2.51 (dd, J = 4.1 Hz, J = 10.5 Hz, 1 H, CHH), 3.05-3.16 (m, 4 H), 3.42(d, J = 12.9 Hz, 1 H, CHH-Ph), 4.01 (d, J = 12.8 Hz, 1 H, CHH-Ph)Ph), 4.59–4.70 (m, 2 H, CH–O), 7.23–7.27 (m, 8 H, CH_{ar}), 7.44– 7.48 (m, 2 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 25.3 (CH₃), 27.4 (CH₃), 29.2 (CH₂Se), 57.1 (CH₂Ph), 58.2 (CH₂N), 67.7 (CHN), 78.0 (CHCH₂), 84.3 (CHCH), 113.0 [C(CH₃)₂], 126.8 (CH_{ar}), 127.3 (CH_{ar}), 128.4 (2 CH_{ar}), 128.8 (2 CH_{ar}), 129.2 (2 CH_{ar}), 130.9 (C_{ar}Se), 132.4 (2 CH_{ar}), 138.4 (C_{ar}) ppm.

(3a*R*,4*R*,6a*S*)-Tetrahydro-2,2-dimethyl-5-[(*R*)-1-phenylethyl]-4-(phenylselenylmethyl)-3a*H*-[1,3]dioxolo[4,5-*c*]pyrrole [(*R*)-20d] and (3a*R*,4*S*,6a*S*)-Tetrahydro-2,2-dimethyl-5-[(*R*)-1-phenylethyl]-4-(phenylselenylmethyl)-3a*H*-[1,3]dioxolo[4,5-*c*]pyrrole [(*S*)-20d]: Starting material 18b (0.52 g, 2.0 mmol) and phenylselenyl bromide (0.65 g, 2.7 mmol). Diastereomers were not separated. Yield: 0.42 g (1.01 mmol, 50%), *dr* = 16:84, white solid, m.p. 117–120 °C, *R*_f = 0.27 (cyclohexane/EtOAc, 9:1). C₂₂H₂₇NO₂Se (416.415): calcd. C 63.45, H 6.54, N 3.36; found C 63.28, H 6.43, N 3.37. Major isomer (*R*)-20d: ¹H NMR (300 MHz, CDCl₃): δ = 1.32 (s, 3 H, *CH*₃), 1.43 (d, *J* = 7.2 Hz, 3 H, CH–CH₃), 1.54 (s, 3 H, *CH*₃), 2.16 (dd, *J* = 4.7 Hz, *J* = 10.3 Hz, 1 H, *CH*H), 2.32–2.38 (m, 1 H, *CH*N), 3.12 (d, *J* = 10.9 Hz, 2 H, CH*H*, CH*HS*e), 3.25 (dd, *J* = 3.4 Hz, *J* =

12.0 Hz, 1 H, CHHSe), 3.99 (q, J = 7.2 Hz, 1 H, CH-Ph), 4.44 $(dd, J = 4.5 Hz, J = 6.4 Hz, 1 H, CH_2CH_0), 4.62 (dd, J = 4.9 Hz)$ J = 6.4 Hz, 1 H, CH–CH–O), 6.89–7.59 (m, 10 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 19.0 (CH–CH₃), 25.9 (C_q–CH₃), 26.1 (CH₂Se), 26.5 (C_q-CH₃), 52.4 (CH₂N), 55.3 (CH-Ph), 63.5 (CHN), 77.1 (CHCH₂), 81.0 (CHCH), 111.5 [C(CH₃)₂], 127.1 (CH_{ar}), 127.3 (CH_{ar}), 128.0 (2 CH_{ar}), 128.5 (2 CH_{ar}), 129.1 (2 CH_{ar}), 130.9 (C_{ar}Se), 134.0 (2 CH_{ar}), 137.9 (C_{ar}) ppm. Minor isomer (S)-20d: ¹H NMR (300 MHz, CDCl₃): $\delta = 1.24$ (d, J = 6.4 Hz, 3 H, CH-CH₃), 1.31 (s, 3 H, CH₃), 1.55 (s, 3 H, CH₃), 2.84 (dd, J = 9.4 Hz, J = 12.0 Hz, 1 H, CHHSe), 3.04–3.08 (m, 3 H, CHHSe, CH_2), 3.41 (dd, J = 2.3 Hz, J = 8.7 Hz, 1 H, CHN), 3.88 (q, J =6.5 Hz, 1 H, CH-Ph), 4.54-4.63 (m, 1 H, CH₂CH-O), 4.66 (dd, J $= 1.9 \text{ Hz}, J = 6.4 \text{ Hz}, 1 \text{ H}, \text{CH-CH-O}, 6.89-7.59 \text{ (m, 10 H, CH_{ar})}$ ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 18.5 (CH–CH₃), 25.4 (C_q– CH₃), 27.1 (C_q-CH₃), 27.2 (CH₂Se), 53.8 (CH₂N), 57.8 (CH-Ph), 65.0 (CHN), 78.9 (CHCH₂), 84.4 (CHCH), 112.1 [C(CH₃)₂], 127.1 (3 CHar), 127.4 (3 CHar), 128.4 (2 CHar), 129.2 (2 CHar), 133.1 $(C_{ar}Se)$, 144.8 (C_{ar}) ppm.

