

# FLUORIDE ION INDUCED REACTIONS OF ORGANOSILANES WITH ELECTROPHILES

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Summary: Benzyl- 4-picolyl- and phenylallyl(trimethyl)silanes react with electrophiles in the presence of KF/18-crown-6 or silica-TBAF under mild conditions.

Fluoride ion is known to display a marked nucleophilic affinity for silicon in organosilanes<sup>1)</sup> due to the high Si-F bond energy, and this fact has been used to cleave X-Si bonds<sup>1)</sup>. The proposed intermediacy of functionalized carbanions in F<sup>-</sup> induced reactions, already reported for acyl<sup>2)</sup>, alkynyl<sup>3)</sup>, allyl<sup>4)</sup> and oxiranyl<sup>5)</sup> silanes, supports the contention that suitable organosilicon compounds containing a C-SiR<sub>3</sub> moiety can be considered as masked carbanions, which can be generated under relatively mild conditions, provided a suitable catalyst is present.

In this Letter, we describe the transfer of benzyl-, 4-picolyl- and phenylallyl- groups from the corresponding labile organosilanes to various weak electrophiles in the presence of KF/18-crown-6 or silica-supported tetrabutylammonium fluoride (silica-TBAF). Details are given in the Table. Reactions are rapid and high-yielding.

Table. F<sup>-</sup> catalysed reactions of organosilanes with electrophiles<sup>a)</sup>

R in R-SiMe <sub>3</sub>	Electrophile	Product <sup>b)</sup>	Yield (%)
o-O <sub>2</sub> NC <sub>6</sub> H <sub>4</sub> CH <sub>2</sub>	PhCHO	o-O <sub>2</sub> NC <sub>6</sub> H <sub>4</sub> CH <sub>2</sub> CH(OH)Ph	70
	PhCH <sub>2</sub> Br	o-O <sub>2</sub> NC <sub>6</sub> H <sub>4</sub> CH <sub>2</sub> CH <sub>2</sub> Ph	60
3,5-Cl <sub>2</sub> C <sub>6</sub> H <sub>4</sub> CH <sub>2</sub>	PhCHO	3,5-Cl <sub>2</sub> C <sub>6</sub> H <sub>4</sub> CH <sub>2</sub> CH(OH)Ph	70
	PhCH <sub>2</sub> Br	3,5-Cl <sub>2</sub> C <sub>6</sub> H <sub>4</sub> CH <sub>2</sub> CH <sub>2</sub> Ph	50
4-Picolyl	PhCHO	4-picolylCH(OH)Ph	80
	PhCH <sub>2</sub> Br	4-picolylCH <sub>2</sub> CH <sub>2</sub> Ph	70
	MeCHCH <sub>2</sub> O	4-picolylCH <sub>2</sub> (OH)Me	30 <sup>c)</sup>
PhCH=CHCH <sub>2</sub>	PhCHO	PhCH=CHCH <sub>2</sub> CH(OH)Ph	50
	PhCH <sub>2</sub> Br	PhCH=CHCH <sub>2</sub> CH <sub>2</sub> Ph	40

a) THF, 12h, 20°, 10 mol% catalyst, under argon.

b) Accompanied by small amounts (<10%) of RH. Product obtained after treating the reaction mixture with dilute HCl.

c) Low yield due to competing F<sup>-</sup> catalysed opening of the oxirane ring<sup>6)</sup>.

Organosilanes corresponding to carbon acids, RH, with  $pK_a < 37$  (Ref 7) appear therefore to be effective agents for transferring the organic group, R, to an electrophilic centre. A recent report<sup>8)</sup> on the other hand suggests that even this  $pK_a$  value may not be limiting.

The method described above is of preparative advantage in those cases where more reactive organometallic reagents cannot be used, *e.g.* with nitro-substituted or with pyridine-type derivatives. Products resulting from allyl-type rearrangements (leading to *o*-substituted toluenes) commonly encountered in Grignard reactions are also absent.

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