Syntheses of substituted 2,4-dioxo-thienopyrimidin-1-acetic acids and their evaluation as aldose reductase inhibitors

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(Received 9 November 1992; accepted 3 May 1993)

Summary — A series of 2,4-dioxo-thieno[2,3-d], [3,2-d] and [3,4-d]pyrimidin-1-acetic acids (2) with a benzyl moiety at the N-3 position were prepared and tested *in vitro* for aldose reductase inhibitory activity against partially purified enzyme from rat lens. Some of these compounds were also evaluated for inhibition of sorbitol accumulation in the sciatic nerve or lens of streptozotocin-induced diabetic rats *in vivo*. Among the synthesized compounds, several showed potent aldose reductase inhibitory activity with IC₅₀s in the 10^{-8} M range. Particularly, the potencies of non-substituted thieno- (2a and 2aa), 5-methylthieno- (2c), 5,6-dimethylthieno-(2g), 6-isopropylthieno- (2j and 2k), 6-chlorothienopyrimidine (2q) and benzothienopyrimidine (2ac) analogs were approximately equipotent to FK-366 (1A) and Ponalrestat (1B) as references. Although most compounds were inactive *in vivo*, 2 compounds, 2k and 2q, possessed moderate *in vivo* activity.

2,4-dioxo-thieno[2,3-d], [3,2-d] and [3,4-d]pyrimidin-1-acetic acids / aldose reductase inhibitory activity / sorbitol accumulation inhibition

Introduction

Aldose reductase (AR), the first enzyme in the polyol pathway, catalyses the NADPH-dependent reduction of D-glucose to its corresponding sugar alcohol, Dsorbitol. Under normal physiological conditions AR participates in osmoregulation, but under hyperglycemic conditions it also contributes to the onset and development of severe diabetic complications [1], ie, retinopathy, neuropathy, cataract, nephropathy and angiopathy [2, 3]. Experimental evidence indicates that inhibition of AR constitutes a possible approach to the treatment or prevention of certain secondary complications [4]. The development of clinically useful aldose reductase inhibitors (ARIs) has led to the identification of a large number of structurally diverse compounds with AR inhibitory activity [4, 5]. One of these ARIs, FK-366 [6] exhibits potent AR inhibitory activity and prevents the development of diabetic complications in animals. Clinical studies of FK-366 performed in Japan and the United States suggest it to be a promising ARI. FK-366 belongs to

the carboxylic acid class of ARIs and bears a quinazolinedione nucleus. Bioisosteric replacement of the benzene ring with a thiophene ring frequently leads to compounds with improved potency and selectivity [7-11], therefore leading us to focus on the preparation of novel carboxylic acid thienopyrimidine derivatives (2) that are the thiophene isosteres of FK-366 (1A) [6]. In the synthesized compounds, several members of the thieno[2,3-d] and [3,2-d] pyrimidine series displayed potent inhibitory activity against rat lens AR in vitro. Their in vitro potency was equivalent to that of FK-366 and Ponalrestat and they moderately reduced sorbitol accumulation in the sciatic nerve and lens of hyperglycemic rats in vivo. Here, we describe the preparation and inhibitory activity of substituted 2,4-dioxothienopyrimidin-1-acetic acids (2) on rat lens AR in vitro, and also describe the lowering effects of some of these compounds on sorbitol accumulation in the sciatic nerve and lens in the streptozotocin-induced diabetic rat [12] in comparison to the carboxylic acids FK-366 (1A) [6] and Ponalrestat (**1B**) [13].

Chemistry

The 2,4-dioxothieno[2,3-d], [3,2-d] and [3,4-d]pyrimidin-1-acetic acids (2) were prepared using the 5 gen-

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Scheme 2. Reagents: a) trichloromethylchloroformate (TCF), dioxane; b) $NH_2CH_2C_6H_4$ -R₃; c) NaOCH₃, MeOH; d) BrCH₂CO₂Et, NaH, DMF; e) conc HCl, AcOH; *Method B*.



Scheme 3. Reagents: a) COCl₂, dioxane; b) BrCH₂CO₂Et, NaH, DMF; c) NH₄OH; d) 1,1-carbonyldiimidazole (CDl), Δ ; e) BrCH₂C₆H₄-R₃, NaH, DMF; f) conc HCl, AcOH; *Method C*.

eral synthetic methods (Methods A,B,C,D and E) as shown in schemes 1–5, where R_1 , R_2 and R_3 represent the appropriate substituents as shown in tables I-III. In the first method (*Method A*) outlined in scheme 1, the amino group of 3 was ethoxycarbonylated with ethyl chlorocarbonate to give 4. Condensation of 4 with an appropriate benzylamine at a reaction temperature of 230-250°C produced the cyclized thienopyrimidine 5, which was subsequently subjected to alkylation with ethyl bromoacetate at the 1-position of 5 to give the 1,3-disubstituted thienopyrimidine (6), which was then hydrolyzed to the acetic acids 2. Due to the strenuous high reaction temperatures (> 240°C) required for the cyclization of the carbamate 4 with amines (formation of compound 5) as shown in scheme 1, the preparation Method B as shown in scheme 2 was adapted for most compounds (2). Refluxing isomeric aminothiophene carboxylates 3 with trichloromethylchloroformate (TCF) followed by treatment with appropriate benzylamines afforded the corresponding ureas (7). Ring closure of 7 with sodium methoxide in methanol proceeded easily to give thienopyrimidines 5 in fair yield. The alkylation of 5 with ethyl bromoacetate in the presence of sodium hydride in DMF produced 6, followed by