X-ray Crystal Analysis of 20d:^[79] Empirical formula: C₂₂H₂₇NO₂Se, formula weight: 416.41 g/mol, temperature: 150(2) K, wavelength: 0.71073 Å, crystal system: orthorhombic, space group: P 21 21 21, unit cell dimensions: a = 9.568(4) Å, b = 9.999(3) Å, c =20.383(3) Å, V = 1950.1(12) Å³, Z = 4, density (calculated): 1.418 Mg/m³, absorption coefficient: 1.941 mm⁻¹, F(000): 864, crystal size: $0.80 \times 0.80 \times 0.60$ mm, Theta range for data collection: 2.00 to 27.48 deg, limiting indices: $-12 \le h \le 12, -12 \le k \le 12$, $-26 \le l \le 26$; reflections collected: 12622, unique: 4461 [R(int.) = 0.0234], completeness to $\theta = 27.48$ is 99.9%, absorption correction: empirical (psi-scan), max./min. transmission: 0.3888/0.3057, refinement method: full-matrix least-squares on F^2 data/restraints/parameters: 4461/0/344, goodness-of-fit on F²: 1.067, final R indices $[I > 2\sigma(I)]$: $R_1 = 0.0217$, $wR_2 = 0.0440$, R indices (all data): R_1 = 0.0283, wR_2 = 0.0467, absolute structure parameter: -0.011(6), extinction coefficient: 0.0020(3), largest diff. peak and hole: 0.240 and -0.319 e/Å³.

Methyl (1S)-2-Phenyl-2-{(3aR,4R,6aS)-tetrahydro-2,2-dimethyl-4-(phenylselenylmethyl)[1,3]dioxolo[4,5-c]pyrrol-5-yl}acetate [(R)-20f] and Methyl (1S)-2-Phenyl-2-{(3aR,4S,6aS)-tetrahydro-2,2-dimethyl-4-(phenylselenylmethyl)[1,3]dioxolo[4,5-c]pyrrol-5-yl}acetate [(S)-20f]: Starting material 18d (0.50 g, 1.6 mmol) and phenylselenyl bromide (0.65 g, 2.7 mmol). Diastereomers were separated, dr =19:81. Major isomer (R)-20f: yield 0.14 g (0.30 mmol, 18%), colourless oil, $R_{\rm f} = 0.12$ (cyclohexane/EtOAc, 9:1). $[a]_{\rm D}^{23} = 201.8$ (c = 0.82, CHCl₃). ¹H NMR (300 MHz, CDCl₃): δ = 1.34 (s, 3 H, CH_3), 1.58 (s, 3 H, CH_3), 2.17 (dd, J = 4.9 Hz, J = 10.9 Hz, 1 H, N–CHH), 2.57 (ddd, J = 4.1 Hz, J = 4.4 Hz, J = 10.9 Hz, 1 H, CHN), 3.03 (dd, J = 3.8 Hz, J = 11.9 Hz, 1 H, CHHSe), 3.12 (dd, J = 10.8 Hz, J = 11.9 Hz, 1 H, CHHSe), 3.35 (d, J = 10.9 Hz, 1 H, N–CHH), 3.66 (s, 3 H, COOCH₃), 4.48 (dd, J = 4.9 Hz, J =6.4 Hz, 1 H, CH₂CH-O), 4.50 (s, 1 H, CH-Ph), 4.70 (dd, J = 4.9 Hz, J = 6.4 Hz, 1 H, CH–CH–O), 6.98–7.02 (m, 2 H, CH_{ar}), 7.18–7.32 (m, 6 H, CH_{ar}), 7.55–7.58 (m, 2 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 25.8 (CH₂Se), 26.1 (C_q-CH₃), 26.5 (C_q-CH₃), 52.0 (COOCH₃), 55.4 (CH₂N), 64.9 (CHN), 65.7 (CH-Ph), 77.1 (CHCH₂), 80.5 (CHCH), 111.8 [(C(CH₃)₂], 127.4 (CH_{ar}), 128.2 (CH_{ar}), 128.4 (2 CH_{ar}), 129.2 (2 CH_{ar}), 129.4 (2 CH_{ar}), 130.6 (CarSe), 133.0 (Car), 133.7 (2 CHar), 171.9 (COO) ppm. HRMS (EI): calcd. for C₂₃H₂₇O₄NSe 461.11053; found 461.11055. Minor isomer (S)-20f: yield of 33 mg (0.072 mmol, 4%), colourless oil, $R_{\rm f}$ = 0.08 (cyclohexane/EtOAc, 9:1). $[a]_{D}^{23} = -31.7$ (c = 1.16, CHCl₃). ¹H NMR (300 MHz, CDCl₃): δ = 1.32 (s, 3 H, CH₃), 1.59 (s, 3 H, CH_3), 2.98–3.04 (m, 3 H, CH_2 Se, CHH), 3.09 (dd, J = 2.9 Hz, J =

