STEREOCONTROLLED SYNTHESIS OF A TRANS-ANTI-TRANS TRICYCLE VIA A TRANSANNULAR DIELS-ALDER STRATEGY

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Transannular Diels-Alder reaction of trans-cis-cis macrocyclic triene 20 vields trans-syn-ABSTRACT: cis tricycle 21 which is converted into trans-anti-trans tricycle 24 by epimerization at C₉ after appropriate functional group transformation.

In earlier communications, 1,2,3 we have described the synthesis and transannular Diels-Alder reaction of three specific examples of 13-membered macrocyclic trienes, demonstrating a potentially powerful strategy for the construction of complex polycyclic molecules. More recently. Takahashi and co-workers⁴ have reported the synthesis of a steroid using the above strategy. We have also reported a study^{5,6} on the transannular Diels-Alder reaction of the eight possible geometric isomers of a 14-membered macrocyclic triene. In this general study, we have shown that direct construction of the trans-anti-trans (TAT) ring junction which corresponds to the configuration of most natural products is not possible by this synthetic approach.

The trans-cis-cis (TCC) olefin geometry of macrocyclic triene 20 allows direct access, by transannular Diels-Alder reaction, to tricyclic adduct 21 having a trans-syn-cis (TSC) ring junction.⁵ The protected alcohol of 21 can be transformed into a ketone, which allows subsequent epimerization at Ca and thus entry to tricycle 24 having the TAT stereochemistry. We wish now to report this work which also constitutes a preliminary study for the synthesis of corticoids.



As previously described,^{3,6} we chose a convergent approach for the synthesis of triene 20. The preparation of the *cis* dienophile is summarized in Scheme I. The known alcohol 1⁷ was benzylated and then treated with dilute acid to give diol 3 (85% overall). After monosilylation 8 (51%), the resulting alcohol 4 was oxidized ⁹ to aldehyde 5; subsequent condensation with the anion of phosphonium bromide 7¹⁰ provided a mixture [55:45] of cis and trans olefins (63% from 4). Chromatographic separation gave pure cis olefin 8 which was desilylated¹¹ (9, 93%), mesylated (10) and alkylated with dimethyl malonate (84%) to give 11. Methanolysis provided alcohol 12 which was then transformed into the mesylate ester 13. Intramolecular cyclization of 13 (HMDSNa, DMF/THF 1:1, 80°C) gave the cycloheptene 14, proving the cis geometry of the former.

The coupling between the dienophile 13 and the trans-cis diene 15⁶ was accomplished (NaH, THF/DMF 1:1) to give triene 16 in 84% yield (Scheme II). Alkylation of 16 with dimethyl malonate



(a) PhCH₂Br, NaH, THF, 0°C to r.t. (b) HCl 10%, H₂O (85%) (c) t-BuPh₂SiCl, imidazole, THF, r.t. (51%) (d) (COCl)₂, DMSO, Et₃N, CH₂Cl₂, -78°C to r.t. (e) DHP, PPTS, CH₂Cl₂, r.t. (82%) (f) i) BrPh₃P+CH₂CH₃, HMDSK, toluene, THF, 0°C ii) **6**, THF, 48 h, r.t. (81%) (g) i) **7**, BuLi, hexane, THF, -78°C to 0°C ii) **5**, THF, chromatography (*cis* = 36%, *trans* = 27% from alcohol **4**) (h) n-Bu₄NF, THF, 0°C to r.t. (93%) (i) MeSO₂Cl, Et₃N, CH₂Cl₂, 0°C (j) CH₂(CO₂Me)₂, NaH, THF/DMF 1:1, Kl, 85°C (84%) (k) MeOH, PPTS, reflux (92%) (l) HMDSNa, THF/DMF 1:1, 80°C, 4.5 x 10⁻³ M (77%)

SCHEME I

provided **17** (80%) which was then desilylated¹¹ to give **18** (91%). Conversion of this allylic alcohol to the corresponding chloride¹² **19** was followed by macrocyclization (Cs₂CO₃, THF/DMF 1:1, 70°C, 10 h, 2.5 x 10⁻³ M, slow addition) to the 14-membered ring **20** (75% from alcohol **18**).

When heated at 270°C for 2.75 hours, macrocycle **20** underwent transannular Diels-Alder reaction to give, in 84% yield, the tricycle **21**¹³ having five asymmetric centers. We can predict easily four of these five centers by analysis of molecular models;⁵ with a *trans-cis* diene and a *cis* dienophile, the transannular Diels-Alder reaction is expected to give a tricycle having TSC relative stereochemistry.