hydrolysis to yield the desirable thienopyrimidin-1acetic acids 2. Of these compounds 2, the preparation of 2 compounds (20 and 2aa) was carried out via the intermediate thienoxazine dione 9 as shown in scheme 3 (Method C). Treatment of aminothiophene carboxylic acid 8 with excess phosgene gave thieno [2,3-d] [1,3]oxazine dione 9 in high yield. Alkylation of 9 with ethyl bromoacetate in NaH/DMF and consequent ammonolysis with aqueous ammonia afforded the corresponding glycine anologues 10. These intermediates were subjected to ring closure by condensation with 1,1-carbonyldiimidazole (CDI) to yield the thienopyrimidines 11. Alkylation at the 3-position of 11 was accomplished by treatment with appropriate benzyl halides in NaH/DMF to give 1,3-disubstituted thienopyrimidines 6. Compounds 6 were also subjected to hydrolysis under acidic conditions to yield thienopyrimidin-1-acetic acids 2. The halogenated compounds on the thiophene ring were prepared by the general synthetic method as shown in scheme 4 (Method D). Halogenation at the 2-position of the thiophene ring of **6** with SO_2Cl_2 or *N*-bromosuccimimide (NBS) in CH₂Cl₂ or CCl₄ afforded the halogenated thienopyrimidines, which through hydrolysis gave the corresponding halogenated thienopyrimidin-1acetic acids 2. The (4-methylpiperazin-1-yl)methyl, (imidazol-1-yl)methyl or (morphorin-4-yl)methyl substituents at the 6-position of thieno[2,3-d]pyrimidines 2 were introduced by the synthetic method, as shown in scheme 5 (*Method E*). Bromination of 6methyl-thienopyrimidine 6h was carried out with NBS in the presence of benzoylperoxide in CCl₄ to yield bromomethyl 12. Subsequent treatment of 12 with appropriate amines gave 6z, which was hydrolyzed under acidic conditions to afford thienopyrimidines 2y. The compounds synthesized and their physical properties have been summarized in tables I–III.

Results and discussion

The synthesized compounds were assessed for the inhibitory activity against rat lens AR [14, 15]. Several compounds demonstrating potent *in vitro* inhibitory activity were also evaluated *in vivo* for their ability to reduce sorbitol accumulation in rat sciatic

nerve according to Peterson *et al* [16] and Clements *et al* [17]. Moreover, in order to investigate the relation between *in vitro* and *in vivo* activity, the plasma half-lives of compounds 2k and 2q were compared in rats with those of FK-366 and Ponalrestat as references. The biological results have been summarized in tables I, IV and figure 1.

A. Thieno[2,3-d]pyrimidine series

As shown in table I, the *in vitro* AR inhibitory activity of the compounds with either no or short alkyl substituents (excluding bulky substituents, such as a *tert*butyl group) on the thiophene ring of thieno[2,3-*d*]pyrimidines was equipotent with FK-366 and Ponalrestat with IC₅₀s in the 10-8 M range. Compounds **20–2t**, containing phenyl or cyclopentyl groups or halogen atom on the thiophene ring of thieno[2,3-*d*]pyrimidine, also retained intrinsic activity. Compound **2x**, containing a bicyclic structure, also showed potent AR inhibitory activity. *In vitro* inhibitory activity



Scheme 4. Reagents: a) SO₂Cl₂ or NBS, CCl₄; b) conc HCl, AcOH; c) SO₂Cl₂, CH₂Cl₂; Method D.



(2Y)

Scheme 5. Reagents: a) NBS, benzoylperoxide, CCl₄; b) AH, benzene; c) conc HCl, AcOH; *Method E*.

 Table I. Chemical and biological data of 3-substituted 2,4-dioxo-thienopyrimidin-1-acetic acid 2.

Compd	R_1	R_2	R_3	Synth method ^a	Yield (%)	mp (%)	Formula ^b	RLAR inhibition ^c IC ₅₀ (10-8 M)
[2,3-d] Series								
2a	Н	Н	2-F, 4-Br	В	77	200-202	$C_{15}H_{10}FBrN_2O_4S$	2.0
2b	CH ₃	Н	3,4-(Cl) ₂	В	48	>300	$C_{16}H_{12}Cl_2N_2O_4S$	3.0
2c	CH ₃	Н	2-F, 4-Br	В	87	207-208	$C_{16}H_{12}FBrN_2O_4S$	1.5
2d	$CH(CH_3)_2$	Н	2-F, 4-Br	В	84	155-157	$C_{18}H_{16}FBrN_2O_4S$	3.3
2e	CH ₃	CH_3	4-Cl	А	61	174–176	$C_{17}H_{15}ClN_2O_4S$	7.5
2f	CH ₃	CH ₃	$2,4-(Cl)_2$	А	53	266-267	$C_{17}H_{14}Cl_2N_2O_4S$	2.0
2g	CH_3	CH ₃	2-F, 4-Br	В	80	233-235	$C_{17}H_{14}FBrN_2O_4S$	2.1
2h	Н	CH ₃	2-F, 4-Br	В	66	242-243	$C_{16}H_{12}FBrN_2O_4S$	4.1
2i	Н	$CH(CH_3)_2$	3,4-(Cl) ₂	В	61	189–191	$C_{18}H_{16}Cl_2N_2O_4S$	3.0
2ј	Н	$CH(CH_3)_2$	2,4-(Cl) ₂	В	44	228-229	$C_{18}H_{16}Cl_2N_2O_4S$	1.7
2k	Н	$CH(CH_3)_2$	2-F, 4-Br	В	55	203-205	$C_{18}H_{16}FBrN_2O_4S$	2.2
21	Н	$CH(CH_3)_2$	$2,4-(F)_2$	В	75	194–196	$C_{18}H_{16}F_2N_2O_4S$	10.0
2m	Н	$C(CH_3)_3$	2,4-(Cl) ₂	В	67	195–197	$C_{19}H_{18}Cl_2N_2O_4S$	10.0
2n	Н	C(CH ₃) ₃	2-F, 4-Br	В	48	217–218	$\mathrm{C_{19}H_{18}FBrN_2O_4S}$	10.0
20	Н	Ph	2-F, 4-Br	С	71	249–251	$C_{21}H_{14}FBrN_2O_4S$	3.4

 Table I. (continued).