11.5 Hz, 1 H, N–C*H*H), 3.38 (ddd, J = 2.3 Hz, J = 3.6 Hz, J = 8.2 Hz, 1 H, C*H*N), 3.64 (s, 3 H, COOC*H*₃), 4.62 (ddd, J = 3.0 Hz, J = 5.1 Hz, J = 6.4 Hz, 1 H, CH₂C*H*–O), 4.68 (dd, J = 2.3 Hz, J = 6.4 Hz, 1 H, CH–C*H*–O), 4.80 (s, 1 H, C*H*–Ph), 7.19–7.42 (m, 10 H, C*H*_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): $\delta = 25.2$ (C_q–CH₃), 27.3 (C_q–CH₃), 28.7 (CH₂Se), 52.0 (COOCH₃), 54.6 (CH₂N), 65.3 (CH–Ph), 67.0 (CHN), 79.2 (CHCH₂), 84.7 (CHCH), 112.5 [C(CH₃)₂], 126.9 (CH_{ar}), 128.4 (CH_{ar}), 128.7 (4 CH_{ar}), 129.2 (2 CH_{ar}), 132.5 (2 CH_{ar}), 133.8 (C_{ar}Se), 136.7 (C_{ar}), 171.8 (COO) ppm. HRMS (EI): calcd. for C₂₃H₂₇O₄NSe 461.11053; found 461.11055.

General Procedure for the Selenocyclisation and Reduction to Pyrrolidines 21a, 21c and 21d: The appropriate N(alkenylidene)alk-ylamine 9 (1.64 mmol) was dissolved in dry CH₂Cl₂ (15 mL) at -78 °C under argon in a flame-dried flask. A solution of camphorselenyl bromide 22 (0.96 mmol), generated in situ, in dry CH₂Cl₂ (30 mL) was added over 30 min. After 2 h at -78 °C, MeOH (5 mL) and NaBH₄ (73 mg, 1.93 mmol) were added. The mixture was warmed slowly to room temperature. The organic layer was separated and washed with water (10 mL). After separating the layers, the aqueous layer was washed with CH₂Cl₂ (2 × 5 mL). The combined organic layers were dried with MgSO₄ and concentrated under vacuum. The remaining yellow oil was purified by column chromatography (40 g of silica gel, 2.3 cm column diameter, cyclohexane/EtOAc mixture).