The benzyl protective group was removed $(SnCl_4, CH_2Cl_2)^{14}$ and the β -orientation of the hydroxy group in the resulting alcohol **22** was established by NMR spectroscopy (4.02 ppm, doublet of doublet of doublet, 9.5 Hz, 9.5 Hz, 4.8 Hz). Then, oxidation (PCC, $CH_2Cl_2)^{15}$ gave ketone **23**.¹⁶ The predicted TSC relative stereochemistry of this ketone was confirmed by single crystal X-ray diffraction analysis.¹⁷ Treatment with sodium carbonate in methanol at 50°C over nine hours transformed **23** into another ketone **24**,¹⁸ spectroscopically different from the former. This ketone **24** must therefore have the desired stereochemistry TAT by isomerization at position C₉.





20



21 R=CH₂Ph 22 R=H g



(a) i) 13, NaH, THF/DMF 1:1, 0°C to r.t. ii) 15, THF/DMF 1:1, r.t. (84%) (b) i) CH₂(CO₂Me)₂, NaH, THF/DMF 1:1 ii) 16, THF/DMF, 80°C (80%) (c) n-Bu₄NF, THF, 0°C to r.t. (91%) (d) MeSO₂Cl, s-collidine, LiCl, DMF, r.t. (e) Cs₂CO₃, DMF/THF 1:1, 2.5 x 10⁻³ M, 75°C, slow addition (75% from alcohol 18) (f) sealed tube, toluene, 270°C, 2.75 h (84%) (g) SnCl₄, CH₂Cl₂, r.t. (80%)²⁰ (h) PCC, CH₂Cl₂, r.t. (80%)²⁰ (l) Na₂CO₃, MeOH, 50°C, 9h (60%)²⁰

SCHEME II

The fact that only one diastereoisomer was obtained in the transannular Diels-Alder reaction shows a complete control of stereochemistry at position C11. This indicates that it will be possible to realize a non-racemic synthesis of a given tricycle starting from an optically active precursor which has the appropriate absolute configuration for the C₁₁ secondary benzyl ether.

In addition, we noted that acidic treatment of alcohol 22 (PTSA, benzene, 60°C) gave lactone 25.19 The two ester functions of the C13 malonate are thus easily differentiated and this type of

process could be useful in the synthesis of corticoids. For instance, the carbonyl carbon of the lactone moiety could represent the C₁₈ hemi-acetal function of aldosterone.

In conclusion, this preliminary work demonstrates that a transannular Diels-Alder reaction followed by epimerization provides ready access to tricycles of TAT stereochemistry. In addition, this method appears very promising for the synthesis of optically active corticoids. Work in this direction is currently underway in our laboratory.

REFERENCES AND NOTES

- (1) K. Baettig, C. Dallaire, R. Pitteloud and P. Deslongchamps. Tetrahedron Lett. 28, 5249 (1987).
- (2) K. Baettig, A. Marinier, R. Pitteloud and P. Deslongchamps. Tetrahedron Lett. 28, 5253 (1987).
- (3) G. Bérubé and P. Deslongchamps. Tetrahedron Lett. 28, 5255 (1987).
- (4) T. Takahashi, K. Shimizu, T. Doi and J. Tsuji. J. Am. Chem. Soc. 110, 2674 (1988).
- (5) S. Lamothe, A. Ndibwami and P. Deslongchamps. Tetrahedron Lett. 29, 1639 (1988).
- (6) S. Lamothe, A. Ndibwami and P. Deslongchamps. Tetrahedron Lett. 29, 1641 (1988).
- (7) C. Piatandosi, C.E. Anderson, A.E. Brecht and C.L. Yarbro. J. Am. Chem. Soc. 80, 6613 (1958).
- (8) S. Hanessian and P. Lavallée. Can. J. Chem. 53, 2975 (1975).
- (9) K. Omura and D. Swern. Tetrahedron 34, 1651 (1978).
- (10) Phosphonium bromide 7 was synthesized from the THP-derivative²¹ 6 of 2-bromoethanol and ethyl(triphenyl phophonium) bromide.²²
- (11) E.J. Corey and A. Venkateswarlu. J. Am. Chem. Soc. 94, 6190 (1972).
- (12) E.W. Collington and A. I Meyers. J. Org. Chem. <u>36</u>, 3044 (1971).
- (13) NMR 21: δ (ppm): 0.92 (3H, s, -CH₃-C₁₀); 1.28-2.93 (13H, m, H-C₁, H-C₂, H-C₄, H-C₉, H-C₁₂, H-C₁₄, H-C₅ and H-C₈); 3.65 (3H, s, -CO₂C<u>H₃</u>); 3.69 (6H, s, 2 -CO₂C<u>H₃</u>) and 3.70 (3H, s, -CO₂C<u>H₃</u>); 3.76 (1H, ddd, 9.9 Hz, 9.9 Hz, 4.2 Hz, H-C₁₁); 4.43 (1H, d, J_{AB} = 10.8 Hz, -O-C<u>H₂-Ph</u>); 4.61 (1H, d, J_{AB} = 10.8 Hz, -O-C<u>H₂Ph</u>); 5.27 (2H, s, H-C₆ and H-C₇); 7.24-7.39 (5H, m, -O-CH₂-C₆<u>H</u>₅).