Compd	R ₁	<i>R</i> ₂	R_3	Synth methodª	Yield (%)	mp (%)	Formula ^b	RLAR inhibition ^c IC ₅₀ (10 ⁻⁸ M)
2р	H	-	2-F, 4-Br	В	49	160–162	$C_{20}H_{18}FBrN_2O_4S$	3.0
2q	Н	Cl	2-F, 4-Br	D	96	221-223	C ₁₅ H ₉ FClBrN ₂ O ₄ S	2.5
2r	Н	Br	2-F, 4-Br	D	93	245-247	$C_{15}H_9FBr_2N_2O_4S$	5.6
2s	CH_3	Cl	3.4-(Cl) ₂	D	19	>300	$C_{16}H_{11}Cl_3N_2O_4S$	2.5
2t	CH_3	∧ ^{Br}	3.4-(Cl) ₂	D	37	>300	$C_{16}H_{11}Cl_2BrN_2O_4S$	2.6
2u		Ç	$4-CH_3$	Α	65	253-256	$C_{20}H_{20}N_2O_4S$	26.0
2v		Ç	4-OCH ₃	А	65	232–234	$C_{20}H_{20}N_2O_5S$	18.0
2w		Ç	2,4-(Cl) ₂	Α	63	265–266	$C_{19}H_{16}Cl_2N_2O_4S$	4.2
2x		\bigcirc	2-F, 4-Br	В	85	288–290	$C_{19}H_{16}FBrN_2O_4S$	2.3
2ya	Н	-CH2N_0	2-F, 4-Br	Ε	79	197–199	$\begin{array}{c} C_{20}H_{19}FBrN_{3}O_{5}S\\ \bullet HCl\bullet l/5H_{2}O\end{array}$	12.0
2yb	Н	-CH ₂ N NCH ₃	2-F, 4-Br	Ε	30	282–285	$C_{21}H_{22}FBrN_4O_4S$ •2(2H ₂ O)	12.0
2yc	Н	-CH2N N	2-F, 4-Br	Ε	92	178–180	$\begin{array}{c} C_{19}H_{14}FBrN_2O_4S\\ \bullet HCl\bullet H_2O \end{array}$	14.0
[3,2-d] Se	eries							
2aa	Н	Н	2-F, 4-Br	С	93	214–216	$C_{15}H_{10}FBrN_2O_4S$	2.0
2ab	Cl	Н	2-F, 4-Br	D	92	197–198	C15H9FClBrN2O4S	3.0
2ac		\sim	2-F, 4-Br	В	74	239–240	C ₁₉ H ₁₂ FBrN ₂ O₄S •1/2H ₂ O	1.5
2ad	c		2-F, 4-Br	В	95	126–128	$C_{19}H_{11}FClBrN_2O_4S$	10.0
2ae	C		2-F, 4-Br	В	96	270–272	$\mathbf{C_{19}H_{11}FClBrN_2O_4S}$	> 100
2af		cı	2-F, 4-Br	В	85	242–243	$\begin{array}{c} C_{19}H_{11}FClBrN_2O_4S\\ \bullet 1/2H_2O\end{array}$	> 100
[3,4-d] Se	eries							
2ba	Н	Н	2-F, 4-Br	В	75	200–202	$C_{15}H_{10}FBrN_2O_4S$	40.0
2bb	Н	Cl	2-F, 4-Br	D	53	239–241	$C_{15}H_9FClBrN_2O_4S$	> 100
FK-366								1.5
Ponalrest	at							1.7

^aSee scheme and *Experimental protocols*; ^bthe analysis was within $\pm 0.4\%$ of the theoretical values; ^crat lens aldose reductase inhibitory activity.

decreased with the introduction of hydrophobic substituents such as morpholinomethyl, N-methylpiperazinomethyl and imidazolylmethyl groups on the thiophene ring as shown in compounds **2ya**, **2yb** and **2yc**. In examining substituents on the benzyl ring at the 3-position of thieno[2,3-d]pyrimidine, compounds containing halogen substituents generally displayed

potent AR inhibitory activity *in vitro*, especially 2a, 2c, 2f, 2g, 2j, 2k, 2q, 2s, 2t and 2x containing 2,4and 3,4-dichloro or 4-bromo-2-fluoro substituents on the benzyl ring. Interestingly, these observations are similar to those for the FK-366 and Ponalrestat series. *In vitro* AR inhibitory activity was significantly influenced by substituents on the benzyl ring at the 3position of thieno[2,3-d]pyrimidine. Namely, **2u** and **2v** with an electron-donating group such as methyl or methoxy groups on benzyl ring showed a significant decrease of AR inhibitory activity in comparison with those compounds with an electron-withdrawing group such as halogen, especially dihalogen. From these observations, the substituent at the 3-position of thieno[3,2-d]pyrimidines and thieno[3,4-d]pyrimidines was focused on the representative 4-bromo-2-fluorobenzyl group with an electron-withdrawing property.

B Thieno [3,2-d]pyrimidine series

The *in vitro* AR inhibitory activity for compounds **2aa**, **2ab** and **2ac** of this series was equipotent to that of the thieno[2,3-*d*]pyrimidine series with IC₅₀s in the 1.5–3.0 x 10⁻⁸ M range. Introduction of a chlorine atom on the benzene ring of benzothieno[3,2-*d*]pyrimidine significantly reduced inhibitory activity.

C Thieno[3,4-d]pyrimidine series

Although the structure-activity relationships in this series cannot be fully elucidated due to an inadequate number of synthesized compounds, 2 compounds (**2ba** and **2bb**) failed to show significant AR inhibitory activity *in vitro* in comparison with **2a** and **2aa** in



the thieno[2,3-d]pyrimidine and thieno[3,2-d]pyrimidine series respectively. In *in vivo* evaluations, the compounds **2a**, **2c**, **2g**, **2j**, **2k**, **2q**, **2aa** and **2ac** were compared with FK-366 and Ponalrestat for their ability to reduce galactitol accumulation. Most of the compounds failed to show statistically significant *in vivo* activity, while **2k** and **2q** moderately reduced

Compd	R_{I}	<i>R</i> ₂	<i>R</i> ³	Synth method ^a	Yield (%)	mp (°C)	Formula ^b
[2,3- <i>d</i>] Series						1	
6a	Н	Н	2-F, 4-Br	В	63	117-119	$C_{17}H_{14}FBrN_2O_4S$
6b	CH_3	Н	$3,4-(Cl)_2$	В	46	124-126	$C_{18}H_{16}Cl_2N_2O_4S$
6c	CH ₃	Н	2-F, 4-Br	В	87	153–154	$C_{18}H_{16}FBrN_2O_4S$
6d	$CH(CH_3)_2$	Н	2-F, 4-Br	В	98	148–149	$C_{20}H_{20}FBrN_2O_4S$
6e	CH_3	CH ₃	4-C1	Α	92	151-152	$C_{19}H_{19}ClN_2O_4S$
6f	CH_3	CH ₃	$2,4-(Cl)_2$	А	96	150-151	$C_{19}H_{18}Cl_2N_2O_4S$
6g	CH ₃	CH ₃	2-F, 4-Br	В	88	144–146	$C_{19}H_{18}FBrN_2O_4S$
6h	Н	CH ₃	2-F, 4-Br	В	58	145-147	$C_{18}H_{16}FBrN_2O_4S$
6i	Н	$CH(CH_3)_2$	$3,4-(Cl)_2$	В	89	91–92	$C_{20}H_{20}Cl_2N_2O_4S$
6ј	Н	$CH(CH_3)_2$	3,4-(Cl) ₂	В	61	146–148	$C_{20}H_{20}Cl_2N_2O_4S$
6k	Н	$CH(CH_3)_2$	2-F, 4-Br	В	88	99 –101	$C_{20}H_{20}FBrN_2O_4S$
61	Н	$CH(CH_3)_2$	$2,4-(F)_2$	В	50	111-112	$C_{20}H_{20}F_2N_2O_4S$
6m	Н	$C(CH_3)_3$	$2,4-(Cl)_2$	В	68	115-116	$C_{21}H_{22}Cl_2N_2O_4S$
6n	Н	$C(CH_3)_3$	2-F, 4-Br	В	90	137–139	$C_{21}H_{22}FBrN_2O_4S$
60	Н	Ph	2-F, 4-Br	С	70	234-235	$C_{23}H_{18}FBrN_2O_4S$