(4aR,4bR,8aS,9aS)-9-{2-[(2R)-N-Benzyl-4,4-dimethylpyrrolidin-2ylselenylmethyl|benzyl}-3,6-diphenyl-hexahydro-2,4,5,7-tetraoxa-9azafluoren and (4aR,4bR,8aS,9aS)-9-{2-[(2S)-N-Benzyl-4,4-dimethylpyrrolidin-2-ylselenylmethyl]benzyl}-3,6-diphenyl-hexahydro-2,4,5,7-tetraoxa-9-azafluoren (21b): The diselenide (0.32 g, 0.32 mmol) was dissolved in dry CH2Cl2 (10 mL) at -78 °C under argon in a flame-dried flask. A 0.1 M solution of bromine (0.32 mmol) in dry CH₂Cl₂ was added over 5 min to generate 23. After 20 min at -78 °C, AgPF₆ (0.19 g, 0.75 mmol) was added. The mixture was stirred for 20 min and cooled to -100 °C. N-Benzyl(2,2-dimethylpent-4-enylidene)amine (9a) (0.36 g, 1.79 mmol) in dry CH₂Cl₂ (5 mL) was added. The mixture was warmed slowly to room temperature. MeOH (5 mL) and sodium borohydride (73 mg, 1.93 mmol) were added, and the mixture was stirred for 1 h. The organic layer was separated and washed with water (10 mL). After separating the layers, the aqueous layer was washed with CH₂Cl₂ $(2 \times 5 \text{ mL})$. The combined organic layers were dried with MgSO₄ and concentrated under vacuum. The remaining oil was purified by column chromatography (40 g silica gel, 2.3 cm column diameter, cyclohexane/EtOAc, 9:1), yielding 0.33 g (0.46 mmol, 72% yield) of a colourless oil. Diastereomers were not separated, dr = 48:52, $R_{\rm f} =$ 0.53 (hexane/EtOAc, 9:1). HRMS (ESI): calcd. for C₄₁H₄₇O₂N₂Se 711.27010; found 711.27013. Major isomer: ¹H NMR (300 MHz, CDCl₃): $\delta = 0.97$ (s, 3 H, CH₃), 1.10 (s, 3 H, CH₃), 1.70 (dd, J =8.3 Hz, J = 9.7 Hz, 1 H, C_q–CH*H*–CH), 1.80 (dd, J = 3.9 Hz, J =8.3 Hz, 1 H, C_q -CH*H*-CH), 1.99 (d, J = 8.8 Hz, 1 H, C_q -CH*H*-N), 2.64 (d, J = 9.0 Hz, 1 H, C_q-CH*H*-N), 2.84–2.92 (m, 1 H, C*H*-CH2-Se), 3.05-3.18 (m, 3 H, Se-CH2, CH), 3.53 (s, 2 H, CH2-Ar), 3.90-4.17 (m, 6 H, 2 CH2-O, 2 CH), 4.40 (s, 2 H, CH2-Ph), 4.75 (d, J = 15.8 Hz, 1 H, CH), 5.49 (s, 2 H, 2O–CH–O), 7.11–7.15 (m, 1 H, CH_{ar}), 7.19–7.54 (m, 17 H, CH_{ar}), 7.64 (d, J = 7.2 Hz, 1 H, SeC_{ar}-CH_{ar}) ppm. ¹³C NMR (75.5 MHz, CDCl₃): δ = 28.9 (CH₃), 30.2 (CH₃), 32.6 (CH₂Se), 35.9 [C(CH₃)₂], 46.7 (CH-CH₂-C_q), 53.1 (CH₂-Ph), 53.8 (CH₂-Ar), 58.3 (2 NCH-CH₂O), 63.6 (CH-CH₂Se), 67.1 (2 CH₂-O), 68.3 [N-CH₂-C(CH₃)₂], 79.7 (2 CH-O), 99.9 (2 Ph-CH-O), 126.5 (4 CHar), 126.7 (CHar), 126.8 (CHar), 127.4 (CHar), 128.2 (4 CHar), 128.5 (4 CHar), 128.8 (CHar), 129.2 $(2 CH_{ar}), 131.3 (SeC_{ar}), 131.8 (CH_{ar}), 138.5 (2 CH-C_{ar}), 139.9$

 (CH_2-C_{ar}) , 142.5 (CH_2-C_{ar}) ppm. Minor isomer: ¹H NMR $(300 \text{ MHz}, \text{ CDCl}_3): \delta = 0.93 \text{ (s, 3 H, CH}_3), 1.08, (s, 3 H, CH}_3),$ 1.66 (dd, J = 8.1 Hz, J = 10.0 Hz, 1 H, C_a-CH*H*-CH), 1.84 (dd, J = 3.7 Hz, J = 7.9 Hz, 1 H, C_q–CH*H*–CH), 1.99 (d, J = 8.8 Hz, 1 H, C_q-CH*H*-N), 2.64 (d, J = 9.0 Hz, 1 H, C_q-CH*H*-N), 2.84-2.92 (m, 1 H, CH-CH₂-Se), 3.05-3.18 (m, 3 H, Se-CH₂, CH), 3.53 (s, 2 H, CH₂-Ar), 3.90-4.17 (m, 6 H, 2 CH₂-O, 2 CH), 4.40 (s, 2 H, CH_2 -Ph), 4.77 (d, J = 16.2 Hz, 1 H, CH), 5.50 (s, 2 H, 2O-CH-O), 7.11-7.15 (m, 1 H, CH_{ar}), 7.19-7.54 (m, 17 H, CH_{ar}), 7.69 (d, J = 7.5 Hz, 1 H, SeC_{ar}-CH_{ar}) ppm. ¹³C NMR (75.5 MHz, $CDCl_3$): $\delta = 28.8 (CH_3), 30.2 (CH_3), 32.5 (CH_2Se), 35.9 [C(CH_3)_2],$ 46.6 (CH-CH₂-C_a), 53.1 (CH₂-Ph), 53.8 (CH₂-Ar), 58.6 (2 NCH-CH₂O), 63.7 (CH-CH₂Se), 67.0 (2 CH₂-O), 68.3 [N-CH₂-C-(CH₃)₂], 79.7 (2 CH–O), 99.9 (2 Ph–CH–O), 126.5 (4 CH_{ar}), 126.6 (CH_{ar}), 126.8 (CH_{ar}), 127.3 (CH_{ar}), 128.2 (4 CH_{ar}), 128.5 (4 CH_{ar}), 128.7 (CHar), 129.2 (2 CHar), 131.6 (SeCar), 131.8 (CHar), 138.5 (2 CH-C_{ar}), 139.9 (CH₂-C_{ar}), 142.6 (CH₂-C_{ar}) ppm.

