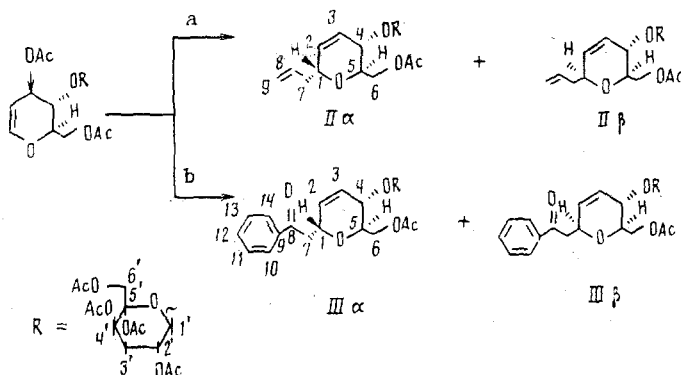


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The reaction of allyltrimethylsilane and 1-trimethylsiloxystyrene catalyzed by Lewis acids has been described only for monosaccharides glycals [1, 2].

In a continuation of a study of the transformations of glycals from oligosaccharides [3, 4], we are the first to report that the reaction of hexa-O-acetyl-D-lactal (I) with a 1.5-molar excess of allyltrimethylsilane or 1-trimethylsiloxystyrene catalyzed by $\text{BF}_3 \cdot \text{OEt}_2$ leads to the corresponding C₁-glycosides (II) (88%) and (III) (80%) in a mixture of α/β -anomers. High-efficiency liquid chromatography gave the C₁-glycoside ratios: (II α):(II β) = 20:1 and (III α):(III β) = 5:1:



In pathway a, allyltrimethylsilane was used with $\text{BF}_3 \cdot \text{OEt}_2$ in CH_2Cl_2 upon warming from -50 to 0°C over 0.5 h. In pathway b, 1-trimethylsiloxystyrene was used with $\text{BF}_3 \cdot \text{OEt}_2$ in CH_2Cl_2 upon warming from -30 to 0°C over 1 h. The physical constants of the three pure isomers are given below.

(II α): $[\alpha]_D^{20} +20.96^\circ$ (c 2.29 CHCl_3). ^{13}C NMR spectrum (δ , ppm): 20.51, 20.71 q ($5\text{CH}_3\text{CO}$), 37.44 t (C^7), 61.33 t ($\text{C}^{6'}$), 63.35 t (C^6), 67.0 d ($\text{C}^{4'}$), 68.86 d ($\text{C}^{2'}$), 69.2 d (C^5), 70.83 d ($\text{C}^{3'}$), 70.87 d ($\text{C}^{5'}$), 72.27 d, 73.02 d (C^1 , C^4), 101.99 d ($\text{C}^{1'}$), 117.44 t (C^9), 126.66 d, 131.36 d (C^2 , C^3), 134.14 d (C^8), 169.36, 169.91, 170.10, 170.22, 170.65 s ($5\text{CH}_3\text{CO}$).

(III α): $[\alpha]_D^{20} +9.24^\circ$ (c 2.26 CHCl_3). ^{13}C NMR spectrum (δ , ppm): 20.50, 20.57, 20.95 q ($5\text{CH}_3\text{CO}$), 41.44 t (C^7), 61.62 t ($\text{C}^{6'}$), 63.19 t (C^6), 66.96 d ($\text{C}^{4'}$), 68.86 d ($\text{C}^{2'}$), 69.8 d (C^1), 69.39 d (C^5), 70.83 d ($\text{C}^{3'}$, $\text{C}^{5'}$), 72.91 d (C^4), 102.14 d ($\text{C}^{1'}$), 127.03 d, 131.19 d (C^2 , C^3), 128.14 d (C^{10} , C^{14}), 128.68 d (C^{11} , C^{13}), 133.34 d (C^{12}), 136.99 s (C^9), 169.35, 169.49, 169.96, 170.30, 170.68 s ($5\text{CH}_3\text{CO}$), 197.15 s (C^8).

(III β): $[\alpha]_D^{20} +64.6^\circ$ (c 0.46 CHCl_3). ^{13}C NMR spectrum (δ , ppm): 20.69, 20.89 q ($5\text{CH}_3\text{CO}$), 44.0 t (C^7), 61.43 t ($\text{C}^{6'}$), 63.58 t (C^6), 67.06 d ($\text{C}^{4'}$), 69.01 ($\text{C}^{2'}$), 70.92 d ($\text{C}^{5'}$), 71.01 d ($\text{C}^{3'}$), 71.76 d (C^5), 73.64 d (C^1), 75.20 d (C^4), 102.38 d ($\text{C}^{1'}$), 127.72, 131.71 d (C^2 , C^3), 128.28 d (C^{10} , C^{14}), 128.66 d (C^{11} , C^{13}), 133.34 d (C^{12}), 137.14 s (C^9), 169.46, 170.11, 170.27, 170.39, 170.83 s ($5\text{CH}_3\text{CO}$), 197.28 s (C^8).

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