

Communications to the Editor

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THE SYNTHESIS AND ANTITUMOR ACTIVITIES OF TROPOLONE AND
8-HYDROXYQUINOLINE DERIVATIVES

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The bis-derivatives (4-6) of 8-hydroxyquinoline, which,
like tropolones, readily form a chelate, were synthesized and
found to be actively antitumorous in tests of survival using
P388 mice. 4 was almost as potent as bistropolone (2a).

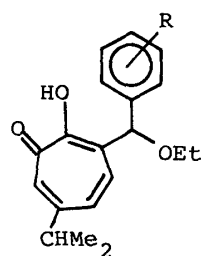
KEYWORDS ——— 8-hydroxyquinoline; tropolone; KB
cell; leukemia P388; antitumor activity

We¹⁾ previously synthesized 3-(α -ethoxycarbonyl)-6-isopropyltropolones (1)
and α,α -bis(2-hydroxy-6-isopropyltropon-3-yl)toluenes (2) by treating hinoki-
tiol²⁾ with *o*-, *m*-, and *p*-substituted benzaldehyde diethyl acetals and tested
their antitumor activities. Although both types of tropolone derivatives (1 and
2) were almost equally potent inhibiting the growth of KB cells (*in vitro* system),
generally, the bistropolones (2) were much more potent than the monotropolones
(1) in survival tests with P388 mice (*in vivo* system).

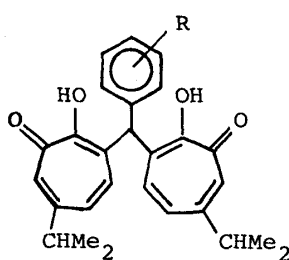
A subsequent study³⁾ of the structure-activity relationship of monotro-
polones (1) and bistropolones (2) demonstrated that two pairs of an acidic
hydroxyl and a proton-accepting group situated in neighboring positions in a
molecule are necessary to produce potent activity in the *in vivo* system.
Tropolones are known to be potent chelators of various metal ions. Consequently,
we considered that the ability of the two carbonyl-hydroxyl pairs to form a
chelate in 2 is closely related to its strong antitumor activity.

From these considerations, we have prepared 7-[α -(2-hydroxy-6-isopropyl-
tropon-3-yl)-4-methoxybenzyl]-8-hydroxyquinoline (3)³⁾ or α,α -bis(8-hydroxy-
quinolin-7-yl)-4-methoxytoluene (4), in which one or two tropolone rings in 2a
is replaced by 8-hydroxyquinoline, a strong chelating agent. Compound 4 was
synthesized by treating *p*-anisaldehyde diethyl acetal with 8-hydroxyquinoline
in the presence of a catalytic amount of potassium *tert*-butoxide in refluxing
cymene. Compounds 3 and 4 were relatively potent at low doses even in the *in*
vivo system.⁴⁾ Similarly, the thiophene and furan analogues (5 and 6) were syn-
thesized and found to be more active at low doses than monotropolone (1a).

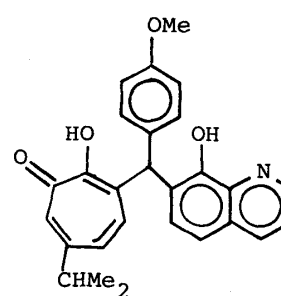
Various types of antitumor agents are applied clinically. The effects of
most of these are a consequence of their ability to interact reversibly or ir-
reversibly with DNA by intercalation, alkylation, oxidative cleavage, etc. Bis-



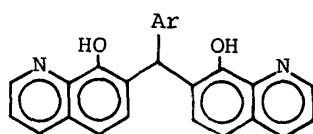
1

a: R = *p*-OMe

2

a: R = *p*-OMe

3



4: Ar = 4-anisyl

5: Ar = 2-thiophenyl

6: Ar = 2-furyl

Table I. Antitumor Activities of Tropolone and 8-Hydroxy-quinoline Derivatives

Compd. No.	Inhibition of KB ^{a)} Cell growth	Antitumor Act. in P388 mice, i.p.	
	IC ₅₀ (μg/ml)	Doses (mg/Kg) ^{b)}	T/C (%) ^{c)}
1a	0.5	100	140
		200	128
		400	140
2a	0.5	0.6	127
		2.5	134
		5	173
3	N. T. ^{d)}	3.1	144
		6.3	151
		12.5	141
4	< 0.3	3.1	111
		6.3	128
		12.5	164
5	< 0.3	2.5	108
		10	125
		20	136
6	< 0.3	10	113
		20	120
		40	138

a) See Ref. 1c. b) The dose listed was given once a day for 1 and 5 days. c) T/C: medium survival time of the treated animals/that of untreated controls x 100. A compound is considered to demonstrate antitumor activity if the test gives T/C values equal to or greater than 120%. d) N. T.: not tested.

tropolones (2) and bis-8-hydroxyquinolines (4-6) were thought to lack such ability because of their structures. On the other hand, α -N-heterocyclic carboxaldehyde thiosemicarbazones⁵⁾ with potent antitumor activity were reported to inhibit ribonucleoside diphosphate reductase,⁶⁾ which catalyzes the conversion of ribonucleotides to deoxyribonucleotides on the DNA biosynthetic pathways and requires iron ions for the activity.

Present results led us to conclude that antitumor-active tropolones and bis-8-hydroxyquinolines chelate the iron necessary for the enzyme thus inhibiting enzyme activity. A study of the antitumor mechanism is in progress and the results will be reported in the near future.

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