Methyl (*R*)-2-{(2*R*)-4,4-Dimethyl-2-[(1*R*,2*S*,4*R*)-4,7,7-trimethyl-3oxobicyclo[2.2.1]hept-2-yl-selenylmethyl]pyrrolidin-1-yl}-2-phenylacetate and Methyl (R)-2-{(2S)-4,4-Dimethyl-2-[(1R,2S,4R)-4,7,7trimethyl-3-oxobicyclo[2.2.1]hept-2-yl-selenylmethyl]pyrrolidin-1yl}-2-phenylacetate (21c): Starting material 9j (270 mg, 1.04 mmol). Diastereomers were isolated as a mixture of rotamers. dr = 38:62. Major isomer: yield of 122 mg (0.25 mmol, 29%), colourless oil, $R_{\rm f}$ = 0.21 (hexane/EtOAc, 9:1). $[a]_{D}^{22} = -7.4$ (c = 1.00, CHCl₃). ¹H NMR (300 MHz, CDCl₃): $\delta = 0.75, 0.77$ (s, 3 H, CH₃), 0.82 (s, 3 H, CH₃), 0.84, 085 (s, 3 H, CH₃), 0.94 (s, 3 H, CH₃), 1.03, 1.04 (s, 3 H, CH₃), 1.32-1.63 (m, 4 H, 2 CH₂), 1.67-1.74 (m, 3 H, CHH-N, CH₂), 1.99–2.04 (m, 1 H, SeCH₂–CH–N), 2.19, 2.21 (d, J =8.6 Hz, 1 H, CHH-N), 2.72-2.83 (m, 2 H, CHH-Se, CH-CHSe), 2.86-2.92 (m, 1 H, CHH-Se), 2.94-3.15 (m, 1 H, CH-Se), 3.60, 3.61 (s, 3 H, COOCH₃), 4.51, 4.52 (s, 1 H, CH-COO), 7.24-7.29 (m, 5 H, CH_{ar}) ppm. ¹³C NMR (75.5 MHz, $CDCl_3$): $\delta = 9.9 (CH_3)$, 19.8 (2 CH₃), 23.4, 23.5 (CH₂), 28.4, 28.5 (CH₃), 29.1, 29.2 (CH₃), 29.5, 29.6 (CH₂Se), 30.5, 30.6 (CH₂), 36.1, 36.2 [C(CH₃)₂], 46.0, 46.2 (CH-CH2-Cq), 46.6, 46.7 (CH-Se), 46.9, 47.0 (Cq-CO), 48.4, 48.5 (CH-CHSe), 52.0, 52.1 (COOCH₃), 58.1, 58.2 (C_q-CO), 60.6, 60.9 (CH-Ph), 63.7, 63.8 (N-CH2-Cq), 67.6, 67.8 (CH-N), 128.2 (CH_{ar}), 128.4, 128.5 (2 CH_{ar}), 129.3, 129.4 (2 CH_{ar}), 135.2, 135.3 (CH-Car), 172.7, 172.8 (COO), 218.2, 218.3 (C=O) ppm. HRMS (ESI⁺) calcd. for $C_{26}H_{37}O_3NSeNa$ 514.1831; found 514.1831. Minor isomer: yield of 74 mg (0.15 mmol, 17%). $R_f = 0.26$ (hexane/ EtOAc, 9:1). $[a]_{D}^{22} = 19.8 \ (c = 1.25, \text{ CHCl}_3)$. ¹H NMR (300 MHz, CDCl₃): $\delta = 0.80, 0.82$ (s, 3 H, CH₃), 0.84 (s, 3 H, CH₃), 0.94, 0.95 (s, 3 H, CH₃), 0.96 (s, 3 H, CH₃), 0.97, 0.98 (s, 3 H, CH₃), 1.32-1.41 (m, 1 H, CHH), 1.52–1.64 (m, 3 H, 3 CHH), 1.71–1.80 (m, 3 H, 2 CHH, CHH-N), 2.01-2.11 (m, 1 H, CHH-N), 2.41-2.49 (m, 1 H, SeCH₂-CH-N), 2.64-2.70 (m, 1 H, CH-CHSe), 2.78-2.81 (m, 1 H, CHH–Se), 3.17-3.29 (m, 1 H, CHH–Se), 3.45, 3.51 (d, J =4.9 Hz, 1 H, CH-Se), 3.65, 3.66 (s, 3 H, COOCH₃), 4.65, 4.70 (s, 1 H, CH–COO), 7.19–7.36 (m, 5 H, CH_{ar}) ppm. ¹³C NMR $(75.5 \text{ MHz}, \text{CDCl}_3): \delta = 9.8 (CH_3), 19.8 (2 CH_3), 23.4, 23.5 (CH_2),$ 28.7, 28.8 (CH₃), 29.4 (CH₃), 30.0 (CH₂Se), 30.5, 30.6 (CH₂), 36.0 $[C(CH_3)_2]$, 46.3, 46.4 (CH-CH₂-C_q), 46.6, 47.7 (CH-Se), 46.9 (C_q-CO), 48.4, 48.5 (CH–CHSe), 51.5, 51.6 (COOCH₃), 58.1, 58.2 (C_q– CO), 60.6, 60.9 (CH-Ph), 62.3, 62.8 (N-CH₂-C_q), 65.7, 66.4 (CH-N), 127.8, 127.9 (CH_{ar}), 128.4 (2 CH_{ar}), 128.6, 128.7 (2 CH_{ar}), 137.4 (CH-Car), 172.2 (COO), 218.3, 218.4 (C=O) ppm. HRMS (ESI⁺): calcd. for C₂₆H₃₈O₃NSe 492.2011; found 492.2011.