Table II. Chemical and biological data of 3-substituted 2,4-dioxo-thienopyrimidin-1-acetic acid ethyl esters 6.

Compd	R	R_2	R_3	Synth methodª	Yield (%)	тр (°С)	Formula ^b
6n	Н	C(CH ₃) ₃	2-F, 4-Br	В	90	137–139	$C_{21}H_{22}FBrN_2O_4S$
60	Н	Ph	2-F, 4-Br	С	70	234–235	$C_{23}H_{18}FBrN_2O_4S$
6р	Н	-	2-F, 4-Br	В	83	9697	$C_{22}H_{22}FBrN_2O_4S$
6q	Н	CI	$3,4-(Br)_2$	D	73	167–168	$C_{17}H_{13}ClFBr_2N_2O_4S$
6r	Н	Cl	2-F, 4-Br	D	93	133–134	C ₁₇ H ₁₃ FClBrN ₂ O ₄ S
6s	H	Br	2-F, 4-Br	D	62	169–170	$C_{17}H_{13}FBr_2N_2O_4S$
6t	CH ₃	Cl	$3,4-(Cl)_2$	D	65	142-145	$C_{18}H_{15}Cl_{3}N_{2}O_{4}S$
6u	CH ₃	Br	3,4-(Cl) ₂	D	85	166–167	$C_{18}H_{15}Cl_2BrN_2O_4S$
6v	(\sim	4-CH ₃	Α	65	147–149	$C_{22}H_{24}N_2O_4S$
6w	(\sim	4-OCH ₃	Α	60	153–154	$C_{22}H_{24}N_2O_5S$
6x	(\sim	2,4-(Cl) ₂	Α	66	169-170	$C_{21}H_{20}Cl_2N_2O_4S$
бу	(\sim	2-F, 4-Br	В	60	141–143	$C_{21}H_{20}FBrN_2O_4S$
6za	Н	-CH2NO	2-F, 4-Br	Ε	42	178–180	$\begin{array}{c} C_{22}H_{23}FBrN_{3}O_{5}S\\ \bullet 2H_{2}O\end{array}$
6zb	Н	-CH2N_NCH3	2-F, 4-Br	Ε	31	105–106	$C_{23}H_{26}FBrN_4O_4S$
6zc	Н	-CH2N N	2-F, 4-Br	Ε	46	141–142	$C_{21}H_{18}FBrN_4O_4S$
[3,2-d] Series							
баа	H	Н	2-F, 4-Br	С	45	162-163	$C_{17}H_{14}FBrN_2O_4S$
6ab	Cl	Н	2-F, 4-Br	D	39	125-126	$C_{17}H_{13}FClBrN_2O_4S$
бас	(2-F, 4-Br	В	82	201-203	$C_{21}H_{14}FBrN_2O_4S$
6ad	(2-F, 4-Br	В	25	136–138	C ₂₁ H ₁₅ FClBrN ₂ O ₄ S
6ae	C1[2-F, 4-Br	В	50	175–176	$C_{21}H_{15}FClBrN_2O_4S$
6af		cı C	2-F, 4-Br	В	63	212–213	$C_{21}H_{15}FClBrN_2O_4S$
[3,4-d] Series							
6ba	Н	Н	2-F, 4-Br	В	81	149–150	$C_{17}H_{14}FBrN_2O_4S$
6bb	Н	Cl	2-F, 4-Br	D	51	171–173	$C_{17}H_{13}FClBrN_2O_4S$

^{a, b}See table I.

sorbitol accumulation in the rat sciatic nerve or lens, however, this reduction was less compared to that observed with FK-366 or Ponalrestat. Examination of plasma concentrations of **2k**, **2q**, FK-366 and Ponalrestat after 10 mg/kg *po* administration in rat as shown in figure 1 indicated that all 4 compounds displayed essentially equal C_{max} but reduced plasma half-lives for compounds **2k** and **2q** (3.1 and 6.1 µg/ ml at 24 h, respectively) compared to FK-366 and Ponalrestat (11.2 and 7.8 μ g/ml at 24 h, respectively). Though the data on plasma concentration for other compounds have not been shown here, the plasma concentration of all compounds subjected to *in vivo* examination was measured in the same manner as that for **2k** and **2q**, and all compounds with poor *in vivo* activity resulted in low C_{max} in plasma. These observations suggest that the *in vivo* activity is proportional to the plasma half-life of these ARIs.

Conclusion

Based on the bioisosteric replacement of the benzene ring on the quinazolidine nucleus with a thiophene ring, a series of thienopyrimidines have been synthesized and evaluated both *in vitro* and *in vivo* for their ability to inhibit aldose reductase *in vitro* and to reduce sorbitol accumulation in sciatic nerve and lens *in vivo*. Although a number of *in vitro* active ARIs have been obtained, none possess *in vivo* inhibitory activity equivalent to that of FK-366 or Ponalrestat. The search for potent *in vivo* active ARIs continues.

Table III. Chemical and biological data of 3-substituted thienopyrimidine-2,4 (1H, 3H)-diones 5.

