This article was downloaded by: [McMaster University] On: 08 July 2013, At: 12:33 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

The Efficient Synthesis of N⁴-Substituted 1-Methylcytosines

Lech Celewicz^a, Krzysztof Ciszewski^a & Krzysztof Golankiewicz^a

^a Faculty of Chemistry, Adam Mickiewicz University, Grunwaldzka 6, 60-780, Poznan, Poland Published online: 24 Sep 2006.

To cite this article: Lech Celewicz , Krzysztof Ciszewski & Krzysztof Golankiewicz (1991) The Efficient Synthesis of N⁴-Substituted 1-Methylcytosines, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 21:14, 1489-1500, DOI: <u>10.1080/00397919108016423</u>

To link to this article: http://dx.doi.org/10.1080/00397919108016423

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages,

and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions

THE EFFICIENT SYNTHESIS OF N⁴-SUBSTITUTED 1-METHYLCYTOSINES

Lech Celewicz, Krzysztof Ciszewski and Krzysztof Golankiewicz

Faculty of Chemistry, Adam Mickiewicz University, Grunwaldzka 6, 60-780 Poznan, Poland

ABSTRACT: An improved procedure for the preparation of 4-chloro-1,2-dihydro-1-methyl-2-oxopyrimidine is described. This compound appeared to be an excellent substrate for the synthesis of N^4 -substituted 1-methylcytosines.

Cytosine derivatives have attracted considerable attention due to their various biological effects, especially, antiviral and anticancer properties of appropriate nucleosides.¹⁻⁵ Among the large number of known cytosine analogs, N⁴-substituted cytosine derivatives are of special interest as naturally occurring constituents of ribonucleic acids (RNA's).⁶ 1-Methylcytosine and its derivatives have proved to be useful model compounds in the

Io whom correspondence should be addressed.

studies of physicochemical properties and chemical reactions of the corresponding cytosine nucleosides.^{7,8} The preparation of N^4 -substituted 1-methylcytosines has been usually achieved via nucleophilic displacement of alkoxy or alkylthio substituents at the 4 position of appriopriate pyrimidine precursors with the use of amines.⁸⁻¹⁰ These methods however, require heating of reagents at high temperatures (often in sealed tubes or autoclaves), particularly, when aromatic amines are used as substrates.^{11,12} We have reported that N-(1H-2-oxo-4-pyrimidinyl)amino acids undergo photochemical decarboxylation and this process can be used for the facile synthesis of N⁴-substituted cytosines¹³ and N⁴-(1-deuterioalkyl)cytosines¹⁴ with substituents derived from amino acids.

In continuation of our studies of biologically active pyrimidines we present an improved method for preparation of 4-chloro-1,2-dihydro-1-methyl-2-oxopyrimidine 1 and an efficient method of synthesis of N^4 -substituted 1-methylcytosines, employing as the synthetic precursor 1 and aliphatic as well as aromatic amines. To our knowledge, little information concerning the reactions of 1-alkyluracils with phosphoryl halides has been published. 4-Chloro-1,2-dihydro-1-methyl-2-oxopyrimidine 1 was originally prepared by Todd and co-workers¹⁵ in 28% yield from 1-methyluracil and phosphoryl chloride heated under reflux. We have found that the reaction of 1-methyluracil with phosphoryl chloride under reflux gives three products: 4-chloro-1-methyl-2-cxopyrimidine 1 (35% isolated yield), 2-chloro-1-methyl-4oxcpyrimidine 2 (20% isolated yield) and 2,4-dichloropyrimidine 3 (20% isolated yield). When the reaction temperature is lowered to 70° C, predominantly product **1** is formed in 75% yield, product **2** in 20% yield, whereas product **3** is not observed.



It should be mentioned that the increased yield of **1** is also due to change of work-up procedure of the reaction mixture.

The 4-chloro-1-methyl-2-oxopyrimidine 1 undergoes efficient reaction with amines in acetonitrile in the presence of potassium carbonate.