Supporting Information (see also the footnote on the first page of this article): Experimental and spectroscopic data of imines 9, 10c, 10d,10c, 14c, 14d, 14e, 15, 18, 20c, 20e, 21a, and 21d.

Acknowledgments

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- H. Takahata, Y. Banba, M. Tajima, T. Momose, J. Org. Chem. 1991, 56, 240–245.
- [2] L. A. Wessjohann, U. Sinks, J. Prakt. Chem. 1998, 340, 189– 203.
- [3] T. Wirth, Top. Curr. Chem. 2000, 208, 1-5.
- [4] N. Petragnani, H. A. Stefani, C. J. Valduga, *Tetrahedron* 2001, 57, 1411–1448.
- [5] M. A. Cooper, A. D. Ward, Austr. J. Chem. 1997, 50, 181-187.
- [6] G. Cardillo, M. Orena, *Tetrahedron* **1990**, *46*, 3321–3407.
- [7] J. E. Balwin, J. Chem. Soc., Chem. Commun. 1976, 734-736.
- [8] X.-F. Ren, E. Turos, C. H. Lake, M. R. Churchill, J. Org. Chem. 1995, 60, 6468–6483.
- [9] E. Turos, M. Parvez, R. S. Garigipati, S. M. Weinreb, J. Org. Chem. 1988, 53, 1116–1118.
- [10] R. Guzman Bernett, J. T. Doi, W. K. Musker, J. Org. Chem. 1985, 50, 2048–2050.
- [11] X.-F. Ren, E. Turos, Tetrahedron Lett. 1993, 34, 1575–1578.
- [12] G. Jana, A. Viso, Y. Díaz, S. Castillón, Eur. J. Org. Chem. 2003, 209–216.
- [13] K. C. Nicolaou, N. A. Petasis, D. A. Claremon, in Organoselenium Chemistry (Ed.: D. Liotta), John Wiley & Sons, 1987, pp. 127–162.
- [14] D. L. J. Clive, V. Farina, A. Singh, C. K. Wong, W. A. Kiel, S. M. Menchen, J. Org. Chem. 1980, 45, 2120–2126.
- [15] R. R. Webb, S. Danishefsky, *Tetrahedron Lett.* 1983, 24, 1357– 1360.
- [16] A. Toshimitsu, K. Terao, S. Uemura, J. Org. Chem. 1986, 51, 1724–1792.
- [17] C. G. Francisco, E. I. León, J. A. Salazar, E. Suárez, *Tetrahe*dron Lett. **1986**, 27, 2513–2516.
- [18] C. Betancor, E. I. León, T. Prange, J. A. Salazar, E. Suárez, J. Chem. Soc., Chem. Commun. 1989, 450–452.
- [19] R. Freire, E. I. León, J. A. Salazar, E. Suárez, J. Chem. Soc., Chem. Commun. 1989, 452–454.
- [20] M. A. Cooper, A. D. Ward, *Tetrahedron Lett.* 1992, 33, 5999– 6002.
- [21] W. Hümmer, E. Dubois, T. Gracza, V. Jäger, *Synthesis* **1997**, 634–642.
- [22] D. N. A. Fox, D. Lathbury, M. F. Mahon, K. C. Molloy, T. Gallgher, J. Chem. Soc., Chem. Commun. 1989, 1073–1075.
- [23] M. Tiecco, L. Testaferri, F. Marini, C. Santi, L. Bagnoli, A. Temperini, *Tetrahedron* 1997, 53, 10591–10602.
- [24] M. Gulla, L. Bierer, S. Schmidt, L. Redcliffe, V. Jäger, Z. Naturforsch., Teil B 2006, 61, 471–485.
- [25] R. Grigg, M. Hadjisoteriou, P. Kennewell, J. Markandu, J. Chem. Soc., Chem. Commun. 1992, 1537–1538.
- [26] H. A. Dondas, R. Grigg, J. Markandu, T. Perrior, T. Suzuki, S. Thibault, W. A. Thomas, M. Thornton-Pett, *Tetrahedron* 2002, 58, 161–174.
- [27] R. Grigg, J. Markandu, T. Perrior, Z. Qiong, T. Suzuki, J. Chem. Soc., Chem. Commun. 1994, 1267–1268.
- [28] A. Hall, K. P. Meldrum, P. R. Therond, R. H. Wightman, Synlett 1997, 123–125.
- [29] R. Grigg, M. Hadjisoteriou, P. Kennewell, J. Markandu, M. Thornton-Pett, J. Chem. Soc., Chem. Commun. 1993, 1340– 1342.
- [30] V. Jäger, L. Bierer, H.-Q. Dong, A. M. Palmer, D. Shaw, W. Frey, J. Heterocycl. Chem. 2000, 37, 455–465.
- [31] M. Gulla, L. Bierer, L. Redcliffe, S. Schmidt, V. Jäger, ARKI-VOC 2006, 76–88.
- [32] M. Tiecco, L. Testaferri, F. Marini, C. Santi, L. Bagnoli, A. Temperini, *Tetrahedron* 1997, 53, 7311–7318.