Compd	R_I	R_2	R_{3}	Synth method ^a	Yield (%)	mp (°C)	<i>Formula</i> ^b
[2,3-d] Series							
5a	Н	Н	2-F, 4-Br	В	71	290–291	$C_{13}H_8FBrN_2O_2S$
5b	CH ₃	Н	3,4-(Cl) ₂	В	70	299–300	$C_{14}H_{10}Cl_2N_2O_4S$
5c	CH ₃	Н	2-F, 4-Br	В	66	296–298	$C_{14}H_{10}FBrN_2O_2S$
5d	$CH(CH_3)_2$	Н	2-F, 4-Br	В	52	271-273	$\mathbf{C_{16}H_{14}FBrN_2O_2S}$
5e	CH ₃	CH ₃	4-Cl	А	68	291–292	$C_{15}H_{13}ClN_2O_2S$
5f	CH ₃	CH ₃	$2,4-(Cl)_2$	Α	70	275–276	$C_{15}H_{12}Cl_2N_2O_2S$
5g	CH ₃	CH ₃	2-F, 4-Br	В	47	258-259	$C_{15}H_{12}FBrN_2O_2S$
5h	Н	CH ₃	2-F, 4-Br	В	55	304–305	$C_{14}H_{10}FBrN_2O_2S$
5i	Н	-CH(CH ₃) ₂	3,4-(Cl) ₂	В	43	253-254	$C_{16}H_{14}Cl_2N_2O_2S$
5j	Н	$CH(CH_3)_2$	$2,4-(Cl)_2$	В	47	223-225	$C_{16}H_{14}Cl_2N_2O_2S$
5k	Н	$CH(CH_3)_2$	2-F, 4-Br	В	76	227-228	$C_{16}H_{14}FBrN_2O_2S$
51	Н	$CH(CH_3)_2$	$2,4-(F)_2$	В	64	219–221	$C_{16}H_{14}F_2N_2O_2S$
5m	Н	$C(CH_3)_3$	$2,4-(Cl)_2$	В	43	264–265	$C_{17}H_{16}Cl_2N_2O_2S$
5n	Н	$C(CH_3)_3$	2-F, 4-Br	В	14	240-241	$C_{17}H_{16}FBrN_2O_2S$
50	Н	-<]	2-F, 4-Br	В	69	257-259	$C_{18}H_{16}FBrN_2O_2S$
5p	\bigcirc		4-CH ₃	А	40	240-241	$C_{18}H_{18}N_2O_2S$
5q	\bigcirc		4-OCH ₃	А	29	230–232	$C_{18}H_{18}N_2O_3S$
5r	\bigcirc		2,4-(Cl) ₂	А	45	255-256	$C_{17}H_{14}Cl_{2}N_{2}O_{2}S$
5s	Č		2-F, 4-Br	В	49	228-230	$C_{17}H_{14}FBrN_2O_2S$
[3,2- <i>d</i>] Series 5aa	\bigcirc		2-F, 4-Br	В	90	> 300	$C_{17}H_{10}FBrN_2O_2S$
5ab	Ç	1	2-F, 4-Br	В	78	267–268	C ₁₇ H ₉ FClBrN ₂ O ₂ S
5ac		-	2-F, 4-Br	В	96	> 300	$C_{17}H_9FClBrN_2O_2S$
5ad	c1		2-F, 4-Br	В	59	> 300	$C_{17}H_9FClBrN_2O_2S$
[3,2-d] Series 5ba	Н	Н	2-F, 4-Br	В	32	> 300	$C_{13}H_{18}FBrN_2O_2S$

^{a, b}See table I.

Table IV. Biological data (*in vivo*) of thienopyrimidin-1-acetic acids 2.

Compd	Dose	Reduction (%)				
	(mg/kg)	Sciatic nerve	Lens			
2a	20	33.4**	NTb			
2c	20	12.6	15.2			
2g	20	30.0**c	NT			
2ј	20	14.0	NT			
2k	20	60.1**	60.0**			
2q	20	73.6***	31.7*			
2aa	20	15.7	NT			
2ac	20	21.4*	NT			
FK-366	10	85.7***	NT			
	20	90.4***	91.8***			
Ponalrestat	10	65.6***	43.4**			

^aReduction of sorbitol accumulation in the sciatic nerves and lens on the streptozotocin (70 mg/kg) induced diabetic rats (n = 7). Compounds were orally administered once a day for 4 d; ^bNT: not tested; ^cstatistically significant at *P <0.05; **P < 0.01; ***P < 0.001.

Experimental protocols

Melting points were obtained on a Yanagimoto micromelting apparatus and are uncorrected. ¹H-NMR spectra were recorded on a JEOL JMN-FX 100 spectrometer using tetramethylsilane (TMS) as an internal standard. Chemical shifts are expressed as δ values (ppm). Elemental analyses were carried out with a Yanagimoto CHN Corder MT-2.



Fig 1. Plasma concentrations of 2k and 2q in comparison with that of FK-366 and Ponalrestat after a single oral administration in rats (10 mg/kg) (mean \pm SD of 2 or 3 rats).

Starting materials

Ethyl 2-aminothiophene-3-carboxylate [18], ethyl 2-amino-5methylthiophene-3-carboxylate [19], ethyl 2-amino-4,5-dimethylthiophene-3-carboxylate [19], ethyl 2-amino-5-isopropylthiophene-3-carboxylate [20], ethyl 2-amino-4,5,6,7tetrahydrobenzothiophene-3-carboxylate [19], ethyl 2-amino-5phenylthiophene-3-carboxylate [20], methyl 3-aminothiophene-4-carboxylate [21], methyl 3-aminothiophene-4carboxylate [21], methyl 3-aminobenzothiophene-2carboxylate [22], methyl 3-amino-6-chlorobenzothiophene-2-carboxylate [22] and methyl 3-amino-6-chlorobenzothiophene-2-carboxylate [22] were prepared according to literature procedures. The following aminothiophene carboxylates were prepared in the same manner as described in the literature [18, 19, 22].

Ethyl 2-amino-5-tert-butylthiophene-3-carboxylate

Yield: 23%; bp: 120–121°C/0.012 mmHg. ¹H-NMR (CDCl₃): 1.25 (s, 9H), 1.28 (t, 3H), 4.25 (q, 2H), 5.20–6.00 (brs, 2H) and 6.60 (s, 1H).

Ethyl 2-amino-5-cyclopentylthiophene-3-carboxylate Yield: 75%; bp: $148-150^{\circ}C/0.022 \text{ mmHg}$. ¹H-NMR (CDCl₃): 1.33 (t, 3H), 1.45-1.80 (m, 6H), 1.90-2.10 (m, 2H), 2.90-3.10(m, 1H), 4.25 (q, 2H), 5.60-5.90 (brs, 2H) and 6.64 (s, 1H).

Ethyl 2-amino-4-isopropylthiophene-3-carboxylate Yield: 61%; bp: 115–118°C/0.019 mmHg. ¹H-NMR (CDCl₃): 1.17 (d, 6H), 1.33 (t, 3H), 3.10–3.60 (m, 1H), 4.28 (q, 2H), 5.80 (s, 1H) and 6.00–6.20 (brs, 2H).