2013	
2:33 08 July	
ersity] at 12	
Aaster Univ	
ed by [McN	
Download	

Т	A	BI	_E
---	---	----	----

			Isolated	
	Produ	ct	Yield	(%)
4 a	$R_1 = H$	$R_2 = CH_3$	90	
4b	R ₁ = H	$R_2 = CH_2CH_3$	85	
4c	$R_1 = H$	$R_2 = CH_2CH_2CH_3$	84	
4d	$R_1 = H$	$R_2 = CH_2(CH_2)_2CH_3$	92	
4e	$R_1 = H$	$R_2 = CH_2(CH_2)_4CH_3$	90	
4 f	$R_1 = H$	$R_2 = C(CH_3)_3$	83	
4g	$R_1 = H$	$R_2 = CH_2CH = CH_2$	78	
4h	$R_1 = H$	R ₂ = cyclohexyl	76	
4 i	$R_1 = H$	$R_2 = CH_2Ph$	82	
4 j	$R_1 = CH_3$	$R_2 = CH_3$	72	
4k	$R_1 = CH_2CH_3$	$R_2 = CH_2CH_3$	84	
41	$R_1 = H$	$R_2 = Ph$	61	
4m	$R_1 = H$	$R_2 = p-tolyl$	60	

The reaction proceeds with aliphatic non-branched chain amines even in temperatures 25-50°C within 1 hour to give N⁴-substituted cytosines in high yields. In the case of sterically hindered t-butylamine and aromatic amines the reaction is performed at reflux (about 82°C). In comparison with the widely used synthetic precursors for N⁴-substituted 1-methylcytosines 4-methoxy- and 4-methylthio-1-methyl-2-oxopyrimidines, 1 is much more reactive and gives desired products in higher yields. In conclusion, we highly recommend the presented method as the most convenient and effective for the synthesis of N^4 -substituted 1-methylcytosines. The scope of the method was demonstrated by the synthesis of thirteen compounds **4a-m**.

EXPERIMENTAL

Melting points were determined on a Boetius apparatus and are uncorrected. Satisfactory microanalyses were obtained on an elemental analyser Perkin-Elmer 240. UV spectra were recorded on a Shimadzu UV-160 spectrophotometer. ¹H- and ¹³C-NMR spectra were determined on Jeol FX 90 Q spectrometer. Mass spectra were made on a Jeol JMS-D-100 mass spectrometer.

Synthesis of 4-chloro-1,2-dihydro-1-methyl-2-oxopyrimidine 1

A mixture of 1-methyluracil (2 g, 15.9 mmol) and phosphoryl chloride (8 ml, 85.8 mmol) was stirred at 70° C for 4 h. The excess of phosphoryl chloride was removed under reduced pressure. The residue was cooled in dry ice-isopropanol bath and neutralized with saturated solution of sodium bicarbonate. The resulting precipitate was filtered off, dissolved in acetonitrile and dried with magnesium sulphate. The solution was evaporated and dried in vacuo over phosphorus pentaoxide to give 1.72 g of product 1, yield 75%. Analytically pure samples of 1 were obtained by crystallization from acetonitrile. The filtrate (remaining after removal of the precipitate of product 1) was extracted with

chloroform (15 x 30 ml), the first two extracts were discarded and the further were combined, dried over magnesium sulphate and evaporated to give 0.46 g of product 2, yield 20%. It appeares that compound 2, whose structure was established by NMR and MS spectroscopy, is very sensitive to moisture. During standard purification by column chromatography or crystallization small amounts of 1-methyluracil always were detected.

1¹⁵: m.p. 207-208°C

¹H-NMR (CDC1₃) δ 3.51 (s, 3H, CH₃), 6.37 (d, 1H, J = 6.8 Hz, C⁵-H), 7.63 (d, 1H, J = 6.8 Hz, C⁶-H).

¹³(-NMR (CDCl₃) δ 38.7, 102.1, 105.1, 145.0, 148.4.

MS, m/z (rel. int.) 146 (32), 145 (19), 144 (100), 143 (37), 116 (19), 115 (20), 109 (33), 81 (20), 80 (13).

2: ¹H-NMR (CD₃CN) δ 3.61 (s, 3H, CH₃), 6.03 (d, 1H, J = 7.4 Hz, C⁵-H), 7.49 (d, 1H, J = 7.4 Hz, C⁶-H).

MS, m/z (rel. int.) 146 (32), 145 (17), 144 (100), 143 (35), 116 (24), 115 (19), 109 (38), 81 (18), 80 (13).

Synthesis of N⁴-substituted 1-methylcytosines 4a-m

To a solution of 1 (2 g, 13.8 mmol) in acetonitrile (220 ml) were added appropriate amine (41.4 mmol) and potassium carbonate (4 g, 28.9 mmol). The reaction mixture was stirred at 50° C for 1 h (in the case of 4f, 4l and 4m the reaction mixture was refluxed for 6 h). The solids were filtered off and washed with acetonitrile. The filtrate was evaporated and the residue was crystallized from acetonitrile 4a-j or water - methanol 4l and 4m.