- [33] N. De Kimpe, M. Boelens, J. Chem. Soc., Chem. Commun. 1993, 916–918.
- [34] N. De Kimpe, M. Boelens, J. Piqueur, J. Baele, *Tetrahedron Lett.* 1994, 35, 1925–1928.
- [35] D. De Smaele, N. De Kimpe, J. Chem. Soc., Chem. Commun. 1995, 2029–2030.
- [36] N. De Kimpe, M. Boelens, J. Contreras, *Tetrahedron Lett.* 1996, 37, 3171–3174.
- [37] C. V. Stevens, M. Peristeropoulou, N. De Kimpe, *Tetrahedron* 2001, 57, 7865–7870.
- [38] K. Terao, A. Toshimitsu, S. Uemura, J. Chem. Soc., Perkin Trans. 1 1986, 1837–1844.
- [39] H. Takahata, K. Yamazaki, T. Takmatsu, T. Yamazaki, T. Momose, J. Org. Chem. 1990, 55, 3947–3950.
- [40] H. Fujioka, H. Kitagawa, Y. Nagatomi, Y. Kita, J. Org. Chem. 1996, 61, 7309–7315.
- [41] Z. Wu, D. R. Mootoo, B. Fraser-Reid, *Tetrahedron Lett.* 1988, 29, 6549–6552.
- [42] H. Fujioka, H. Kitagawa, N. Matsunaga, Y. Nagatomi, Y. Kita, *Tetrahedron Lett.* 1996, 37, 2245–2248.
- [43] J. M. Takacs, M. A. Helle, *Tetrahedron Lett.* 1989, 30, 7321– 7324.
- [44] J. M. Takacs, M. A. Helle, B. J. Sanyal, T. A. Eberspacher, *Tetrahedron Lett.* 1990, 31, 6765–6768.
- [45] D. A. Berges, J. Fan, L. Nannan, N. K. Dalley, *Tetrahedron* 2001, 57, 9915–9924.
- [46] S. Robin, G. Rousseau, Tetrahedron 1998, 54, 13681–13736.
- [47] A. Toshimitsu, K. Terao, S. Uemura, J. Org. Chem. 1987, 52, 2018–2026.
- [48] H. Takahata, T. Takamatsu, T. Yamazaki, J. Org. Chem. 1989, 54, 4812–4822.
- [49] S. Kano, T. Yokomatsu, H. Iwasawa, S. Shibuya, *Heterocycles* 1987, 26, 359–362.
- [50] K. A. Tehrani, K. Van Syngel, M. Boelens, J. Contreras, N. De Kimpe, D. W. Knight, *Tetrahedron Lett.* 2000, 41, 2507– 2510.
- [51] E. R. Alonoso, K. A. Tehrani, M. Boelens, D. W. Knight, V. Yu, N. De Kimpe, *Tetrahedron Lett.* 2001, *42*, 3921–3923.
- [52] B. H. Lipshutz, T. Gross, J. Org. Chem. 1995, 60, 3572-3573.
- [53] H. Akita, K. Uchida, C. Y. Chen, K. Kato, *Chem. Pharm. Bull.* 1998, 46, 1034–1038.
- [54] L. A. Paquette, S. Bailey, J. Org. Chem. 1995, 60, 7849-7856.
- [55] A. M. Palmer, V. Jäger, Eur. J. Org. Chem. 2001, 1293-1308.
- [56] T. Wirth, Angew. Chem. 2000, 112, 3890-3900.
- [57] T. Wirth, Tetrahedron 1999, 55, 1-28.