Methyl 3-amino-5-chlorobenzothiophene-2-carboxylate Yield:70%; mp: 182–183°C. Anal calcd for $C_{10}H_8CINO_2S$: C, 49.69; H, 3.34; N, 5.80; found: C, 49.95; H, 3.28; N, 5.81.

2-Amino-5-phenylthiophene-3-carboxylic acid 8aYield: 75%; mp: 175–177°C. Anal calcd for C₁₁H₉NO₂S; C, 60.26; H, 4.14; N, 6.39; found: C, 60.53; H, 4.16; N, 6.37.

3-Aminothiophene-2-carboxylic acid **8b** Yield: 30%; mp: 88–89°C. Anal calcd for $C_5H_5NO_2S$; C, 41.95; H, 3.52; N, 9.78; found: C, 41.91; H, 3.60; N, 9.74.

4-Bromo-2-fluorobenzylamine

This was prepared from 4-bromo-2-fluorobenzylbromide [23] according to the method of Calat *et al* [24]. Yield: 66%; bp: $61-64^{\circ}C/0.35$ mmHg. ¹H-NMR (CDCl₃): 3.86 (s, 2H) and 7.10–7.40 (s, 3H). All other aminothiophene carboxylates, benzylamines and benzylhalides were commercially available.

Chemistry

Method A

Ethyl 4,5-Dimethyl-2-ethoxycarbonylaminothiophene-3carboxylate 4a. A mixture of ethyl 2-amino-4,5-dimethylthiophene-3-carboxylate (3.15 g, 15.8 mmol) and ethyl chloroformate (40 ml) was refluxed for 3 h. After cooling, the reaction mixture was evaporated under reduced pressure and the residue recrystallized from ethanol to give 4a (3.7 g, 86%), mp: 68–69°C. Anal calcd for C₁₂H₁₇NO₄S; C, 53.12; H, 6.32; N, 5.16; found: C, 53.49; H, 6.10; N, 5.17. Ethyl 2-ethoxycarbonylamino-4,5,6,7-tetrahydrobenzothiophene-3-carboxylate **4b**. The title compound was prepared in a similar manner to that described for **4a**. Yield: 67%; mp: $65-66^{\circ}$ C. ¹H-NMR (CDCl₃): 1.30 (t, 3H), 1.40 (t, 3H), 1.67– 1.93 (m, 4H), 2.60–3.00 (m, 4H), 4.28 (q, 2H), 4.32 (q, 2H) and 10.40 (brs, 1H).

3-(4-Chlorobenzyl)-5,6-dimethylthieno[2,3-d]pyrimidine-2,4(1H, 3H)-dione 5e. A mixture of 4a (1.0 g, 3.7 mmol) and 4-chlorobenzylamine (1.0 g, 7.1 mmol) was heated to 230–240°C for 8 h. After cooling, the crude solid was recrystallized from a mixture of ethanol and dimethylformamide (DMF) to yield 5e (0.8 g, 68%), mp: 291–292°C. ¹H-NMR(DMSO-d₆): 2.25 (s, 3H), 2.26 (s, 3H), 4.99 (s, 2H), 7.20–7.45 (m, 4H) and 11.00 (brs, 1H). Compounds 5f and 5p–5r were prepared from the corresponding thiophene carboxylate 4 and benzylamine in a similar manner to that described for 5e. The physical data have been listed in table III.

Ethyl 3-(4-chlorobenzyl)-5,6-dimethyl-2,4-dioxo-thieno[2,3d]pyrimidin-1-acetate **6e**. Compound **5e** (0.6 g, 2.1 mmol) was slowly added to a suspension of NaH (60% in oil, 0.1 g, 2.5 mmol) at 0–5°C. After stirring for 0.5 h at rt, ethyl bromoacetate (0.4 g, 3.2 mmol) was added dropwise and the mixture was stirred for 17 h at the same temperature. The reaction mixture was then evaporated under reduced pressure, and a mixture of ice-water and diluted HCl was added to the residue. The resulting precipitate was collected and recrystallized from ethanol to give **6e** (0.7 g, 92.0%), mp: $151-152^{\circ}$ C. ¹H-NMR (CDCl₃): 1.19 (t, 3H), 2.30 (s, 6H), 4.16 (q, 2H), 4.73 (s, 2H), 5.04 (s, 2H) and 7.20–7.50 (m, 4H). Compounds **6f** and **6v–6x** were prepared in a similar manner to that described for **6e**. The physical data have been listed in table II.

3-(4-Chlorobenzyl)-5,6-dimethyl-2,4-dioxo-thieno[2,3-d]pyrimidin-1-acetic acid 2e. A solution of **6e** (0.7 g, 1.7 mmol) and 4 N–NaOH (2 ml) in MeOH (30 ml) was heated at 60°C for 0.5 h. The reaction mixture was evaporated under reduced pressure and a mixture of ice-water and diluted HCl was added to the residue. The resulting precipitate was collected and recrystallized from methanol to give 2e (0.4 g, 61%), mp: 174–176°C. ¹H-NMR(DMSO–d₆): 2.29 (s, 6H), 4.62 (s, 2H), 5.04 (s, 2H) and 7.15–7.45 (m, 4H). Compounds 2f and 2u–2w were prepared in a similar manner to that described for 2e. The physical data have been listed in table I.

Method B

3-(4-Bromo-2-fluorobenzyl)-thieno[2,3-d]pyrimidine-2,4-(1H, 3H)-dione 5a. Trichloromethylchloroformate (TCF) (12 ml) was added dropwise to a solution of ethyl 2-aminothiophene-3-carboxylate (7.5 g, 43.8 mmol) in dioxane (60 ml) at 5°C. The reaction mixture was stirred for 1 h at rt, heated to 60–70°C for 4 h, and then evaporated under reduced pressure. The resulting residue was dissolved in Et₂O (80 ml) and filtered. The filtrate was added to a solution of 4-bromo-2-fluorobenzylamine (10.0 g, 49.0 mmol) in dioxane (50 ml) at rt, after which the reaction mixture was stirred for 12 h at rt and then evaporated under reduced pressure. The residue was excrystallized from ethanol to yield compound 7a.