The product 4m before crystallization was purified by flash chromatography on silica gel column (ethyl acetate - ethanol 50:1). **4a^{8,9,15}**: m.p. 177-178°C UV (CH₃OH) λ_{max} 273 nm, ϵ_{max} 7100 1 H-NMR (CD₂OD) δ 2.88 (s, 3H, N⁴-CH₂), 3.35 (s, 3H, N¹-CH₂), 5.81 (d, 1H, J = 7.3 Hz, C^5 -H), 7.45 (d, 1H, J = 7.3 Hz, C^6 -H). Downloaded by [McMaster University] at 12:33 08 July 2013 ¹³C-NMR (CD₂OD) δ 27.7, 37.6, 96.6, 146.2, 159.8, 166.4. MS, m/z (rel. int.) 139 (100), 109 (21),97 (34). **4b**: m.p. 135-137°C UV (CH_OH) λ_{max} 274 nm, ϵ_{max} 9600 ¹H-NMR (CD₂OD) δ 1.17 (t, 3H, J = 7.2 Hz, CH₃), 3.35 (s, 3H, N^{1} -CH₂), 3.38 (m, 2H, N^{4} -CH₂), 5.79 (d, 1H, J = 7.3 Hz, C⁵-H), 7.44 (d, 1H, J = 7.3 Hz, C^{6} -H). ¹³C-NMR (CD₂OD) δ 14.5, 36.4, 37.6, 96.7, 146.2, 159.9, 165.8. MS, m/z (rel. int.) 153 (100), 138 (28), 125 (44), 109 (15), 95 (29), 83 (37). 4c: m.p. 157-158°C UV (CH₃OH) λ_{max} 274 nm, ϵ_{max} 10000 ¹H-NMR (CD₀OD) δ 0.94 (t, 3H, J = 7.2 Hz, CH₂), 1.6 (m, 2H, CH₂), 3.35 (s, 3H, N^1 -CH₂), 3.37 (m, 2H, N^4 -CH₂), 5.81 (d, 1H, J = 7.3 Hz, C^{5} -H), 7.44 (d, 1H, J = 7.3 Hz, C^{6} -H).

 13 C-NMR (CD₃OD) δ 11.7, 23.2, 37.6, 43.3, 96.7, 146.2, 159.9, 166.0.

MS. m/z (rel. int.) 167 (58), 152 (42), 139 (25), 138 (75), 125 (100), 109 (25), 95 (75), 83 (51).

4d: m.p. 125-126^oC UV (CH₃OH) λ_{max} 274 nm, ϵ_{max} 9400 ¹H-NMR (CD₂OD) δ 0.94 (t, 3H, J = 7.0 Hz, CH₂), 1.51 (m, 4H, $CH_{3}-CH_{2}$), 3.38 (s, 3H, N¹-CH₃), 3.39 (m, 2H, N⁴-CH₂), 5.87 (d, 1H, $J = 7.3 \text{ Hz}, \text{ C}^{5}-\text{H}), 7.46 \text{ (d, 1H, } J = 7.3 \text{ Hz}, \text{ C}^{6}-\text{H}).$ ¹³C-NMR (CD_OD) δ 14.0, 20.8, 31.9, 37.8, 41.2, 96.9, 146.2, 159,9, 165.7. MS m/z (rel. int.) 181 (54), 166 (9), 152 (52), 139 (74), 138 (69), 125 (100), 109 (24), 95 (45), 83 (34). **4e**: m.p. 91−92°C UV (CH₃OH) λ_{max} 274 nm, ϵ_{max} 10800 ¹H-NMR (CD₂OD) δ 0.89 (t, 3H, J = 7.0 Hz, CH₃), 1.33 (m, 8H, CH₂), 3.33 (m, 2H, N^4 -CH₂), 3.36 (s, 3H, N^1 -CH₂), 5.85 (d, 1H, J = 7.2 Hz, C^{5} -H), 7.46 (d, 1H, J = 7.2 Hz, C^{6} -H). ¹³(-NMR (CD₂OD) δ 14.4, 23.6, 27.7, 30.0, 32.7, 37.6, 41.6, 96.7, 1**4**€.2, 160.0, 165.9. MS, m/z (rel. int.) 209 (42), 194 (8), 180 (18), 166 (34), 152 (62), 139 (90), 138 (70), 125 (100), 109 (28), 95 (43), 83 (37). **4f**: m.p. 255-257°C UV (CH₃OH) λ_{max} 272 nm, ϵ_{max} 9900 ¹H-NMR (CD₃OD) δ 1.44 (s, 9H, CH₃), 3.33 (s, 3H, N¹-CH₃), 5.79 (d, 1H, J = 7.3 Hz, C^{5} -H), 7.37 (d, 1H, J = 7.3 Hz, C^{6} -H). ¹³C-NMR (CD₂OD) δ 28.0, 37.6, 52.8, 97.8, 145.6, 159.6, 165.8. MS, m/z (rel. int.) 181 (34), 166 (15), 125 (100), 83 (32). 4g: m.p. 159°C UV (CH₃OH) λ_{max} 275 nm, ε_{max} 10300