- [58] M. Tiecco, L. Testaferri, C. Santi, C. Tomassini, F. Marini, L. Bagnoli, A. Temperini, *Eur. J. Org. Chem.* 2000, 20, 3451–3457.
- [59] M. Tiecco, L. Testaferri, C. Santi, F. Marini, L. Bagnoli, A. Temperini, *Tetrahedron: Asymmetry* 1999, 10, 747–757.
- [60] M. Tiecco, L. Testaferri, C. Santi, F. Marini, L. Bagnoli, A. Temperini, *Tetrahedron Lett.* 1998, 39, 2809–2812.
- [61] M. Tiecco, L. Testaferri, C. Santi, F. Marini, L. Bagnoli, A. Temperini, C. Tomassini, *Eur. J. Org. Chem.* 1998, 11, 2275– 2277.
- [62] T. G. Back, B. P. Dyck, S. Nan, *Tetrahedron* **1999**, *55*, 3191–3208.
- [63] T. G. Back, S. Nan, J. Chem. Soc., Perkin Trans. 1 1998, 3123– 3124.
- [64] T. G. Back, B. P. Dyck, Chem. Commun. 1996, 2567-2568.
- [65] T. G. Back, B. P. Dyck, J. Org. Chem. 1995, 60, 703-710.
- [66] D. Liotta, G. Zima, C. Barnum, M. Saindane, *Tetrahedron Lett.* 1980, 21, 3643–3646.
- [67] T. G. Back, B. P. Dyck, M. Parvez, J. Chem. Soc., Chem. Commun. 1994, 515–516.
- [68] K.-i. Fujita, K. Murata, M. Iwaoka, S. Tomoda, J. Chem. Soc., Chem. Commun. 1995, 1641–1642.
- [69] K.-i. Fujita, K. Murata, M. Iwaoka, S. Tomoda, *Tetrahedron* 1997, 53, 2029–2048.
- [70] K.-i. Fujita, K. Murata, M. Iwaoka, S. Tomoda, *Tetrahedron Lett.* 1995, 36, 5219–5222.

- [71] K.-i. Fujita, M. Iwaoka, S. Tomoda, Chem. Lett. 1994, 23, 923–926.
- [72] N. Baggett, P. Stribblehill, J. Chem. Soc., Perkin Trans. 1 1977, 1123–1126.
- [73] Y. Masaki, H. Oda, K. Kazuta, A. Usui, A. Itoh, F. Xu, *Tetra*hedron Lett. **1992**, 33, 5089–5092.
- [74] S.-H. Moon, S. Lee, Synth. Commun. 1998, 28, 3919-3926.
- [75] A. Duréault, M. Portal, J.-C. Depezay, Synlett 1991, 225–226.
- [76] M. Iwaoka, S. Tomoda, *Phosphorus Sulfur Silicon Relat. Elem.* 1992, 67, 125–130.
- [77] T. K. M. Shing, *Tetrahedron* 1988, 44, 7261–7264.
- [78] S. Masamune, W. Choy, J. S. Peterson, L. R. Sita, Angew. Chem. Int. Ed. 1985, 24, 1–30.
- [79] CCDC-631454 (for 10e), -631455 [for 14f(S)] and -631556 (for 20d) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc. cam.ac.uk/data_request/cif.

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