l-(4-Bromo-2-fluorobenzyl)-3-(3-ethoxycarbonyl-thiophen-2-yl)urea (7*a*). Compound 7a (13.2 g, 75%), mp: 158–159°C. ¹H-NMR (CDCl₃): 1.30 (t, 3H), 4.20–4.50 (m, 4H), 6.79 (d, 1H), 7.06 (d, 1H), 7.20–7.65 (m, 3H), 8.44 (t, 1H) and 10.25 (brs, 1H). The obtained urea (12.65 g, 31.5 mmol) was added to a solution of Na (1.45 g, 63 mmol) in methanol (300 ml) and the reaction mixture refluxed for 2 h and then evaporated under reduced pressure. The residue was dissolved in water (80 ml) and acidified (pH = 1) with diluted HCl. The resulting precipitate was collected and recrystallized from ethanol to yield **5a** (7.9 g, 71%), mp: 290–291°C. ¹H-NMR (DMSO–d₆): 5.03 (s, 2H), 7.00–7.65 (m, 5H) and 12.35 (brs, 1H). Compounds **5b–5d**, **5g–5o**, **5s**, **5aa–5ad** and **5ba** were prepared in a similar manner to that described for **5a**. The physical data have been listed in table III.

Ethyl 3-(4-bromo-2-fluorobenzyl)-2,4-dioxo-thieno[2,3-d]pyrimidin-1-acetate 6a. The title compound was prepared for **5a** and ethyl bromoacetate in a similar manner to that described for **6e**. Yield: 63%, mp: 117–119°C. ¹H-NMR (CDCl₃): 1.19 (t, 3H), 4.18 (q, 2H), 4.80 (s, 2H), 5.09 (s, 2H) and 7.00–7.65 (m, 5H). Compounds **6b–6d**, **6g–6n**, **6p**, **6y**, **6ac–6af** and **6ba** were prepared in a similar manner to that described for **6a**. The physical data have been listed in table II.

3-(4-Bromo-2-fluorobenzyl)-2,4-dioxo-thieno[2,3-d]pyrimidinl-acetic acid 2a. A solution of 6a (3.0 g, 6.8 mmol), conc HCl (20 ml) in AcOH (50 ml) was refluxed for 6 h. After cooling, the resulting precipitate was collected, washed with water and recrystallized from 50% ethanol to give 2a (2.2 g, 77%), mp: 200–202°C. ¹H-NMR (DMSO–d₆): 4.70 (s, 2H), 5.09 (s, 2H), 7.00–7.65 (m, 5H) and 13.50 (brs, 1H). Compounds 2b–2d, 2g–2n, 2p, 2x, 2ac–2af and 2ba were prepared in a similar manner to that described for 2a. The physical data have been listed in table I.

Method C

6-Phenyl-2H-thieno[2,3-d][1,3]oxazine-2,4-(1H)-dione 9a. 15 ml phosgene (2.7 M solution in CCl₄) was added dropwise to a solution of **8a** (5.0 g, 22.8 mmol) in dioxane (50 ml) at 5–10°C. The reaction mixture was refluxed for 4 h and then evaporated under reduced pressure. The residue was recrystallized from a mixture of ethanol and acetone to yield **9a** (5.0 g, 89%), mp: 225–227°C. Anal calcd for C₁₂H₇NO₃S: C, 58.77; H, 2.88; N, 5.71; found: C, 59.04; H, 2.84; N, 5.73. Compound **9b** was prepared from **8b** in a similar manner to that described for **9a**; yield: 90%, mp: 204°C. Anal calcd for C₆H₃NO₃S: C, 42.60; H, 1.79; N, 8.28; found: C, 42.51; H, 1.85; N, 8.00.

2-Ethoxycarbonylmethylamino-5-phenyl-thiophene-3-carboxamide 10a. A solution of 9a (5.0 g, 4.1 mmol) in DMF (1 ml) was added dropwise to a suspension of NaH (60% in oil, 1.2 g) in DMF (40 ml) under ice-cooling. The reaction mixture was stirred for 0.5 h at rt and ethyl bromoacetate (4.0 g, 24 mmol) was then added to the mixture at 10°C. After the reaction mixture had been stirred for 2 h at rt, 28% aqueous ammonia (15 ml) was added dropwise at 5°C and stirring was continued for 0.5 h at same temperature. The reaction mixture was then poured into a mixture of ice and diluted HCl, and extracted with ethyl acetate (50 ml x 2). The organic layer was dried over sodium sulfate and evaporated under reduced pressure. The obtained residue was recrystallized from ethanol to yield 10g (3.0 g, 48%), mp: 155–156°C. Anal calcd for C₁₅H₁₆N₂O₃S: C, 59.19; H, 5.30; N, 9.20; found: C, 59.28; H, 5.22; N, 9.10. Compound 10b was prepared in a similar manner to that described for 10a; yield: 44%, mp: 135-136°C. Anal calcd for C₉H₁₂N₂O₃S: C, 47.36; H, 5.30; N, 12.27; found: C, 47.28; H, 5.36; N, 12.27.

Ethyl 6-phenyl-2,4-dioxo-thieno[2,3-d]pyrimidin-1-acetate **11a**. A mixture of **10a** (2.8 g, 9.2 mmol) and 1,1-carbonyldiimidazole (CDI) (3.1 g, 19.1 mmol) was heated to 120–130°C for 2 h. Dioxane (30 ml) was then added and the mixture refluxed for 1.5 h. After cooling, the reaction mixture was evaporated under reduced pressure and the residue was recrystallized from a mixture of ethanol and acetone to give **11a** (2.3 g, 76%), mp: 267°C. Anal calcd for C₁₆H₁₄N₂O₄S; C, 58.17; H, 4.27; N, 8.48; found: C, 58.24; H, 4.25; N, 8.76. Compound **11b** was prepared in a similar manner to that described for **11a**; yield: 80%, mp: 253–254°C. Anal calcd for C₁₀H₁₀N₂O₄S; C, 47.24; H, 3.96; N, 11.02; found: C, 47.38; H, 3.99; N, 10.98.

Ethyl 3-(4-bromo-2-fluorobenzyl)-2,4-dioxo-6-phenylthieno[2,3-d]pyrimidin-1-acetate **60**. A solution of **11a** (2.0 g, 6.1 mmol) in DMF (10 ml) was added dropwise to a suspension of NaH (60% in oil, 0.4 g) in DMF (40 ml) under ice-cooling. After stirring for 1 h at room temperature, the mixture was cooled to 5°C and a solution of 4-bromo-2-fluorobenzylbromide (1.9 g, 7.1 mmol) was added dropwise. The reaction mixture was then stirred for 16 h at rt, evaporated under reduced pressure, and poured into a mixture of ice and diluted HCI. The resulting precipitate was collected and recrystallized from dioxane to yield **60** (2.2 g, 70%). Compound **6aa** was prepared in a similar manner to that described for **60**. The physical data have been listed in table II.