¹H-NMR (CD₂OD) δ 3.35 (s, 3H, N¹-CH₃), 3.99 (m, 2H, N⁴-CH₂), 5.04-5.34 (m, 2H, =CH₂), 5.71-6.13 (m, 1H, -CH=), 5.83 (d, 1H, $J = 7.3 \text{ Hz}, \text{ C}^{5}-\text{H}), 7.46 \text{ (d, 1H, } J = 7.3 \text{ Hz}, \text{ C}^{6}-\text{H}).$ ¹³C-NMR (CD_OD) δ 37.6, 43.8, 96.5, 116.4, 135.3, 146.6, 159.8, 165.9. MS, m/z (rel. int.) 165 (100), 164 (69), 150 (32). **4h**: m.p. 193^OC UV (CH₃OH) λ_{max} 274 nm, ε_{max} 10500 ¹H-NMR (CD₂OD) δ 1.25-2.03 (m, 10H, CH₂), 3.33 (s, 3H, N¹-CH₃), 3.91 (m, 1H, CH), 5.76 (d, 1H, J = 7.2 Hz, C^{5} -H), 7.41 (d, 1H, J =7.2 Hz, C⁶-H). ¹³C-NMR (CD₃OD) δ 26.0, 26.7, 33.6, 37.6, 50.2, 96.9, 146.2, 160.0, 165.1. MS, m/z (rel. int.) 207 (36), 178 (6), 164 (13), 150 (19), 126 (52), 125 (100), 83 (17). **4i**: m.p. 195-196°C UV (CH₃OH) λ_{max} 274 nm, ϵ_{max} 11200 1 H-NMR (CD₃OD) & 3.32 (s, 3H, N¹-CH₃), 4.55 (s, 2H, CH₂), 5.84 (d, 1H, J = 7.3 Hz, C^{5} -H), 7.22-7.35 (m, 5H, Ph), 7.42 (d, 1H, J = 7.3Hz, C^6 -H). ¹³C-NMR (CD₂OD) δ 37.7, 45.2, 96.6, 128.2, 128.7, 129.4, 139.8, 146.6, 159.9, 166.0. MS, m/z (rel. int.) 215 (45), 106 (37), 91 (100). **4**,j^{8,9,15}: m.p. 181°C UV (CH₃OH) λ_{max} 281 nm, ϵ_{max} 11300 ¹H-NMR (CD₂OD) δ 3.12 (s, 6H, N⁴-CH₂), 3.39 (s, 3H, N¹-CH₂), 6.08 (d, 1H, J = 7.6 Hz, C^5 -H), 7.60 (d, 1H, J = 7.6 Hz, C^6 -H).

```
<sup>13</sup>C-NMR (CD<sub>2</sub>OD) δ 37.8, 38.0, 93.5, 147.5, 159.2, 165.4.
MS, m/z (rel. int.) 153 (100), 138 (36), 124 (31), 123 (45), 109
(12), 95 (17).
4k: m.p. (oil)
UV (CH<sub>3</sub>OH) \lambda_{max} 281 nm, \epsilon_{max} 12600
<sup>1</sup>H-NMR (CD<sub>3</sub>OD) \delta 1.18 (t, 3H, J = 7.0 Hz, CH<sub>3</sub>), 3.38 (s, 3H,
N^{1}-CH<sub>2</sub>), 3.58 (m, 2H, N^{4}-CH<sub>2</sub>), 6.02 (d, 1H, J = 7.5 Hz, C^{5}-H),
7.61 (d, 1H, J = 7.5 \text{ Hz}, C^6-H).
<sup>13</sup>C-NMR (CD<sub>2</sub>OD) δ 13.3, 37.7, 43.7, 93.5, 147.5, 159.4, 164.1.
MS, m/z (rel. int.) 181 (49), 166 (8), 152 (100), 138 (33).
41<sup>11,12</sup>: m.p. 265-267°C (lit. <sup>11,12</sup> 256-258°C)
UV (CH<sub>3</sub>OH) \lambda_{max} 297 nm, \varepsilon_{max} 17300
<sup>1</sup>H-NMR (DMSO-d<sub>z</sub>) \delta 3.29 (s, 3H, N<sup>1</sup>-CH<sub>2</sub>), 5.91 (d, 1H, J = 7.2 Hz,
C^{5}-H), 6.93-7.42 (m, 3H, CH), 7.71-7.82 (m, 2H, CH), 7.72 (d, 1H,
J = 7.2 \text{ Hz}, \text{ C}^{6}\text{-H}, 9.57 (br s, 1H, NH).
<sup>13</sup>C-NMR (DMSO-d<sub>2</sub>) δ 37.0, 94.3, 120.0, 122.7, 128.4, 139.3, 146.7,
155.7, 162.2.
MS, m/z (rel. int.) 201 (89), 200 (100), 159 (21).
4m: m.p. 269-270<sup>°</sup>C
UV (CH<sub>3</sub>OH) \lambda_{max} 298 nm, \varepsilon_{max} 20400
<sup>1</sup>H-NMR (DMSO-d<sub>2</sub>) 2.26 (s, 3H, CH<sub>2</sub>), 3.28 (s, 3H, N^{1}-CH<sub>2</sub>), 5.88 (d,
1H, J = 7.4 Hz, C^{5}-H), 7.11 (d, 2H, J = 8.4 Hz, CH), 7.62 (d, 2H,
J = 8.4 Hz, CH), 7.69 (d, 1H, J = 7.4 Hz, C<sup>6</sup>-H), 9.49 (br s, 1H,
NH).
^{13}C-NMR (DMSO-d<sub>2</sub>) \delta 20.5, 37.0, 94.3, 120.1, 128.9, 131.7, 136.8,
146.5, 155.8, 162.1.
```