IR 21: v (cm⁻¹) (CHCl₃): 3020, 3030 and 2955 (aliphatic and aromatic C-H); 1730 (C=O); 1435 and 1455 (-CH₂-, -CH₃); 1220, 1250 and 1265 (C-O).

MS 21: m/e 542 (M⁺). Mol. Wt. (ms) calcd. for $C_{29}H_{35}O_8$ (M⁺-OMe) = 511.2332; found = 511.2319. Melting point 21: 104-106°C.

- (14) M.H. Park, R. Takeda, K. Nakanishi. Tetrahedron Lett. 28, 3823 (1987).
- (15) E.J. Corey and J.W. Suggs. Tetrahedron Lett., 2647 (1975).
- (16) NMR 23: δ (ppm): 0.90 (3H, s, -CH₃-C₁₀); 1.13-2.55 and 3.05-3.17 (11H, m, H-C₁, H-C₂, H-C₄, H-C₅, H-C₉, H-C₁₄); 2.64 (1H, d, J_{AB} = 14.6 Hz, H-C₁₂); 2.86 (1H, dd, J_{AB} = 14.7 Hz, J_W = 1.0 Hz, H-C₁₂); 3.69, 3.70, 3.74 and 3.82 (12H, 4s, 4 -CO₂CH₃); 5.33-5.47 (2H, m, H-C₆ and H-C₇).
 IR 23: v (cm⁻¹) (CHCl₃): 3020, 2955 (aliphatic C-H); 1730 (C=O); 1435 (-CH₂-, -CH₃); 1260, 1220 (C-O).
 MS 23: m/e 450 (M⁺). Mol. Wt. (ms) calcd. for C₂₃H₃₀O₉= 450.1890; found = 450.1884. Melting point 23: 150-151°C.
- (17) A. Michel and G. Boulay. Unpublished results. Université de Sherbrooke, Sherbrooke, QC, Canada.
- (18) NMR 24: δ (ppm): 0.95 (3H, s, -CH₃-C₁₀); 1.05-2.61 (11H, m, H-C₁, H-C₂, H-C₄, H-C₅, H-C₈, H-C₉, H-C₁₄); 2.62 (1H, dd, J_{AB} = 14.0 Hz, J_W = 1.1 Hz, H-C₁₂); 2.91 (1H, dd, J_{AB} = 14.0 Hz, J_W = 2.2 Hz, H-C₁₂); 3.70, 3.73, 3.74 and 3.75 (12H, 4s, 4 -CO₂CH₃); 5.37 (1H, dm, J_{AB} = 10.2 Hz, H-C₆ or H-C₇); 5.47 (1H, ddd, J_{AB} = 10.2 Hz, J_{BX} = 2.2 Hz, H-C₆ or H-C₇).

IR 24: v (cm⁻¹) (CHCl₃): 2955, 2920, 2850 (aliphatic C-H); 1730 (C=O); 1435, 1455, 1465 (-CH₂-, -CH₃); 1250, 1220 (C-O).

MS 24: m/e 450 (M⁺). Mol. Wt. (ms) calcd. for C₂₃H₃₀O₉ = 450.1890; found = 450.1884. Melting point 24: 114-115°C.

- (19) NMR 25: δ (ppm): 0.94 (3H, s, -CH₃-C₁₀); 1.28-2.81 (13H, m, H-C₁, H-C₂, H-C₄, H-C₅, H-C₈, H-C₉, H-C₁₂, H-C₁₄); 3.72, 3.77 and 3.78 (9H, 3s, 3 -CO₂CH₃); 4.82 (1H, d, 4.3 Hz, H-C₁₁); 5.35 (1H, dm, J_{AB} = 9.8 Hz, H-C₆ or H-C₇); 5.58 (1H, ddd, J_{AB} = 9.8 Hz, J_{BX} = 2.3 Hz, J_{BX} = 2.3 Hz, H-C₆ or H-C₇).
 IR 25: v (cm⁻¹) (CHCl₃): 3015, 2950, 2920 (aliphatic C-H); 1780 (C=O lactone); 1730 (C=O ester); 1450, 1435 (-CH₂-, -CH₃); 1255, 1220 (C-O).
 MS 25: m/e 420 (M⁺). Mol. Wt. (ms) calcd. for C₂₂H₂₈O₈ = 420.1784; found = 420.1781. Melting point 25: 149-150°C.
- (20) Not optimized yields, reactions made on scale of 5 mg.
- (21) N. Miyashita, A. Yoshikoshi and P. A. Grieco. J. Org. Chem. 42, 3772 (1977).
- (22) A. Maercker. Org. React. 14, 293 (1965).
- (23) Support for this work by NSERCC (Ottawa), "FCAR" (Québec) and MERCK FROSST CANADA INC. (Montréal) is gratefully acknowledged.