3-(4-Bromo-2-fluorobenzyl)-2,4-dioxo-6-phenyl-thieno[2,3d]pyrimidin-1-acetic acid 20. The title compound was prepared from 60 by a similar manner to that described for 2a. Compound 2aa was also prepared in a similar manner to that described for 2a. The physical data have been listed in table I.

Method D

Ethyl 3-(4-bromo-2-fluorobenzyl)-6-chloro-2,4-dioxothieno[2,3-d]pyrimidin-1-acetate 6r. Sulfuryl chloride (0.41 ml, 5.0 mmol) was added dropwise to a solution of 6a (2.2 g, 5.0 mmol) in CCl₄ (60 ml) at rt. The reaction mixture was stirred for 4 h at 60°C and evaporated under reduced pressure to give a crude oil which was dissolved in Et₂O and then triturated with *n*-hexane. The resulting precipitate was collected and recrystallized from isopropyl ether to give 6r (2.1 g, 93%), mp: 133–134°C. ¹H-NMR(CDCl₃): 1.21 (t, 3H), 4.26 (q, 2H), 4.59 (s, 2H), 5.21 (s, 2H), 7.20 (s, 2H) and 7.00-7.40 (m, 3H). Compounds 6q and 6t were prepared in a similar manner to that described for 6r, while compounds 6s and 6u were prepared using NBS instead of SO₂Cl₂ and then worked up in a similar manner to that described for 6r. Compounds 6ab and **6bb** were prepared using CH_2Cl_2 instead of CCl_4 as a solvent and worked up in a similar manner to that described for 6r. **6ab**: ¹H-NMR(CDCl₃): 1.20 (t, 3H), 4.20 (q, 2H), 5.20 (s, 2H), 6.90–7.35 (m, 3H) and 7.50 (s, 1H). **6bb**: ¹H-NMR(CDCl₃): 1.29 (t, 3H), 4.27 (q, 2H), 5.13 (s, 2H), 5.24 (s, 2H), 6.80–7.10 (m, 1H) and 8.14 (s, 1H). The physical data have been listed in table II.

3-(4-Bromo-2-fluorobenzyl)-6-chloro-2,4-dioxo-thieno[2,3-d]pyrimidin-1-acetic acid 2q. The title compound was prepared from 6r in a similar manner to that described for 2a. Compounds 2r-2t were also prepared by a similar manner to that described for 2a.

Method E

Ethyl 3-(4-bromo-2-fluorobenzyl)-6-bromomethyl-2,4-dioxothieno[2,3-d]pyrimidin-1-acetate 12. A mixture of 6h (0.88 g, 1.9 mmol), NBS (0.38 g, 2.1 mmol) and benzoyl peroxide (75% wet with H_2O , 0.062 g) in CCl₄ (20 ml) was stirred for 6 h at 40°C. After cooling, the reaction mixture was filtered and evaporated under reduced pressure to give an oil which was purified by silica gel chromatography with a mixture of chloroform/ether (50:1) to yield **12** (0.35 g, 34%). ¹H-NMR (CDCl₃): 1.28 (t, 3H), 4.20 (q, 2H), 4.56 (s, 2H), 4.70 (s, 2H), 5.26 (s, 2H) and 6.89–7.43 (m, 4H).

Ethyl 3-(4-bromo-2-fluorobenzyl)-2,4-dioxo-6-(morphorin-1yl)methyl-thieno[2,3-d]pyrimidin-1-acetate **6za**. A solution of morphorine (0.96 g, 11.2 mmol) in dioxane (5 ml) was added dropwise to a solution of **12** (2.0 g, 3.7 mmol) in dioxane (20 ml) under ice-cooling. The reaction mixture was stirred for 4 h at 80°C and evaporated under reduced pressure to give an oil which was extracted with chloroform (50 ml x 2), washed with water and dried over sodium sulfate. The organic layer was evaporated under reduced pressure and the residue was purified by silica gel chromatography with a mixture of chloroform/ether (4:1) to yield **6za** (0.84 g, 42%), mp: 178–180°C. Compounds **6zb–6zc** were prepared in a similar manner to that described for **6za**. The physical data have been listed in table II.

3-(4-Bromo-2-fluorobenzyl)-2,4-dioxo-6-(morphorin-1yl)methyl-thieno[2,3-d]pyrimidin-1-acetic acid 2ya. The title compound was prepared from 6za in a similar manner to that described for 2a. The obtained crude product was recrystallized from ethanol to give 2ya. Compounds 2yb and 2yc were also prepared in a similar manner to that described for 2a. The physical data have been listed in table I.

Pharmacology

Inhibition of AR in vitro [14]

The preparation of AR and determination of IC_{50} activity were carried out as previously described [14]. AR activity was assayed by spectrophotometrically following the 340-nm oxidation of reduced nicotinamide adenine dinucleotide phosphate (NADPH) to NADP using DL-glyceraldehyde as substrate. The reaction mixture contained 0.1 M phosphate buffer (pH = 6.2), 0.25 mM NADPH, 1.5 mM DL-glyceraldehyde and the enzyme in a total vol of 1 ml. Inhibitor effects on enzymatic activity were determined by adding the inhibitors to the reaction mixture in the desired concentration. The % inhibition for each inhibitor was calculated by comparing the reaction rate of the solution containing the inhibitor with that of control. IC₅₀ values were obtained by graphic estimation from the log concentration–response curves.

Inhibition of sorbitol accumulation in vivo [16, 17]

Six-wk-old male Wistar rats were rendered diabetic by an *iv* injection of streptozotocin (70 mg/kg body weight). One wk after injection, rats with plasma glucose levels > 400 mg/dl were grouped into either control or experimental groups, the latter being given a test compound (10 or 20 mg/kg) in suspension in 0.5% methyl cellulose orally once per d for 4 d. All rats were kept in identical cages and had free access to laboratory chow and water. Eighteen h after the final administration of the compound, all rats were anesthetized with ether and the sciatic nerves removed. Sorbitol was extracted from the sciatic nerve by the method of Peterson *et al* [16] and measured enzymatically by the method of Clements *et al* [17]. Sorbitol content was compared with that obtained for the control group given vehicle only.

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