MS, m/z (rel. int.) 215 (81), 214 (100), 173 (31).

ACKNOWLEDGEMENTS

The authors are indebted to Mrs Halina Thiel-Pawlicka for technical assistance. This work was supported by the Grant II/2.

REFERENCES

- 1. De Clercq E., J. Med. Chem., 1986, 29, 1561.
- 2. Shugar D., Pure Appl. Chem., 1985, 57, 423.
- 3. Robins R. K., Chem. Eng. News, 1986, 64, 28.
- Mansuri M. M., Starrett J. E., Wos J. A., Tortolani D. R., Brodfuehrer P. R., Howell H. G., Martin J. C., J. Org. Chem., 1989, 54, 4780.
- Matsuda A., Yasuaka J. Ueda T., Chem. Pharm. Bull., 1989, 37, 1659.
- Hall R. H., "The Modified Nucleosides in Nucleic Acids", Columbia University Press, New York, 1971, p. 295.
- 7. Fox J. J., Shugar D., Biochim. Biophys. Acta, 1952, 9, 369.
- 8. Szer W., Shugar D., Acta Biochim. Polon., 1966, 13, 177.
- 9. Katritzky A. R., Waring A. J., J. Chem. Soc., 1963, 177.
- 10. Ueda T., Fox J. J., J. Org. Chem., 1964, 29, 1762.
- Martirosyan Z. A., Gunar V.I., Zavyalov S. I., Izv. Akad. Nauk SSSR, Ser. Khim., 1970, 1127.
- Zavyalov S. I., Gunar V. I., Martirosyan Z. A., Ovechkina L.
 F., Izv. Akad. Nauk SSSR, Ser. Khim., 1972, 2530.
- Celewicz L., Spychala J., Golankiewicz K., Synth. Commun., 1987, 17, 1939.

- 14 Celewicz L., Spychala J., Golankiewicz K., J. Labelled Compd. Radiopharm., 1988, 25, 1401.
- 15 Kenner G. W., Reese C. B., Todd A. R., J. Chem. Soc., 1955, 855.

(Received in UK 8 April, 1991)